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Two novel tricyclic diterpenoids from *Isodon rubescens* var. *taihangensis*

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Abstract—Two novel tricyclic diterpenoids rubescensins U (1) and V (2) were isolated from the leaves of *Isodon rubescens* var. *taihangensis*. They were elucidated as a 8,15-*seco-ent*-kauranoid and an *ent*-abietanoid, respectively, by 1D and 2D NMR spectra, and single crystal X-ray analysis. Compound 1 is the first example of an 8,15-*seco-ent*-kaurane from the plants genus *Isodon*. A discussion of their biogenesis is described.

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1. Introduction

In recent years, a series of tricyclic diterpenoids were reported from the genus Isodon, which was well-known to be abundant in tetracyclic ent-kaurane diterpenoids.¹ Among them, adenanthin L (3) from I. adenantha,² laxiflorin O (4) from I. eriocalyx var. laxioflora,³ and eriocaside A (5) from I. eriocalyx,⁴ were elucidated as entabietanoids; Melissoidesin L (6) from *I. Melissoides*,⁵ was an abietanoid, and taibaihenryiin C (7) from I. henryi was even regarded as having a novel skeleton,⁶ on the basis of their tricyclic skeleton. In our continuing research for more bioactive substances from the Isodon plants, two tricyclic diterpenoids (1 and 2) were isolated from Isodon rubescens var. *taihangensis* Z. Y. Gao and Y. R. Li,⁷ a famous folk herbal medicine for treatment of cancers.⁸ Compounds 1 and 2 were determined as a 8,15-seco-ent-kauranoid and an ent-abietanoid by the key H-8 β of 1 and H-8 α of 2, respectively. From the similarity in the structures of compounds 1 and 2, a brief discussion of their biogenesis is described.

2. Results and discussion

Compound 1 was obtained as colorless, prismatic crystals

with a molecular formula $C_{20}H_{28}O_6$ determined by the HREIMS. The 20 carbon atoms found in the ¹³C and DEPT NMR spectra of 1 consisted of a ketonic carbon, an aldehydic carbon, an olefinic quaternary carbon, an olefinic methylene carbon, a hemiacetal carbon, seven methine carbons including three oxygenated ones, four methylene carbons, two quaternary carbons, and two methyl carbons, which obviously suggested a diterpene skeleton. Compound 1 was further deduced to be a tricyclic diterpenoid by the absence of a quaternary carbon found in other typical entkauranoids also isolated from the same plant, such as lasiodonin (8),⁹ and the presence of H-8 clearly exhibiting HMBC correlations with C-10, C-11, and C-13 (Table 1). Because H-8 has been determined to be of a β orientation by the ROESY correlations of H-8/H-5ß and H-8/H-9ß, and considering the structures of diterpenoids isolated from this plant, compound 1 was deduced to be a 8,15-seco-entkauranoid, instead of an ent-abietanoid.

The remaining oxygenated functionalities of **1** were established accordingly. OH-1 α and OH-7 α were deduced by the HMBC correlations of H-1/C-5 and C-9, H-7/C-5 and C-9 (Table 1), and the ROESY correlations of H-1 β /H-5 β and H-7 β /H-8 β (Fig. 1). The ketonic carbon was assigned as C-6 by the long-range correlations of H-5 and H-7 with C-6 in the HMBC spectrum. Based on the analysis of the relational HMBC correlations of **1** (Table 1), the olefinic bond conjugated with the aldehydic group was located at C-13. The 11,20-epoxy group was also deduced in the same way. Consequently, with the aid of the NOEs of H-11 α /H-13 α and H-20/Me-19 in the ROESY spectrum, compound **1**

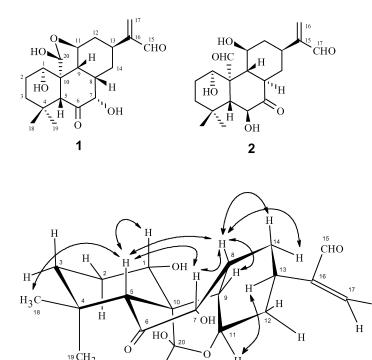
Keywords: Labiatae; Isodon rubescens var. taihangensis; 8,15-seco-ent-Kauranoid; ent-Abietanoid; Rubescensin U; Rubescensin V.

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| No. | 1 | | | 2 | | |
|-----------|------------------------------------|-----------------|---------------------|----------------------------|----------------------|---------------------|
| | $^{1}\mathrm{H}$ | ¹³ C | HMBC ^b | ¹ H | ¹³ C | HMBC ^b |
| 1 | 3.92-3.95 m | 76.9 d | 9, 20 | 4.22–4.27 m | 74.7 d | 2, 3, 5, 9, 20 |
| 1-OH | 6.15 d, 8.0 | | 1, 10 | 7.95 s | | |
| 2 | 2.80–2.84 m 2.00–2.06 m | 29.8 t | 4, 10 | 2.10–2.15 m 2.03–2.08 m | 28.9 t | 1, 3, 4, 10 |
| 3 | 1.51 overlap 1.38 dt, 4.0, 13.2 | 41.4 t | 1, 4, 5, 18, 19 | 1.60–1.65 m 1.39–1.45 m | 39.7 t | 1, 2, 4, 5 |
| 4 | | 32.7 s | | | 34.9 s | |
| 5 | 2.49 s | 57.8 d | 6, 7, 9, 18, 19, 20 | 1.66 d, 10.0 | 57.5 d | 6, 7, 9, 10, 18, 19 |
| 6 6-OH | | 214.8 s | | 4.67 d, 10.0 8.02 s | 75.2 d | 4, 5, 7 |
| 7 | 4.52 d, 8.0 | 77.7 d | 5, 6, 8, 9 | | 210.8 s | |
| 7-OH | 6.88 s | | 6, 8 | | | |
| 8 | 3.08-3.11 m | 37.2 s | 10, 11, 13 | 3.37 dt, 2.2, 12.0 | 47.5 d | 7, 9, 11, 14 |
| 9 | 2.68 dd, 3.6, 11.0 | 52.6 d | 1, 5, 7, 8, 12 | 1.71 br t, 12.0 | 59.9 d | 5, 8, 10, 11, 20 |
| 10 | | 54.4 s | | | 58.7 s | |
| 11 | 3.66 dt, 3.0, 11.0 | 72.5 d | 8, 10, 20 | 4.40–1.45 m | 70.3 d | 8, 9, 12 |
| 11-OH | | 24.5 | | 5.79 br s | 11.0 | |
| 12 | 2.32 overlap 1.58–1.63 m | 36.7 t | 9, 13, 14, 16 | 2.36–2.41 m 1.51–1.55 m | 41.9 t | 9, 11, 13, 14, 16 |
| 13 | 3.81–3.85 m | 33.4 d | 8, 11, 14, 15, 17 | 2.72 br t, 12.5 | 32.8 d | 16 |
| 14 | 2.28 overlap | 31.9 t | 8, 9, 13, 16 | 2.25 dd, 2.2, 13.2 | 31.8 t | 7, 8, 9, 12, 16 |
| | 1.41–1.46 m | | .,,,, | 1.45–1.49 m | | ., ., , ,, |
| 15 | 9.53 s | 194.5 d | 13, 17 | | 153.6 s | |
| 16 | | 154.8 s | | 6.19, 5.89 (each 1H, s) | 133.4 t | 13, 15, 17 |
| 17 | 6.12, 5.84 (each 1H, s) s | 133.6 t | 13, 15 | 9.55 s | 194.4 d | 13, 16 |
| 18 | 1.01 s (3H) | 30.3 q | 3, 4, 5, 19 | 1.40 s (3H) | 34.0 q | 3, 4, 5, 19 |
| 19 | 1.67 s (3H) | 21.0 g | 3, 4, 5, 18 | 1.27 s (3H) | 23.1 q | 3, 4, 5, 18 |
| 20 | 5.73 d, 8.0 | 103.2 d | 1, 5, 9, 11 | 10.73 s | 207.9 [°] d | 10 |
| 20-OH | 8.47 d, 8.0 | | 10, 20 | | | |

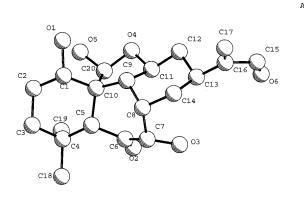
Table 1. NMR spectral data and HMBC correlations for 1 and $2^{\rm a}$

 $^{\rm a}$ $^{\rm 1}{\rm H}$ NMR, 400 MHz; $^{\rm 13}{\rm C}$ NMR, 100 MHz, pyridine- d_5 ; data in ppm (J in Hz). $^{\rm b}$ From H to C.



Η

19 ĊH3



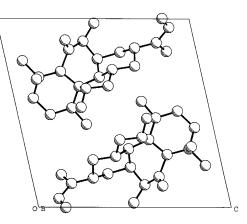


Figure 2. Crystal structure of 1.

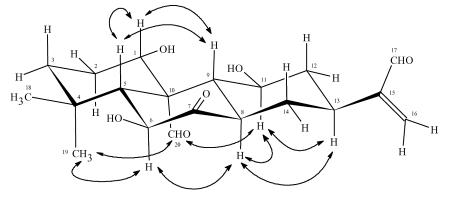


Figure 3. Selected ROESY correlations for 2.

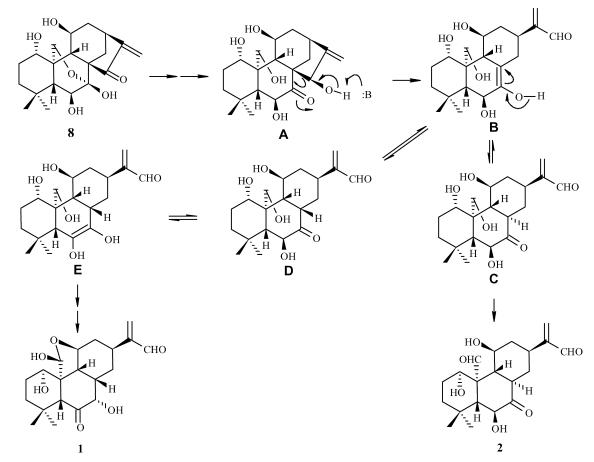


Figure 4. A plausible biogenetic pathway to account for the formation of compounds 1 and 2.

was elucidated as $20(S)-1\alpha,7\alpha,20$ -trihydroxy-6,15-dioxo-11 $\beta,20$ -epoxy-8,15-*seco-ent*-kaur-16(17)-ene, named rubescensin U (1). Finally, the X-ray crystallographic analysis of 1 (Fig. 2) confirmed stereochemically that rings A and C were in chair forms, and ring B showed a twist boat conformation. This 8,15-*seco-ent*-kaurene skeleton was proved and reported from the genus *Isodon* plants for the first time.¹⁰

Similarly, compound **2** was also educed to be a tricyclic diterpenoid. A comparison of the ¹H and ¹³C NMR data of compounds **2** and **1** suggested that **2** was derived from the hydrolysis of the hemiacetal group at C-20 of **1**. Further analysis of the ROESY spectrum of **2** revealed a key difference between **2** and **1**, in that the H-8 of **2** showed NOEs with H-13 α and H-11 α (Fig. 3) instead of correlating with H-5 β and H-9 β , indicating the α -orientation of H-8. This presence of H-8 α was confirmed by the coupling constant (*J*=12.0 Hz) between H-8 and H-9 β , and indicated that **2** is an *ent*-abietanoid. Accordingly, by the ROESY correlation of H-6 α /Me-19, compound **2** was established as $1\alpha, 6\beta, 11\beta$ -trihydroxy-7,17,20-trioxo-*ent*-abiata-15(16)-ene, and named rubescensin V.

The biogenesis from lasiodonin 8, one of the major *ent*-kauranoids of this plant, to compounds 1 and 2 was postulated (Fig. 4) to explain their origins. In the proposed biogenetic pathway, a retroaldol reaction from A to B resulted in the key transformation that converted a tetracyclic *ent*-kaurane to a tricylic diterpenoid.^{11,12} The keto-end equilibration of B gave C and D, and determined the key stereochemical difference between 1 and 2. The subsequent enolization from D to E, oxidation and hemiacetalization yielded 1. Compound 2 was derived from the oxidation of C. Thus, the *ent*-abietanoid 2 could have originated from an *ent*-kaurane.

3. Experimental

3.1. General procedures

Melting points were measured on an XRC-1 micro melting point apparatus and were uncorrected. Optical rotations were measured on a JASCO DIP-370 digital polarimeter. IR spectra were obtained on a Bio-Rad FtS-135 spectrophotometer with KBr pellets. MS were recorded on a VG Auto Spec-3000 spectrometer. 1D- and 2D NMR spectra were obtained on the Bruker AM-400 and DRX-500 instruments with TMS as an internal standard.

3.2. Plant material

The leaves of *Isodon rubescens* var. *taihangensis* were collected from Hebi Prefecture, Henan Province, in August 2000, and identified by Professor Z. W. Lin, Kunming Institute of Botany. A voucher specimen has been deposited in the Herbarium of the Kunming Institute of Botany, Chinese Academy of Sciences.

3.3. Extraction and isolation

The 70% Me₂CO extracts of the air-dried and powdered

leaves of *I. rubescens* var. *taihangensis* (10 kg) were partitioned with EtOAc to afford the EtOAc extract (400 g), which was subjected to silica gel column chromatography using CHCl₃, CHCl₃–Me₂CO (9:1, 8:2, 7:3, 6:4) and Me₂CO as eluents. Compounds **1** and **2** (14 and 6 mg) were obtained from the CHCl₃–Me₂CO (7:3) fraction after repeated silica gel column chromatographic separations, followed by preparative TLC and recrystallization from MeOH.

3.3.1. Compound 1. Colorless prismatic crystals. Mp 202–204 °C; $[\alpha]_D^{21.6}$ =-60.0 (*c*=0.1, acetone); IR (KBr) ν_{max} : 3433, 2928, 1716, 1683, 1683, 1124 cm⁻¹. ¹H NMR (C₅D₅N, 400 MHz) and ¹³C NMR (C₅D₅N, 100 MHz): see Table 1; EI-MS (70 eV) *m/z* (%): 364 (M⁺, 3), 346 (20), 328 (8), 318 (40), 300 (15); HREIMS *m/z*: [M]⁺ 364.1897 (calcd for C₂₀H₂₈O₆ 364.1886).

Crystal data for 1. Crystals of 1, crystallized from methanol, belong to the monoclinic space group $P2_1$. Crystal data: $C_{20}H_{28}O_6 H_2O, M=364.43, a=12.368(2), b=6.275(1),$ c=12.289(2) Å, $\beta=102.76(1)^\circ$, V=930.2(3) Å³, Z=2, d=1.301 g/cm⁻³, Mo K α radiation, linear absorption coefficient $\mu = 1.0 \text{ cm}^{-1}$. A colorless quadrate lumpish crystal of dimensions 0.02×0.15×0.60 mm³ was used for X-ray measurements on a MAC DIP-2030 diffractometer with a graphite monochromator, maximum 2θ value of 50.0° was set. The total number of independent reflections measured was 1530, 1431 of which were considered to be observed $(|F|^2 \ge 8\sigma |F|^2)$. The structure was solved by the direct method SHELX-86 and expanded using difference Fourier techniques, refined by the program and method NOMCSDP¹³ and full-matrix least-squares calculations. Hydrogen atoms were fixed at calculated positions. The final indices were $R_f = 0.071$, $R_w = 0.070$ (w=1/ $\sigma |F|^2$).

3.3.2. Compound **2.** White amorphous powder; $[\alpha]_D^{21.4} = -5.0$ (c=0.2, acetone); IR (KBr) ν_{max} : 3441, 2928, 1705, 1683, 1084 cm⁻¹. ¹H NMR (C₅D₅N, 400 MHz) and ¹³C NMR (C₅D₅N, 100 MHz): see Table 1; (+) FAB-MS m/z: 365 ([M+1]⁺); (+) HRFABMS m/z: [M+H]⁺ 365.1987 (calcd for C₂₀H₂₉O₆ 364.1964).

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