



## Gelegamines A–E: five new oxindole alkaloids from *Gelsemium elegans*

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### ABSTRACT

Five new oxindole alkaloids, gelegamines A–E (**1–5**), were isolated from the roots of *Gelsemium elegans*. Their structures were extensively elucidated on the basis of spectroscopic analysis. Among them, the epoxy ring (C-19/C-20) of gelegamine A (**1**) was assigned as  $\alpha$ -orientation by ROESY experiment and DFT method at B3LYP/6-31g(d) level, and gelegamine B (**2**) is the first humantenine-type alkaloid with 19-(E) ethylidene configuration. The absolute configurations of gelegamines A–E (**1–5**) were established on biosynthetic consideration coupled with CD experiments.

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## 1. Introduction

*Gelsemium elegans* (Loganiaceae) is a liane native to Southeast Asia, where it is used in folk medicine for the treatment of pain, spasticity, and skin ulcers.<sup>1</sup> In a previous chemical investigation, a number of indole alkaloids based on six different structural skeletons were reported from *G. elegans*.<sup>2</sup> Some of them showed the interesting pharmacological effects, such as analgesic, anti-inflammatory, and antitumor activities.<sup>3</sup> As a part of our ongoing research into alkaloids of chemical, pharmacological, and clinical significance,<sup>4</sup> we investigated the chemical constituents of *G. elegans*, which led to the isolation of two new humantenine-type alkaloids, gelegamines A–B (**1–2**) and three new gelsedine-type alkaloids, gelegamines C–E (**3–5**), together with 10 known ones. In this paper, we describe the isolation and structural identification of gelegamines A–E. The possible biogenetic relationships of these compounds are discussed.

## 2. Results and discussion

### 2.1. Structural elucidation of gelegamines A–E (**1–5**)

Investigation of the MeOH extract of the roots of *G. elegans* (11.2 kg) resulted in the isolation of 5 new compounds named gelegamines A–E (**1–5**) together with 10 known ones: 19(Z)-akuammidine,<sup>5</sup> 19(Z)-16-*epi*-voacarpine,<sup>6</sup> *N*<sub>a</sub>-methoxytaberpsychine,<sup>7</sup> humantenirine,<sup>8</sup> 11-methoxygelsedamide,<sup>9</sup> gelsenicine,<sup>10</sup> 14-hydroxygelsenicine,<sup>11</sup> 19-oxo-gelsenicine,<sup>2b</sup> koumine,<sup>12</sup> and gelsevirine<sup>13</sup> by comparison of their spectral data with those reported in the literature.

The molecular formula of compound **1** was established as C<sub>21</sub>H<sub>24</sub>N<sub>2</sub>O<sub>5</sub> with HRMS (*m/z* 385.1770) [M+H]<sup>+</sup>. The UV absorption at 218 and 266 nm showed the characteristics of an oxindole nucleus.<sup>2b,8</sup> The <sup>1</sup>H NMR spectrum indicated the presence of three aromatic protons attributed to ring A of the oxindole system ( $\delta$  7.36, d, *J*=8.5 Hz; 6.59, dd, *J*=8.5, 2.5 Hz; 6.50, d, *J*=2.5 Hz), an *N*<sub>a</sub>-O-methyl group at  $\delta$  3.94 (3H, s), an *O*-methyl group on the aromatic ring at  $\delta$  3.81 (s), an oxymethine proton at  $\delta$  3.69 (d, *J*=7.0 Hz, H-3), and oxymethylene protons (H<sub>2</sub>-17) at  $\delta$  4.36 (m) and  $\delta$  4.05 (dd, *J*=10.5, 3.5 Hz) (Table 1).

Twenty-one carbon resonances were also resolved in the <sup>13</sup>C NMR spectrum (Table 2), and were further classified via DEPT experiments into 1 carbonyl, 5 quaternary carbons, 9 methines, 3

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**Table 1**  
<sup>1</sup>H (500 MHz, *J* in Hz) NMR data for compounds **1–5**

Position	<b>1</b> <sup>a</sup>	<b>2</b> <sup>a</sup>	<b>3</b> <sup>b</sup>	<b>4</b> <sup>a</sup>	<b>5</b> <sup>a</sup>
3	3.69 (d, 7.0)	3.75 (d, 6.5)	3.56 (d, 6.0)	3.71 (m)	3.73 (m)
5	4.37 (m)	4.07 (br s)	4.26 (m)	4.42 (m)	4.72 (m)
6	2.54 (dd, 15.5, 7.5), 2.23 (m)	2.24 (dd, 15.5, 5.5), 2.14 (dd, 15.5, 3.0)	2.61 (dd, 15.6, 4.6), 2.52 (dd, 15.6, 2.4)	2.27 (m), 2.35 (m)	2.52 (m), 2.31 (m)
9	7.36 (d, 8.5)	7.26 (d, 8.5)	7.48 (d, 7.5)	7.41 (d, 10.0)	7.42 (d, 10.0)
10	6.59 (dd, 8.5, 2.5)	6.59 (dd, 8.5, 2.5)	7.20 (t, 7.5)	6.57 (dd, 10.0, 2.5)	6.59 (dd, 10.0, 2.5)
11			7.39 (t, 7.5)		
12	6.50 (d, 2.5)	6.54 (d, 2.5)	7.10 (d, 7.5)	6.47 (d, 2.5)	6.48 (d, 2.5)
14	2.44 (m), 2.27 (m)	2.57 (m), 2.32 (dd, 15.0, 4.0)	3.04 (m), 2.31 (m)	2.35 (m), 2.14 (m)	2.29 (m), 2.21 (m)
15	2.09 (m)	3.19 (br m)	2.78 (d, 9.5)	2.59 (br t, 10.0)	2.60 (br t, 10.0)
16	2.41 (m)	2.49 (br s)	3.02 (m)	2.89 (t, 11.5)	3.43 (t, 11.5)
17	4.36 (m), 4.05 (dd, 10.5, 3.5)	4.25 (d, 11.0), 4.18 (dd, 11.0, 2.5)	4.37 (d, 12.0), 4.18 (dd, 12.0, 3.0)	4.28 (m), 4.27 (m)	4.31 (m), 4.29 (m)
18	1.43 (d, 5.5)	1.87 (d, 7.5)	1.39 (d, 6)	1.30 (t, 9.5)	2.66 (s)
19	3.36 (q, 5.5)	7.11 (q, 7.5)	4.61 (m)	2.74 (m), 2.46 (m)	
21	7.32 (1H, s)		4.03 (m), 3.81 (m)		
N <sub>a</sub> -Ome	3.94 (s)	3.94 (s)	4.01 (s)	3.94 (s)	3.93 (s)
Ar-OCH <sub>3</sub>	3.81 (s)	3.82 (s)		3.81 (s)	3.82 (s)
N <sub>b</sub> -Me			3.07 (s)		

<sup>a</sup> Measured in CDCl<sub>3</sub>.<sup>b</sup> Measured in CD<sub>3</sub>OD.

methylenes, 1 methyl, and 2 *O*-methyl carbons. A comprehensive analysis of the 1D and 2D NMR (<sup>1</sup>H-<sup>1</sup>H COSY, HSQC, and HMBC) spectra indicated that **1** is an analogue of humanenirine.<sup>8</sup> Two oxygenated carbons at  $\delta$  59.6 and 58.5 were assigned to C-19 and C-20, respectively, as part of an epoxide, by the HMBC correlations from H-15 and H<sub>3</sub>-18 to C-19 and C-20, whereas an imine carbon at  $\delta$  162.4 was assigned to C-21 by the HMBC correlations from H-21 to C-15, C-19, and C-20. Therefore, the chemical structure of **1** was established as shown in Figure 1.

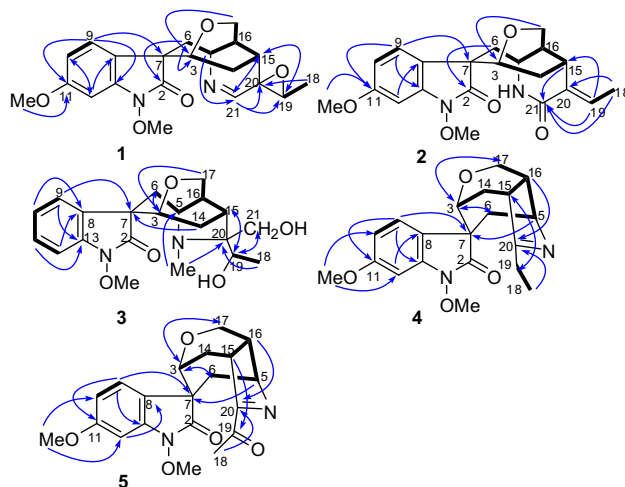
The relative configuration of **1** was deduced from ROESY experiments and molecular modeling (Gaussian D.01)<sup>14</sup> using ab initio calculations. In the ROESY spectrum, the cross peaks observed between the proton pairs H-3/H<sub>2</sub>-14, H<sub>2</sub>-14/H-15, H-9/H-6, H-6/H-5, and H-5/H-16 indicated that the relative configuration of C-3, C-5, C-15, and C-16 in **1**, as shown (Fig. S6, Supplementary data), is identical to that in humanenirine.<sup>8</sup> The ROESY correlation between H-19 and H-21 indicated that both protons were on the same side. Although ROESY correlations of H<sub>3</sub>-18/H-14a, H<sub>3</sub>-18/H-14b, H<sub>3</sub>-18/H-15, and H-19/H-21 were unambiguous, it is not sufficient to determine the orientation of the C-19/C-20 epoxy ring. Therefore, DFT

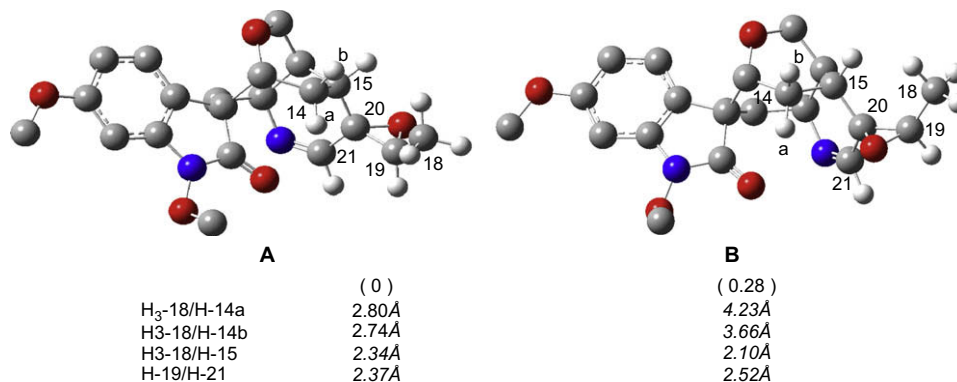
calculations at the B3LYP/6-31G (d) level were made for the two possible structures of **1**, corresponding to the  $\alpha$  (**A**) or  $\beta$  (**B**) orientations of the C-19/C-20 epoxy ring as shown in Figure 2. Two optimized structures were obtained, in which the calculated distance of the proton pairs near the epoxide oxygen of **A** was fully consistent with the corresponding ROESY data. Therefore, the orientation of the C-19/C-20 epoxy ring was determined to be  $\alpha$ .

The molecular formula of compound **2** was established as C<sub>21</sub>H<sub>24</sub>N<sub>2</sub>O<sub>5</sub> with HRMS (*m/z* 385.1756) [M+H]<sup>+</sup>, UV absorption revealed an oxindole nucleus.<sup>2b,8</sup> The <sup>13</sup>C NMR (Table 2) and DEPT spectra showed the presence of 21 carbon signals composed of 2 carbonyl carbons, 8 double-bond carbons (6 aromatic and 2 vinylic), and 11 sp<sup>3</sup> carbons (2 *O*-methyls, 1 methyl, 3 methylenes, 4 methines, and 1 quaternary carbon). The spectroscopic properties of **2** were reminiscent of those of humanenirine.<sup>8</sup> The main difference from humanenirine was a methylene at C-21 in the later was replaced by a conjugated carbonyl carbon ( $\delta$ <sub>C</sub> 166.5) due to UV absorption band at 256 nm. The HMBC correlations from H-15, H-19 and H<sub>3</sub>-18 to C-21 verified this deduction. In addition, by comparison of chemical shift of C-15 in **2** with that of humanenirine, a obvious upfield-shift about 6.8 ppm was observed, this suggested the configuration of C-18 in **2** should be *E*-geometry due to  $\gamma$ -gauche effect between C-15 and C-18 in **2**, which was further confirmed by strong ROESY correlation of H-15/H<sub>3</sub>-18. Based on further 2D-NMR analysis, the structure of **2**, gelegamine B, was determined as shown

**Table 2**  
<sup>13</sup>C (100 MHz) NMR data for compounds **1–5**

Position	<b>1</b> <sup>a</sup>	<b>2</b> <sup>a</sup>	<b>3</b> <sup>b</sup>	<b>4</b> <sup>a</sup>	<b>5</b> <sup>a</sup>
2	172.4	172.6	177.2	171.7	171.6
3	73.5	73.5	76.2	75.2	75.5
5	59.2	53.5	72.4	72.0	74.4
6	35.9	38.0	29.8	37.8	38.2
7	55.6	55.0	58.2	55.4	55.9
8	123.2	121.3	131.5	123.9	123.6
9	125.3	125.7	126.7	125.4	125.3
10	107.8	107.8	125.3	107.7	107.9
11	160.2	160.4	130.2	160.1	160.3
12	94.3	94.7	109.2	93.9	94.0
13	139.2	139.9	139.9	139.1	139.1
14	25.1	29.8	24.0	26.9	27.4
15	28.0	27.5	38.0	39.6	39.3
16	31.3	34.8	37.5	42.5	38.9
17	66.4	65.9	63.4	62.0	61.7
18	14.4	14.2	20.5	9.9	26.1
19	59.6	136.9	65.6	25.6	197.6
20	58.9	133.4	78.7	185.1	178.0
21	162.4	166.5	62.4		
N <sub>a</sub> -Ome	63.4	63.6	64.5	63.4	63.4
Ar-OCH <sub>3</sub>	55.6	55.6		55.5	55.6
N <sub>b</sub> -Me			35.0		

<sup>a</sup> Measured in CDCl<sub>3</sub>.<sup>b</sup> Measured in CD<sub>3</sub>OD.**Figure 1.** <sup>1</sup>H-<sup>1</sup>H COSY (bold) and HMBC (arrow, H → C) correlations for compounds **1–5**.



**Figure 2.** Two DFT-optimized possible structures (A and B) were found for gelegamine A (1). Energies in kcal/mol relative to the more stable one (A) are given in parentheses. Calculated distance of the proton pairs near the epoxide in the two structures is also available.

(Figs. 1 and 3). To best of our knowledge, compound **2** is the first humantenine-type alkaloid with 19-(*E*) ethylidene configuration.

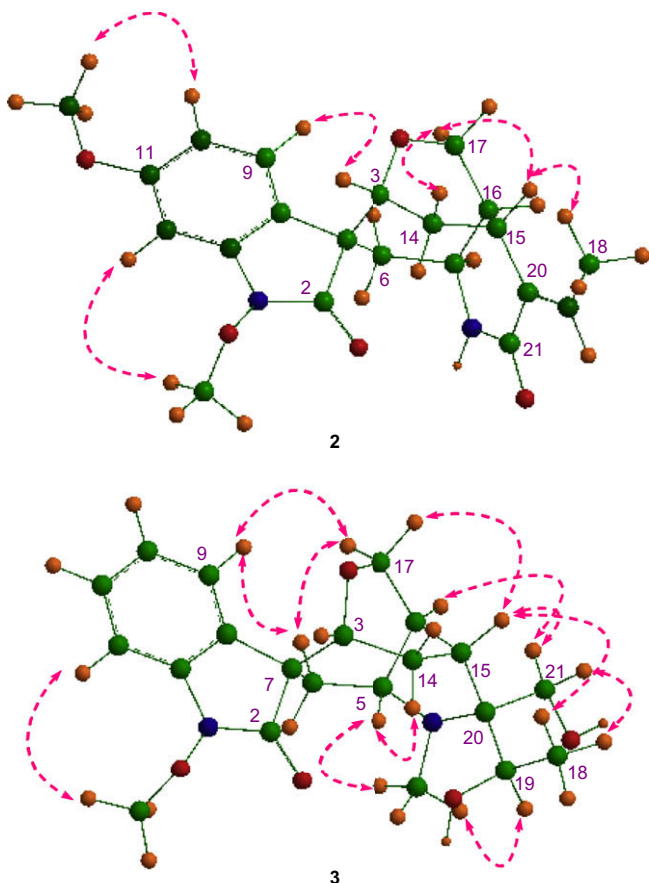
The molecular formula of the alkaloid **3** was established as C<sub>21</sub>H<sub>28</sub>N<sub>2</sub>O<sub>5</sub> with HRMS ( $m/z$  389.2077 [M+H]<sup>+</sup>). The <sup>1</sup>H and <sup>13</sup>C NMR data (Tables 1 and 2) showed the presence of 1 carbonyl, 3 double-bonds (2 trisubstituted and 1 disubstituted), and 14 sp<sup>3</sup> carbons (1 N<sub>a</sub>-O-methyl, 2 methyls, 4 methylenes, 5 methines, and 2 quaternary carbons). Comparison of the NMR data for **3** with those of 19-hydroxy-11-methoxygelselegine,<sup>15</sup> which showed that their chemical shifts were similar, except for the lack of the O-methyl signal at C-11 and the presence of a methyl carbon signal at  $\delta$  35.0 ppm. The downfield shift of  $\sim$ 10 ppm for C-5 and C-20 in **3** relative to those in 19-hydroxy-11-methoxygelselegine suggests

that the methyl group is connected to N<sub>b</sub>. All the NMR data imply that gelegamine C (**3**) is 19-hydroxy-N<sub>b</sub>-methylgelselegine, which was confirmed with 2D NMR experiments. The planar structure and the relative configuration of **3** are shown in Figures 1 and 3, respectively.

The HRMS analysis of the new alkaloid **4**, showed a protonated molecular ion peak at  $m/z$  357.1817 ([M+H]<sup>+</sup>), corresponding to the molecular formula C<sub>20</sub>H<sub>24</sub>N<sub>2</sub>O<sub>4</sub> ( $m/z$  356.1736). The UV absorption at 261 (3.55), 240 (3.67), and 217 (3.49) nm revealed the presence of an oxindole nucleus.<sup>2b</sup> The <sup>1</sup>H NMR spectrum of **4** (Tables 1 and 2) exhibited characteristic signals for a 1,2,4-trisubstituted phenyl group ( $\delta$  6.47, d,  $J$ =2.5 Hz; 6.57, dd,  $J$ =10.0, 2.5 Hz; 7.41, d,  $J$ =10.0 Hz), an N<sub>a</sub>-O-methyl group ( $\delta$  3.94, s), an O-methyl group ( $\delta$  3.81, s), an oxymethine group ( $\delta$  3.71, m, H-3), an oxymethylene ( $\delta$  4.28, m; 4.27, m, H<sub>2</sub>-17), and an ethyl group ( $\delta$  1.30 [3H, t,  $J$ =9.5 Hz, H<sub>3</sub>-18], 2.74, m, 2.46, m [each 1H, H<sub>2</sub>-19]). Comparison of the NMR data of **4** with those of the known alkaloid gelsenicine<sup>10</sup> indicated that their chemical shifts were similar, except for the signals of the aromatic protons, suggesting that compound **4** is a gelsenicine derivative, with an O-methyl group attached to either C-10 or C-11. The presence of an HMBC correlation between H-10 ( $\delta$  6.57, dd,  $J$ =10.0, 2.5 Hz) and O-methyl carbon ( $\delta$  55.5) and a ROESY correlation between H-3 and H-9 unambiguously led to the conclusion that the O-methyl group is located at C-11. Thus, the structure of gelegamine D (**4**) was assigned as shown in Figure 1.

The molecular formula of alkaloid **5** was established as C<sub>20</sub>H<sub>22</sub>N<sub>2</sub>O<sub>5</sub> with HRMS ( $m/z$  393.1420 [M+Na]<sup>+</sup>). The UV spectrum displayed the characteristics of an oxindole nucleus, and the IR spectrum displayed absorption of a carbonyl group at 1714 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectrum showed the signals of a methine proton at  $\delta$  4.72 (m) connected to the nitrogen atom of an imine functional group, three aromatic protons attributed to ring A of the oxindole system ( $\delta$  7.42 [d,  $J$ =10.0 Hz]; 6.59 [dd,  $J$ =10.0, 2.5 Hz]; 6.48 [d,  $J$ =2.5 Hz]), and oxymethylene protons at  $\delta$  4.29 (m) and 4.31 (m) (H<sub>2</sub>-17) (Table 1). The <sup>13</sup>C and DEPT NMR spectra revealed the existence of a conjugated ketone carbonyl carbon at  $\delta$  197.6, an imine carbon at  $\delta$  178.0, an N<sub>a</sub>-O-methyl carbon at  $\delta$  63.4, a regular O-methyl carbon at  $\delta$  55.6, a methyl carbon at  $\delta$  26.1, a methine carbon at  $\delta$  74.4, and a carbonyl carbon at  $\delta$  171.6 (Table 2). The <sup>1</sup>H and <sup>13</sup>C NMR spectra of **5** are very similar to those of GS-1,<sup>16</sup> except for the lack of a hydroxyl group at C-14. Thus, the structure of **5** was deduced to be 14-deoxy GS-1, namely gelegamine E, which was confirmed by 2D NMR experiments.

The similar patterns of Cotton effects in the CD spectra corresponding to the UV absorption maxima of alkaloids **1**–**4** (Fig. 4) indicate that the chiral centers have an absolute configuration identical to that of other known analogues.<sup>2b,10,17</sup>



**Figure 3.** ROESY (dashed) correlations for gelegamines B and C (**2** and **3**).

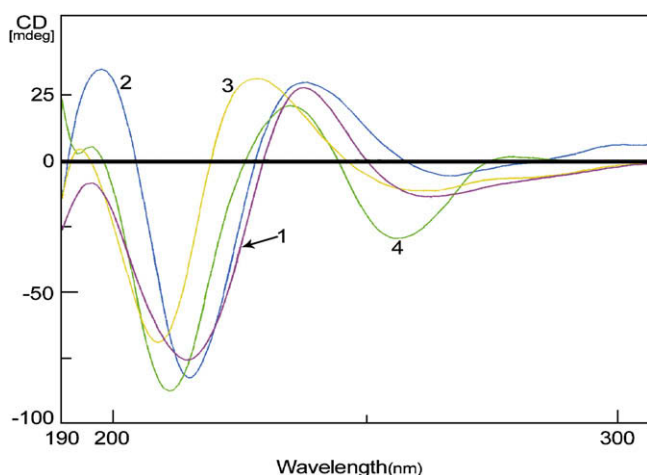


Figure 4. CD spectra for gelegamines A–D (1–4) (in MeOH).

## 2.2. Possible biogenetic route for gelegamines A–E (1–5)

With consideration of the new compounds gelegamines A–E (1–5) isolated from the title plant, a systematic biogenetic pathway from 19-(*E*) ethylidene configuration sarpagine-type alkaloids to gelsedine-type alkaloids was proposed via the sequence as described in Scheme 1: (i) transformation of the sarpagine-type indole alkaloid, gardnerine,<sup>2b</sup> into a C/D ring opening compound,

11-methoxytaberpsychine;<sup>2b</sup> (ii) oxidation rearrangement and methoxylation of *N*<sub>a</sub> function<sup>17,18a</sup> from 11-methoxytaberpsychine to 19*E*-humantenine-type intermediate **7**; (iii) epoxidation from **7** to the key intermediate **9**; (iv) semipinacol rearrangement of **9** to form intermediate **11** with a tetrahydropyrrole and aldehyde functionality; (v) oxidative cleavage of C-21 in **11** to form a gelsedine-type alkaloid **12**, which could be further converted to **4** or **5** via reduction and oxidation, respectively.<sup>18</sup>

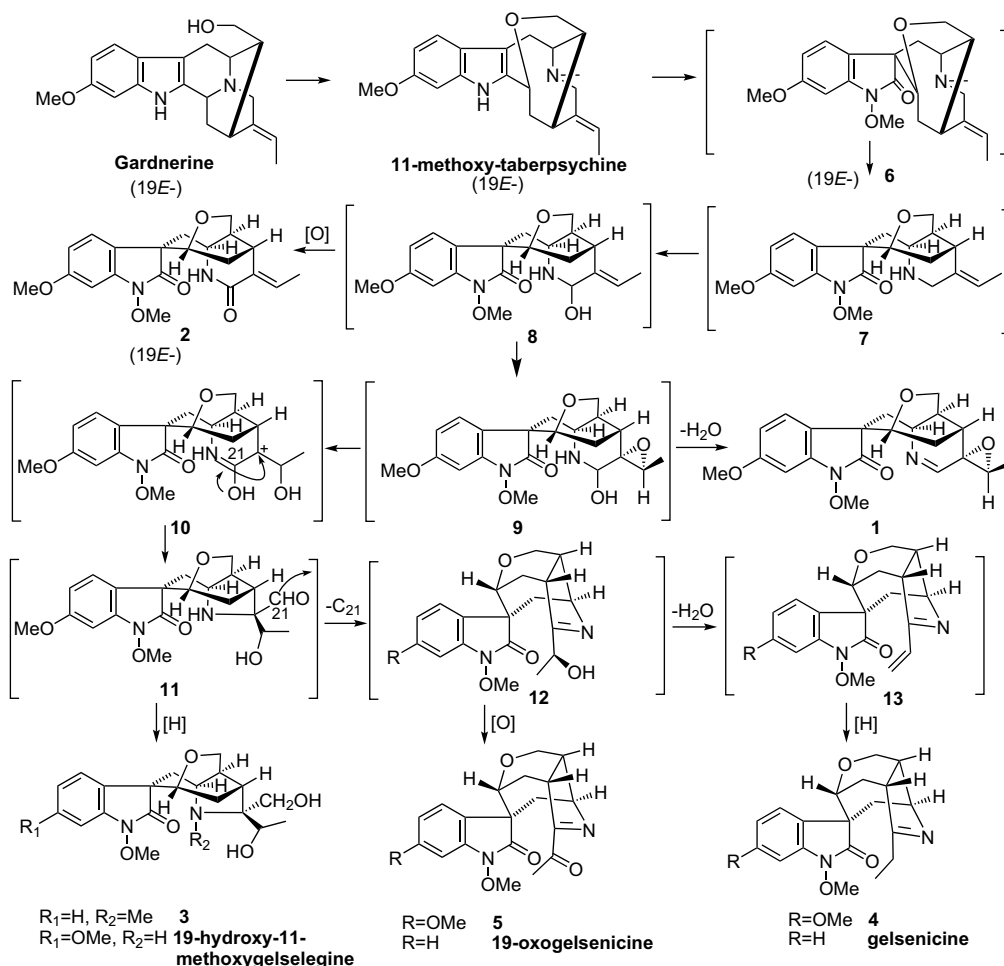
## 2.3. Activity test

The cytotoxic activity of compounds **3–5** and 10 known ones against HL-60 human leukemia and A-549 human lung cancer cell lines were evaluated by follow the standard protocols MTT<sup>19</sup> and SRB<sup>20</sup> methods, respectively. And pseudolaric acid B<sup>21</sup> was used as a positive control. But none showed obvious effects ( $IC_{50} > 1.0 \times 10^{-5}$  M).

## 3. Experimental section

### 3.1. General

Optical rotations were measured on a Perkin–Elmer 241 polarimeter. UV spectra were obtained on a UV-210A spectrometer. CD spectra were recorded on a JASCO J-810 spectropolarimeter. IR spectra were recorded on a Bio-Rad FTS-135 spectrometer with KBr pellets. NMR spectra were recorded on a Bruker AM-500



Scheme 1. Hypothetical biogenetic pathway for *Gelsemium* alkaloids.



spectrometer with TMS as the internal standard. ESIMS were carried out on a Finnigan MAT 90 mass spectrometer and VG Auto Spec-3000 instrument, respectively. Chromatographic separations were performed on silica gel (90–150  $\mu\text{m}$ ; Qingdao Marine Chemical Plant, Qingdao, China) columns, Sephadex LH-20 (40–70  $\mu\text{m}$ ; Amersham Pharmacia Biotech AB, Uppsala, Sweden) columns, or Lichroprep RP-18 gel (40–63  $\mu\text{m}$ ; Merck, Darmstadt, Germany) columns. Semipreparative HPLC was performed on a Zorbax SB-C18 (10  $\mu\text{m}$ ; Agilent Co., Ltd. Wilmington, Delaware) column (i.d.  $9.4 \times 250$  mm), eluted with  $\text{CH}_3\text{OH}$ – $\text{H}_2\text{O}$  (50:50; for 30 min at a flow rate of 3.0 mL/min; detection, UV 254 nm, 280 nm) at 30 °C. Pre-coated silica gel GF<sub>254</sub> and HF<sub>254</sub> plates (Qingdao Haiyang Chemical Plant, Qingdao, China) were used for TLC.

### 3.2. Plant material

The roots of *G. elegans* were collected in Xishuangbanna Tropical Botanical Garden, Chinese Academy of Sciences, Yunnan Province, China, in April 2006. The plant was identified by Prof. De-Ding Tao and a voucher specimen (KIB 06051011) was deposited in the Herbarium of Kunming Institute of Botany, Chinese Academy of Sciences.

### 3.3. Extraction and isolation

The air-dried roots of *G. elegans* (11.2 kg) were percolated with MeOH three times at room temperature and twice under reflux. The combined MeOH extracts were concentrated and the residue (135 g) was suspended in 1.5 L of water, acidified with 5% HCl to pH 3, and partitioned with EtOAc to remove the neutral components. The aqueous phase was then adjusted to pH 9 with a saturated solution of  $\text{Na}_2\text{CO}_3$  before  $\text{CHCl}_3$  extraction. The combined organic phases were concentrated to yield a crude alkaloid mixture (44.6 g), which was subjected to silica gel open-column chromatography, eluted with a  $\text{CHCl}_3$ –MeOH gradient to yield seven fractions (Fr. A–G). Fraction C (160 mg) was subjected to silica gel column chromatography using a gradient solvent system containing petroleum ether–acetone– $\text{Et}_2\text{NH}$  (30:1:0.1 to 5:1:0.1) to yield gelsenicine (23 mg), 14-hydroxygelsenicine (7 mg), 19-oxo-gelsenicine (9 mg), and an alkaloid mixture (Fr. C1). The subfraction Fr. C1 was separated on a silica gel column eluted with petroleum ether–acetone– $\text{Et}_2\text{NH}$  (5:1:0.1 to 3:1:0.1) to yield gelegamine D (**4**, 9 mg), gelegamine E (**5**, 4 mg), 19(*Z*)-akuammidine (3 mg), 19(*Z*)-16-*epi*-voacarpine (7 mg), and another alkaloid mixture, which was further separated by semipreparative HPLC ( $\text{C}_{18}$  reversed-phase silica gel column, MeOH– $\text{H}_2\text{O}$  [50:50]) to yield gelegamine C (**3**, 6.8 mg). Fraction D (120 mg) was separated chromatographically on a silica gel column, eluted with petroleum ether–acetone– $\text{Et}_2\text{NH}$  (15:1:0.1 to 3:1:0.1), to yield six compounds: gelegamine A (**1**, 10 mg), gelegamine B (**2**, 14 mg), gelsevirine (6 mg), humantenirine (7 mg), 11-methoxygelsemamide (8 mg), koumine (11 mg), and  $N$ -a-methoxyanhydrovobasinediol (14 mg), in that order.

### 3.4. Characteristics of alkaloids 1–5

#### 3.4.1. Gelegamine A (**1**)

Amorphous;  $[\alpha]_D^{25} -373.3$  (c 0.60, MeOH); UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 266 (3.42), 218 (3.23) nm; CD (c 1.43 mmol/L, MeOH, 25 °C)  $\Delta\epsilon$  (nm)  $-13.81$  (263),  $+27.49$  (238),  $-75.76$  (215),  $-8.68$  (196); IR (KBr)  $\nu_{\text{max}}$  3426, 2926, 1721, 1628, 1497, 1216, 1109, 1040, 997, 877  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (measured in  $\text{CDCl}_3$ ), see Tables 1 and 2; ESI-MS  $m/z$  385.3  $[\text{M}+\text{H}]^+$ ; HRESIMS  $m/z$  385.1770 (calcd for  $\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_5$ , 384.1685).

#### 3.4.2. Gelegamine B (**2**)

Amorphous;  $[\alpha]_D^{25} -186.1$  (c 0.60, MeOH); UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 256 (3.50), 219 (4.47) nm; CD (c 38.03 mmol/L, MeOH, 25 °C)  $\Delta\epsilon$  (nm)  $-1.01$  (268),  $+5.80$  (238),  $-15.87$  (215),  $+6.78$  (198); IR (KBr)  $\nu_{\text{max}}$  3433, 2927, 1722, 1628, 1441, 1217, 1078, 966, 772  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (measured in  $\text{CDCl}_3$ ), see Tables 1 and 2; ESI-MS  $m/z$  385.4  $[\text{M}+\text{H}]^+$ ; HRESIMS  $m/z$  384.1756 (calcd for  $\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_5$ , 384.1763).

#### 3.4.3. Gelegamine C (**3**)

Amorphous;  $[\alpha]_D^{25} -61.5$  (c 0.13, MeOH); UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 256 (4.05), 208 (4.71) nm; CD (c 70.62 mmol/L, MeOH, 25 °C)  $\Delta\epsilon$  (nm)  $-8.71$  (262),  $+23.50$  (229),  $-52.16$  (209),  $+3.38$  (194); IR (KBr)  $\nu_{\text{max}}$  3421, 2925, 1710, 1618, 1464, 1238, 1038, 997, 750  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (measured in  $\text{CD}_3\text{OD}$ ), see Tables 1 and 2; ESI-MS  $m/z$  388  $[\text{M}+\text{H}]^+$ ; HRESIMS  $m/z$  389.2077  $[\text{M}+\text{H}]^+$  (calcd for  $\text{C}_{21}\text{H}_{28}\text{N}_2\text{O}_5$ , 388.1998).

#### 3.4.4. Gelegamine D (**4**)

Amorphous;  $[\alpha]_D^{25} -87.0$  (c 0.09,  $\text{CHCl}_3$ ); UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 261 (3.55), 240 (3.67), 223 (3.50), 217 (3.49) nm; CD (c 1.40 mmol/L, MeOH, 25 °C)  $\Delta\epsilon$  (nm)  $-24.62$  (257),  $+17.72$  (236),  $-73.28$  (211),  $+4.58$  (196); IR (KBr)  $\nu_{\text{max}}$  3436, 2924, 1724, 1598, 1461, 1216, 879, 715  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (measured in  $\text{CDCl}_3$ ), see Tables 1 and 2; ESI-MS  $m/z$  357.0  $[\text{M}+\text{H}]^+$ ; HRESIMS  $m/z$  357.1517 (calcd for  $\text{C}_{20}\text{H}_{24}\text{N}_2\text{O}_4$ , 356.1736).

#### 3.4.5. Gelegamine E (**5**)

Amorphous;  $[\alpha]_D^{25} -303.3$  (c 0.05,  $\text{CHCl}_3$ ); UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 268 (3.89), 240 (4.07), 223 (4.82), 216 (3.82) nm; IR (KBr)  $\nu_{\text{max}}$  3429, 2923, 1714, 1626, 1496, 1215, 1038, 988, 825  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR (measured in  $\text{CDCl}_3$ ), see Tables 1 and 2; ESI-MS  $m/z$  371.3  $[\text{M}+\text{H}]^+$ ; HRESIMS  $m/z$  393.1420  $[\text{M}+\text{Na}]^+$  (calcd for  $\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_5$ , 370.1529).

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### Supplementary data

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### References and notes

- Editorial Committee of Chinese Materia Medica; the Administration Bureau of Traditional Chinese Medicine. In *Chinese Materia Medica (Zhonghua Bencao)*; Shanghai Science & Technology: Shanghai, 2000; Vol. 6, pp 213–215.
- (a) Kitajima, M. *J. Nat. Med.* **2007**, *61*, 14–23; (b) Ponglux, D.; Wongseripipatana, S.; Subhadhirasakul, S.; Takayama, H.; Yokota, M.; Ogata, K.; Phisalaphong, C.; Aimi, N.; Sakai, S. *Tetrahedron* **1988**, *44*, 5075–5094; (c) Xu, Y. K.; Yang, S. P.; Liao, S. G.; Zhang, H.; Ling, L. P.; Ding, J.; Yue, J. M. *J. Nat. Prod.* **2006**, *69*, 1347–1350; (d) Yin, S.; He, X. F.; Wu, Y.; Yue, J. M. *Chem. Asian J.* **2008**, *3*, 1824–1829.
- (a) Rujjanawate, C.; Kanjanapothi, D.; Panthong, A. *J. Ethnopharmacol.* **2003**, *89*, 91–95; (b) Zhang, L. L.; Lin, J. M.; Wu, Z. J. *Chin. Med. Mat.* **2003**, *26*, 451–453; (c) Chi, D. B.; Lei, L. S.; Pang, J. X.; Jiang, Y. P. *Acad. J. First Mil. Med. Univ.* **2003**, *23*, 911–913.
- (a) Li, C. S.; Di, Y. T.; Mu, S. Z.; He, H. P.; Zhang, Q.; Fang, X.; Zhang, Y.; Li, S. L.; Lu, Y.; Gong, Y. Q.; Hao, X. J. *J. Nat. Prod.* **2008**, *71*, 1202–1206; (b) Mu, S. Z.; Wang, J. S.; Yang, X. S.; He, H. P.; Li, C. S.; Di, Y. T.; Wang, Y.; Zhang, Y.; Fang, X.; Huang, L. J.; Hao, X. J. *J. Nat. Prod.* **2008**, *71*, 564–569; (c) Di, Y. T.; Liu, L. L.; Li, C. S.; Zhang, Y.; Zhang, Q.; Mu, S. Z.; Sun, Q. Y.; Yang, F. M.; Liu, H. Y.; Hao, X. J. *Helv. Chim. Acta* **2008**, *91*, 838–843; (d) Tan, C. J.; Di, Y. T.; Wang, Y. H.; Wang, Y.; Mu, S. Z.; Gao, S.; Zhang, Y.; Kong, N. C.; He, H. P.; Zhang, J. X.; Fang, X.; Li, C. S.; Lu, Y.; Hao, X. J. *Tetrahedron Lett.* **2008**, *49*, 3376–3379; (e) Zhang, Y.; He, H. P.; Di, Y. T.; Mu, S. Z.; Wang, Y. H.; Wang, J. S.; Li, C. S.; Kong, N.; Gao, S.; Hao, X. J. *Tetrahedron Lett.* **2007**, *48*, 9104–9107; (f) Li, C. S.; Di, Y. T.; He, H. P.; Gao, S.; Wang, Y. H.; Lu, Y.; Zhong, J. L.; Hao, X. J. *Org. Lett.* **2007**, *9*, 2509–2512; (g) Di, Y. T.; He, H. P.; Wang, Y. S.; Li, L. B.;

- Lu, Y.; Gong, J. B.; Fang, X.; Kong, N. C.; Li, S. L.; Zhu, H. J.; Hao, X. J. *Org. Lett.* **2007**, *9*, 1355–1358; (h) Mu, S. Z.; Wang, Y.; He, H. P.; Yang, X. W.; Wang, Y. H.; Di, Y. T.; Lu, Y.; Chang, Y.; Hao, X. J. *J. Nat. Prod.* **2006**, *69*, 1065–1069; (i) Di, Y. T.; He, H. P.; Liu, H. Y.; Du, Z. Z.; Tian, J. M.; Yang, X. W.; Wang, Y. H.; Hao, X. J. *Tetrahedron Lett.* **2006**, *47*, 5329–5331; (j) Di, Y. T.; He, H. P.; Lu, Y.; Yi, P.; Li, L.; Wu, L.; Hao, X. J. *J. Nat. Prod.* **2006**, *69*, 1074–1076; (k) Li, L.; He, H. P.; Di, Y. T.; Tian, J. M.; Hao, X. J. *Helv. Chim. Acta* **2006**, *89*, 1457–1462; (l) Li, L.; He, H. P.; Di, Y. T.; Gao, S.; Hao, X. J. *Tetrahedron Lett.* **2006**, *47*, 6259–6262; (m) Di, Y. T.; He, H. P.; Li, C. S.; Tian, J. M.; Mu, S. Z.; Li, S. L.; Gao, S.; Hao, X. J. *J. Nat. Prod.* **2006**, *69*, 1745–1748; (n) Hu, X. J.; He, H. P.; Zhou, H.; Di, Y. T.; Yang, X. W.; Hao, X. J. *Helv. Chim. Acta* **2006**, *89*, 1344–1349; (o) He, H. P.; Shen, Y. M.; Zhang, J. X.; Zuo, G. Y.; Hao, X. J. *J. Nat. Prod.* **2001**, *64*, 379–380.
5. Sakai, S. I.; Wongseripipatana, S.; Dhavadee Ponglux, D. *Chem. Pharm. Bull.* **1987**, *35*, 4668–4671.
6. Kogure, N.; Nishiya, C.; Kitajima, M.; Takayama, H. *Tetrahedron Lett.* **2005**, *46*, 5857–5861.
7. Lin, L. Z.; Cordell, G. A.; Ni, C. Z.; Clerdy, J. *Phytochemistry* **1989**, *28*, 2827–2831.
8. Lin, L. Z.; Cordell, G. A. *J. Nat. Prod.* **1989**, *52*, 588–594.
9. Lin, L. Z.; Cordell, G. A. *Tetrahedron Lett.* **1989**, *30*, 1177–1180.
10. Lin, L. Z.; Cordell, G. A.; Ni, C. Z.; Clerdy, J. *J. Org. Chem.* **1989**, *54*, 3199–3202.
11. Yang, J. S.; Chen, Y. W. *Acta Pharmacol. Sin.* **1982**, *17*, 633–634.
12. Lin, L. Z.; Cordell, G. A.; Ni, C. Z.; Clerdy, J. *Phytochemistry* **1990**, *29*, 965–968.
13. Schun, Y.; Cordell, G. A. *J. Nat. Prod.* **1986**, *49*, 483–487.
14. Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Montgomery, J. A. Jr.; Vreven, T.; Kudin, K. N.; Burant, J. C.; Millam, J. M.; Iyengar, S. S.; Tomasi, J.; Barone, V.; Mennucci, B.; Cossi, M.; Scalmani, G.; Rega, N.; Petersson, G. A.; Nakatsuji, H.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Klene, M.; Li, X.; Knox, J. E.; Hratchian, H. P.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Ayala, P. Y.; Morokuma, K.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Zakrzewski, V. G.; Dapprich, S.; Daniels, A. D.; Strain, M. C.; Farkas, O.; Malick, D. K.; Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; Ortiz, J. V.; Cui, Q.; Baboul, A. G.; Clifford, S.; Cioslowski, J.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Martin, R. L.; Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Challacombe, M.; Gill, P. M. W.; Johnson, B.; Chen, W.; Wong, M. W.; Gonzalez, C.; Pople, J. A. *Gaussian 03, Revision D.01*; Gaussian: Wallingford, CT, 2005.
15. Lin, L. Z.; Cordell, G. A.; Ni, C. Z.; Clerdy, J. *Phytochemistry* **1990**, *29*, 3013–3017.
16. Kitajima, M.; Urano, A.; Kogure, N.; Takayama, H. *Chem. Pharm. Bull.* **2003**, *51*, 1211–1214.
17. Takayama, H.; Odaka, H.; Aimi, N.; Sakai, S. I. *Tetrahedron Lett.* **1990**, *31*, 5483–5486.
18. (a) Takayama, H.; Masubuchi, K.; Kitajima, M.; Aimi, N.; Sakai, S. I. *Tetrahedron* **1989**, *45*, 1327–1336; (b) Kitajima, M.; Kogure, N.; Yamaguchi, K.; Takayama, H.; Aimi, N. *Org. Lett.* **2003**, *5*, 2075–2078; (c) Takayama, H.; Kitajima, M.; Ogata, K.; Sakai, S. I. *J. Org. Chem.* **1992**, *57*, 4583–4584.
19. Liu, W. J.; Jiang, J. F.; Xiao, D.; Ding, J. *Biochem. Pharmacol.* **2002**, *64*, 1677–1687.
20. Xiao, D.; Zhu, S. P.; Gu, Z. L. *Acta Pharmacol. Sin.* **1997**, *18*, 280–283.
21. Pan, J. P.; Li, Z. L.; Hu, C. Q.; Chen, K.; Chang, J. J.; Lee, K. H. *Planta Med.* **1990**, *56*, 383–385.