# Three New Dimeric Orcinol Glucosides from Curculigo orchioides 

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#### Abstract

Three new phenolic glucosides named orcinosides $\mathrm{A}, \mathrm{B}$, and $\mathrm{C}(\mathbf{1}, \mathbf{2}$, and $\mathbf{3}$, resp. $)$ were isolated in low yields $\left(4.0 \times 10^{-6}, 11.5 \times 10^{-6}, 4.5 \times 10^{-6} \%\right.$, resp. ) from the rhizomes of Curculigo orchioides. Their structures were elucidated by comprehensive spectroscopic analyses including FAB-MS, HR-ESI-MS, IR, and 1D- and 2D-NMR (HSQC, HMBC) data. Compounds $\mathbf{1 - 3}$ contained two orcinol-glucoside moieties linked through a $\mathrm{CH}_{2}$ group.


Introduction. - Curculigo orchioides Gaertn., which belongs to the Amaryllidaceae family and named 'Xian-Mao' in Pharmacopoeia of China [1], is a multipurpose drug with numerous pharmacological activities. It has been employed as an analeptic agent for the treatment of decline in strength, and against jaundice and asthma [2]. Previous studies on the rhizomes of this species revealed the presence of cycloartane saponins [3], phenolic glycosides [4], and chlorophenyl glucosides [5]. Lakshmi et al. reported that phenols and phenolic glycosides from this plant were responsible for the stimulation of the immune response by acting both on macrophages and lymphocytes [6]. In addition, Wu et al. reported the potent antioxidative activities of some phenolic glycosides [7]. We had found that orcinol derivatives from C. orchioides showed antidepression activity [8]. The interest in biologically active substances from this medicinal plant encouraged us to further explore its phytochemical composition. Our investigation resulted in the isolation of three new phenolic glucosides in trace amounts, orcinosides $\mathrm{A}, \mathrm{B}$, and $\mathrm{C}(\mathbf{1}, \mathbf{2}$, and $\mathbf{3}$, resp.). We describe the isolation and structural elucidation of $\mathbf{1 - 3}$ (Fig. 1).

Results and Discussion. - The $70 \%$ EtOH extract of the roots of C. orchioides was applied to $D_{101}$ macroporous resin eluted with $10 \% \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$. Further fractionation by a combination of column chromatography on silica gel, PR-18, and Sephadex LH-20 afforded compounds $\mathbf{1}-\mathbf{3}$ in low yields.

Compound $\mathbf{1}$ was obtained as colorless needles with an optical rotation of $[\alpha]_{D}^{28.0}=$ $-65.6(c=0.61, \mathrm{MeOH})$. The FAB mass spectrum (negative-ion mode) exhibited a quasi-molecular-ion peak and fragment-ion peaks at $m / z 583\left([M-\mathrm{H}]^{-}\right), 421$ ([M$\left.\left.\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{O}_{5}\right]^{-}\right)$, and $259\left(\left[M-\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{O}_{10}\right]^{-}\right)$, suggesting the presence of two hexose moieties. The HR-ESI-MS (negative-ion mode) analysis provided the molecular



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Fig. 1. The structures of compounds $\mathbf{1}-\mathbf{3}$
formula $\mathrm{C}_{27} \mathrm{H}_{36} \mathrm{O}_{14}$ from the quasi-molecular-ion peak at $m / z 583.2025\left([M-\mathrm{H}]^{-}\right)$. The IR spectrum showed absorptions for OH groups ( $3417 \mathrm{~cm}^{-1}$ ) and aromatic rings (1621, $1590 \mathrm{~cm}^{-1}$ ), and a strong absorption at $1074 \mathrm{~cm}^{-1}$ due to a glucosidic linkage in the molecule. Hydrolysis of compound 1 with $10 \% \mathrm{H}_{2} \mathrm{SO}_{4}$ in MeOH furnished glucose, which was identified by comparison with an authentic sample on PC. In the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum (Table 1), two aromatic H -atom signals ( $\delta(\mathrm{H}) 6.49$ (br. $s$ ), 6.37 (br.s)), a $\mathrm{CH}_{2}$ $(\delta(\mathrm{H}) 3.93(s))$, and a Me signal $(\delta(\mathrm{H}) 2.19(s))$ were observed, together with a signal of an anomeric H -atom at $\delta(\mathrm{H}) 4.72(d, J=7.6)$ due to the $\beta$-linked glucose moiety in the molecule. The ${ }^{13} \mathrm{C}$-NMR spectrum (Table 1) displayed 14 C -atom signals, including those for one Me and one $\mathrm{CH}_{2}$ group, and six aromatic C -atoms assignable to a benzene ring, and a set of $\beta$-D-glucopyranose C -atom signals [9]. The above NMR data were similar to those of orcinol glucopyranoside [10] except that there were signals of a

Table 1. ${ }^{1} H$ - and ${ }^{13} C$-NMR Data of Compound 1. At $500 / 125 \mathrm{MHz}$, in $\mathrm{CD}_{3} \mathrm{OD}, \delta$ in ppm, $J$ in Hz. For positions, see Fig. 1.

| Position | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ | Position | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1a (1b) | 158.3 (s) | - | 1'a (1'b) | 103.1 (d) | $4.72(d, J=7.6)$ |
| 2a (2b) | 115.3 (s) | - | $2^{\prime} \mathrm{a}$ ( $2^{\prime} \mathrm{b}$ ) | 74.6 (d) | 3.45-3.46 (m) |
| 3a (3b) | 156.1 (s) | - | 3'a (3'b) | 78.1 (d) | 3.37-3.39 (m) |
| 4a (4b) | 108.8 (d) | 6.49 (br. s) | 4'a (4'b) | 71.3 (d) | 3.34-3.36 (m) |
| 5 a (5b) | 138.2 (s) | - | 5'a ( $5^{\prime} \mathrm{b}$ ) | 77.4 (d) | 3.47-3.48 (m) |
| 6a (6b) | 111.5 (d) | 6.37 (br. s) | $6^{\prime} \mathrm{a}$ ( $6^{\prime} \mathrm{b}$ ) | 62.4 (t) | $3.86(d d, J=12.4,1.5), 3.68(d d, J=12.1,4.0)$ |
| 7a (7b) | 21.5 (q) | 2.19 (s) | $\mathrm{CH}_{2}$ | 18.3 (t) | 3.93 (s) |

quaternary C -atom at $\delta(\mathrm{C}) 115.3$ and of a $\mathrm{CH}_{2}$ group at $\delta(\mathrm{C}) 18.3$ in the ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum of compound 1. Analyses of the FAB-MS, HR-ESI-MS and the NMR data indicated that compound $\mathbf{1}$ was composed of two orcinol glucopyranoside units and a $\mathrm{CH}_{2}$ group, with a symmetrical structure. The HMBCs between $\delta(\mathrm{H}) 3.93$ (br. $s, \mathrm{CH}_{2}$ ), and $\delta(\mathrm{C}) 158.3(\mathrm{C}(1 \mathrm{a})$ and $\mathrm{C}(1 \mathrm{~b})), 115.3(\mathrm{C}(2 \mathrm{a})$ and $\mathrm{C}(2 \mathrm{~b}))$, and $156.1(\mathrm{C}(3 \mathrm{a})$ and C (3b)) established the connection between the two orcinol glucopyranoside units through the $\mathrm{CH}_{2}$ group. Therefore, the structure of compound $\mathbf{1}$ was assigned as shown in Fig. 1 and named orcinoside A (1).

Compound 2 was isolated as colorless needles and had a molecular formula of $\mathrm{C}_{27} \mathrm{H}_{36} \mathrm{O}_{14}$ as determined by the HR-ESI-MS (negative-ion mode) peak at $\mathrm{m} / \mathrm{z} 583.2040$ ( $[M-H]^{-}$). The FAB-MS showed quasi-molecular-ion peak and fragment-ion peaks at $m / z 583\left([M-H]^{-}\right), 421\left(\left[M-\mathrm{C}_{6} \mathrm{H}_{11} \mathrm{O}_{4}\right]^{-}\right)$, and $259\left(\left[M-\mathrm{C}_{12} \mathrm{H}_{21} \mathrm{O}_{10}\right]^{-}\right)$, similarly to compound 1. The IR spectrum displayed absorptions at 3417 (OH), 1613, 1593 (aromatic ring), and $1075 \mathrm{~cm}^{-1}$ (glucosidic bond). Acidic hydrolysis of compound 2 with $10 \% \mathrm{H}_{2} \mathrm{SO}_{4}$ liberated glucose identified by comparison with an authentic sample on PC. The ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum (Table 2) exhibited signals for four aromatic H -atoms at $\delta(\mathrm{H}) 6.40$ (br. $s, 1 \mathrm{H}$ ), 6.31 (br. $s, 1 \mathrm{H}), 6.44(d, J=2.4,1 \mathrm{H}), 6.25(d, J=2.4,1 \mathrm{H})$, two Me-group singlets at $\delta(\mathrm{H}) 2.20,2.17$, and signals for two $\beta$-linked anomeric H -atoms at $\delta(\mathrm{H}) 4.78(d, J=7.7)$ and $4.79(d, J=7.6)$. The ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum (Table 2) displayed 27 C -atom signals, corresponding to two Me and one $\mathrm{CH}_{2}$ groups, and twelve aromatic C -atoms and two sets of $\beta$-D-glucopyranose C -atoms. Comparison of the NMR data of compound $\mathbf{2}$ with those of compound $\mathbf{1}$ revealed a high similarity except that signals of all C-atoms in the ${ }^{13} \mathrm{C}$-NMR spectrum of compound $\mathbf{2}$ appeared in pairs, suggesting that compound 2 should also consist of two orcinol glucopyranoside units and one $\mathrm{CH}_{2}$ group, but that these were arranged asymmetrically. Compound 2 differed from $\mathbf{1}$ mainly in the linkage mode between the two orcinol glucopyranoside units and the $\mathrm{CH}_{2}$ group. As shown in Fig 2, the HMBC correlations of the $\mathrm{CH}_{2} \mathrm{H}$-atoms $(\delta(\mathrm{H}) 3.97$ (d, $J=9.4), 3.87(d, J=9.4)$ ) with $\mathrm{C}(1 \mathrm{a}), \mathrm{C}(2 \mathrm{a}), \mathrm{C}(3 \mathrm{a}), \mathrm{C}(3 \mathrm{~b}), \mathrm{C}(4 \mathrm{~b})$, and $\mathrm{C}(5 \mathrm{~b})$ indicated that the $\mathrm{C}(2 \mathrm{a})$ of part A was linked to $\mathrm{C}(4 \mathrm{~b})$ of part B through the $\mathrm{CH}_{2}$ group. Consequently, the structure of compound 2 was determined as depicted in Fig. 1 and named orcinoside B (2).

Compound $\mathbf{3}$ was obtained as colorless needles. Its molecular formula $\mathrm{C}_{27} \mathrm{H}_{36} \mathrm{O}_{14}$, deduced from the HR-ESI-MS peak at $m / z 583.2041\left([M-H]^{-}\right)$, was the same as those of compounds $\mathbf{1}$ and $\mathbf{2}$. Analyses of the NMR data (Table 2) revealed that compound $\mathbf{3}$ also contained two orcinol glucopyranoside units and one $\mathrm{CH}_{2}$ group. The main difference between compounds $\mathbf{3}$ and $\mathbf{2}$ were the linkage positions of the two orcinol glucopyranoside units through the $\mathrm{CH}_{2}$ group. The correlations of the $\mathrm{CH}_{2}$ group $(\delta(\mathrm{H})$ $4.02(d, J=9.1), 3.92(d, J=9.1))$ with $\mathrm{C}(3 \mathrm{a}), \mathrm{C}(4 \mathrm{a}), \mathrm{C}(5 \mathrm{a}), \mathrm{C}(1 \mathrm{~b}), \mathrm{C}(5 \mathrm{~b})$, and $\mathrm{C}(6 \mathrm{~b})$ in the HMBC experiment (Fig. 2) suggested that the linkage between the two orcinol glucopyranoside fragments and the $\mathrm{CH}_{2}$ group was $\mathrm{C}(6 \mathrm{a})-\mathrm{CH}_{2}-\mathrm{C}(4 \mathrm{~b})$. The other correlations in the HMBC spectrum (Fig. 2) confirmed the structure. Thus, the structure of compound $\mathbf{3}$ was characterized as displayed in Fig. 1 and named orcinoside C (3).

Compounds 1-3 were obtained in trace amount from the rhizome of $C$. orchioides. The three compounds had the same molecular formula and almost the same NMR spectra. They differ in the linkage positions of the two orcinol glucopyranoside units



Fig. 2. The key HMBCs of compounds 1-3
with the $\mathrm{CH}_{2}$ group. As shown in Fig. 1, compounds $\mathbf{1}$ and $\mathbf{2}$ possessed the same part A moieties, and compounds $\mathbf{2}$ and $\mathbf{3}$ shared the same part B. Orcinol glucoside is the main phenolic glycoside in this plant [10]. Additionally, several orcinol derivatives had been isolated from C. orchioides [11]. However, dimeric orcinol glucosides were isolated for the first time from this plant and even from the Amaryllidaceae family. From the combinatorial view, the other three orcinol glucopyranoside derivatives with $\mathrm{C}(2 \mathrm{a})-\mathrm{CH}_{2}-\mathrm{C}(6 \mathrm{~b}), \mathrm{C}(4 \mathrm{a})-\mathrm{CH}_{2}-\mathrm{C}(4 \mathrm{~b})$, and $\mathrm{C}(6 \mathrm{a})-\mathrm{CH}_{2}-\mathrm{C}(6 \mathrm{~b})$ connections may also exist in this plant, although we could not detect them during our investigation.

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## Experimental Part

General. Column chromatography (CC): silica gel $\left(\mathrm{SiO}_{2} ; 200-300\right.$ mesh, Qingdao Meigao Chemical Co., Ltd., Qingdao, P. R. China), $\mathrm{Al}_{2} \mathrm{O}_{3}$ (Shanghai Wusi Chemical Reagents Company), $D_{101}$ macroporous resins (Tianjin Pesticide Chemical Company), Sephadex LH-20 (Pharmacia, Fine Chemical Co. Ltd.) , and Lichroprep RP-18 (40-63 $\mu \mathrm{m}$; Merck, D-Darmstadt). M.p.: XRC-1 micro-melting-point apparatus (Sichuang University, P. R. China); uncorrected. Fractions were monitored by TLC, visualizing by spraying with $10 \% \mathrm{H}_{2} \mathrm{SO}_{4}$ in EtOH followed by heating. Paper chromatography (PC):
Table 2. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-NMR Data of Compounds $\mathbf{2}$ and 3. At 400 and 100 MHz , respectively, in $\mathrm{CD}_{3} \mathrm{OD}, \delta$ in ppm, $J$ in Hz . For positions, see Fig. 1 .

| Position | 2 |  | 3 |  | Position | 2 |  | 3 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ |  | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ | $\delta(\mathrm{C})$ | $\delta(\mathrm{H})$ |
| 1a | 157.9 (s) | - | 157.1 (s) | - | 1b | 157.7 (s) | - | 158.0 (s) | - |
| 2a | 115.5 (s) | - | 101.5 (d) | $6.52(d, J=2.4)$ | 2b | 100.8 (d) | $6.44(d, J=2.4)$ | 102.5 (d) | $6.50(d, J=2.4)$ |
| 3 a | 156.5 (s) | - | 156.5 (s) | - | 3b | 156.8 (s) | - | 156.5 (s) | - |
| 4 a | 107.8 (d) | 6.40 (br. $s$ ) | 121.1 (s) | - | 4b | 121.9 (s) | - | 111.5 (d) | $6.37(d, J=2.4)$ |
| 5a | 137.7 (s) | - | 140.5 (s) | - | 5b | 140.7 (s) | - | 140.3 (s) | - |
| 6a | 111.1 (d) | 6.31 (br. $s$ ) | 112.3 (d) | $6.30(d, J=2.4)$ | 6b | 112.2 (d) | $6.25(d, J=2.4)$ | 122.5 (s) | - |
| 7a | 21.6 (q) | 2.20 (s) | 20.5 (q) | 2.11 (s) | 7 b | 20.5 (q) | 2.17 (s) | 20.4 (q) | 2.00 (s) |
| 1'a | 101.8 (d) | $4.78(d, J=7.7)$ | 102.2 (d) | $4.85(d, J=7.6)$ | 1'b | 102.8 (d) | $4.79(d, J=7.6)$ | 102.9 (d) | $4.77(d, J=7.5)$ |
| 2'a | 74.4 (d) | 3.38-3.39 (m) | 74.7 (d) | 3.43-3.44 (m) | 2'b | 74.6 (d) | 3.39-3.40 (m) | 74.8 (d) | 3.43-3.44 (m) |
| 3'a | 77.8 (d) | 3.36-3.38 (m) | 77.9 (d) | 3.42-3.43 (m) | 3'b | 78.0 (d) | 3.40-3.41 (m) | 77.9 (d) | 3.44-3.45 (m) |
| $4{ }^{\text {a }}$ | 71.2 (d) | 3.34-3.35 (m) | 71.3 (d) | 3.40-3.41 (m) | 4'b | 71.4 (d) | 3.28-3.29 (m) | 71.2 (d) | 3.40-3.41 (m) |
| 5'a | 77.8 (d) | 3.30-3.31 (m) | 77.6 (d) | 3.44-3.46 (m) | 5 'b | 77.8 (d) | 3.37-3.39 (m) | 77.9 (d) | 3.44-3.46 (m) |
| 6'a | 62.5 (t) | $\begin{aligned} & 3.65(d d, J=15.0,6.7) \\ & 3.84-3.85(\mathrm{~m}) \end{aligned}$ | 62.4 (t) | $\begin{aligned} & 3.91 \text { (overlapped), } \\ & 3.73-3.74(\mathrm{~m}) \end{aligned}$ | $6{ }^{\text {'b }}$ | 62.3 (t) | $\begin{aligned} & 3.75(d d, \\ & J=15.3,5.0), \\ & 3.93-3.94(m) \end{aligned}$ | 62.4 (t) | $\begin{aligned} & 3.91 \text { (overlapped) } \\ & 3.74-3.75(\mathrm{~m}) \end{aligned}$ |
| $-\mathrm{CH}_{2}-$ | 21.1 (t) | $\begin{aligned} & 3.97(d, J=9.4), \\ & 3.87(d, J=9.4) \end{aligned}$ | 23.4 (t) | $\begin{aligned} & 4.02(d, J=9.1), \\ & 3.92(d, J=9.1) \end{aligned}$ |  |  |  |  |  |

visualization by spraying with phthalic acid/aniline reagent, followed by heating. Optical rotations: Horiba SEPA-300 polarimeter. UV Spectra: $U V-210 A$ spectrometer; $\lambda_{\max }(\log \varepsilon)$ in nm. IR Spectra: Shimadzu IR-450 instrument; KBr pellets; in $\mathrm{cm}^{-1} .{ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-NMR spectra: Bruker $A V-400$ or DRX500 spectrometers; with TMS as internal standard; $\delta$ in ppm, $J$ in Hz. FAB-MS (neg.): VG-Auto-spec3000 mass spectrometer, glycerol as matrix. ESI- and HR-ESI-MS: API Qstar-Pulsar-1 mass spectrometer; in $m / z$ (rel. \%).

Plant Material. The rhizomes of Curculigo orchioides Gaertn. were collected in Wenshan County, Yunnan Province, P. R. China, in November 2005, and authenticated by Prof. Dr. Li-Gong Lei, Kunming Institute of Botany, Chinese Academy of Sciences. There were no microbial contamination or other impurities found in the collected samples. A voucher specimen (No. 20051106) had been deposited with the State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences.

Extraction and Isolation. The air-dried and powdered rhizomes of C. orchioides ( 200 kg ) were extracted with $70 \% \mathrm{EtOH}$ three times under reflux (each $10001,2 \mathrm{~h}$ ). The extract was concentrated to a small volume (2001) and submitted to CC on $D_{101}$ macroporous resin with gradient elution $\left(\mathrm{H}_{2} \mathrm{O} \rightarrow 10 \%\right.$ $\left.\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O} \rightarrow 40 \% \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O} \rightarrow 70 \% \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O} \rightarrow 90 \% \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}\right)$ to afford five fractions: Frs. $I-$ V. Fr. II $\left(10 \% \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}\right.$ fraction, 8000 g$)$ was subjected to $\mathrm{Al}_{2} \mathrm{O}_{3} \mathrm{CC}$ and eluted with $\mathrm{AcOEt} / \mathrm{EtOH} /$ $\mathrm{H}_{2} \mathrm{O}(90: 10: 1 \rightarrow 80: 20: 2 \rightarrow 70: 30: 3)$ to afford Frs. II.A - II.C. Fr. II.A $(100 \mathrm{~g})$ was subjected to $\mathrm{SiO}_{2} \mathrm{CC}$ with $\mathrm{CHCl}_{3} / \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}$ 85:15:1.5 to give three fractions Fr. II.A.1 - II.A.3. Fr. II.A. 3 was separated by $R P-18 \mathrm{CC}\left(\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O} 10: 90\right)$, and subsequently subjected to $\mathrm{SiO}_{2} \mathrm{CC}$ with $\mathrm{AcOEt} / \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}$ 80:20:1 and purified by Sephadex LH-20 (MeOH) to yield compounds $\mathbf{1}(8 \mathrm{mg}), \mathbf{2}(23 \mathrm{mg})$, and $\mathbf{3}$ ( 9 mg ).

Orcinoside A (=2-[2-( $\beta$-D-Glucopyranosyloxy)-6-hydroxy-4-methylbenzyl]-3-hydroxy-5-methylphenyl $\beta$-D-Glucopyranoside; 1). Colorless needles. M.p. $179-180^{\circ} .[\alpha]_{\mathrm{D}}^{28.0}=-65.6(c=0.61, \mathrm{MeOH})$. UV ( MeOH ): 270 (3.68). IR (KBr): 3417, 2921, 2883, 1621, 1590, 1515, 1458, 1420, 1074, 1034, 894, 534. NMR: Table 1. FAB-MS (neg.): $583\left([M-\mathrm{H}]^{-}\right), 421$ ([ $\left.\left.M-\mathrm{H}-\mathrm{Glc}\right]^{-}\right), 259\left([M-\mathrm{H}-2 \mathrm{Glc}]^{-}\right)$. HR-ESI-MS (neg.): $583.2025\left([M-\mathrm{H}]^{-}, \mathrm{C}_{27} \mathrm{H}_{35} \mathrm{O}_{14}^{-}\right.$; calc. 583.2026).

Orcinoside $B$ (=2-[4-( $\beta$-D-Glucopyranosyloxy)-2-hydroxy-6-methylbenzyl]-3-hydroxy-5-methylphenyl $\beta$-D-Glucopyranoside; 2). Colorless needles. M.p. $175-177^{\circ} .[\alpha]_{\mathrm{D}}^{27.2}=-231.5(c=0.72, \mathrm{MeOH})$. UV (MeOH): 278 (3.63). IR (KBr): 3417, 2921, 2887, 1613, 1593, 1492, 1075, 1033, 531. NMR: Table 2. FAB-MS (neg.): $583\left([M-\mathrm{H}]^{-}\right), 421\left([M-\mathrm{H}-\mathrm{Glc}]^{-}\right), 259$ ([ $M-\mathrm{H}-2$ Glc $\left.]^{-}\right)$. HR-ESI-MS (neg.): $583.2040\left([M-H]^{-}, \mathrm{C}_{27} \mathrm{H}_{35} \mathrm{O}_{14}^{-}\right.$; calc. 583.2026).

Orcinoside $C$ (=2-[4-( $\beta$-D-Glucopyranosyloxy)-2-hydroxy-6-methylbenzyl]-5-hydroxy-3-methylphenyl $\beta$-D-Glucopyranoside; 3). Colorless needles. M.p. $210-211^{\circ}$. $[\alpha]_{\mathrm{D}}^{27.4}=-26.9(c=0.62, \mathrm{MeOH})$. UV (MeOH): 280 (3.73). IR (KBr): 3418, 2922, 1612, 1592, 1489, 1460, 1074, 1036, 531. NMR: Table 2. FAB-MS (neg.): $583\left([M-H]^{-}\right)$. HR-ESI-MS (neg.): $583.2041\left([M-H]^{-}, \mathrm{C}_{27} \mathrm{H}_{35} \mathrm{O}_{14}^{-}\right.$; calc. 583.2026).

Acidic Hydrolysis. A soln. of $\mathbf{1 - 3}$ (each 3 mg ) in a mixture of $\mathrm{MeOH}(1.0 \mathrm{ml})$ and $10 \% \mathrm{H}_{2} \mathrm{SO}_{4}$ $(1.0 \mathrm{ml})$ was refluxed for 2 h . The hydrolysate was allowed to cool, diluted with 2 ml of $\mathrm{H}_{2} \mathrm{O}$, and extracted with 4 ml of AcOEt . The aq. layer was neutralized with aq. $\mathrm{Ba}(\mathrm{OH})_{2}$ and concentrated in vacuum to give a residue, in which glucose (from $\mathbf{1 - 3}$ ) was identified by comparison with an authentic sample ( $\mathrm{BuOH} / \mathrm{AcOEt} / \mathrm{H}_{2} \mathrm{O} 4: 1: 5$, upper layer, $\left.R_{\mathrm{f}} 0.45 ; \mathrm{PhOH} / \mathrm{H}_{2} \mathrm{O} 4: 1, R_{\mathrm{f}} 0.50\right)$ on PC : visualized by spraying with phthalic acid/aniline reagent ( 1.66 g phthalic acid and 0.93 g aniline dissolved in 100 ml $\mathrm{H}_{2} \mathrm{O} /$ sat. BuOH ), followed by heating.

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