



Sensory-guided isolation and identification of new sweet-tasting dammarane-type saponins from Jiaogulan (*Gynostemma pentaphyllum*) herbal tea

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ABSTRACT

The purpose of this study was to elucidate the chemical basis for the sweet property produced by *Gynostemma pentaphyllum* and find new natural high-potency (HP) sweeteners. Sixteen new compounds (gypenosides YN 1–16) were obtained by sensory-guided isolation and identification, in which fifteen of them were sweet-tasting constituents with sweetness intensities 10–100 times higher than that of sucrose evaluated by human sensory panel test. Their structures were established by 1D and 2D nuclear magnetic resonance spectra, mass spectroscopy, infrared spectroscopy, UV–visible spectroscopy, and chemical method. Gypenoside YN 4 was the sweetest compound with a concentration of 15.504 ± 1.343 mg/kg, while gypenoside YN 12 has the highest concentration (1397.674 ± 12.948 mg/kg), as shown by ultra-performance liquid chromatography–tandem mass spectrometry (UPLC-MS/MS). Structure-activity relationship analysis implied that the compounds' sweetness intensity was associated with side-chain substitutions at C-20 or the number of glucosyl groups at C-3. These new plant-derived natural products may be potential natural sweeteners.

1. Introduction

The gustation of humans, which is the ability to perceive food taste, plays an essential role in food selection, preferences, and intake (Ben Shoshan-Galeczki & Niv, 2020). Sweet, umami, bitter, salty and sour are the five basic tastes that humans can detect (Zhang, Sun, Gu, & Du, 2017). Among them, the sweet taste is an indicator of nutrient-rich food and so is innately attractive to humans with the most pleasant gustatory sensations. In contrast the bitter taste is a warning of toxic ingredients within food, which is linked to food rejection due to unpleasant gustatory sensations (Temussi, 2009). Even babies of a few days old can show their pleasure in response to sweetness but aversion to bitterness (Lindemann, 2001).

With the improvement of people's standard of living, sugar sweeteners such as sucrose, glucose, and fructose have become increasingly

widely used in the food and pharmaceutical industries (Goel, Gajula, Gupta, & Rai, 2021), resulting in excessive intake of sugar sweeteners. This has caused some health concerns, as health problems, such as dental caries, hyperglycemia, cardiovascular diseases, and obesity have increased annually (Acevedo, Ramirez-Sarmiento, & Agosin, 2018; Vos et al., 2016). These health issues have motivated people to look for natural non-sugar sweeteners, leading to the discovery and application of steviolosides and mogrosides (Soejarto, Addo, & Kinghorn, 2019).

Gynostemma pentaphyllum (Thunb.) Makino, which belongs to Cucurbitaceae family, is commonly known as “Jiaogulan” in China, also called “Southern ginseng”. It is a perennial liana plant which is mainly distributed in China, Korea, Japan, and other Southeast Asian countries. It is a popular folk medicine which has been attributed with anticancer, hypolipidemic, hypoglycemic and immunomodulation properties (Zhai et al., 2021). It also has been widely used as herbal tea, vegetables, and

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function food for centuries (Yang et al., 2013). It was first recorded in the book “Herbs for Famine” of Ming Dynasty in ancient China, in which it is listed as a vegetable.

In 1976, reports of panaxadiol and 2 α -hydroxypanaxadiol from a hydrolysate of the saponin-rich fraction of *G. pentaphyllum*, of which the panaxadiol is the aglycone of characteristic saponins from *Panax* species, promoted the interest of domestic and international scholars into phytochemical investigation of *G. pentaphyllum* (Nagai, Izawa, Nagumo, Sakurai, & Inoue, 1981). The dammarane-type saponins from “Jiaogulan”, known as gypenosides, are one of its main functional constituents. So far, more than 300 gypenosides have been isolated and identified from *G. pentaphyllum* and its processed samples (Nguyen, Ha, Yang, Pham, & Oh, 2021). In-depth study of the chemical basis of “Jiaogulan”, has scientifically verified its medicinal efficacy. In recent years, products of *G. pentaphyllum* have become more popular in European and North American countries for their pharmacological activity (Nguyen et al., 2021), such as lowering serum lipid and cholesterol (Weng et al., 2021; Wu et al., 2011), hypoglycaemic (Gao et al., 2016), anti-inflammatory (Wang, Wang et al., 2020), anticancer (Liu, Li, Duan, Xie, & Piao, 2021), hepatoprotective activities (Jia et al., 2018), and neuroprotection (Wang, Zhao et al., 2020; Zhai et al., 2021).

Literature research combined with ethnobotanical investigation showed that *G. pentaphyllum* has sweet and bitter taste varieties (Lu et al., 2013; Wu et al., 2011). The sweet variety has been used as a dietary sweetener in Japan (Nagai et al., 1981). Scholars have carried out systematic and in-depth research on the medicinal substances and health care effects of *G. pentaphyllum*. However, there is scant research into the chemical basis of its tastes and flavors. In previous investigations, the contents of ginsenosides Rb1, Rb3, Rd, F2, and 20(S)-panaxadiol in the sweet taste “Jiaogulan” variant were obviously higher than those in the bitter variant, according to ultra-performance liquid chromatography quadrupole time-of-flight mass spectrometry (UPLC-Q-TOF-MS) and high-performance liquid chromatography coupled with evaporative light scattering detection (HPLC-ELSD) (Yang et al., 2013). Their homology in the ribosomal ITS-1 region shares only 69.01% (Wu et al., 2011). Kinghorn, Chin, Pan, and Jia (2010) reported that gypenoside XX was a sweet compound that could be used as a natural sweetener, but did not report data regarding its sweetness intensity. Overall, the compounds responsible for the sweet-taste of *G. pentaphyllum* remain unclear.

Sensory-guided isolation strategy can quickly locate the critical taste compounds from a complex extraction (Frank, Ottinger, & Hofmann, 2001), avoiding more time-consuming methods or substances being separated with no taste activity by blind separation (Shi et al., 2009). This method was successfully applied to elucidate the new taste activity substances of a variety of food materials, such as the bitter compound of Huangjiu (Lu et al., 2021), the sweet taste glycosides of *Myriopterion extensum* (Sun et al., 2016, 2018), the sweet taste saponins of “Tugancao” (Zhang et al., 2017), and the sweetness compounds of wine (Cretin, Waffo-Teguo, Dubourdieu, & Marchal, 2019). Therefore we employed this strategy to locate and identify the sweet taste constituents of “Jiaogulan”.

According to our research team’s preliminary sensory evaluation of “Jiaogulan” from several areas, the sample from Yunnan province of China showed a pleasant sweet taste, and was therefore selected for further investigation. Aiming to identify the sweet-tasting compounds to find new natural high-potency (HP) and low-calory sweeteners, we carried out sensory-guided fractionation and purification of this plant extract. As a result, we obtained sixteen new gypenosides involving three structure types (type I-III) according to the side chain at C-20 of aglycone. Fifteen of them showed a sweet taste, with sweetness intensities ranging from 10 to 100 times greater than that of sucrose. One of them showed a bitter taste. Preliminary structure-activity relationship research showed that the sweet intensity of the type III compounds is higher than that of the other two types of compounds when sugar chains of C-3 and C-20 are the same. The contents of these sweet

compounds were quantitatively analyzed by ultra-performance liquid chromatography-electron spray ionization-mass spectrometer/mass spectrometer (UPLC-MS/MS).

2. Material and methods

2.1. Chemicals, instruments and procedures

The materials used in extraction and purification were: acetonitrile (HPLC grade) (Shanghai Xingke, Ltd., Shanghai, China), chloroform (Rionlon, Tianjin, China), dioxane (Sinopharm chemical reagent Co. Ltd., Shanghai, China), sodium dicarbonate (Damao, Tianjin, China). Methanol (HPLC grade) and water (LC-MS grade) were purchased from Fisher Scientific (Fisher Scientific, Pittsburgh, Pennsylvania, USA). The materials used in the absolute configurations of sugar units were: hydrochloric acid (Xilong Chemical Co. Ltd., Guangdong, China), *N*-trimethylsilylimidazole (Sangon Biotech, Shanghai, China), *L*-cysteine methyl hydrochloride (Sigma, Shanghai, China), *n*-hexane (Damao, Tianjin, China), and D-(+)-glucose (D-Glc), D-xylose (D-Xyl), *L*-rhamnose (*L*-Rha), purchased from J & K Scientific Ltd (Guangzhou, China).

Spectra of nuclear magnetic resonance (NMR) were recorded by a Bruker Avance III spectrometer (Bruker, Bremen, Germany). The mass spectra were recorded by a Shimadzu UPLC-Q-TOF-MS (Shimadzu, Tokyo, Japan). The optical rotation was tested on an Autopol VI polarimeter (Rudolph Research Analytical, Hackettstown, NJ, USA). A UV-2401PC spectrophotometer produced by Shimadzu (Shimadzu, Tokyo, Japan) was used to obtain ultraviolet (UV) spectra. Infrared (IR) spectra using KBr pellets were scanned on a NICOLET iS10 spectrometer (Thermo Fisher Scientific, Waltham, Massachusetts, USA).

Silica gel (200 ~ 300 mesh, Qingdao Marine Chemical Co., Ltd., Qingdao, China), Sephadex gel LH-20 (GE Healthcare Biosciences AB, Uppsala, Sweden), macroporous resins column chromatography HP-20 (Yunkai, Tianjin, China), reversed phase (RP) C-18 silica gel (40~63 μ m, Merck, Darmstadt, Germany) were employed to conduct column chromatography. A semi-preparative HPLC, Newstyle liquid chromatography (LC) system (Hanbon, Jiangsu, China) equipped with a UV-vis detector (NU3000) and a series of YMC-Pack ODS columns were used to purify the compounds. Thin-layer chromatography (TLC) was performed on silica gel GF₂₅₄ (Qingdao Marine Chemical Co., Ltd., Qingdao, China). Spots were detected under UV light and visualized by heating after spraying with 5% vanillin-H₂SO₄ reagent (vanillin: Shanghai Ryon Biological Technology CO., Ltd., Shanghai, China; H₂SO₄: Damao, Tianjin, China). All samples intended for tasting were dried by a VirTis freezing dryer (SP Scientific, Warminster, Pennsylvania, USA) to eliminate the residual solvents.

2.2. Plant materials

The aerial parts of *Gynostemma pentaphyllum* were collected from Yuxi City in Yunnan province of China and identified by Dr. Wen-Yun Chen in Kunming Institute of Botany. Its voucher specimen (JGL-YN-S-001) was kept in Yunnan Key Laboratory for Wild Plant Resources, Kunming Institute of Botany, Chinese Academy of Sciences.

2.3. Extraction, sensory-guided fractionation and purification

The powdered aerial parts of *G. pentaphyllum* (4.3 kg) were extracted with 95% ethanol at 50 °C for 72 h. The filtrates were concentrated under vacuum to produce a thick, dark extract (257 g), which was partitioned by silica gel column chromatography (CC) eluted with petroleum ether-acetone solvent system (10:1, 7:1, 3:1, 1:1, 0:1; v/v) and methanol to produce eight fractions (Fr. 1 ~ Fr.8), respectively. Fraction 8 (200 g), which showed a sweet taste in taste evaluation, was conducted on macroporous resins HP-20 CC (2 kg), eluting by MeOH-H₂O gradient (0:100, 30:70, 50:50, 70:30, 100:0, v/v) to produce six fractions (Fr.8a ~ Fr.8f), according to TLC analyses. Fr.8c ~Fr.8e were sweet-activity

parts according to the sensory test.

Fr.8c (10 g) was separated by a RP C-18 CC system (φ 4.0 \times 45 cm), gradient eluted with water, MeOH-H₂O (30%, 50%, 70% and 100% MeOH, to 8 fractions (Fr.8c-1 ~ Fr.8c-8). Fr.8c-2 (4.58 g) was further divided to seven fractions (Fr.8c-2-a ~ Fr.8c-2-g) by RP C-18 CC system (φ 2.0 \times 45 cm, MeOH-H₂O 30%, 40%, 50%, 60%, and 100% MeOH). Compounds **3** (239 mg, t_R = 11.30 min) and **9** (357 mg, t_R = 15.00 min) were obtained by semi-preparative HPLC (ODS-AQ column, 250 \times 10 mm) with 69% MeOH-H₂O from Fr.8c-2-d, and **4** (24 mg, t_R = 16.4 min) and **7** (13 mg, t_R = 20.6 min) were purified by semi-preparative HPLC (ODS-AQ column, 250 \times 10 mm) with 62% MeOH-H₂O from Fr.8c-2-e. Fr.8c-4 (1.08 g) was separated over a RP C-18 CC (φ 2.0 \times 45 cm), eluted with MeOH-H₂O (50%, 52%, 54%, 56%, 58%) and 100% MeOH, to yield six fractions (Fr.8c-4-a ~ Fr.8c-4-f). Fr.8c-4-e (0.14 g) was purified by semi-preparative HPLC on an ODS-AQ column (250 \times 10 mm, MeOH/H₂O = 70/30) to gain **5** (39 mg, t_R = 25.3 min) and **14** (11 mg, t_R = 13.5 min). Fr.8c-5 (0.38 g) was subjected to RP C-18 CC (φ 2.0 \times 45 cm), eluted with MeOH-H₂O (50%, 58%, 60%) and 100% MeOH, to produce five subfractions (Fr.8c-5-a ~ Fr.8c-5-e). Fr.8c-5-d (0.10 g) was separated by semi-preparative HPLC on a YMC-Pack CN column (250 \times 10 mm, MeOH/H₂O = 53/47) to obtain **1** (47 mg, t_R = 10.6 min). Fr.8c-8 (1.05 g) was submitted to a RP C-18 CC (φ 2.0 \times 45 cm), eluted with MeOH-H₂O (45%, 47%, 49%, 51%, 70%) and 100% MeOH, to give four fractions (Fr.8c-8-a ~ Fr.8c-8-d). Fr.8c-8-c (0.43 g) was further separated by semi-preparative HPLC on an ODS-AQ column (250 \times 10 mm, 3.0 mL/min, MeOH/H₂O = 68/32) to yield **2** (216 mg, t_R = 20.6 min). Compound **8** (136 mg, t_R = 16.05 min) was divided from Fr.8c-8-d (0.20 g) by semi-preparative HPLC (ODS-AQ column, 250 \times 10 mm, 3.0 mL/min, MeOH/H₂O = 70/30).

Fr.8d (30 g) was chromatographed on RP C-18 CC (φ 4.0 \times 25 cm), eluted with water, MeOH-H₂O (30%, 50%, 60%, 70%) and 100% MeOH, to yield nine fractions (Fr.8d-1 ~ Fr.8d-9). Fr.8d-7 (1.70 g) was submitted to a silica gel CC, eluted by chloroform-methanol (8:2, 7:3, 6:4), to obtain nine fractions (Fr.8d-7-a ~ Fr.8d-7-i). Fr.8d-7-b (0.04 g) was divided by semi-preparative HPLC on an ODS-AQ column (250 \times 10 mm, 3.0 mL/min, MeOH/H₂O = 60/40) to produce **6** (6 mg, t_R = 25.3 min). Compound **15** (13 mg, t_R = 18. min) was obtained by a semi-preparative HPLC (ODS-AQ column, 250 \times 10 mm, 3.0 mL/min, MeOH/H₂O = 60/40) from Fr.8d-7-e (0.14 g). Fr.8d-7-g (0.51 g) was eluted with MeOH-H₂O (55%, 60%, 65%, 70%) and 100% MeOH on RP C-18 CC (φ 2.0 \times 25 cm), to yield three subfractions (Fr.8d-7-g-1 ~ Fr.8d-7-g-3). Fr.8d-7-g-1 (0.25 g) and Fr.8d-7-g-3 (0.07 g) were further separated by semipreparative HPLC (ODS-AQ column, 250 \times 10 mm, 3.0 mL/min) with 70% MeOH-H₂O to yield **13** (195 mg, t_R = 13.2 min) and **16** (12 mg, t_R = 14.6 min), respectively. Compound **11** (68 mg) was obtained from Fr.8d-8 (0.37 g) by recrystallization in methanol.

Fr.8e (38.93 g) was eluted with CHCl₃-MeOH (8:1, 8:2), CHCl₃-MeOH-H₂O (7:3:0.5) on a silica gel CC, successively, to produce seven fractions (Fr.8e-1 ~ Fr.8e-7). Fr.8e-2 (12.0 g) was separated again using a silica gel CC (CHCl₃-MeOH-H₂O system 3:1:0.1) to yield nine fractions (Fr.8e-2-a ~ Fr.8e-2-i), based on TLC analysis. Fr.8e-2-d (2.1 g) was isolated by RP C-18 CC (φ 2.0 \times 25 cm), eluted with MeOH-H₂O (70%, 75%, 78%) and 100% MeOH, to produce three fractions (Fr.8e-2-d-1 ~ Fr.8e-2-d-3). Compound **10** (54 mg, t_R = 6.8 min) was obtained from Fr.8e-2-d-2 (1.03 g) by semi-preparative HPLC (ODS-A column, 250 \times 10 mm, 3.0 mL/min, MeCN/H₂O = 50/50). Fr.8e-2-f (1.3 g) was subjected to RP C-18 CC (φ 2.0 \times 25 cm), eluted with MeOH-H₂O (65%, 70%, 72%, 75%) and 100% MeOH, to produce four fractions (Fr.8e-2-f-1 ~ Fr.8e-2-f-4). Compound **12** (75 mg, t_R = 5.30 min) was obtained from Fr.8e-2-f-3 (0.55 g) by semi-preparative HPLC (ODS-A column, 250 \times 10 mm, 3.0 mL/min, MeCN/H₂O = 50/50).

Gyenosides YN **1**, white amorphous powder, $[\alpha]_D^{19.3}$ - 13.39 (c 0.102, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 202.00 (3.14); IR (KBr) ν_{max} cm⁻¹: 3428, 2973, 2926, 1631, 1450; HR-ESI-MS 1093.5805 [M - H]⁻ (calcd. for C₅₃H₈₉O₂₃ m/z 1093.5800 [M - H]⁻); NMR data, see

table S1 and S2.

Gyenosides YN **2**, white amorphous powder, $[\alpha]_D^{24.8}$ - 11.93 (c 0.110, pyridine); UV (MeOH) λ_{max} (nm) (log ϵ): 195.50 (3.37); IR (KBr) ν_{max} cm⁻¹: 3417, 2939, 2882, 1633, 1450; HR-ESI-MS 1255.6323 [M - H]⁻ (calcd. for C₅₉H₉₉O₂₈ m/z 1255.6328 [M - H]⁻); NMR data, were shown in table S1 and S2.

Gyenosides YN **3**, white amorphous powder, $[\alpha]_D^{19.7}$ - 12.41 (c 0.108, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 202.00 (3.15); IR (KBr) ν_{max} cm⁻¹: 3406, 2928, 1635, 1450, 1416; HR-ESI-MS 708.3394 [M-2H]²⁻ (calcd. for C₆₅H₁₀₉O₃₃ m/z 708.3392 [M-2H]²⁻); NMR data, see table S1 and S2.

Gyenosides YN **4**, white amorphous powder, $[\alpha]_D^{23.4}$ - 10.98 (c 0.106, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 197.00(3.74); IR (KBr) ν_{max} cm⁻¹: 3418, 2927, 1667, 1631, 1453; HR-ESI-MS 1415.6706 [M - H]⁻ (calcd. for C₆₅H₁₀₇O₃₃ m/z 1415.6700 [M - H]⁻); NMR data, see table S1 and S2.

Gyenosides YN **5**, white amorphous powder, $[\alpha]_D^{23.5}$ + 2.22 (c 0.108, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 196.50 (3.51); IR (KBr) ν_{max} cm⁻¹: 3418, 2969, 2939, 1632, 1458; HR-ESI-MS 1093.5801 [M - H]⁻ (calcd. for C₅₃H₈₉O₂₃ m/z 1093.5801 [M - H]⁻); NMR data were shown in S1 and S2.

Gyenosides YN **6**, white amorphous powder, $[\alpha]_D^{19.5}$ - 2.96 (c 0.106, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 202.00 (3.39); IR (KBr) ν_{max} cm⁻¹: 3853, 3727, 3425, 2926, 2068, 1632, 1458; HR-ESI-MS 553.2940 [M-2H]²⁻ (calcd. for C₅₄H₉₀O₂₃ m/z 553.2942 [M-2H]²⁻); NMR data, see table S1 and S2.

Gyenosides YN **7**, white amorphous powder, $[\alpha]_D^{24.8}$ + 6.14 (c 0.104, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 195.50 (3.63); IR (KBr) ν_{max} cm⁻¹: 3419, 2968, 2927, 2880, 1636, 1454; HR-ESI-MS 1139.5859 [M - H]⁻ (calcd. for C₅₄H₉₁O₂₅ m/z 1139.5859 [M - H]⁻); NMR data, see table S1 and S2.

Gyenosides YN **8**, white amorphous powder, $[\alpha]_D^{24.8}$ - 12.26 (c 0.107, pyridine); UV (MeOH) λ_{max} (nm) (log ϵ): 195.50 (3.65); IR (KBr) ν_{max} cm⁻¹: 3417, 2970, 2935, 1633, 1454; HR-ESI-MS 1255.6326[M - H]⁻ (calcd. for C₅₉H₉₉O₂₈ m/z 1255.6328 [M - H]⁻); NMR data were shown in table S1 and S2.

Gyenosides YN **9**, white amorphous powder, $[\alpha]_D^{24.9}$ - 2.93 (c 0.124, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 195.50 (3.76); IR (KBr) ν_{max} cm⁻¹: 3421, 2969, 2931, 1636, 1455; HR-ESI-MS 1417.6858 [M - H]⁻ (calcd. for C₆₅H₁₀₉O₃₃ m/z 1417.6857 [M - H]⁻); NMR data, see table S1 and S3.

Gyenosides YN **10**, white amorphous powder, $[\alpha]_D^{27.4}$ - 3.9 (c 0.107, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 196.0 (3.94); IR (KBr) ν_{max} cm⁻¹: 3403, 2967, 2941, 1629, 1451, 1049; HR-ESI-MS 1077.5852 [M - H]⁻ (calcd. for C₅₃H₈₉O₂₂ m/z 1077.5851 [M - H]⁻); NMR data, see table S1 and S3.

Gyenosides YN **11**, white amorphous powder, $[\alpha]_D^{27.0}$ - 32.62 (c 0.100, pyridine); UV (MeOH) λ_{max} (nm) (log ϵ): 202.50 (3.51); IR (KBr) ν_{max} cm⁻¹: 3424, 2975, 2924, 2029, 1639, 1550, 1453; HR-ESI-MS 1239.6373 [M - H]⁻ (calcd. for C₅₉H₉₉O₂₇ m/z 1239.6379 [M - H]⁻); NMR data were exhibited in table S1 and S3.

Gyenosides YN **12**, white amorphous powder, $[\alpha]_D^{27.0}$ - 4.33 (c 0.108, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 196.5 (3.80); IR (KBr) ν_{max} cm⁻¹: 3392, 2926, 1739, 1633, 1445; HR-ESI-MS 1281.6489 [M - H]⁻ (calcd. for C₆₁H₁₀₁O₂₈ m/z 1281.6485 [M - H]⁻); NMR data, see table S1 and S3.

Gyenosides YN **13**, white amorphous powder, $[\alpha]_D^{19.8}$ - 23.64 (c 0.110, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 202.50 (3.50); IR (KBr) ν_{max} cm⁻¹: 3441, 2970, 2925, 1638, 1453; HR-ESI-MS 700.3419 [M-2H]²⁻ (calcd. for C₆₅H₁₀₉O₃₂ m/z 700.3417 [M-2H]²⁻); NMR data, see table S1 and S3.

Gyenosides YN **14**, white amorphous powder, $[\alpha]_D^{24.9}$ - 4.33 (c 0.108, methanol); UV (MeOH) λ_{max} (nm) (log ϵ): 195.50 (3.76); IR (KBr)

ν_{\max} cm^{-1} : 3425, 2925, 1632, 1450; HR-ESI-MS 1431.7014 [M – H][–] (calcd. for C₆₆H₁₁₁O₃₃ m/z 1431.7013 [M – H][–]); NMR data were presented in table S1 and S3.

Gypenosides YN 15, white amorphous powder, $[\alpha]_{\text{D}}^{19.8} + 8.33$ (c 0.108, methanol); UV (MeOH) λ_{\max} (nm) (log ϵ): 202.50 (3.32); IR (KBr) ν_{\max} cm^{-1} : 3440, 2924, 2856, 1632; HR-ESI-MS 1123.5902 [M – H][–] (calcd. for C₅₄H₉₁O₂₄ m/z 1123.5906 [M – H][–]); NMR data, see table S1 and S3.

Gypenosides YN 16, white amorphous powder, $[\alpha]_{\text{D}}^{29.4} - 15.29$ (c 0.107, pyridine); UV (MeOH) λ_{\max} (nm) (log ϵ): 202.50 (3.46); IR (KBr) ν_{\max} cm^{-1} : 3440, 2970, 2926, 1634, 1550, 1453; HR-ESI-MS 692.8447 [M–2H]^{2–} (calcd. for C₆₅H₁₀₉O₃₁ m/z 692.8443 [M–2H]^{2–}); NMR data were shown in table S1 and S3.

2.4. Acidic hydrolysis of compounds and absolute configuration determination of sugar units in compounds

Each compound was accurately weighed as 2 mg, placed in a round-bottom flask, and was acid hydrolyzed to obtain sugar residues as previously described (Liang et al., 2011; Zhang et al., 2017). Anhydrous pyridine (1 mL) and L-cysteine methyl ester hydrochloride (2 mg) were added into each residue and then stirred at 60 °C for 2 h to gain the mixture solution. Then, the *N*-trimethylsilylimidazole (0.2 mL) was mixed into the mixture solution to react at 60 °C for 2 h. The final reaction solution was partitioned by *n*-hexane/H₂O (1/1, 3 mL each). After concentration, the *n*-hexane part was analyzed by Agilent 7890A gas chromatography (GC) with a flame ionization detector (FID) and an HP-5 capillary column, 50 m × 0.32 mm × 0.52 μm (Agilent Technologies, Santa Clara, CA, U.S.A.). The carrier gas was helium (99.999%); injector and FID temperature always held at 250 °C. By comparison of the retention time of derivatives of the gypenoside hydrolyzates with derivatives of L-rhamnose (13.0 min), D-xylose (14.8 min), and D-glucose (19.9 min) standards, the absolute configurations of sugar units were determined.

2.5. Sensory evaluations of fractions and isolated gypenosides

Our research team's evaluation panel was created as previously described (Jia & Yang, 2009; Zhang et al., 2017). They evaluated the taste character and sweetness intensity of fractions and pure gypenosides. This panel consists of seven sweet gustation sensitive panelists (three men and four women, 24 to 45 ages, all Chinese).

The water solutions of crude extract and fractions were made at 0.04 mg/mL by distilled water. Sensory panelists evaluated their taste characters to find the sweet fractions and sub-fractions.

The stock solution of each compound with a concentration of 1 mg/mL was made with distilled water. Stock solution was diluted to obtain a descending series of lower concentration solutions (0.50, 0.25, 0.20, 0.125, 0.10, 0.04, and 0.02 mg/mL). The sucrose solutions with concentrations of 10 mg/mL and 20 mg/mL were prepared. Sixteen compounds, Fr.8 and all sub-fractions from Fr. 8 are soluble in distilled water at room temperature. Saturated solutions of Fr. 1 ~ Fr. 7 with poor solubility were used for evaluation after filtering. The panelists tasted the gypenoside solutions and compared their sweetness with the sucrose solution to determine which solutions had a similar taste to that of the sucrose solution. They needed to rinse their mouths with water in between samples and rest for some time after tasting several cups of samples. The ratio between the concentration of the sucrose solution and that of the gypenosides solution was considered the relative sweetness intensity.

2.6. Quantitation of gypenosides by UPLC–MS/MS

Standard solutions of obtained compounds were prepared in 70% methanol–water by accurately weighing at concentrations of 1.0, 2.0,

and 5.0 μg/mL for compounds 1–5, 10–13, and 15 and 5.0, 10.0, and 25.0 μg/mL for compound 8 and 50.0, 100.0, and 250.0 μg/mL for compound 9 to make working curves. The fraction 8, which includes all of the obtained compounds, was accurately weighed into a vial (2 mL) to prepare the sample solution at 2.0 mg/mL. All standard and sample solutions were filtrated through a 0.45 μm filter membrane.

Chromatographic analysis was performed in a Waters Acquity UPLC system (Waters Corp., Milford, MA) on the column BEH C18 (50 mm × 2.1 mm, 1.7 μm, Waters Corp., Milford, MA) with the flow rate of 0.3 mL/min, using a linear gradient composed of water (mobile phase A) and acetonitrile (mobile phase B): 0–6 min, 10–20% B; 6–50 min, 20–40% B; 50–51 min, 40–10% B; 51–55 min, 10% B, and each injection volume was 10 μL.

MS analysis was operated using a Xevo Triple Quadrupole MS (Waters Corporation, Milford, MA, USA) equipped with an ESI source set in negative ionization mode. The parameters were set as follows: capillary voltage, 2.4 kV; cone voltage, 40 V; desolvation temperature, 200 °C; desolvation gas flow, 550 L/h; collision gas, argon; collision energy, 55 eV; desolvation gas, nitrogen; cone gas flow, 150 L/h; source temperature, 150 °C. The quantitation analysis data was collected using the multiple reaction monitoring (MRM) modes (Sun et al., 2022) by simultaneously screening parent and daughter ions. The MassLynx software (version 4.1, Waters, Milford, MA, USA) was used to control the instruments and data acquisition and processing.

2.7. Statistical analysis

Statistical analysis was performed using Microsoft Excel 2019. The data are shown as mean value ± standard deviation (SD) with triplicates.

3. Results and discussion

3.1. Sweet taste-guided isolation and structure determination of compounds 1 ~ 16

The 95% ethanol extraction of *G. pentaphyllum* was fractionated by silica gel CC to yield eight fractions (Fr. 1 ~ Fr. 8). The sweet taste active portion Fr. 8 by sweet sensory evaluation was further fractionated by macroporous resins column chromatography to obtain six subfractions (Fr.8a ~ Fr.8f), of which the sweet taste active subfractions (Fr.8c ~ Fr.8e) were subjected to further isolation and purification on silica gel column, RP C-18 CC, and semipreparative HPLC to yield fifteen sweet-tasting gypenoside (named as gypenosides YN 1–9, 11–16), one bitter-tasting gypenoside (named as gypenosides YN 10). All sixteen compounds obtained as a white amorphous powder are new natural products (Figs. 1–3), and they could be divided into three types according to the side chain at C-20 of aglycone. Their structures were elucidated as dammarane-type tetracyclic triterpenoids based on their experimental spectroscopic data from one-dimensional (1D) and two-dimensional (2D) nuclear magnetic resonance (NMR) spectra, mass spectroscopy (MS), infrared spectroscopy (IR), and UV–visible spectroscopy (UV). The data of heteronuclear single-quantum correlation spectroscopy (HSQC), heteronuclear multiple-bond correlation spectroscopy (HMBC), ¹H–¹H correlation spectroscopy (COSY), rotating frame nuclear overhauser effect spectroscopy (ROESY) spectra, and heteronuclear single quantum coherence-total correlation spectroscopy (HSQC-TOCSY) analysis ascertain the assignments of the sugar units and aglycone. Fig. 4 exhibits the selected HMBC, ROESY, and ¹H–¹H COSY correlations of each representative compound of the three types (3, 6, and 12). The Supporting Information has exhibited all the detailed structure determinations of the sixteen compounds.

3.2. Sensory evaluations of extraction, fractions and isolated gypenosides

The sweetness intensities of gypenosides YN 1–16 were tested by

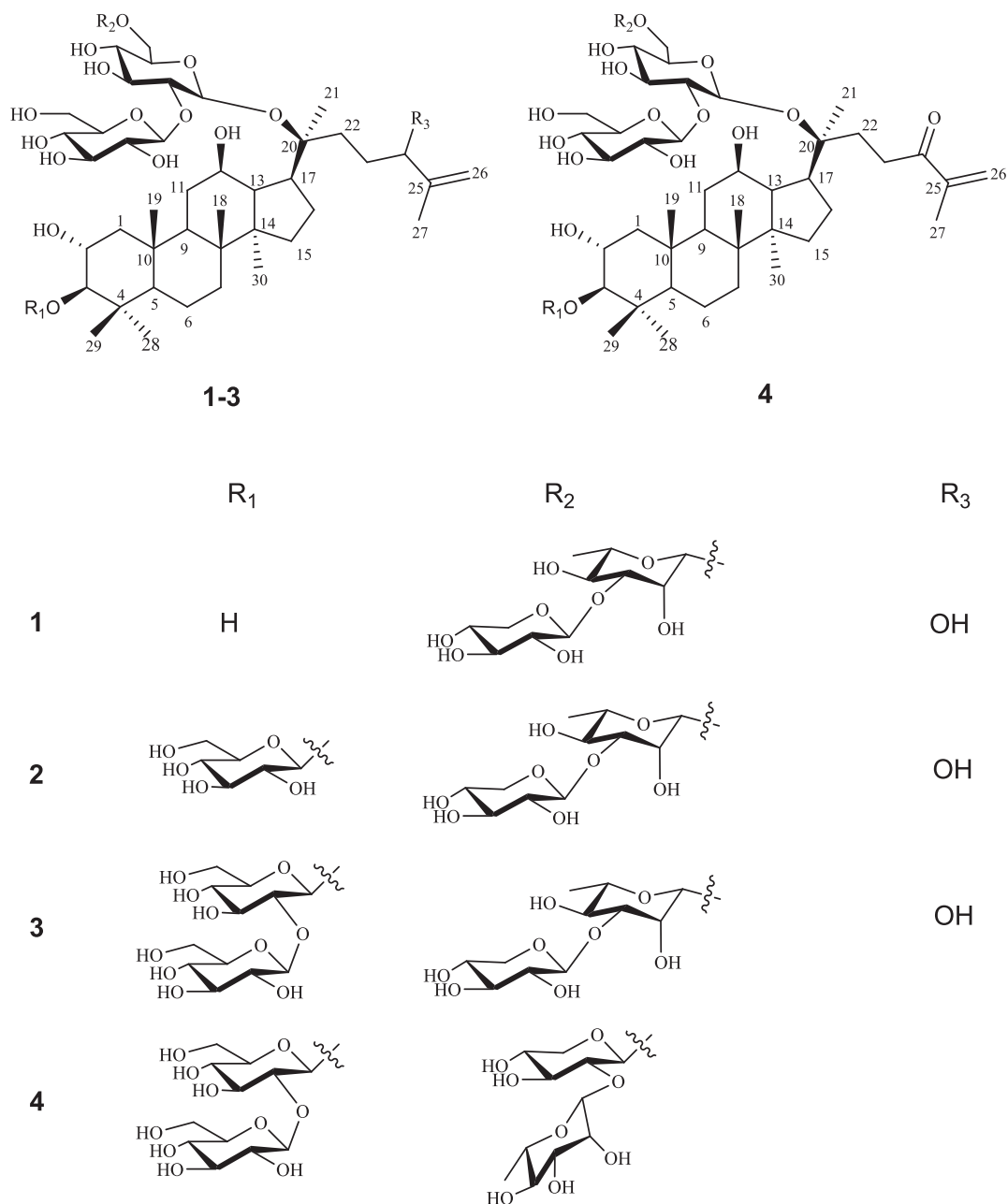


Fig. 1. The structures of compounds 1-4.

seven tasters (see Table 1). It indicated that fifteen of them (1-9, and 11-16) have a sweet taste, and one compound (10) has a bitter taste. Among them, compound 4 has the highest sweet intensity (100 times more than that of sucrose) and the lowest taste thresholds of 0.03 mg/mL, followed by 12, 13, 14, and 16 with moderate sweet intensities and sweetness threshold. Ten gypenosides, 1-3, 5-9, 11, and 15, have the lower sweet intensities and the higher sweetness thresholds. Of these compounds, 1, 4, 7-9 have better gustatory sensations, no after-taste or no bitter taste, and 2, 3, 5, 16 have a slightly astringent taste after-taste, and 6, 11-15 have a slightly bitter after-taste.

During the research process, the sensory evaluation method was used to discover the sweet "Jiaogulan" varieties from many samples. It is a method that can quickly locate fractions containing sweet substances, but at the same time, it is more time-consuming and labor-intensive than mechanized methods. Currently, the electronic tongue, an instrumental method, has been successfully applied to the qualitative and quantitative study of food taste characteristics (Velo, Dias, Rodrigues, Pereira,

& Peres, 2016). Replacing sensory evaluation during sweet variety discovery and sensory-guided separation with an electronic tongue will provide rapid, reproducible results, with continuous operation, especially when used in industrial production it will be cost effective (Phat, Moon, & Lee, 2016; Veloso et al., 2016), and avoid limitations on the number of samples a trained sensory panel can assess per day (Velo et al., 2016). However, gustatory property evaluation of the isolated compounds is more advantageous for sensory evaluation, and remains the only method by which to provide integrated, direct measurements of peoples' perceived intensities of target attributes (Phat et al., 2016).

3.3. Quantitation analysis of gypenosides by UPLC-MS/MS

The quantities of compounds 1-5, 8-13, and 15 were analyzed by UPLC-MS/MS using the MRM mode. The test data is shown in Table 2. Compound 4 had the highest sweetness, although its concentration was slightly lower, analyzed quantitatively as 15.504 ± 1.343 mg/kg.

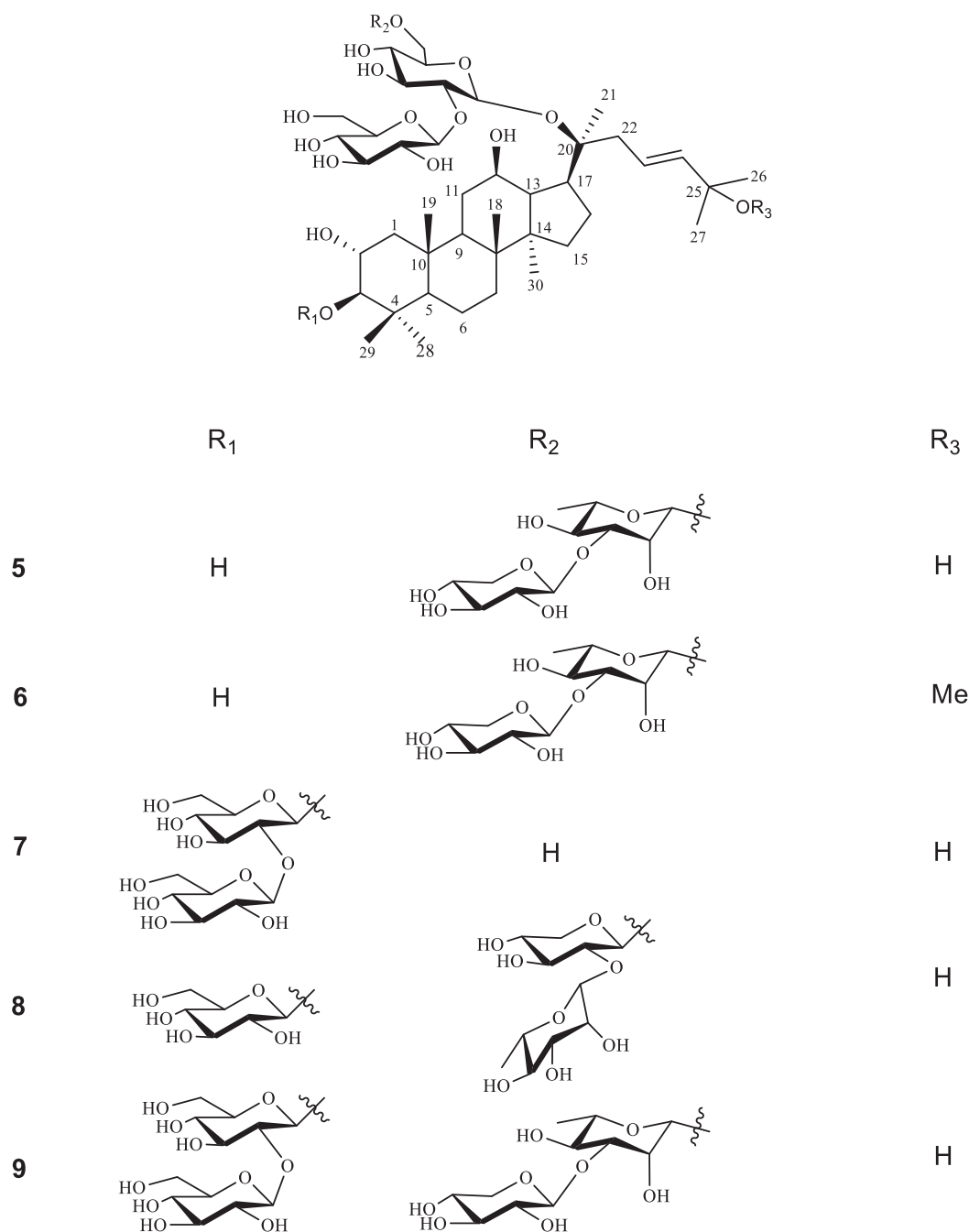


Fig. 2. The structures of compounds 5–9.

Compound 12 was the most abundant substance analyzed quantitatively, reaching 1397.674 ± 12.948 mg/kg. As its sweetness is 50 times higher than that of sucrose, it can provide a preliminary explanation for the especially sweet taste of this sweet variety. The content of compound 13, whose sweetness is also 50 times higher than that of sucrose, was quantified as 69.767 ± 2.326 mg/kg. Compounds 6, 7, 14, and 16 were used up in the process of structure elucidation and taste activity evaluation due to the small amount of isolated compounds, resulting in no standard substances to make quantitative curves, so these four compounds were not assayed. According to UPLC analysis, compounds 1 and 5, 2 and 8, 3 and 9 could not be separated from each other for the same molecular, parent ions, and daughter ions. Therefore compounds 1, 2, and 3 were randomly selected as standard substances to quantify the contents for each of these three pairs of compounds,

respectively.

3.4. Structure–activity relationship discussions of isolated gypenosides

Sixteen isolated gypenosides could be divided into three types according to the side chain at C-20 of aglycone. Compounds 1–4 belong to type I, of which the side chain at C-20 is 4-methyl-4-pentenyl group and its derivatives. Compounds 5–9 belong to type II, of which the side chain is 4-methyl-2-pentenyl and its derivatives. Compounds 10–16 belong to type III, of which the side chain is 4-methyl-3-pentenyl. The other structural differences of the sixteen compounds include the sequence and the number of sugar chains linked at C-3 and C-20, the side chain at C-20, and the hydroxylation of C-2. When the sugar chains at C-3 and C-20 are the same, it could be inferred that compounds belonging to type

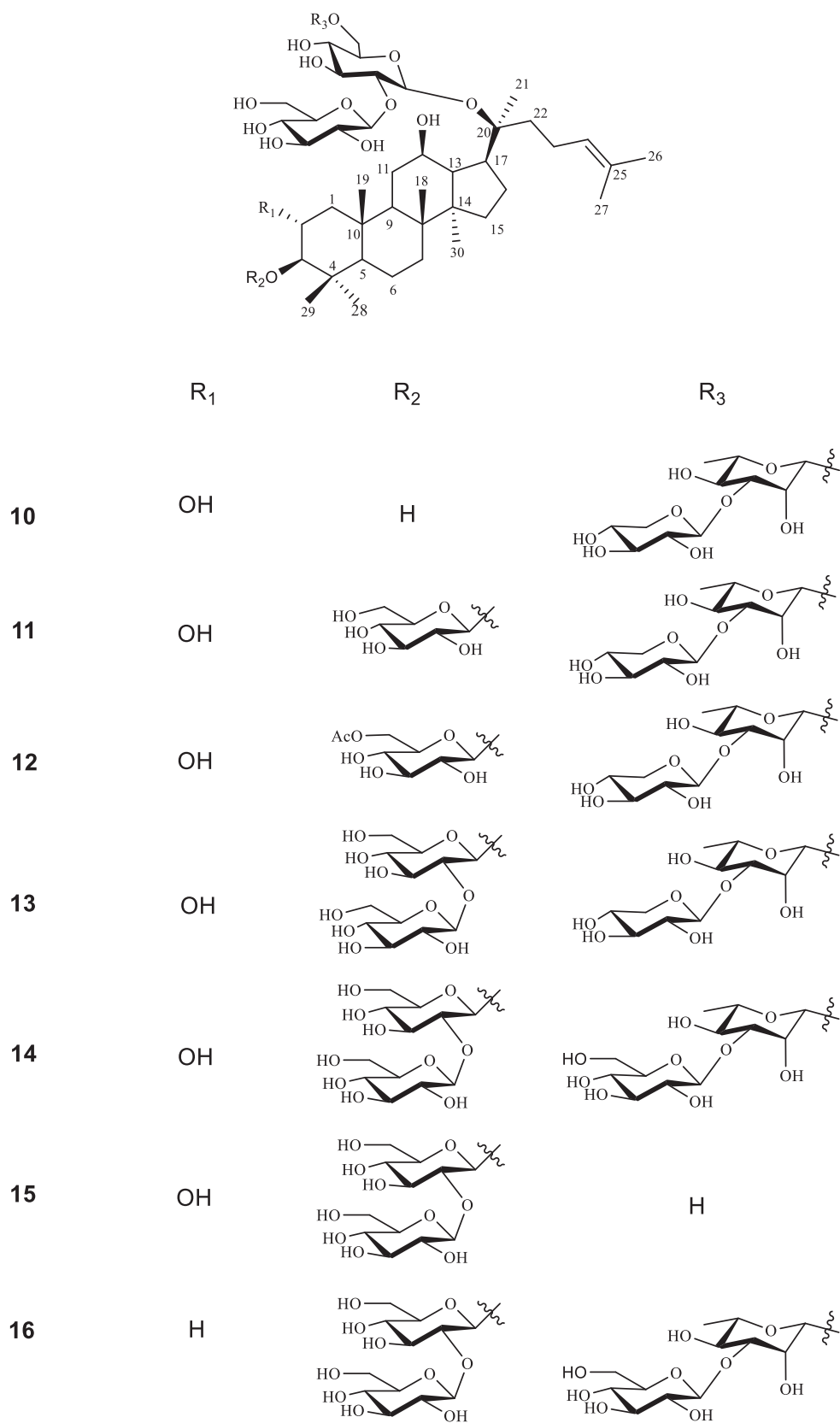


Fig. 3. The structures of compounds 10–16.

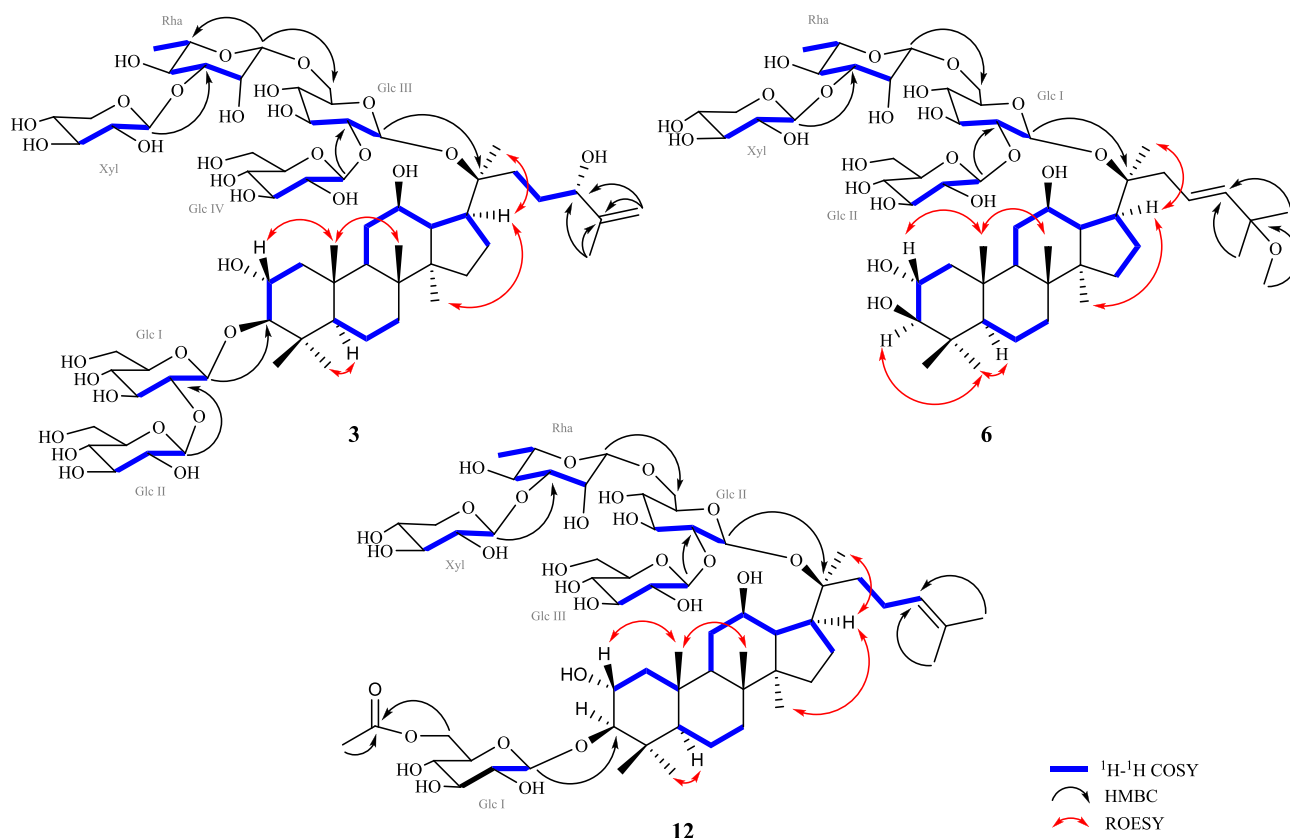


Fig. 4. Selected HMBC, ^1H - ^1H COSY, and ROESY correlations of **3**, **6**, **12** (The sixteen isolated compounds are divided into three types according to the side chain at C-20, and the three compounds selected here are the representatives of the three types.).

Table 1

The sensory evaluation of compounds 1–16.

Compounds	Type	R ₁	R ₂	R ₃	Taste characters	Sweet intensity ^a	Sweet taste threshold (mg/mL) ^b
1	I	-H	α -L-Rha ³ - β -D-Xyl	-OH	Sweet, better gustatory sensation	20	0.17
2	I	β -D-Glc	α -L-Rha ³ - β -D-Xyl	-OH	Sweet, astringent taste after-taste	20	0.17
3	I	β -D-Glc ² - β -D-Glc	α -L-Rha ³ - β -D-Xyl	-OH	Sweet, astringent taste after-taste	20	0.17
4	I	β -D-Glc ² - β -D-Glc	β -D-Xyl ² - α -L-Rha	=O	Sweet, better gustatory sensation	100	0.03
5	II	-H	α -L-Rha ³ - β -D-Xyl	-H	Sweet, astringent taste after-taste	10	0.1
6	II	-H	α -L-Rha ³ - β -D-Xyl	-CH ₃	Sweet, a slightly bitter after-taste	30	0.07
7	II	β -D-Glc ² - β -D-Glc	-H	-H	Sweet, better gustatory sensation	20	0.07
8	II	β -D-Glc	β -D-Xyl ² - α -L-Rha	-H	Sweet, better gustatory sensation	10	0.3
9	II	β -D-Glc ² - β -D-Glc	α -L-Rha ³ - β -D-Xyl	-H	Sweet, better gustatory sensation	10	0.3
10	III	-OH	-H	α -L-Rha ³ - β -D-Xyl	Bitter	—	—
11	III	-OH	β -D-Glc	α -L-Rha ³ - β -D-Xyl	Sweet, a slightly bitter after-taste	30	0.1
12	III	-OH	β -D-Glc ⁶ -acetyl	α -L-Rha ³ - β -D-Xyl	Sweet, a slightly bitter after-taste	50	0.1
13	III	-OH	β -D-Glc ² - β -D-Glc	α -L-Rha ³ - β -D-Xyl	Sweet, a slightly bitter after-taste	50	0.10
14	III	-OH	β -D-Glc ² - β -D-Glc	α -L-Rha ³ - β -D-Glc	Sweet, a slightly bitter after-taste	80	0.03
15	III	-OH	β -D-Glc ² - β -D-Glc	-H	Sweet, a slightly bitter after-taste	10	0.07
16	III	-H	β -D-Glc ² - β -D-Glc	α -L-Rha ³ - β -D-Xyl	Sweet, astringent taste after-taste	50	0.07

^a Sweet intensity relative to the sucrose solution (1%, w/w).

^b Minimum concentration that human can percept.

III have higher sweet intensity than the other two types (**3**, **9**, **13**, and **16**; **2**, **8**, and **11**). In type I, the amount of glucosyl group has almost no effect on sweet intensity (**1**, **2**, and **3**), this phenomenon is also reflected in Type II (**5**, **8**, and **9**). Compound **4** has the highest sweet intensity (100 times that of sucrose solution), which may be due to differences in the order of the sugar chains connected to the sixth position of Glc III, or may be related to the oxidation of the 24-position hydroxyl group to a carbonyl group. In contrast, the sweet intensity increases in type III as the amount of glucosyl group of C-3 (**10**, **11**, **12**, and **13**), especially compound **10**, with C-3 free hydroxyl, which is a bitter compound with no sweet taste. In addition, because xylosyl group of sugar chain at C-20

is replaced by glucosyl, the sweetness of compound **14** is significantly increased (**13**, **14**, and **16**). Acetylation may also increase sweetness (**11** and **12**).

In summary, we speculate that sweetness intensity may be affected by oxidation, glycosylation and acetylation in the chemical structures of gypenosides. This phenomenon may be caused by changes in the position or number of binding sites and binding energy between gypenosides and human taste receptors after oxidation, glycosylation, and acetylation, which would provide more opportunities for the hydrogen bonding which is responsible for sweetness signaling (Mayank & Jaitak, 2015). This speculation would be worth elucidating in future research.

Table 2

Parameters of MRM mode to acquire data and content of compounds 1–5, 8–13, and 15 in the aerial parts of Jiaogulan.

Compd.	Trace ion (m/z)		RT (min)	Cone voltage	Collision energy	Dwell (secs)	Regression equation (n = 3)	R ²	Area	Content in aerial parts of Jiaogulan (mg/kg) (Mean ± SD, n = 3)
	parent	daughter								
1 + 5	1093.25	337.00	21.34	70.0	65.0	0.003	y = 23532x + 349.14	0.9997	78043.971 ± 911.823	76.744 ± 0.000
2 + 8	1255.20	337.04	15.27	40.0	30.0	0.003	y = 2143.5x - 165.91	0.9997	3207.196 ± 156.784	101.550 ± 0.000
3 + 9	1417.25	337.00	9.00	60.0	55.0	0.003	y = 438.48x - 1783.9	0.9994	3207.196 ± 16.957	186.047 ± 0.000
4	1415.30	337.00	12.18	70.0	60.0	0.003	y = 14548x - 402.11	0.9998	76886.977 ± 667.026	15.504 ± 1.343
10	1077.25	337.00	37.61	50.0	45.0	0.003	y = 98126x - 5646.8	1.0000	693364.084 ± 3451.011	166.667 ± 0.343
11	1239.30	337.00	30.75	70.0	50.0	0.003	y = 66994x - 6115.3	1.0000	76886.977 ± 2512.974	27.132 ± 1.343
12	1281.25	1221.33	37.07	40.0	35.0	0.003	y = 2723.1x - 240.15	0.9998	159863.339 ± 1458.450	1397.674 ± 12.948
13	1401.25	337.00	22.09	80.0	65.0	0.003	y = 12670x + 318.92	1.0000	39178.426 ± 1134.742	69.767 ± 2.326
15	1123.25	961.20	22.43	70.0	60.0	0.003	y = 1295.6x - 17.212	1.0000	6342.653 ± 35.647	113.953 ± 0.000

4. Conclusions

Food choices and preferences are generally influenced by palatability. For the use of “Jiaogulan”, as an herbal tea, its sweet varieties are significantly more popular with consumers. This paper used gustatory-guided isolation to study the compound responsible for the sweet taste in a variety of “Jiaogulan” from Yunnan province of China. This led to the first discovery of fifteen sweet-tasting compounds and one bitter-tasting compound. All of them are new dammarane-type tetracyclic triterpenoids. These compounds are the main constituents responsible for the sweet tastes of *G. pentaphyllum*. They are characterized by a pleasant sweet taste without any bitter taste or by only a subtle bitter after-taste. This study clarified the chemical basis for the sweet taste of this plant. The newly discovered compounds are potential natural sweeteners and thereby provide new lead substances for the study of natural non-sugar sweeteners. Compound 4 is especially promising for use as a sweetener, due to its high sweetness intensity. Compound 12 has the highest content of the sweet-tasting compounds. The results of this study provide leading molecules for the development of new natural non-sugar sweet substances. It also enriches the natural sweet compound library, and provides a theoretical basis and technical support for discovering natural sweeteners from this extraordinary sweet “Jiaogulan” variety.

In addition, “Jiaogulan” herbal tea is produced by manufacturing processes like green tea, including plucking, pan firing or steaming, rolling, and drying. Whether differences of these sweetness compounds (including structure and content) emerge during high-temperature processing should be studied in the further. Further research is also needed into the differences in molecular mechanisms between this sweet variety from Yunnan province of China and other sweet varieties or bitter varieties elsewhere.

CRedit authorship contribution statement

Hong-Xia Zhang: Validation, Investigation, Writing – original draft. **Zhong-Ze Wang:** Investigation, Formal analysis, Writing – original draft. **Zhi-Zhi Du:** Supervision, Conceptualization, Writing – review & editing, Project administration, Funding acquisition.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.foodchem.2022.132981>.

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