Isolation and Characterization of New Bitter Diterpenoids from the Basidiomycete Sarcodon scabrosus

by Bing-Ji Ma, Hua-Jie Zhu, and Ji-Kai Liu*

Kunming Institute of Botany, the Chinese Academy of Sciences, Kunming 650204, P. R. China (e-mail: jkliu@mail.kib.ac.cn)

The novel cyathane-type diterpenoids scabronine G and H (1 and 2, resp.) were isolated from the fruiting bodies of the basidiomycete *Sarcodon scabrosus* together with four known compounds, allocyathin B_2 (3), sarcodonin A (4), sarcodonin G (5), and scabronine F (6). Their structures were determined by spectroscopic means, including 2D-NMR (HMBC, HMQC, ROESY, ¹H, ¹H-COSY).

Introduction. – Sarcodon scabrosus is a mushroom belonging to the family Thelephoraceae and has a strongly bitter taste. Diterpenoids, sarcodonins A-H and scabronines A-F, have previously been isolated from this mushroom as the bitter principles [1–3]. All these diterpenoids possess a cyathane skeleton consisting of angularly condensed five-, six-, and seven-membered rings and show stimulating activity on nerve-growth-factor (NGF) synthesis *in vitro*. In continuing our studies on basidiomycete-derived bioactive secondary metabolites, we investigated the chemical constituents of the mushroom Sarcodon scabrosus from Yunnan, China. This report describes the structural elucidation of two new compounds, named scabronines G and H (1 and 2, resp.).

Results and Discussion. – The CHCl₃-soluble fraction of the EtOH extract from the fruiting bodies of *S. scabrosus* was subjected to repeated chromatography to give compounds 1 and 2^{1}) (*Fig. 1*).

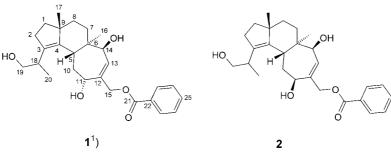


Fig. 1. Structures of scabronines G(1) and H(2)

1) Arbitrary numbering; for systematic names, see Exper. Part.

^{© 2004} Verlag Helvetica Chimica Acta AG, Zürich

Compound **1** was obtained as an optically active oily solid. High-resolution ESI-MS (pos.) gave a pseudomolecular-ion peak at m/z 463.2471 (C₂₇H₃₆O₅Na⁺; calc. 463.2460). Further spectral data (¹H- and ¹³C-NMR (*Table 1*) ¹H,¹H-COSY (*Table 1*), HMQC, HMBC (*Table 1*), and ROESY) established the structure and relative configuration of **1** as shown in *Fig. 1*.

	$\delta(C)$ (DEPT)	$\delta(\mathrm{H})$	¹ H, ¹ H-COSY (selected)	HMBC (selected)
CH ₂ (1)	37.6 (CH ₂)	1.32 (<i>m</i>)		C(17)
$CH_2(2)$	28.6 (CH ₂)	2.15(m)		
C(3)	143.2 (C)			
C(4)	134.5 (C)			
H-C(5)	39.9 (CH)	2.95 (br. $d, J = 11.3$)		
C(6)	42.1 (C)			
$CH_{2}(7)$	32.2 (CH ₂)	$1.20 (m, H_{\alpha}), 2.10 (m, H_{\beta})$		
$CH_{2}(8)$	36.3 (CH ₂)	1.46 (<i>m</i>)		C(4)
C(9)	49.4 (C)			
$CH_{2}(10)$	33.4 (CH ₂)	2.20 (<i>m</i>)	H - C(11)	
H - C(11)	74.1 (CH)	6.02 (<i>m</i>)	$CH_{2}(10)$	
C(12)	142.1 (C)			
H - C(13)	127.1 (CH)	5.82(d, J = 6.8)	H - C(14)	C(11), C(15)
H - C(14)	75.5 (CH)	3.90 (d, J = 6.8)	H-C(13)	C(5), C(7), C(12)
$CH_2(15)$	62.7 (CH ₂)	4.05 (d, J = 12.8),		
		4.02 (d, J = 12.8)		
Me(16)	16.4 (Me)	0.72 (s)		C(5)
Me(17)	24.4 (Me)	1.02 (s)		C(1)
H - C(18)	34.9 (CH)	2.85 (<i>m</i>)	CH ₂ (19), Me(20)	C(4)
$CH_{2}(19)$	65.7 (CH ₂)	3.35 (<i>m</i>), 3.28 (<i>m</i>)	H - C(18)	
Me(20)	15.6 (Me)	0.86 (d, J = 6.7)	H - C(18)	
C(21)	166.4 (C)			
C(22)	129.8 (C)			
H-C(23),	129.4 (CH)	7.91 $(t, J = 7.2)$	H-C(24), H-C(26)	
H - C(27)				
H-C(24),	128.2 (CH)	7.32(t, J = 7.2)	H-C(23), H-C(27),	
H - C(26)			H-C(25)	
H - C(25)	133.0 (CH)	7.46 $(t, J = 7.2)$	• •	

Table 1. ¹*H*- and ¹³*C*-*NMR Data* (CDCl₃) of **1**. δ in ppm, *J* in Hz; arbitrary numbering¹).

The ¹H-NMR spectrum of **1** showed signals for one secondary and two tertiary Me groups at δ 0.86 (d, J = 6.7 Hz), and 0.72 and 1.02 (each *s*, each 3 H), respectively. The secondary Me and a CH₂ group at δ (H) 3.35 and 3.28 (2m) were coupled with a CH proton at δ (H) 2.85 (m), demonstrating the presence of an isolated system CH(Me)CH₂OH. ¹H-NMR Signals of 5 aromatic H-atoms at δ (H) 7.32–7.91 and six signals in the ¹³C-NMR between (δ (C) 128.2 and 133.0, besides a carbonyl signal at δ (C) 166.4, were consistent with a benzoate moiety. Moreover, the ¹³C-NMR of **1** showed two O–CH and one O–CH₂ signals δ (C) 75.5 and 74.1, and δ (C) 62.7, resp.), and signals of a tetrasubstituted δ (C) 143.2 and 134.5) and trisubstituted C=C bond (δ 142.1 (C) and 127.1 (CH)). Based on these partial structures, the construction of the molecular framework was deduced from ¹H, ¹H COSY, HMQC, and HMBC experiments. The HMBC correlations C(2) (δ 28.6)/H–C(18) (δ 2.85), C(5), C(7), and C(12) (δ 3.9, 32.2, and 142.1)/H–C(14) (δ 3.90), C(17) (δ 24.4)/CH₂(1) (δ 1.32), C(4) (δ 134.5)/CH₂(8) (δ 1.46) and H–C(18) (δ 2.85), C(5) (δ 39.9)/Me(16) (δ 0.72), and C(11) and C(15) (δ 74.1 and 62.7)/H₋C(7) and H_{β}–C(5)/H–C(11) indicated that these H-atoms were situated on the same side. In addition, the ROESY correlations H_{α}–C(7)/H–C(14), CH₂(8)/Me(16) and H–C(13)/Me(16) confirmed the proposed relative configuration of **1**.

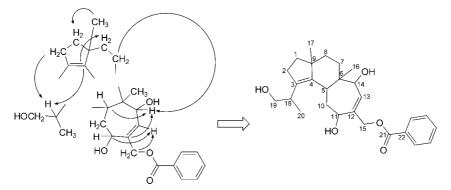


Fig. 2. HMBC Correlations and planar structure of scabronine G (1)

Compound **2** was obtained as an optically active powder. High-resolution ESI-MS (pos.) gave an pseudomolecular-ion peak at m/z 463.2464 ($C_{27}H_{36}O_5Na^+$; calc. 463.2460). The NMR spectra of **2** were similar to those of **1** but on TLC (silica gel), **1** was less polar than **2**. Interpretation of the spectral data (¹H- and ¹³C-NMR (*Table 2*),

Table 2. ¹ <i>H</i> - and ¹³ <i>C</i> - <i>NMR Data</i> (CD ₃ OD) of 2 . δ in ppm, <i>J</i> in Hz; arbitrary numbering ¹).

	$\delta(C)$ (DEPT)	$\delta(\mathrm{H})$	¹ H, ¹ H-COSY (selected)	HMBC (selected)
CH ₂ (1)	39.2 (CH ₂)	1.40 (<i>m</i>)		C(17)
$CH_2(2)$	30.0 (CH ₂)	2.30(m)		
C(3)	143.7 (C)			
C(4)	136.1 (C)			
H-C(5)	41.4 (CH)	2.78 (br. $d, J = 11.2$)	$CH_{2}(10)$	
C(6)	43.8 (C)			
$CH_{2}(7)$	33.0 (CH ₂)	$1.20 (m, H_a), 1.90 (m, H_\beta)$		
$CH_{2}(8)$	38.0 (CH ₂)	1.50 (<i>m</i>)		C(4)
C(9)	50.6 (C)			
$CH_{2}(10)$	37.0 (CH ₂)	2.20 (<i>m</i>)	H - C(11)	
H - C(11)	71.6 (CH)	4.72 (<i>m</i>)	$CH_{2}(10)$	
C(12)	141.2 (C)			
H - C(13)	131.6 (CH)	5.87 $(d, J = 6.8)$	H - C(14)	C(6), C(11), C(15)
H - C(14)	76.5 (CH)	4.02 (d, J = 6.8)	H - C(13)	C(5), C(7), C(12)
$CH_{2}(15)$	67.2 (CH ₂)	4.93 (d, J = 12.8),		C(13), C(21)
		4.90 (d, J = 12.8)		
Me(16)	17.3 (Me)	0.83(s)		C(5), C(7), C(14)
Me(17)	25.0 (Me)	1.03 (s)		C(1), C(4)
H - C(18)	36.5 (CH)	3.03 (<i>m</i>)	CH ₂ (19), Me(20)	C(2), C(4)
$CH_{2}(19)$	67.4 (CH ₂)	3.49 (dd), 3.35 (dd)	H - C(18)	C(3)
Me(20)	16.4 (Me)	0.97 (d, J = 6.7)	H - C(18)	
C(21)	167.8 (C)			
C(22)	128.6 (C)			
H-C(23),	130.6 (CH)	8.03 (t, J = 7.2)	H-C(24), H-C(26)	
H - C(27)				
H-C(24),	129.6 (CH)	7.46 $(t, J = 7.2)$	H-C(23), H-C(27),	
H - C(26)			H-C(25)	
H - C(25)	134.2 (CH ₂)	7.58 $(t, J = 7.2)$		

¹H,¹H-COSY (*Table 2*), HMQC, and HMBC (*Table 2*)) revealed that **2** had the same gross structure as **1**. However, a detailed comparative study of the NMR and ROESY data of **1** and **2** suggested that **2** was the 11-epimer of **1** (*Fig. 1*)

The NMR signals of C(10), C(11), C(13), C(15), and H-C(11), and $CH_2(15)$ of **1** and **2** were observed at different positions. The ROESY correlations $H_a-C(7)/H-C(14)$, $CH_2(8)/Me(16)$ and H-C(13)/Me(16) of **2** indicated that these H-atoms were situated on the same side. The correlation $H_\beta-C(5)/H-C(11)$ was not observed in the case of **2**.

Comparison of the physicochemical properties with the reported data allowed us to identify compounds 3-6, isolated from the same fungus, as allocyathin B₂ [4], sarcodonin A, sarcodonin G [1], and scabronine F [3], respectively (*Fig. 3*).

This project was supported by the National Natural Science Foundation of China (30225048).

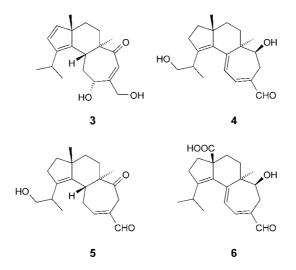


Fig. 3. Structures of allocyathin B (3), sarcodonin A (4), sarcodonin G (5), and scabronine F (6)

Experimental Part

General. Column chromatography CC. Optical rotations: Horiba SEPA-300 digital polarimeter. IR Spectra: Bio-Rad FTS-135 spectrometer; KBr pellets; \tilde{v} in cm⁻¹. ¹H- and ¹³C-NMR Spectra: Bruker AM-400 and DRX-500 spectrometers, SiMe₄ as internal standard; δ in ppm, J in Hz. MS: VG Auto-Spec-3000 and API QSTAR-Pulsar-i spectrometer; m/z (rel. int.).

Mushroom Material. The fresh fruiting bodies of *Sarcodon scabrosus* were collected at Ailao Mountain of Yunnan Province, P. R. China, in July 2003. The botanical identification was made by Prof. *Mu Zang*, Kunming Institute of Botany, the Chinese Academy of Sciences. A voucher specimen was deposited in the Herbarium of the Kunming Institute of Botany, the Chinese Academy of Sciences.

Extraction and Isolation. The entire freshly collected fruiting bodies of *S. scabrosus* (dry weight after extraction 150 g) were immersed in 95% EtOH and left at r.t. for several days. Then the EtOH extract was decanted and evaporated. The residue was extracted with $CHCl_3 (4 \times)$. The extract (70 g) was fractionated by CC (silica gel, petroleum ether/acetone 9:1, 8:2, 7:3, and 6:4). *Sarcodonin G* (5; 12 mg) and *scabronine F* (6; 10 mg) were obtained by prep. TLC (CHCl₃/MeOH 9:1) of *Fr. 1* (eluted with petroleum ether/acetone 6:4). *Fr. 2* (eluted with petroleum ether/acetone 6:4).

(*RP-8*, MeOH/H₂O 8:2): scabronine G (1; 10 mg), scabronine H (2; 200 mg), allocyathin B_2 (3; 50 mg), and sarcodonin A (4, 150 mg).

 $\begin{array}{l} Scabronine \ G \ \ (= rel-(3aR,5aR,68,9R,10aR)-2,3,3a,4,5,5a,6,9,10,10a-Decahydro-8-(hydroxymethyl)-1-(2-hydroxy-1-methylethyl)-3a,5a-dimethylcyclohept[e]indene-6,9-diol; 1): Red oil. [a]_D^{20} = -18.53 (c=0.2, CHCl_3). UV (CHCl_3): 207.8, 233.6. IR (KBr): 3430, 2931, 2867, 1713, 1452, 1375, 1315, 1275, 1177, 1116, 1027, 714. ^{1}H- and ^{13}C-NMR: Table 1. HR-ESI-MS: 463.2471 (C_{27}H_{36}O_{5}Na^{+}; calc. 463.2460). \end{array}$

Scabronine H (= rel-(3aR,5aR,6\$,9\$,10aR)-2,3,3a,4,5,5a,6,9,10,10a-Decahydro-8-(hydroxymethyl)-1-(2-hydroxy-1-methylethyl)-3a,5a-dimethylcydohept[e]indene-6,9-diol; **2**): Yellow powder. M.p. 165 – 168° (MeOH). $[a]_{20}^{20} = -9.08 (c = 0.3, MeOH)$. UV (MeOH): 204.4, 228.0. IR (KBr): 3421, 2934, 2867, 1719, 1452, 1373, 1315, 1272, 1176, 1114, 1016, 712. ¹H- and ¹³C-NMR: *Table 2.* HR-ESI-MS: 463.2464 (C₂₇H₃₆O₅Na⁺; calc. 463.2460).

REFERENCES

- H. Shibata, T. Tokunaga, D. Karaswa, A. Hirota, M. Nakayama, H. Nozaki, T. Tada, Agric. Biol. Chem. 1989, 53, 3373.
- [2] T. Ohta, T. Kita, N. Kobayashi, Y. Obara, N. Nakahata, Tetrahedron Lett. 1998, 39, 6229.
- [3] T. Kita, Y. Takaya, Y. Oshima, Tetrahedron 1998, 54, 11877.
- [4] W. A. Ayer, H. Taube, *Tetrahedron Lett.* 1972, 19, 1917.

Received July 16, 2004