

## Polyynes from *Toona ciliata* var. *ciliata* and Related Cytotoxic Activity

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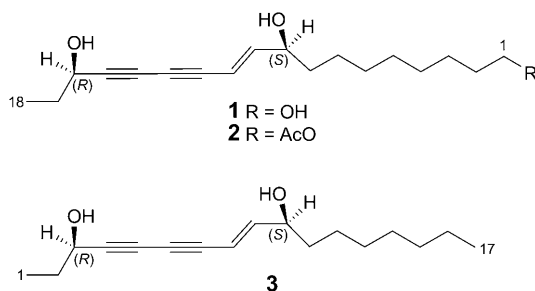
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A phytochemical investigation of *Toona ciliata* var. *ciliata* afforded three new polyynes, **1–3**. Their structures were elucidated on the basis of spectroscopic analysis and chemical methods. Only compound **3** exhibited potent cytotoxicity against the HL-60 cell line with an  $IC_{50}$  value of  $6.7 \pm 0.27 \mu\text{M}$ .

**Introduction.** – *Toona ciliata* var. *ciliata* (Meliaceae), a good timber tree, is widely distributed in the south of China, such as Yunnan, Sichuan, Guangdong, Hainan provinces [1]. Previous chemical investigations on *Toona ciliata* and its varieties have led to the isolation of a series of bioactive compounds, especially limonoids [2]. In the course of our search for structurally unique and potentially bioactive natural products from the Meliaceae family, three new polyynes, (9*S*,10*E*,16*R*)-octadec-10-ene-12,14-diyne-1,9,16-triol (**1**), (9*S*,10*E*,16*R*)-9,16-dihydroxyoctadec-10-ene-12,14-diyne-1-yl acetate (**2**), and (3*R*,8*E*,10*S*)-heptadec-8-ene-4,6-diyne-3,10-diol (**3**), were isolated from the leaves of *Toona ciliata* var. *ciliata*. Here, we report the isolation and structure elucidation of the new compounds, and their cytotoxicity.



**Results and Discussion.** – The AcOEt extract of the leaves of *Toona ciliata* var. *ciliata* was subjected to  $\text{SiO}_2$  and *Sephadex LH-20* column chromatography, as well as semi-preparative HPLC to afford three new compounds, **1–3**.

Compound **1** was obtained as a colorless oil. The molecular formula was determined as  $\text{C}_{18}\text{H}_{28}\text{O}_3$  by a *pseudo*-molecular-ion peak in the HR-ESI-MS ( $m/z$  315.1935 ( $[M + \text{Na}]^+$ ; calc. 315.1936)). The IR absorption band at  $3424 \text{ cm}^{-1}$  implied the presence of an OH group. Its UV spectrum exhibited absorption maxima at 215, 241, 254, 268, and

284 nm, suggesting a typical ene-diyne system [3]. Obvious signals in the  $^1\text{H-NMR}$  spectrum (Table 1) were those of two olefinic H-atoms ( $\delta(\text{H})$  6.31 (*dd*,  $J = 6.4, 16.0$ , H–C(10)); 5.75 (*d*,  $J = 16.0$ , H–C(11)), two O-bearing CH groups ( $\delta(\text{H})$  4.20 (*dd*,  $J = 6.4, 12.4$ , H–C(9)); 4.42 (*t*,  $J = 9.2$ , H–C(16)), one O-bearing  $\text{CH}_2$  group ( $\delta(\text{H})$  3.64 (*t*,  $J = 6.4$ ,  $\text{CH}_2(1)$ ), and one terminal Me group ( $\delta(\text{H})$  1.01 (*t*,  $J = 3.6$ , Me(18)). The  $^{13}\text{C-NMR}$  (Table 2) and DEPT spectra further showed signals for four quaternary acetylenic C-atoms ( $\delta(\text{C})$  77.3 (*s*, C(12)), 73.7 (*s*, C(13)), 69.5 (*s*, C(14)), 83.0 (*s*, C(15))), as well as for eight non-O-bearing  $\text{CH}_2$  groups. Detailed information on the structure was provided by HMBC and  $^1\text{H},^1\text{H-COSY}$  data (Fig. 1). A detailed analysis of the  $^1\text{H},^1\text{H-COSY}$  spectrum of **1** established the three fragments **1a** (C(1)–C(2)), **1b** (C(8) to C(11)) and **1c** (C(16) to C(18)). Further HMBCs of H–C(11) and H–C(16) to C(13) established the connection between fragments **1b** with **1c** via the conjugated diynes, and the connection from **1a** to **1b** via five  $\text{CH}_2$  groups was deduced by the observed HMBCs of H–C(8) and H–C(2) to the relative  $\text{CH}_2$  signals. Thus, **1** was determined as octadec-10-ene-12,14-diyne-1,9,16-triol. The C(10)=C(11) bond was assigned the (*E*)-configuration on the basis of the large vicinal coupling constant ( $J(10,11) = 16.0$  Hz).

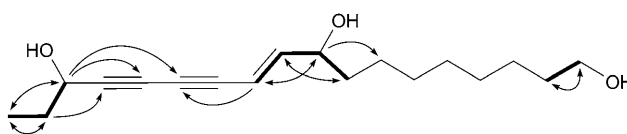


Fig. 1. Key  $^1\text{H},^1\text{H-COSY}$  correlations (—) and HMBCs (---) of **1**

Table 1.  $^1\text{H-NMR}$  Data of Compounds **1** (400 MHz), **2** (500 MHz), and **3** (400 MHz). Measured in  $\text{CDCl}_3$ ;  $\delta$  in ppm

	<b>1</b>	<b>2</b>	<b>3</b>
$\text{CH}_2(1)$ or Me(1)	3.64 ( <i>t</i> , $J = 6.4$ )	4.05 ( <i>t</i> , $J = 6.8$ )	1.01 ( <i>t</i> , $J = 7.2$ )
$\text{CH}_2(2)$	1.54–1.58 ( <i>m</i> )	1.60 ( <i>dd</i> , $J = 6.8, 13.5$ )	1.72–1.76 ( <i>m</i> )
$\text{CH}_2(3)$ or H–C(3)	1.30 ( <i>br. s</i> )	1.30 ( <i>br. s</i> )	4.42 ( <i>t</i> , $J = 6.8$ )
$\text{CH}_2(4)$	1.30 ( <i>br. s</i> )	1.30 ( <i>br. s</i> )	
$\text{CH}_2(5)$	1.30 ( <i>br. s</i> )	1.30 ( <i>br. s</i> )	
$\text{CH}_2(6)$	1.30 ( <i>br. s</i> )	1.30 ( <i>br. s</i> )	
$\text{CH}_2(7)$	1.30 ( <i>br. s</i> )	1.30 ( <i>br. s</i> )	
$\text{CH}_2(8)$ or H–C(8)	1.51–1.55 ( <i>m</i> )	1.53 ( <i>t</i> , $J = 6.5$ )	5.75 ( <i>d</i> , $J = 16.0$ )
H–C(9)	4.20 ( <i>dd</i> , $J = 6.4, 12.4$ )	4.19 ( <i>dd</i> , $J = 6.5, 12.0$ )	6.31 ( <i>dd</i> , $J = 16.0, 5.6$ )
H–C(10)	6.31 ( <i>dd</i> , $J = 6.4, 16.0$ )	6.31 ( <i>dd</i> , $J = 6.5, 16.0$ )	4.17 ( <i>ddd</i> , $J = 12.8, 5.6, 1.6$ )
H–C(11) or $\text{CH}_2(11)$	5.75 ( <i>d</i> , $J = 16.0$ )	5.76 ( <i>d</i> , $J = 16.0$ )	1.51–1.55 ( <i>m</i> )
$\text{CH}_2(12)$			1.26 ( <i>br. s</i> )
$\text{CH}_2(13)$			1.26 ( <i>br. s</i> )
$\text{CH}_2(14)$			1.26 ( <i>br. s</i> )
$\text{CH}_2(15)$			1.26 ( <i>br. s</i> )
H–C(16) or $\text{CH}_2(16)$	4.42 ( <i>t</i> , $J = 9.2$ )	4.42 ( <i>t</i> , $J = 9.2$ )	1.26 ( <i>br. s</i> )
$\text{CH}_2(17)$ or Me(17)	1.73–1.77 ( <i>m</i> )	1.74–1.78 ( <i>m</i> )	0.87 ( <i>t</i> , $J = 6.4$ )
Me(18)	1.01 ( <i>t</i> , $J = 3.6$ )	1.01 ( <i>t</i> , $J = 3.8$ )	
AcO		2.05 ( <i>s</i> )	

<sup>a</sup>) Assignments may be interchanged.

Table 2.  $^{13}\text{C}$ -NMR Data of Compounds **1**–**3**. At 100 MHz,  $\delta$  in  $\text{CDCl}_3$ , in ppm.

C-Atom	<b>1</b>	<b>2</b>	<b>3</b>
1	63.0 ( <i>t</i> )	64.6 ( <i>t</i> )	9.4 ( <i>q</i> )
2	32.6 ( <i>t</i> )	28.5 ( <i>t</i> )	30.6 ( <i>t</i> )
3	29.4 ( <i>t</i> ) <sup>a</sup>	25.8 ( <i>t</i> ) <sup>a</sup>	64.1 ( <i>d</i> )
4	29.3 ( <i>t</i> ) <sup>a</sup>	29.1 ( <i>t</i> ) <sup>a</sup>	83.0 ( <i>s</i> )
5	29.2 ( <i>t</i> ) <sup>a</sup>	29.3 ( <i>t</i> ) <sup>a</sup>	69.5 ( <i>s</i> )
6	25.6 ( <i>t</i> ) <sup>a</sup>	29.3 ( <i>t</i> ) <sup>a</sup>	73.7 ( <i>s</i> )
7	25.1 ( <i>t</i> )	25.1 ( <i>t</i> )	77.3 ( <i>s</i> )
8	36.8 ( <i>t</i> )	36.8 ( <i>t</i> )	108.1 ( <i>d</i> )
9	72.0 ( <i>d</i> )	72.0 ( <i>d</i> )	149.6 ( <i>d</i> )
10	149.6 ( <i>d</i> )	149.6 ( <i>d</i> )	72.1 ( <i>d</i> )
11	108.1 ( <i>d</i> )	108.1 ( <i>d</i> )	36.8 ( <i>t</i> )
12	77.3 ( <i>s</i> )	77.3 ( <i>s</i> )	25.2 ( <i>t</i> )
13	73.7 ( <i>s</i> )	73.7 ( <i>s</i> )	29.4 ( <i>t</i> ) <sup>a</sup>
14	69.5 ( <i>s</i> )	69.5 ( <i>s</i> )	29.2 ( <i>t</i> ) <sup>a</sup>
15	83.0 ( <i>s</i> )	82.9 ( <i>s</i> )	31.8 ( <i>t</i> )
16	64.1 ( <i>d</i> )	64.1 ( <i>d</i> )	22.6 ( <i>t</i> )
17	30.6 ( <i>t</i> )	30.6 ( <i>t</i> )	14.1 ( <i>q</i> )
18	9.3 ( <i>q</i> )	9.3 ( <i>q</i> )	
AcO		21.0 ( <i>q</i> ), 171.3 ( <i>s</i> )	

<sup>a</sup>) Assignments may be interchanged.

Compound **2** was isolated as a colorless oil. The molecular formula was established as  $\text{C}_{20}\text{H}_{30}\text{O}_4$  deduced by the *pseudo*-molecular-ion peak in the HR-ESI-MS ( $m/z$  357.2041 ( $[M + \text{Na}]^+$ ); calc. 357.2041). The IR spectrum showed absorptions for OH ( $3419\text{ cm}^{-1}$ ) and CO ( $1718\text{ cm}^{-1}$ ) groups. The UV spectrum and 1D-NMR data were quite similar to those of **1**, except for the occurrence of an additional AcO group ( $\delta(\text{H})$  2.05 (*s*);  $\delta(\text{C})$  21.0 (*q*) and 171.3 (*s*)) in **2**. The downfield shift of  $\text{CH}_2(1)$  ( $\delta(\text{H})$  4.05 (*t*,  $J = 6.8$ )) indicated acylation of the OH group at C(1), which was identified by the HMBC of H–C(1) to ester CO ( $\delta(\text{C})$  171.3 (*s*)). Further 2D-NMR (HSQC, HMBC, and  $^1\text{H}, ^1\text{H}$ -COSY) data confirmed the structure of **2** as 9,16-dihydroxyoctadec-10-ene-12,14-diyn-1-yl acetate. The (*E*)-configuration of the C(10)=C(11) bond was also deduced from the  $J(10,11)$  value (16.0 Hz).

Compound **3** was obtained as a colorless oil. The molecular formula was deduced as  $\text{C}_{17}\text{H}_{26}\text{O}_2$  by a *pseudo*-molecular-ion peak in the HR-ESI-MS ( $m/z$  285.1830 ( $[M + \text{Na}]^+$ ; calc. 285.1830). Comparison of its spectroscopic data with those of **1** and **2** revealed an overall similarity, except for the presence of an additional terminal Me group ( $\delta(\text{H})$  1.01 (*t*,  $J = 7.2$ )), and the absence of two  $\text{CH}_2$  groups (one O-bearing) in **3**. Extensive 2D-NMR (HSQC, HMBC, and  $^1\text{H}, ^1\text{H}$ -COSY) data identified the planar structure of **3** as (*E*)-heptadec-8-en-4,6-diyne-3,10-diol [4]. However, the obvious different chemical shift of C(2) ( $\delta(\text{C})$  30.6 (*t*)) in **3** indicated that **3** was an epimer of the latter, with the only difference being the different absolute configuration at C(3). The Mosher's method was applied for determining its absolute configuration [5].

To this end, **3** was treated with (–)-(*R*)- and (+)-(*S*)-MTPA (=  $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetic acid) chloride to give (*S*)- and (*R*)-MTPA diesters of **3**,

**3a** and **3b**, respectively.  $^1\text{H},^1\text{H}$ -COSY Data were used for the assignment of H-atom signals of **3a** and **3b**. Analysis of the chemical shift differences ( $\Delta\delta = \delta_S - \delta_R$ ) of the H-atoms neighboring the O-bearing CH groups according to the *Mosher* model allowed the assignment of the (*R*)- and (*S*)-configuration at C(3) and C(10) [5], respectively (Fig. 2). Accordingly, compound **3** reported here was a new compound, (3*R*,8*E*,10*S*)-heptadec-8-ene-4,6-diyne-3,10-diol, while the compound reported previously should be (3*S*,8*E*,10*R*)-heptadec-8-ene-4,6-diyne-3,10-diol [4].

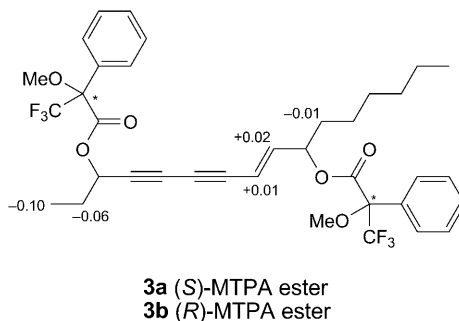


Fig. 2. Application of the modified Mosher's method for secondary alcohols on the MTPA esters of **3** (**3a** and **3b**).  $\Delta\delta$  ( $\delta_S - \delta_R$ ) are given in ppm.

Due to the little amount of **1** and **2**, the absolute configuration of these two compounds could not be determined directly by *Mosher*'s method, but considering that compounds **1** and **2** were isolated from the same extract, and their chemical shifts and optical rotations were quite similar to those of **3**. Thus, the absolute configurations of **1** and **2** are assumed to be the same as **3**.

Compounds **1–3** were tested for *in vitro* inhibitory activities against HL-60, SMMC-7721, A549, SK-BR-3, and PANC-1 human tumor cell lines (details are available as *Supplementary Material*<sup>1)</sup>), using DDP (*cis*-diammineplatinum(II) dichloride) as a positive control. Significant cytotoxicity was only observed for compound **3** against the HL-60 cells with an  $IC_{50}$  value of  $6.7 \pm 0.27 \mu\text{M}$ .

Polyacetylenes are uncommon in Meliaceae. To our knowledge, they had been found within this plant family only in *Swietenia mahagoni* [6]. In this context, an endophytic origin of the compounds cannot be excluded.

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### Experimental Part

*General.* CC: silica gel *H* ( $\text{SiO}_2$ , 10–40  $\mu\text{m}$ ; *Qingdao Marine Chemical Ltd. Co.*); *MCI* gel *CHP 20P* (75–150  $\mu\text{m}$ , *Mitsubishi Chemical Industries Ltd.*); *Sephadex LH-20* (40–70  $\mu\text{m}$ , *Pharmacia*), or *RP-18* gel (40–63  $\mu\text{m}$ , *Merck*). TLC: silica-gel plates (size: 50  $\times$  100 mm, thickness: 0.20–0.25 mm, *Qingdao Marine Chemical Ltd. Co.*), detection by UV illumination and spraying with 10%  $\text{H}_2\text{SO}_4$  in EtOH, followed by heating. HPLC: *Zorbax SB-C-18* column (i.d. 9.4  $\times$  250 mm; *Agilent Co. Ltd.*). Optical rotations: *Perkin–Elmer* model 241 polarimeter. UV Spectra: *Shimadzu UV-2401* spectrophotometer. IR

<sup>1)</sup> *Supplementary Material* may be obtained upon request from the authors.

Spectra: *Bio-Rad FTS-135* spectrometer, with KBr pellets. 1D- and 2D-NMR spectra: *Bruker AM-400* (400 and 100 MHz, resp.) or *DRX-500* (500 and 125 MHz, resp.) instrument with TMS as an internal standard. ESI-MS: *Finnigan MAT 90* instrument. HR-ESI-MS: *API Qstar Pulsar LC/TOF* instrument.

*Plant Material.* The leaves of *Toona ciliata* var. *ciliata* were collected from the area of Gaoligongshan, Yunnan Province, P. R. China, in July 2008, and were identified by Prof. H. Li (Kunming Institute of Botany, Chinese Academy Sciences). A voucher specimen (KUN No. 080426) was deposited with the State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy Sciences.

*Extraction and Isolation.* The air-dried and powdered leaves of the plant (3.5 kg) were extracted with EtOH 95% (3 × 6 l, 5 h each). The extracts were then suspended in H<sub>2</sub>O and further extracted with petroleum ether (PE; 3 × 2 l) and AcOEt (4 × 3 l). The AcOEt extracts (92 g) were first subjected to CC (SiO<sub>2</sub>; gradient of PE/acetone 10:0, 8:1, 7:3, and 6:4) to afford *Fractions 1–10*. *Fr. 3* (6.8 g) was first subjected to CC (*MCI* gel; gradient of MeOH/H<sub>2</sub>O (60:40 to 100:0 (v/v))) to afford eight fractions, *Fr. A1–A8*. *Fr. A4* (654 mg) was subjected to CC (*Sephadex LH-20*; acetone 100%) to afford *Fr. B1–B3*. *Fr. B1* (32 mg) was further purified by HPLC (MeOH/H<sub>2</sub>O 70:30; flow rate: 3.0 ml/min; detection: UV 254, 230, 210 nm; *t<sub>R</sub>*: 9 min (**2**), 14 min (**1**)) at 30°, yielding compounds **1** (8 mg) and **2** (6 mg). *Fr. B2* (60 mg) was subjected to CC (SiO<sub>2</sub>; gradient of CHCl<sub>3</sub>/acetone 100:1) to afford **3** (20 mg).

(*9S,10E,16R*)-*Octadec-10-ene-12,14-diyne-1,9,16-triol* (**1**). Colorless oil.  $[\alpha]_D^{25} = -25.8$  ( $c = 0.070$ , MeOH). UV (MeOH): 208 (1.5), 215 (1.8), 228 (0.2), 241 (0.3), 254 (0.5), 268 (0.7), 284 (0.6). IR (KBr): 3425, 2931, 2856, 1629. <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Table 1* and 2, resp. ESI-MS: 583 ( $[2M - 1]^-$ ). HR-ESI-MS: 315.1935 ( $[M + Na]^+$ , C<sub>18</sub>H<sub>28</sub>NaO<sub>3</sub>; calc. 315.1936).

(*9S,10E,16R*)-*9,16-Dihydroxyoctadec-10-ene-12,14-diyn-1-yl Acetate* (**2**). Colorless oil.  $[\alpha]_D^{25} = -12.8$  ( $c = 0.2$ , MeOH). UV (MeOH): 208 (1.4), 215 (1.8), 228 (0.2), 241 (0.3), 254 (0.5), 268 (0.7), 284 (0.6). IR (KBr): 3419, 2932, 2857, 1738, 1718, 1248, 1048. <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Table 1* and 2, resp. ESI-MS: 357 ( $[M + Na]^+$ ). HR-ESI-MS: 357.2050 ( $[M + Na]^+$ , C<sub>20</sub>H<sub>30</sub>NaO<sub>4</sub>; calc. 357.2041).

(*3R,8E,10S*)-*Heptadec-8-ene-4,6-diyne-3,10-diol* (**3**). Colorless oil.  $[\alpha]_D^{25} = -10.9$  ( $c = 0.2$ , MeOH). UV (MeOH): 208 (1.9), 215 (2.4), 229 (0.2), 241 (0.4), 254 (0.7), 268 (1.0), 284 (0.8). IR (KBr): 3405, 2929, 2857, 1016, 956. <sup>1</sup>H- and <sup>13</sup>C-NMR: see *Table 1* and 2, resp. ESI-MS: 285 ( $[M + Na]^+$ ). HR-ESI-MS: 285.1830 ( $[M + Na]^+$ , C<sub>17</sub>H<sub>26</sub>NaO<sub>2</sub>; calc. 285.1830).

*Determination of the Absolute Configuration of Compound 3 by the Mosher's Method.* Compound **3** (2.0 mg) was dissolved in 250 μl of dry pyridine and treated with 4-(dimethylamino)pyridine (DMAP; a spatula tip) and (–)-(R)-MTPA (= *α*-methoxy-*α*-(trifluoromethyl)phenylacetic acid) chloride (10 μl). The mixture was stirred at r.t. for 1 h. After removal of the solvent, the mixture was purified by CC (*RP-18*; acetone/H<sub>2</sub>O 50:50) to afford the (*S*)-MTPA diester **3a** (4.6 mg). The same procedure afforded the (*R*)-MTPA diester **3b** (5.1 mg).

(*3R,8E,10S*)-*Heptadec-8-ene-4,6-diyne-3,10-diyl (2S,2'S)-Bis[3,3,3-trifluoro-2-methoxy-2-phenylpropanoate]* (**3a**). Colorless oil. <sup>1</sup>H-NMR (500 Hz, CDCl<sub>3</sub>): 7.53, 7.49, 7.41 (MTPA H-atoms); 6.20 (*ddd*,  $J = 39.0, 16.0, 7.5$ , H–C(9)); 5.72 (*dd*,  $J = 62.5, 16.0$ , H–C(8)); 5.59 (*t*,  $J = 6.5$ , H–C(3)); 5.48 (*t*,  $J = 7.5$ , H–C(10)); 3.58, 3.53 (MTPA MeO, overlapped); 1.83–1.87 (*m*, CH<sub>2</sub>(2)); 1.64–1.72 (*m*, CH<sub>2</sub>(11)); 1.28 (*d*,  $J = 6.6$ , H<sub>a</sub>–C(12)); 1.25 (H<sub>b</sub>–C(12), overlapped); 1.25 (CH<sub>2</sub>(13), overlapped); 1.20 (CH<sub>2</sub>(14), overlapped); 1.20 (CH<sub>2</sub>(15), overlapped); 1.20 (CH<sub>2</sub>(16), overlapped); 0.94 (*t*,  $J = 7.0$ , Me(1)); 0.87 (*t*,  $J = 3.0$ , Me(17)). The assignments of CH<sub>2</sub>(13), CH<sub>2</sub>(14), CH<sub>2</sub>(15), and CH<sub>2</sub>(16) may be interchanged.

(*3R,8E,10S*)-*Heptadec-8-ene-4,6-diyne-3,10-diyl (2R,2'R)-Bis[3,3,3-trifluoro-2-methoxy-2-phenylpropanoate]* (**3b**). Colorless oil. <sup>1</sup>H-NMR (500 Hz, CDCl<sub>3</sub>): 7.53, 7.49, 7.41 (MTPA H-atoms); 6.18 (*ddd*,  $J = 39.0, 16.0, 7.0$ , H–C(9)); 5.71 (*dd*,  $J = 62.5, 16.0$ , H–C(8)); 5.55 (*t*,  $J = 6.5$ , H–C(3)); 5.48 (*t*,  $J = 7.0$ , H–C(10)); 3.55 (MTPA MeO, overlapped); 1.89–1.93 (*m*, CH<sub>2</sub>(2)); 1.64–1.74 (*m*, CH<sub>2</sub>(11)); 1.28 (*d*,  $J = 6.6$ , H<sub>a</sub>–C(12)); 1.25 (H<sub>b</sub>–C(12), overlapped); 1.25 (CH<sub>2</sub>(13), overlapped); 1.20 (CH<sub>2</sub>(14), overlapped); 1.20 (CH<sub>2</sub>(15), overlapped); 1.20 (CH<sub>2</sub>(16), overlapped); 1.04 (*t*,  $J = 7.0$ , Me(1)); 0.87 (*t*,  $J = 3.0$ , Me(17)). The assignments of CH<sub>2</sub>(13), CH<sub>2</sub>(14), CH<sub>2</sub>(15), and CH<sub>2</sub>(16) may be interchanged.

*Cytotoxicity Assays.* IC<sub>50</sub> Values of compounds **1–3** against HL-60, SMMC-7721, A549, SK-BR-3, and PANC-1 human tumor cell lines were determined by the MTT (= 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-2H-tetrazolium bromide) method [7]. Briefly, cells were plated in 96-well plates 24 h before treatment and continuously exposed to different concentrations of compounds for 72 h. The percentage

of viable cells was quantified at 595/630 nm with an ELISA reader. The cytotoxic concentration that caused the reduction of viable cells by 50% was determined from dose–response curve, and data were obtained from triplicate experiments.

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