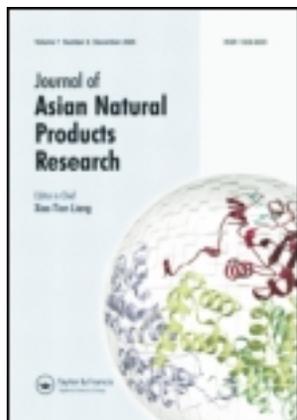


This article was downloaded by: [Kunming Institute of Botany]

On: 13 February 2012, At: 00:24

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/ganp20>

Three new secoiridoid glycoside dimers from *Swertia mileensis*

Chang-An Geng^{a b}, Xue-Mei Zhang^a, Yun-Bao Ma^a, Zhi-Yong Jiang^a, Ji-Feng Liu^a, Jun Zhou^a & Ji-Jun Chen^a

^a State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming, 650204, China

^b Graduate University of Chinese Academy of Sciences, Beijing, 100039, China

Available online: 15 Jun 2010

To cite this article: Chang-An Geng, Xue-Mei Zhang, Yun-Bao Ma, Zhi-Yong Jiang, Ji-Feng Liu, Jun Zhou & Ji-Jun Chen (2010): Three new secoiridoid glycoside dimers from *Swertia mileensis*, *Journal of Asian Natural Products Research*, 12:6, 542-548

To link to this article: <http://dx.doi.org/10.1080/10286020.2010.491477>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

ORIGINAL ARTICLE

Three new secoiridoid glycoside dimers from *Swertia mileensis*

Chang-An Geng^{ab}, Xue-Mei Zhang^a, Yun-Bao Ma^a, Zhi-Yong Jiang^a, Ji-Feng Liu^a,
Jun Zhou^{a*} and Ji-Jun Chen^{a*}

^aState Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650204, China; ^bGraduate University of Chinese Academy of Sciences, Beijing 100039, China

(Received 18 March 2010; final version received 3 May 2010)

Three new secoiridoid glycoside dimers named swerilactosides A–C (**1–3**) were isolated from *Swertia mileensis*. Their structures were elucidated based on extensive spectral analyses (1D and 2D NMR, MS, and IR spectroscopic means).

Keywords: secoiridoid glycoside dimers; swerilactosides A–C; *Swertia mileensis*; Gentianaceae

1. Introduction

The family Gentianaceae, annual or perennial herbs, contains about 80 genera and 700 species, of which 22 genera and 427 species are distributed in China [1]. Many species mainly belonging to the *Gentiana* and *Swertia* genus are used as traditional Chinese herbs to treat hepatitis, cholecystitis, and digestive system disease [2]. Previous investigation reveals that secoiridoid glycosides, xanthenes, flavones, and triterpenoids are the main constituents of Gentianaceae plants [3]. In 1958, Fu and Sun [4] reported the isolation of three alkaloids from *Gentiana macrophylla* (namely ‘Qin-Jiao’ in Chinese), one of which was identified as gentianine. Afterwards, Prof. Liang *et al.* [5,6] first applied NMR and IR spectral analyses, together with chemical methods, to determine the structures of gentinidine and gentianal. Later, Govindachari *et al.* [7] proved gentianine to be an artificial

product during the extraction with $\text{NH}_3 \cdot \text{H}_2\text{O}$.

Swertia mileensis (= *Swertia leduicii*), well known as ‘Qing-Ye-Dan’ in Chinese, belongs to the *Swertia* genus, the second largest genus next to *Gentiana* of the family Gentianaceae [8]. As a traditional Chinese medicine (TCM), it has long been used to treat viral hepatitis in the Yi and Ha-Ni nationality regions, Mile and Kaiyuan Counties, Yunnan Province. In the 1970s, a large amount of phytochemical and pharmacological investigations on *S. mileensis* was carried out, which promoted it to be documented in *Chinese Pharmacopoeia* (1977–2010 editions) as a new TCM source [9]. Presently, its significantly curative effect on acute viral hepatitis has resulted in wide clinical applications [10–13].

In order to clarify the active components [14], our previous bioassay-guided fractionation has led to the isolation of four types of novel iridoid lactones:

*Corresponding authors. Emails: chenjj@mail.kib.ac.cn; jzhou@mail.kib.ac.cn

swerilactones A and B (C18 skeleton) [15], swerilactones C and D (C20 skeleton) [16], swerilactones E and F (lactones with naphthyl rings), and swerilactone G (a secoiridoid aglycone dimer) [17], with anti-HBV activity *in vitro*, and subsequently, the other three unusual secoiridoid glycoside dimers (two molecules of secoiridoids connected by a molecule of the glycosyl group) were obtained from this plant. Generally, the *Swertia* genus is rich in secoiridoid glycosides; however, the secoiridoid glycoside dimers were seldom reported [3,18]. Herein, we describe the isolation and structural elucidation of swerilactosides A–C based on extensive spectroscopic analyses (Figure 1).

2. Results and discussion

Swerilactoside A (**1**) had a molecular formula of $C_{25}H_{32}O_{13}$ by positive HR-ESI-MS at m/z 563.1728 $[M + Na]^+$. The IR spectrum showed the absorption bands of OH (3423 cm^{-1}), C–O (1698 cm^{-1}), double bond (1620 cm^{-1}), and glycosyl group ($1082, 1027, \text{ and } 1005\text{ cm}^{-1}$).

The ^1H and ^{13}C NMR spectra of compound **1** displayed 25 carbon signals due to 6 quaternary carbons, 11 methines, 7 methylenes, and 1 methyl group, of which two lactone carbonyl carbons, three double bonds, and one glucosyl group were revealed (Table 1).

Detailed analyses of its NMR spectra suggested a swertiamarin fragment (**1a**), which was also supported by the ^1H – ^1H COSY, HMBC, and ROESY spectra. Compared to the known swertiamarin [19], the C-3' in compound **1** was shifted significantly downfield from $\delta_{\text{C}} 77.7$ (d) to 86.3 (d); contrarily, C-4' was shifted slightly upfield from $\delta_{\text{C}} 71.4$ (d) to 69.5 (d), which suggested that another partial structure was linked to C-3' by the glycosidic linkage in compound **1**.

In addition to the swertiamarin fragment (**1a**), the nine residual carbons were ascribed to one lactone carbonyl carbon [$\delta_{\text{C}} 166.5$ (s, C-10'')], one tetra-substituted double bond [$\delta_{\text{C}} 158.3$ (s, C-5'') and 123.2 (s, C-4'')], two oxygenated methines [$\delta_{\text{C}} 95.8$ (d, C-3''), dioxxygenated one) and 63.4 (d, C-1'')], three methylenes [$\delta_{\text{C}} 67.4$ (t, C-7''), oxygenated one), 37.3 (t, C-8''), and 29.0 (t, C-6'')], and one methyl group [$\delta_{\text{C}} 20.6$ (q, C-9'')], which indicated a secoiridoid aglycone-like fragment. In the HMBC spectrum, the correlations of H-7'' with C-5'' and C-10'', H-6'' with C-4'' and C-8'', H-8'' with C-4'' and C-9'', and H-3'' with C-1'' and C-10'', together with the ^1H – ^1H COSY correlations of H-7''/H-6'' and H-8''/H-1''/H-9'', led to the construction of the partial fragment **1b** (Figure 2).

The connection of C-3' and C-3'' by an ether bond was determined by the HMBC correlations of H-3' with C-3'' and H-3''

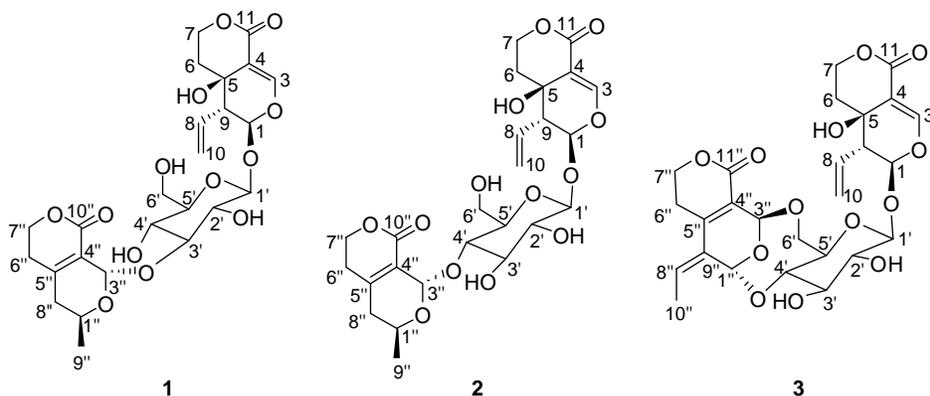


Figure 1. The structures of compounds **1**–**3**.

Table 1. ^1H and ^{13}C NMR spectral data of compounds 1–3 (in pyridine- d_5 , δ in ppm, J in Hz).

| No. | 1 ^a | | 2 ^b | | 3 ^b | |
|------|---------------------|-----------------|---------------------|-----------------|---------------------|-----------------|
| | ^1H | ^{13}C | ^1H | ^{13}C | ^1H | ^{13}C |
| 1 | 5.74, d, 1.2 | 98.7, d | 5.67, d, 1.2 | 99.1, d | 5.56, d, 1.2 | 99.2, d |
| 3 | 7.63, s | 154.4, d | 7.61, s | 154.8, d | 7.62, s | 154.6, d |
| 4 | | 109.2, s | | 108.8, s | | 108.9, s |
| 5 | | 64.3, s | | 64.3, s | | 64.3, s |
| 6a | 1.89, m | 33.5, t | 1.88, m | 33.7, t | 1.90, m | 33.7, t |
| 6b | 1.74, bd, 13.4 | | 1.72, bd, 14.0 | | 1.74, bd, 14.1 | |
| 7a | 4.73, m | 65.9, t | 4.72, m | 66.0, t | 4.74, m | 65.9, t |
| 7b | 4.32, m | | 4.33, m | | 4.33, m | |
| 8 | 5.42, m | 133.7, d | 5.39, m | 133.7, d | 5.42, m | 133.8, d |
| 9 | 2.91, dd, 9.2, 1.3 | 51.8, d | 2.90, dd, 9.2, 1.2 | 51.9, d | 2.90, dd, 9.5, 1.0 | 52.0, d |
| 10a | 5.36, dd, 17.0, 2.6 | 121.3, t | 5.34, dd, 17.0, 2.5 | 121.3, t | 5.37, dd, 17.0, 2.3 | 121.3, t |
| 10b | 5.29, dd, 9.4, 2.6 | | 5.27, dd, 9.4, 2.5 | | 5.28, dd, 9.6, 2.3 | |
| 11 | | 167.9, s | | 168.1, s | | 167.9, s |
| 1' | 4.76, d, 7.9 | 98.7, d | 4.63, d, 8.0 | 100.0, d | 4.64, d, 8.0 | 100.5, d |
| 2' | 3.34, m | 74.4, d | 3.28, m | 74.4, d | 3.26, m | 75.2, d |
| 3' | 3.62, t, 8.7 | 86.3, d | 3.47, t, 9.0 | 86.3, d | 3.44, m | 75.7, d |
| 4' | 3.30, m | 69.5, d | 3.67, t, 9.0 | 78.3, d | 3.77, t, 9.4 | 79.4, d |
| 5' | 3.41, m | 78.6, d | 3.37, m | 77.5, d | 3.40, m | 72.4, d |
| 6'a | 3.90, dd, 12.0, 1.9 | 62.4, t | 3.94, m | 61.8, d | 3.97, m | 67.6, t |
| 6'b | 3.69, dd, 12.0, 5.4 | | | | | |
| 1'' | 4.34, m | 63.4, d | 4.41, m | 63.8, d | 5.78, s | 95.0, d |
| 3'' | 5.50, s | 95.8, d | 5.40, s | 94.5, d | 5.57, s | 92.3, d |
| 4'' | | 123.2, s | | 123.2, d | | 119.8, s |
| 5'' | | 158.3, s | | 157.9, s | | 147.6, s |
| 6''a | 2.53, t, 6.3 | 29.0, t | 2.61, m | 29.4, t | 2.71, m | 23.3, t |
| 6''b | | | 2.38, m | | | |
| 7''a | 4.43, t, 6.4 | 67.4, t | 4.36, m | 66.9, t | 4.44, m | 67.0, t |
| 7''b | | | | | 4.39, m | |
| 8'' | 2.28, m | 37.3, t | 2.27, m | 37.4, t | 6.53, q, 7.3 | 138.4, d |
| 9'' | 1.26, d, 6.2 | 20.6, q | 1.25, d, 6.2 | 20.6, q | | 130.7, s |
| 10'' | | 166.5, s | | 165.2, s | 2.01, d, 7.3 | 14.8, q |
| 11'' | | | | | | 165.7, s |

Notes: ^aData measured at 400 MHz for ^1H NMR, 100 MHz for ^{13}C NMR.^bData measured at 500 MHz for ^1H NMR, 125 MHz for ^{13}C NMR.

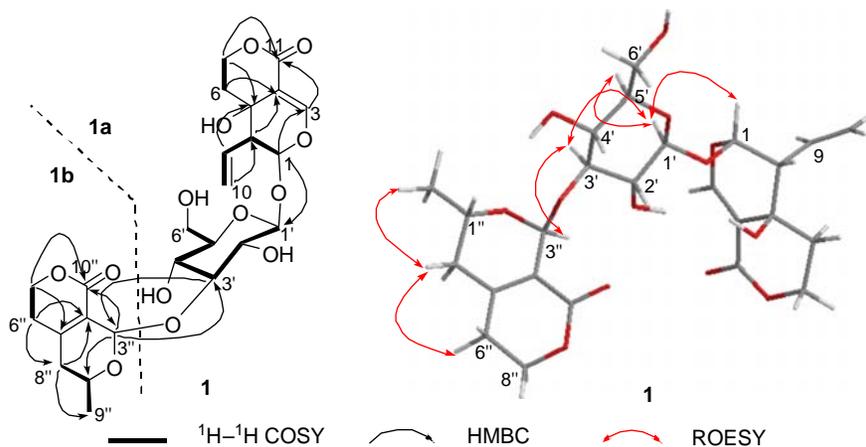


Figure 2. Selected ^1H - ^1H COSY, HMBC, and ROESY correlations of compound **1**.

with C-3'. However, only a weak correlation of H-3'' with H-9'' was detected in the ROESY spectrum, which was insufficient to determine the same orientation of H-3'' and Me-9''. This problem has been encountered in our previous investigation, namely, swerilactone G possessed a similar partial structure with the fragment **1b**, and its relative configuration has been proved by X-ray single-crystal diffraction [16]. Although the Me-9 and H-3 were located at the same side, the ROESY correlation of neither H-3/H-1 nor H-3/H-9 was detected in that the connection of H-C (3)-O-C (1)-C (9) possessed the *W* conformation. In addition, the coupling constant of H-9'' ($J = 6.2$ Hz) in swerilactoside A was also identical to that of H-9 ($J = 6.2$ Hz) in swerilactone G. Thus, it was plausible to deduce that fragment **1b** possessed a similar configuration to that of **3a** in swerilactone G. Consequently, the structure of compound **1** was elucidated to be swerilactoside A, as shown in Figure 1.

Swerilactoside B (**2**) possessed the same molecular structure of $\text{C}_{25}\text{H}_{32}\text{O}_{13}$ as that of compound **1**. The UV, IR, and NMR spectra of compound **2** were very close to those of compound **1**, which suggested that they possessed a similar skeleton. The ^1H - ^1H COSY and HMBC analyses suggested that compound **2**

contained the same partial fragments **2a** and **2b** as those of compound **1**. The HMBC correlations of H-3'' with C-4' and H-4' with C-3'' and the upfield shift of C-3' from δ_{C} 86.3 (d) in compound **1** to 76.5 (d) in compound **2**, as well as the downfield shift of C-4' from δ_{C} 69.5 (d) in compound **1** to 78.3 (d) in compound **2**, corresponding to the variations of $\Delta\delta_{\text{H-3}'}$ (-0.15 ppm) and $\Delta\delta_{\text{H-4}'}$ (+0.37 ppm), proposed that fragment **2b** was linked to **2a** by C (4')-O-C (3''). Similarly, the correlation of neither H-3''/H-1'' nor H-3''/H-9'' was detected in the ROESY spectrum (Figure 3), together with the completely consistent coupling constant of H-9'' ($J = 6.2$ Hz) with that in swerilactoside B and swerilactone G [17], which indicated that fragment **2b** adopted the same relative configuration as that of **1b**. Thus, the structure of compound **2** was elucidated to be swerilactoside B, as shown in Figure 1.

Swerilactoside C (**3**) had a molecular formula of $\text{C}_{26}\text{H}_{30}\text{O}_{13}$ by a quasi-molecular ion peak at m/z 585.1367 [$\text{M} + \text{Cl}$] $^-$ in the negative HR-ESI-MS. The IR spectrum suggested the presence of OH (3436 cm^{-1}), C=O (1703 cm^{-1}), double bond (1620 cm^{-1}), and glycosyl group ($1084, 1057, \text{ and } 1032\text{ cm}^{-1}$).

The ^1H and ^{13}C NMR (DEPT) spectra exhibited 26 carbon resonances due to 7

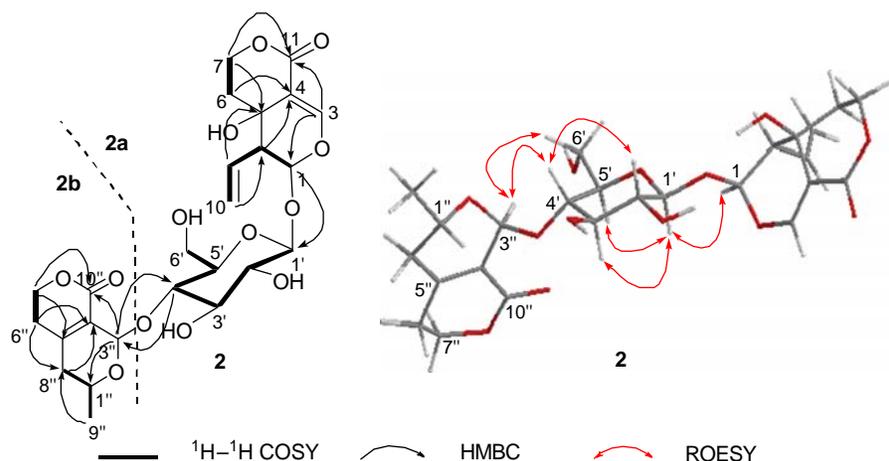


Figure 3. Selected ^1H - ^1H COSY, HMBC, and ROESY correlations of compound **2**.

quaternary carbons, 12 methines, 6 methylenes, and 1 methyl group. The NMR spectral data of compound **3** were similar to those of compound **2**, except for the presence of additional tri-substituted double bond [δ_{C} 138.4 (d, C-8'') and 130.7 (s, C-9'')] and the absence of one methylene [δ_{C} 37.4 (t, C-8'')] observed in compound **3**, together with the obvious downfield shift of C-1'' [from δ_{C} 63.8 (d) in compound **2** to δ_{C} 95.0 (d) in compound **3**]. In addition, the chemical shift variation of C-2' ($\Delta\delta = +0.8$ ppm), C-3' ($\Delta\delta = -0.8$ ppm), C-5' ($\Delta\delta = -5.1$ ppm), C-6' ($\Delta\delta = +5.8$ ppm), and C-10'' [$\Delta\delta = -5.8$ ppm (corresponding to C-9'' in compound **2**)] was observed in Table 1. In addition to the swertiamarin fragment (**3a**), the other partial structure **3b** was constructed based on the ^1H - ^1H COSY correlations of H-6'' with H-7'', and H-8'' with H-10'', and the HMBC correlations of H-7'' with C-5'' and C-11'', H-6'' with C-4'' and C-9'', H-8'' with C-5'' and C-1'', and H-3'' with C-1'', C-5'' and C-11'' (Figure 4). The glycosidic linkage between C-1'' and C-4' was deduced by HMBC correlations of H-1''/C-4' and H-4'/C-1''. Similarly, the connection of C-3'' and C-6' by a glycosidic bond was detected by the HMBC correlations of H-3'' with C-6' and H-6' with C-3''.

The correlations of H-1''/H-4' and H-3''/H-6' in the ROESY spectrum suggested the β -orientation of H-1'' and the α -orientation of H-3''. The *Z*-configuration of the double bond between C-8'' and C-9'' was deduced based on the ROESY correlations of H-8'' with H-6'' and H-10'' with H-1''. Thus, the structure of compound **3** was deduced to be swerilactoside C, as shown in Figure 1.

Swerilactosides A–C were three unusual secoiridoid glycoside dimers obtained from the traditional Chinese herb *S. mileensis*, which further enriched the skeleton type of secoiridoid glycosides.

3. Experimental

3.1 General experimental procedures

Optical rotations were determined on a JASCO model 1020 polarimeter (Horiba, Tokyo, Japan). UV spectra were measured on a Shimadzu UV-2401A spectrophotometer (Shimadzu, Kyoto, Japan). IR (KBr) spectra were recorded on a Bio-Rad FTS-135 spectrometer (Bio-Rad, Hercules, CA, USA). 1D and 2D NMR spectra were recorded on Bruker AM-400 NMR or DRX-500 spectrometers (Bruker, Bremerhaven, Germany) with TMS as an internal standard. MS spectra were run on a VG Auto Spec-3000 spectrometer (VG, Manchester,

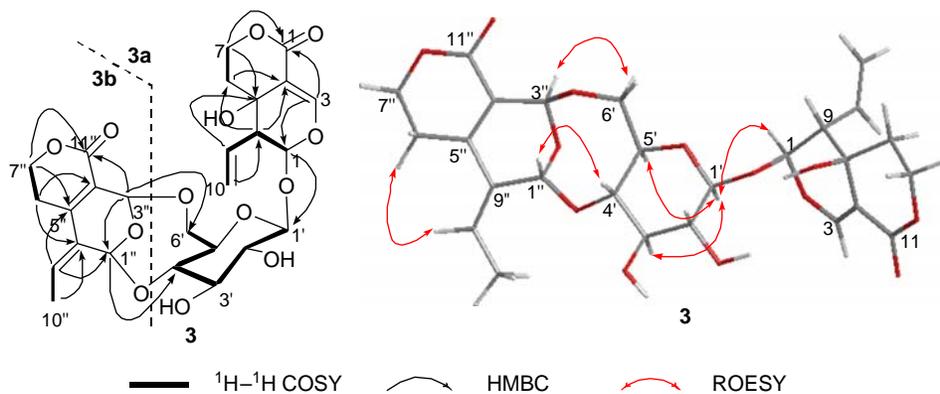


Figure 4. Selected ^1H - ^1H COSY, HMBC, and ROESY correlations of compound **3**.

England). Silica gel (200–300 mesh) for column chromatography was obtained from Qingdao Makall Chemical Company, Qingdao, China. HPLC (Waters Alliance 2695), equipped with a photodiode array detector (Waters 2996) and a Waters 600 pump, was purchased from Waters Co. Ltd, Milford, MA, USA. Sephadex LH-20 (20–150 μm) was purchased from Pharmacia Fine Chemical Co. Ltd, Uppsala, Sweden.

3.2 Plant material

The whole plant of *S. mileensis* was collected in Mile County, Yunnan Province, China, on 6 November 2008, and was identified as *S. mileensis* T. N. Ho et W. L. Shi by Prof. Dr. Li-Gong Lei, Kunming Institute of Botany, Chinese Academy of Sciences. A voucher specimen (No. 2008-11-01) has been deposited in the Laboratory of Antivirus and Natural Medicinal Chemistry, Kunming Institute of Botany.

3.3 Extraction and isolation

The air-dried whole plant (5.0 kg) of *S. mileensis* was powdered and extracted with 90% and 50% EtOH under reflux successively (each time 2 h, 15.0 liters \times 2 times). The combined extracts were concentrated under reduced pressure to give a residue (1.3 kg). The residue was suspended in water and extracted with

petroleum ether (1.0 liters \times 2), ethyl acetate (1.0 liters \times 3), and *n*-butanol (1.0 liters \times 3) successively. The ethyl acetate part (170.5 g) was chromatographed on a silica gel column (2.0 kg, 11.0 \times 50.0 cm) eluted with CHCl_3 -MeOH (from 100:0 to 0:100, v/v) to furnish 10 fractions A–J. Fraction B (8.5 g) was chromatographed on a silica gel column (100.0 g, 3.0 \times 30.0 cm) with a gradient elution of CHCl_3 - Me_2O (90:1 \rightarrow 50:50) to supply four fractions B1–B4. Fraction B4 (3.0 g) was performed on a silica gel column (30.0 g, 1.7 \times 25.0 cm) eluted with CHCl_3 -MeOH (90:1 \rightarrow 80:20) to obtain three subfractions B4-1 to B4-3. Subfraction B4-1 (100.0 mg) was dissolved in MeOH and purified with a semi-preparative HPLC apparatus, using a Waters XTerra Prep RP-18 column (7.8 \times 300 mm, 10 μm), eluted with MeOH- H_2O (35:65, flow rate = 4.5 ml/min), detected at 254 nm, to obtain compound **1** (80.0 mg, R_t = 18.0 min). Subfraction B4-2 (500.0 mg) was subjected to a silica gel column (30.0 g, 1.7 \times 25.0 cm) eluted with CHCl_3 - Me_2CO (80:20), and then further purified with HPLC (the conditions were similar to compound **1**) to supply compound **2** (30.0 mg, R_t = 13.0 min). Subfraction B4-3 (300.0 mg) was first loaded on a silica gel column (30.0 g, 1.7 \times 25.0 cm) and eluted with CHCl_3 - Me_2O (80:20), and then purified

with a Sephadex LH-20 column (50.0 g, 1.4×145.0 cm, MeOH) to give compound **3** (100.0 mg).

3.3.1 Swerilactoside A (1)

A white powder; $[\alpha]_D^{19.8} - 94.04$ ($c = 0.68$, MeOH); UV (MeOH) λ_{\max} (nm) (log ϵ): 231 (4.16); IR (KBr) ν_{\max} (cm^{-1}): 3423, 1698, 1620, 1473, 1419, 1280, 1269, 1082, 1027, 1005, 787; ^1H and ^{13}C NMR spectral data see Table 1; ESI-MS (+) m/z : 563 $[\text{M} + \text{Na}]^+$; HR-ESI-MS (+) m/z : 563.1728 $[\text{M} + \text{Na}]^+$ (calcd for $\text{C}_{25}\text{H}_{32}\text{O}_{13}\text{Na}$, 563.1740).

3.3.2 Swerilactoside B (2)

A white powder; $[\alpha]_D^{19.8} - 127.69$ ($c = 0.20$, MeOH); UV (MeOH) λ_{\max} (nm) (log ϵ): 231 (4.12); IR (KBr) ν_{\max} (cm^{-1}): 3429, 1705, 1620, 1472, 1416, 1326, 1280, 1207, 1154, 1079, 1028, 948, 929, 758; ^1H and ^{13}C NMR spectral data see Table 1; ESI-MS (-) m/z : 575 $[\text{M} + \text{Cl}]^-$; HR-ESI-MS (-) m/z : 575.1517 $[\text{M} + \text{Cl}]^-$ (calcd for $\text{C}_{25}\text{H}_{32}\text{O}_{13}\text{Cl}$, 575.1531).

3.3.3 Swerilactoside C (3)

A white powder; $[\alpha]_D^{20.0} - 67.11$ ($c = 0.14$, MeOH); UV (MeOH) λ_{\max} (nm) (log ϵ): 269 (4.14), 240 (4.14); IR (KBr) ν_{\max} (cm^{-1}): 3436, 1703, 1620, 1472, 1434, 1408, 1273, 1246, 1208, 1159, 1126, 1084, 1057, 1032, 1013, 961, 930, 903, 846, 760; ^1H and ^{13}C NMR spectral data see Table 1; ESI-MS (-) m/z : 585 $[\text{M} + \text{Cl}]^-$; HR-ESI-MS (-) m/z : 585.1367 $[\text{M} + \text{Cl}]^-$ (calcd for $\text{C}_{26}\text{H}_{30}\text{O}_{13}\text{Cl}$, 585.1374).

Acknowledgements

This work was financially supported by the National Natural Science Foundation of China and Yunnan Province (Grant No. U0832603), the Major Project of Innovative Engineering (KSCX2-YW-R-181), Xibu Zhiguang Joint-Scholarship of Chinese Academy of Sciences and the State Key Laboratory of Phytochemistry and Plant Resources in West China. We are grateful to the staff of the analytical group

of the State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences.

References

- [1] T.N. Ho, *Flora of China* (Science Press, Beijing, 1988), Vol. 62.
- [2] W.X. Yang, L. Zhou, H.L. Geng, and B.F. Qin, *Acta Bot. Real. Occident. Sin.* **23**, 2235 (2003).
- [3] S.R. Jensen and J. Schripsema, in *Gentianaceae—Systematic and Natural History*, edited by L. Struwe and V. Albert (Cambridge University Press, Cambridge, 2002), Chap. 6, p. 573.
- [4] F.Y. Fu and N.C. Sun, *Acta Pharm. Sin.* **6**, 198 (1958).
- [5] X.T. Liang, D.Q. Yu, and F.Y. Fu, *Acta Pharm. Sin.* **11**, 412 (1964).
- [6] Z. Xue and X.T. Liang, *Chin. Sci. Bull.* **19**, 378 (1974).
- [7] T.R. Govindachari, S.S. Sathe, and N. Viswanathan, *Indian J. Chem.* **4**, 201 (1966).
- [8] P.T. Li, A.J. Leeuweberg, and D.J. Middleton, *Flora China* **16**, 115 (1995).
- [9] L. Gao, *Yunnan J. Trad. Chin. Med. Mater. Med.* **27**, 65 (2006).
- [10] Y. Du and H.D. Li, *Chin. Pharm.* **17**, 304 (2006).
- [11] Q.S. Liang and X.Y. Gao, *Zhong Cao Yao Tong Xun* **9**, 1 (1979).
- [12] G.M. Du and G.Y. Li, *Yunnan Zhong Yi Za Zhi* **3**, 35 (1981).
- [13] Contagious Department, 59th Hospital of the Chinese PLA, *New Chin. Med.* **4**, 202 (1975).
- [14] X.S. Li, Z.Y. Jiang, F.S. Wang, Y.B. Ma, X.M. Zhang, and J.J. Chen, *China J. Chin. Mater. Med.* **33**, 2790 (2008).
- [15] C.A. Geng, Z.Y. Jiang, Y.B. Ma, J. Luo, X.M. Zhang, H.L. Wang, Y. Shen, A.X. Zuo, J. Zhou, and J.J. Chen, *Org. Lett.* **11**, 4120 (2009).
- [16] C.A. Geng, X.M. Zhang, Y. Shen, A.X. Zuo, J.F. Liu, Y.B. Ma, J. Luo, J. Zhou, Z.Y. Jiang, and J.J. Chen, *Org. Lett.* **11**, 4834 (2009).
- [17] C.A. Geng, X.M. Zhang, Y.B. Ma, Z.Y. Jiang, J. Luo, J. Zhou, H.L. Wang, and J.J. Chen, *Tetrahedron Lett.* **51**, 2483 (2010).
- [18] S. Rodriguez, A. Marston, J.L. Wolfender, and K. Hostettmann, *Curr. Org. Chem.* **2**, 627 (1998).
- [19] C.H. Duan, B.J. Shi, L.H. Wu, G.X. Chou, and Z.T. Wang, *Chin. J. Nat. Med.* **5**, 417 (2007).