# Novel taxoids from the Chinese yew Taxus yunnanensis 

Sheng-Hong Li, ${ }^{\text {a }}$ Hong-Jie Zhang, ${ }^{\text {a,b,* }}{ }^{*}$ Xue-Mei Niu, ${ }^{\text {a }}$ Ping Yao, ${ }^{\text {b }}$ Han-Dong Sun ${ }^{\text {a }}$ and Harry H. S. Fong ${ }^{\text {b,* }}$<br>${ }^{\text {a }}$ Laboratory of Phytochemistry, Kunming Institute of Botany, The Chinese Academy of Sciences, Kunming, 650204 Yunnan, People's Republic of China<br>${ }^{\mathrm{b}}$ Program for Collaborative Research in the Pharmaceutical Sciences, Department of Medicinal Chemistry and Pharmacognosy, College of Pharmacy, University of Illinois at Chicago, 833 S. Wood Street, Chicago, IL 60612, USA

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#### Abstract

Six new taxoids, taxuyunnanines $\mathrm{W}(\mathbf{1}), \mathrm{X}(\mathbf{2}), \mathrm{Y}(\mathbf{3}), \mathrm{Z}(\mathbf{5}), 10$-deacetyl-10-( $\beta$-glucopyranosyl)-yunnanxane (8), and 7-( $\beta$ -xylosyl)- $N$-methyl-taxol C (10), as well as taxuchin B (4) and yunnanxane (7), have been isolated from the bark of Taxus yunnanensis. Compounds 1 and 2 were characterized as $11(15 \rightarrow 1)$-abeo taxoids containing an orthoester group, which have not been reported from any Taxus plants to date; isolates $\mathbf{3}$ and $\mathbf{5}$ are the examples of taxanes having a tetrahydronfuran ring, being reported only for the second time; and $\mathbf{8}$ is the first natural $6 / 8 / 6$ skeletal type taxane glucosidic form. This is also the first report of a co-occurrence of xylosidation and N methylation in a taxane (10). Compound $\mathbf{4}$ is the only known chlorine-containing taxane found in nature thus far, and is being reported from this plant for the first time. The structures were determined by spectroscopic and chemical means including 1D and 2D NMR spectra. © 2002 Elsevier Science Ltd. All rights reserved.


## 1. Introduction

The naturally occurring diterpene, paclitaxel, originally isolated from Taxus brevifolia by Wani et al., ${ }^{1}$ has been considered the most promising natural anticancer agent. Our search for novel and bioactive taxanes have resulted in the isolation of many types of new and known compounds of this class from the roots and barks of Taxus yunnanensis Cheng et L. K. Fu, a species endemic to Southwestern China and Burma. ${ }^{2-11}$ In our continuing work on this plant, we have discovered two new orthoester containing taxoids, taxuyunnanines $\mathrm{W}(\mathbf{1})$ and $\mathrm{X}(\mathbf{2})$, two new tetrahydronfuran ring containing taxoids, taxuyunnanines $Y(\mathbf{3})$ and $Z(5)$, a new $6 / 8 / 6$ skeletal type taxane glucoside, 10-deacetyl-10( $\beta$-glucopyranosyl)yunnanxane (8), and a new $N$-methylated taxane xyloside, 7 -( $\beta$-xylosyl)- $N$-methyltaxol C (10), as well as the known chlorine-containing taxoid, taxuchin $B$ (4) ${ }^{12}$ and the known yunnanxane (7). ${ }^{13}$ The structures were determined by spectroscopic and chemical means including 1D and 2D NMR spectra. The current paper reports the

[^0]isolation and structure elucidation of these novel natural products.

## 2. Results and discussion

Taxuyunnanine W (1), white amorphous solid, showed $[\mathrm{M}+\mathrm{gly}-\mathrm{H}]^{+}$at $\mathrm{m} / \mathrm{z} 721$ and $[\mathrm{M}-\mathrm{H}]^{+}$at $\mathrm{m} / \mathrm{z} 629$ in the negative FABMS, which corresponds to a molecular weight of 630 . A molecular formula of $\mathrm{C}_{33} \mathrm{H}_{42} \mathrm{O}_{12}$ was determined for 1 through high resolution FABMS measurement of the ion at $\mathrm{m} / \mathrm{z} 629\left[\left(\mathrm{~m} / \mathrm{z} 629.2546\right.\right.$, calcd: 629.2598, $\left.[\mathrm{M}-\mathrm{H}]^{+}\right)$. In the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra (Tables 1 and 2 ) of $\mathbf{1}$, the presence of four characteristic taxane-type methyl signals [ $\delta_{\mathrm{H}} 1.81(3 \mathrm{H}, \mathrm{s}), 1.76(3 \mathrm{H}, \mathrm{s}), 1.75(3 \mathrm{H}, \mathrm{s}), 1.33(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}$ $28.1(\mathrm{q}), 28.0(\mathrm{q}), 15.4(\mathrm{q}), 13.2(\mathrm{q})]$, a doublet signal at $\delta_{\mathrm{H}}$ $3.29(J=10.6 \mathrm{~Hz})$ ascribable to $\mathrm{H}-3$, two quaternary carbon signals at $\delta_{\mathrm{C}} 76.2$ and 66.8 and two olefinic quaternary carbons at $\delta_{\mathrm{C}} 148.7$ and 134.4 indicated that $\mathbf{1}$ is a $11(15 \rightarrow 1)$-abeotaxoid. ${ }^{14,15}$ In addition, three ester groups were identified to consist of a benzoxy and two acetoxy groups in the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra. However, unlike normal abeotaxoids, some unusual signals [ $\delta_{\mathrm{C}} 119.8 \mathrm{ppm}$ (s, $\mathrm{CD}_{3} \mathrm{OD}$ was employed as solvent unless otherwise specified), $\delta_{\mathrm{H}} 1.50(3 \mathrm{H}, \mathrm{s})$ and $\left.\delta_{\mathrm{C}} 22.1(\mathrm{q})\right]$ were observed, which suggested the presence of an orthoacetoxy group in 1 . This orthoacetoxy group was thought to be connected to the taxane skeleton either through three oxygen bridges or two oxygen rings according to the number of double bond equivalents in its molecular formula.

Table 1. ${ }^{1} \mathrm{H}$ NMR data of $\mathbf{1}, \mathbf{1 a}$ and $\mathbf{2}(500 \mathrm{MHz}, J$ in $\mathrm{Hz}, \delta$ in ppm)

| Proton | $1^{\text {a }}$ | $1^{\text {b }}$ | $1 \mathrm{a}^{\text {b }}$ | $2^{\text {b }}$ |
| :---: | :---: | :---: | :---: | :---: |
| H-2 | 5.75 (1H, d, 10.6) | 5.16 (1H, d, 10.6) | 5.15 (1H, d, 10.6) | 5.16 (1H, d, 10.8) |
| H-3 | 3.29 (1H, d, 10.6) | 2.72 (1H, d, 10.6) | 2.72 (1H, d, 10.5) | 2.78 (1H, d, 10.8) |
| H-5 | 5.39 (1H, br s) | 4.97 (1H, br d, 2.4) | 4.99 (1H, br d, 2.4) | 3.80 (1H, br d, 2.2) |
| H-6a | 2.31 (1H, m) | 1.98 (1H, dt, 4.0, 13.1) | 2.01 (1H, m) | 1.91 (1H, dt, 4.3, 13.5) |
| H-6b | 1.77 (1H, m) | 1.67 (1H, dt, 1.4, 13.8) | 1.71 (1H, t, 13.8) | 1.59 (1H, t, 12.9) |
| H-7 | 5.70 (1H, overlap) | 5.26 (1H, dd, 4.3, 12.4) | 5.26 (1H, dd, 4.3, 12.5) | 5.59 (1H, dd, 4.5, 12.3) |
| H-9 | 5.69 (1H, overlap) | 5.15 (1H, d, 5.0) | 5.14 (1H, d, 5.2) | 5.07 (1H, d, 5.0) |
| H-10 | 5.56 (1H, d, 4.9) | 4.90 (1H, d, 5.2) | 5.87 (1H, d, 5.4) | 4.92 (1H, d, 5.0) |
| H-13 | 4.80 (1H, t, 7.0) | 4.37 (1H, t, 7.2) | 5.49 (1H, t, 7.2) | 4.32 (1H, t, 7.3) |
| H-14a | 2.66 (1H, m) | 2.22 (1H, dd, 6.9, 14.7) | 2.34 (1H, dd, 6.7, 14.5) | 2.19 (1H, dd, 6.9, 14.7) |
| H-14b | 2.56 (1H, m) | 2.05 (1H, dd, 7.8, 14.7) | 1.93 (1H, dd, 7.4, 14.7) | 2.04 (1H, dd, 7.8, 14.7) |
| $\mathrm{CH}_{3}-16$ | 1.76 (3H, s) | 1.40 (3H, s) | 1.43 (3H, s) | 1.40 (3H, s) |
| $\mathrm{CH}_{3}-17$ | 1.33 (3H, s) | $1.02(3 \mathrm{H}, \mathrm{s})$ | 0.97 (3H, s) | 1.00 (3H, s) |
| $\mathrm{CH}_{3}-18$ | 1.75 (3H, s) | 1.40 (3H, s) | 1.36 (3H, s) | 1.37 (3H, s) |
| $\mathrm{CH}_{3}-19$ | 1.81 (3H, s) | 1.49 (3H, s) | 1.50 (3H, s) | 1.45 (3H, s) |
| H-20a | 4.83 (1H, d, 8.3) | 4.62 (1H, d, 7.9) | 4.65 (1H, d, 8.0) | 4.56 (1H, d, 7.8) |
| H-20b | 3.75 (1H, d, 7.8) | 3.54 (1H, d, 7.4) | 3.57 (1H, d, 7.8) | 3.41 (1H, d, 7.8) |
| OBz-H-3' | 8.44 (2H, d, 7.9) | 8.06 (2H, dd, 1.4, 8.0) | 8.08 (2H, d, 7.6) | 8.05 (2H, d, 7.4) |
| OBz-H-4 ${ }^{\prime}$ | 7.40 (2H, t, 7.4) | 7.49 (2H, dt, 1.5, 8.0) | 7.54 (2H, t, 7.7) | 7.45 (2H, t, 7.7) |
| OBz-H-5' | 7.48 (1H, t, 7.5) | 7.61 (1H, dt, 1.3, 8.7) | 7.67 (1H, t, 7.7) | 7.58 (1H, t, 7.4) |
| $\mathrm{CH}_{3}-2^{\prime \prime}$ | 1.73 (3H, s) | 1.50 (3H, s) | 1.50 (3H, s) | 1.53 (3H, s) |
| 5-OAc | 1.66 (3H, s) | 1.89 (3H, s) | 1.91 (3H, s) |  |
| 7-OAc | 1.93 (3H, s) | 1.97 (3H, s) | 1.97 (3H, s) | $1.95(3 \mathrm{H}, \mathrm{s})$ |
| 10, 13-OAc |  |  | 2.03 (3H, s) |  |
|  |  |  | 2.06 (3H, s) |  |

${ }^{\text {a }}$ Data were measured in pyridine- $d_{5}$ with reference to the most downfield signal of pyridine- $d_{5}(\delta 8.71 \mathrm{ppm})$.
${ }^{\mathrm{b}}$ Data were measured in $\mathrm{CD}_{3} \mathrm{OD}$ with reference to the center peak of $\mathrm{CD}_{3} \mathrm{OD}(\delta 3.30 \mathrm{ppm})$.

Compound $\mathbf{1}$ is a polyoxygenated taxoid based on the NMR data (Fig. 1). By analysis of the 2D NMR $\left({ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}\right.$ COSY, HMQC, HMBC and ROESY, see Table 3) data, the oxygenated carbons in 1 were elucidated to be $\mathrm{C}-2, \mathrm{C}-4$, C-5, C-7, C-9, C-10, C-13, C-15 and C-20, consistent with normal abeotaxoids. The three ester, one benzoxyl and two acetoxyl groups, were assigned to $\mathrm{C}-9, \mathrm{C}-5$ and $\mathrm{C}-7$, respectively, due to the presence of the HMBC correlations
of H-9 [ $\left.\delta_{\mathrm{H}} 5.15(1 \mathrm{H}, \mathrm{d}, J=5.0 \mathrm{~Hz})\right]$ to a benzoyl carbonyl carbon at $\delta_{\mathrm{C}} 166.8(\mathrm{~s}), \mathrm{H}-5\left[\delta_{\mathrm{H}} 4.97(1 \mathrm{H}\right.$, br d, $J=2.4 \mathrm{~Hz})$ ] to an acetyl carbonyl carbon at $\delta_{\mathrm{C}} 171.3$ (s), and H-7 [ $\delta_{\mathrm{H}} 5.26$ $(1 \mathrm{H}, \mathrm{dd}, J=4.3,12.4 \mathrm{~Hz})]$ to an acetyl carbonyl carbon at $\delta_{\mathrm{C}}$ 172.0 (s). Moreover, both $\mathrm{H}_{2}-20$ [ $\delta_{\mathrm{H}} 4.62(1 \mathrm{H}, \mathrm{d}, J=7.9 \mathrm{~Hz})$ and $3.54(1 \mathrm{H}, \mathrm{d}, J=7.4 \mathrm{~Hz})$ ] and the methyl signal at $\delta_{\mathrm{H}} 1.50$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-2^{\prime \prime}$ ) showed long-range correlations to a quaternary carbon at $\delta_{\mathrm{C}} 119.8$ (s, $\mathrm{C}-1^{\prime \prime}$ ), which strongly



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$10 \mathrm{R}_{1}=\mathrm{CH}_{3}, \mathrm{R}_{2}=\mathrm{n}-\mathrm{C}_{5} \mathrm{H}_{11}$

Figure 1. Structures of compounds from T. yunnanensis.

Table 2. ${ }^{13} \mathrm{C}$ NMR data of $\mathbf{1}$ and $\mathbf{2}(125 \mathrm{MHz}, \delta$ in ppm)

| Carbon | $1^{\text {a }}$ | $1^{\text {b }}$ | $1^{\text {c }}$ | $1(\Delta \delta)^{\text {d }}$ | $2{ }^{\text {b }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| C-1 | 66.8 s | 67.055 s | 67.104 s | 0.049 | 67.1 s |
| C-2 | 72.7 d | 72.935 d | 72.949 d | 0.014 | 72.9 d |
| C-3 | 37.3 d | 37.775 d | 37.788 d | 0.013 | 36.6 d |
| C-4 | 81.9 s | 82.453 s | 82.462 s | 0.009 | 84.3 s |
| C-5 | 73.2 d | 73.869 d | 73.875 d | 0.006 | 71.8 d |
| C-6 | 28.1 t | 28.336 t | 28.353 t | 0.017 | 30.9 t |
| C-7 | 70.1 d | 71.165 d | 71.176 d | 0.011 | 71.5 d |
| C-8 | 44.3 s | 44.838 s | 44.852 s | 0.014 | 45.0 s |
| C-9 | 77.1 d | 77.427 d | 77.453 d | 0.026 | 78.1 d |
| C-10 | 67.8 d | 68.161 d | 68.271 d | 0.110 | 68.2 d |
| C-11 | 134.4 s | 134.912 s | 134.966 s | 0.054 | 135.1 s |
| C-12 | 148.7 s | 148.948 s | 148.946 s | -0.002 | 148.6 s |
| C-13 | 76.2 d | 77.105 d | 77.235 d | 0.130 | 77.4 d |
| C-14 | 42.7 t | 42.226 t | 42.309 t | 0.083 | 42.2 t |
| C-15 | 76.2 s | 77.105 s | 77.235 s | 0.130 | 77.1 s |
| C-16 | 28.0 q | 27.520 q | 27.621 q | 0.101 | 27.6 q |
| C-17 | 28.1 q | 27.581 q | 27.621 q | 0.040 | 27.7 q |
| C-18 | 13.2 q | 12.828 q | 12.851 q | 0.023 | 12.9 q |
| C-19 | 15.4 q | 15.357 q | 15.397 q | 0.040 | 15.7 q |
| C-20 | 70.7 t | 71.396 t | 71.410 t | 0.014 | 71.3 t |
| OBz-C-1 ${ }^{\prime}$ | 165.9 s | 166.840 s | 166.829 s | -0.011 | 167.4 s |
| OBz-C-2 ${ }^{\prime}$ | 130.8 s | 131.059 s | 131.086 s | 0.027 | 131.3 s |
| OBz-C-3 ${ }^{\prime}$ | 130.3 d | 130.960 d | 130.974 d | 0.014 | 131.1 d |
| OBz-C-4 ${ }^{\prime}$ | 128.9 d | 129.697d | 129.705 d | 0.008 | 129.5 d |
| OBz-C-5' | 133.5 d | 134.496 d | 134.496 d | - | 134.2 d |
| C-1" | 119.2 s | 119.847 s | 119.856 s | 0.009 | 119.6 s |
| C-2" | 22.3 q | 22.123 q | 22.147 q | 0.024 | 22.3 q |
| 5-OAc | 169.7 s | 171.336 s | 171.307 s | -0.029 |  |
| 7-OAc | 20.7 q | 20.917 q | 20.931 q | 0.014 |  |
|  | 170.1 s | 171.967 s | 171.942 s | -0.025 | 172.0 s |
|  | 20.9 q | 21.958 q | 21.976 q | 0.018 | 21.0 q |

${ }^{\text {a }}$ Signals were measured in pyridine- $d_{5}$ with reference to the center peak of the most downfield signals of pyridine- $d_{5}$ ( $\delta 149.9 \mathrm{ppm}$ ).
${ }^{\mathrm{b}}$ Signals were measured in $\mathrm{CD}_{3} \mathrm{OD}$ with reference to the center peak of $\mathrm{CD}_{3} \mathrm{OD}(\delta 49.0 \mathrm{ppm})$.
${ }^{\mathrm{c}}$ Signals were measured in $\mathrm{CD}_{3} \mathrm{OH}$ with reference to the furthermost carbon ( $\mathrm{C}-5^{\prime}$ ) signal of the benzoyl group obtained in $\mathrm{CD}_{3} \mathrm{OD}$ ( $\delta$ $134.496 \mathrm{ppm})$.
${ }^{\mathrm{d}} \Delta \delta=\delta\left(\mathrm{CD}_{3} \mathrm{OH}\right)-\delta\left(\mathrm{CD}_{3} \mathrm{OD}\right)$.
suggested the connection of the orthoacetoxyl group to C-20 through an oxygen bridge. However, spectral evidence for the presence of the two other oxygen bridges between the orthoacetoxy group and the taxane skeleton were not observed in the HMBC spectrum, even when solvents and NMR parameters were changed. The lack of a suitable crystal for X-ray diffraction compounded the difficulty in determining the structure of this compound.

Nevertheless, the possible formation of the other two oxygen bridges could only occur between the orthoacetoxy group and any two carbons from among C-2, C-4, C-10, C13 and $\mathrm{C}-15$. Acetylation of $\mathbf{1}$ with pyridine $-\mathrm{Ac}_{2} \mathrm{O}$ (1:1) at room temperature for 24 h yielded a 10,13 -diacetate of $\mathbf{1}$, which excluded the possibility of the formation of two oxygen bridges at $\mathrm{C}-10$ and $\mathrm{C}-13$. Although acetylation on C-2 was not observed, the possibility of C-2 being substituted with a hydroxy group could not be excluded due to the inherent difficulty in acetylating the hindered C-2 hydroxy group of rearranged taxoids. ${ }^{10,16}$ In a comparison with the ${ }^{13} \mathrm{C}$ NMR data of taxuyunnanine $\mathrm{K},{ }^{9}$ where the $\mathrm{C}-2$, $\mathrm{C}-4$ and $\mathrm{C}-15$ carbons were hydroxylated, 1 showed downfield chemical shifts at $\mathrm{C}-2$ and $\mathrm{C}-4$ and minimal change at $\mathrm{C}-15$, which suggested that two oxygen atoms are being bridged through the orthoacetoxy group to C-2 and to C-4, respectively. This conclusion was consistent with the Dreiding model of $\mathbf{1}$. Theoretically, direct evidence for C-2 being connected to the orthoacetoxy group could be obtained from the observation of ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ long-range correlation between H-2 and C-1". Unfortunately, such observation was not observed from repeated HMBC experiments under different parameters.

In order to confirm that an orthoacetoxy group is connected to C-2, the ${ }^{13} \mathrm{C}$ NMR spectra of an equal molar of compound 1 were measured in $\mathrm{CD}_{3} \mathrm{OH}$ and $\mathrm{CD}_{3} \mathrm{OD}$, respectively. Theoretically, a protonated hydroxy being exchanged to a deuterated hydroxy will cause a $\beta$-isotope shift of the ${ }^{13} \mathrm{C}$ NMR signal of the carbon to which the hydroxy group is

Table 3. 2D NMR data of $\mathbf{1}$ and $2\left(\mathrm{CD}_{3} \mathrm{OD}, 500 \mathrm{MHz}\right)$

| Proton | 1 |  | 2 |  |
| :---: | :---: | :---: | :---: | :---: |
|  | ROESY | HMBC | ROESY | HMBC |
| H-2 | H-16, 19, 20a | C-1, 3, 4, 8, 14, 15 | H-3, 16, 19, 20a | C-1, 14, 15 |
| H-3 | H-7 | C-1, 2, 4, 7, 8, 9, 19 | H-2, 7 | C-1, 2, 4, 5, 7, 8, 19, 20 |
| H-5 | H-6a, 6b, 20b | C-3, 4, 6, 20, Oac | H-6a, 6b, 20b | C-3, 4, 5 |
| H-6a | H-5, 7 | C-4, 5, 7, 8 | H-5, 7, 6b | C-4, 5, 7 |
| H-6b | H-5, 20a, 20b | C-7, 8 | H-5, 6a | C-5, 7 |
| H-7 | H-3, 6a, 10 | C-6, 8, 9, 19, Oac | H-3, 6a, 10 | C-6, 8, 9, 19, Oac |
| H-9 | H-19 | C-3, 7, 8, 10, 11, 19, $1^{\prime}$ | H-16, 18, 19 | C-3, 8, 10, 11, $1^{\prime}$ |
| H-10 | H-7, 18 | C-1, 8, 9, 11, 12 | H-7, 18 | C-1, 8, 9, 11, 12 |
| H-13 | H-14a, 17 | C-11, 12 | H-14a, 14b, 17, 18 | C-11, 12 |
| H-14a | H-13, 14b | C-1, 11, 12, 13, 15 | H-13, 14b, 16 | C-1, 11, 12, 13, 15 |
| H-14b | H-14a | C-1, 2, 13, 15 | H-13, 14a | C-1, 2, 13, 15 |
| $\mathrm{CH}_{3}-16$ | H-2 | C-1, 15, 17 | H-2, 9, 14a | C-1, 15, 17 |
| $\mathrm{CH}_{3}-17$ | H-13 | C-1, 15, 16 | H-13 | C-1, 15, 16 |
| $\mathrm{CH}_{3}-18$ | H-10 | C-11, 12, 13 | H-9, 10 | C-11, 12, 13 |
| $\mathrm{CH}_{3}$-19 | H-2, 9, 20a | C-3, 7, 8, 9 | H-2, 9, 20a | C-3, 7, 8, 9 |
| H-20a | H-6b, 19, 20b | $\mathrm{C}-3,5,1^{\prime \prime}$ | H-2, 6b, 19, 20b | C-3, $1^{\prime \prime}$ |
| H-20b | H-6b, 20a | C-3, 4, 5, $1^{\prime \prime}$ | H-5, 6b, 20a | C-3, 4, 5, $1^{\prime \prime}$ |
| OBz-H-3' | H-4' | C-1 ${ }^{\prime}, 2^{\prime}, 4^{\prime}, 5^{\prime}$ | H-4' | $\mathrm{C}-1^{\prime}, 2^{\prime}, 4^{\prime}, 5^{\prime}$ |
| OBz-H-4 ${ }^{\prime}$ | H-3', $5^{\prime}$ | C-2 $2^{\prime}, 3^{\prime}, 5^{\prime}$ | H-3', $5^{\prime}$ | C-2 ${ }^{\prime}, 3^{\prime}, 5^{\prime}$ |
| OBz-H-5' | H-4' | $\mathrm{C}-3^{\prime}, 4^{\prime}$ | H-4' | C-3', $4^{\prime}$ |
| $\mathrm{CH}_{3}-2^{\prime \prime}$ |  | $\mathrm{C}-1^{\prime \prime}$ |  | $\mathrm{C}-1^{\prime \prime}$ |

attached. ${ }^{17}$ In compound $\mathbf{1}$, the $\delta_{\mathrm{C}}$ data (Table 2 ) compared between solvents $\mathrm{CD}_{3} \mathrm{OH}$ and $\mathrm{CD}_{3} \mathrm{OD}$ showed small $\beta$ isotope shifts $\left(\Delta \delta_{\mathrm{CD} 3 \mathrm{OH}-\mathrm{D} 3 \mathrm{OD}}=0.026,0.011\right.$ and 0.006 ppm , respectively) for the esterified C-9, C-7 and C5 , while the hydroxylated $\mathrm{C}-10$ and $\mathrm{C}-13$ gave obvious $\beta$ isotope shifts $\left(\Delta \delta_{\mathrm{CD} 3 \mathrm{OH}-\mathrm{CD} 3 \mathrm{OD}}=0.11\right.$ and 0.13 ppm , respectively). In addition, the small $\beta$-isotope shifts of C 2 and C-4 ( 0.014 and 0.009 ppm , respectively) indicated that both carbons were not oxygenated by free hydroxyl groups. On the other hand, an apparent $\beta$-isotope shift was observed on C-15 (Fig. 2), which strongly supported the presence of a hydroxyl group on this carbon. Thus, the C-1 and C-4 oxygen atoms were determined to be bridged through the orthoacetoxy group.

The stereochemistry of $\mathbf{1}$ was established by 2D ROESY data (Table 3). ROEs between $\mathrm{H}-2 / \mathrm{Me}-16$ and -19 , $\mathrm{H}-20 \mathrm{a} /$ $\mathrm{Me}-19, \mathrm{H}-7 / \mathrm{H}-3, \mathrm{H}-9 / \mathrm{Me}-19, \mathrm{H}-10 / \mathrm{H}-7$ and $\mathrm{Me}-18, \mathrm{H}-13 /$ Me-17 fixed the oxy-groups at C-2, C-4, C-7, C-9, C-10 and $\mathrm{C}-13$ to be $\alpha-, \alpha-, \beta-, \alpha$-, $\beta$ - and $\alpha$-oriented, respectively. The ROEs between $\mathrm{H}-5 / \mathrm{H}-20 \mathrm{~b}$, as well as the small coupling constant of H-5 ( $J=2.4 \mathrm{~Hz}$ ), established an $\alpha$ orientation for the acetoxy group at C-5. Accordingly, $\mathbf{1}$ was deduced to be $5 \alpha, 7 \beta$-diacetoxy- $9 \alpha$-benzoxy- $2 \alpha, 4 \alpha, 20$ -orthoacetoxy-10, $13 \alpha$, 15-trihydroxy-11(15 $\rightarrow 1$ )-abeo-taxa-11-ene, and was given the trivial name of taxuyunnanine W .

Taxuyunnanine X (2) was obtained as a white amorphous solid (Fig. 1). The FABMS exhibited [M+gly] ${ }^{+}$at $m / z 680$ and $[\mathrm{M}]^{+}$at $m / z 588$, corresponding to a molecular formula of $\mathrm{C}_{31} \mathrm{H}_{40} \mathrm{O}_{11}$, which was confirmed by HRFABMS $(\mathrm{m} / \mathrm{z}$ $[\mathrm{M}-\mathrm{H}]^{+}$587.2430, calcd: 587.2492). The close resemblance of the NMR spectra (Tables 1 and 2) to those of 1 suggested 2 to be a related abeotaxoid. Compound 2 also contains a benzoxy and an orthoacetoxy group, but differing from 1, it has only one acetoxy group. The H-5 signal shifted dramatically upfield from $\delta_{\mathrm{H}} 4.97(1 \mathrm{H}, \mathrm{br}$ d, $J=2.4 \mathrm{~Hz})$ in 1 to $\delta_{\mathrm{H}} 3.80(1 \mathrm{H}$, br d, $J=2.2 \mathrm{~Hz})$ in $\mathbf{2}$, suggesting 2 to be deacetylated at C-5. Like 1, the acetyl and benzoyl groups were determined to be at C-7 and C-9, respectively, due to the presence of the HMBC correlations (Table 3) of H-7 to an acetyl carbonyl carbon, and H-9 to a benzoyl carbonyl carbon in 2. Although only the HMBC correlations of $\mathrm{H}_{2}-20$ to the quaternary carbon at $\delta_{\mathrm{C}} 119.8$ (s) were observed, the orthoacetoxy group was assigned to C-2, C-4 and C-20 due to the closely similar NMR data to those of $\mathbf{1}$. Additionally, 2 showed very similar 2D NMR ( ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY, HMQC, HMBC and ROESY, Table 3) data to those of $\mathbf{1}$. Compound $\mathbf{2}$ was therefore elucidated to be $7 \beta$-acetoxy- $9 \alpha$-benzoxy- $2 \alpha, 4 \alpha, 20$-orthoacetoxy- $5 \alpha, 10 \beta$,
$13 \alpha$, 15-tetrahydroxy-11(15 $\rightarrow 1$ )-abeo-taxa-11-ene, and assigned the trivial name of taxuyunnanine X .

An orthoesterified taxane, 4-deacetyl-5-epi-20, $O$-seco-taxol-4, 5, 20-orthoacetate had been previously proposed as an intermediate of taxol in its reaction with Meerwein's reagent, which led to a product (20-acetoxy-4-deacetyl-5-epi-20,O-secotaxol) with an opened oxetane ring. ${ }^{18}$ In the present study, however, the orthoesterified taxoids, taxuyunnanines W and X , were obtained. We speculate that these compounds might be the biosynthetic precursors of the opened D-ring $11(15 \rightarrow 1)$-abeotaxoids. Although orthoesters have been reported to occur in daphnane and tigliane diterpenoids, the current paper is the first report of the existence of orthoesters in taxane diterpenoids.

Taxuyunnanine $\mathrm{Y}(\mathbf{3})$ was isolated as white powders (Fig. 1). It has a molecular formula of $\mathrm{C}_{29} \mathrm{H}_{38} \mathrm{O}_{10}$ based on the negative ion HRFABMS peak at $m / z 545.2329\left([\mathrm{M}-\mathrm{H}]^{-}\right.$, calcd 545.2387) and the NMR data (see Tables 4 and 5). In addition the signals for a benzoate and an acetate, its ${ }^{1} \mathrm{H}$ NMR spectrum contained four methyl singlets ( $\delta 1.40,1.76$, $1.84,2.31$ ) and a methine doublet ( $\delta 3.37, J=11.3 \mathrm{~Hz}$ ). The ${ }^{13} \mathrm{C}$ NMR (DEPT) spectra showed 20 carbons, with a quaternary carbon at $\delta 66.7$ and two olefinic quaternary carbons at $\delta 136.3$ and 146.7, respectively. These data, in conjunction with the highly oxygenated nature of the molecule established 3 to be an 11(15 $\rightarrow 1)$-abeotaxoid substituted by a benzoxy group, an acetoxy and several hydroxy groups. However, complete assignments using 2D NMR (see Table 6) experiments indicated that, unlike normal rearranged taxanes, the large coupling constant between H-9 and $\mathrm{H}-10(J=5.1 \mathrm{~Hz})$ was greatly reduced in $\mathbf{3}$. Moreover, ${ }^{3} J$ long-range correlations between H-20b [ $\delta$ $4.08(1 \mathrm{H}, \mathrm{d}, J=7.9 \mathrm{~Hz})]$ and $\mathrm{C}-2[\delta 77.5(\mathrm{~d})]$, and between $\mathrm{H}-2[\delta 5.99(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=11.3 \mathrm{~Hz})]$ and $\mathrm{C}-20[\delta 75.4(\mathrm{t})]$ were observed in the HMBC spectrum, suggesting the presence of an oxygen bridge between C-2 and C-20, forming a C-2(20) tetrahedrofuran ring moiety in $\mathbf{3}$. The reason for the small coupling constant between $\mathrm{H}-9$ and $\mathrm{H}-10$ in 3 could be explained by the conformational transformation of the Bring, which is very common in $5 / 7 / 6$-skeletal type taxanes. The conformational change of the B-ring caused by the formation of the C-2(20) tetrahedrofuran ring in 3 resulted in a change of the dihedral angle between $\mathrm{H}-9$ and $\mathrm{H}-10$, which in turn led to a change of the coupling constant between $\mathrm{H}-9$ and $\mathrm{H}-10$.

The benzoxy and the acetoxy group were assigned to C-7 and C-9, respectively, based on the presence of HMBC correlations of H-7 [ $\delta 6.29(1 \mathrm{H}, \mathrm{dd}, J=4.5,11.9 \mathrm{~Hz})$ to the


Figure 2. Expansion of the ${ }^{13} \mathrm{C}$ NMR spectra of compound 1 recorded in different solvents: (a) signals were measured in $\mathrm{CD}_{3} \mathrm{OD}$; (b) signals were measured in $\mathrm{CD}_{3} \mathrm{OH}$.

Table 4. ${ }^{1} \mathrm{H}$ NMR data of compounds $\mathbf{3}-\mathbf{5}$ ( $J$ in Hz, $\delta$ in ppm)

| Proton | $3^{\text {a,b }}$ | $4^{\text {c,d,e }}$ | $5^{\text {a,d }}$ | $5^{\text {b,c }}$ |
| :---: | :---: | :---: | :---: | :---: |
| H-2 | 5.99 (d, 11.3) | 6.23 (broadened signal) | 4.23 (d, 9.3) | 4.89 (d, 9.8) |
| H-3 | 3.37 (d, 11.3) | 2.46 (d, 6.4) | 2.39 (d, 9.8) | 2.87 (d, 9.8) |
| H-5 | 4.58 (br s) | 3.95 (br t, 3.6) | 4.83 (covered by $\mathrm{H}_{2} \mathrm{O}$ ) | 5.43 (t, 2.8) |
| H-6a | 2.88 (t, 11.9) | 2.09 (m) | 2.19 (m) | 2.88 (m) |
| H-6b | 2.33 (m) | 1.70 (m) | 1.82 (m) | 2.49 (br dd, 4.4, 14.5) |
| H-7 | 6.29 (dd, 4.5, 11.9) | 5.55 (dd, 4.4, 11.6) | 4.14 (dd, 4.6, 11.4) | 4.75 (dd, 4.6, 11.3) |
| H-9 | 5.91 (d, 5.1) | 5.80 (broadened signal) | 4.48 (d, 10.4) | 5.02 (covered by $\mathrm{H}_{2} \mathrm{O}$ ) |
| H-10 | 5.63 (d, 5.1) | 4.72 (d, 8.3) | 6.04 (d, 10.4) | 6.67 (d, 10.3) |
| H-13 | 4.96 (t, 6.9) | 4.55 (t, 6.7) | 5.73 (dt, 3.2, 8.9) | 6.14 (dd, 3.5, 9.4) |
| H-14a | 2.76 (dd, 6.6, 14.0) | 2.30 (m) | 2.38 (m) | 2.93 (10.5, 15.9) |
| H-14b | 2.55 (dd, 7.6, 14.0) | 1.74 (m) | 1.99 (m) | 2.53 (dd, 4.7, 15.6) |
| $\mathrm{CH}_{3}-16$ | 1.40 (s) | 1.26 (s) | 1.07 (s) | 1.32 (s) |
| $\mathrm{CH}_{3}-17$ | 1.84 (s) | 1.02 (s) | 1.40 (s) | 1.71 (s) |
| $\mathrm{CH}_{3}-18$ | 1.76 (s) | 1.61 (br s) | 2.02 (s) | 2.28 (s) |
| $\mathrm{CH}_{3}$-19 | 2.31 (s) | 1.59 (s) | 1.44 (s) | 2.06 (s) |
| H-20a | 4.60 (d, 7.9) | 4.54 (d, 9.6) | 3.43 (s, $\left.\mathrm{H}_{2}-20\right)$ | 3.91 (d, 8.5) |
| H-20b | 4.08 (d, 7.9) | 4.05 (d, 10.4) |  | 3.85 (d, 8.5) |
| OBz-H-3 ${ }^{\prime}$ | 8.57 (d, 8.2) | 7.98 (d, 7.5) |  |  |
| OBz-H-4 ${ }^{\prime}$ | 7.23 (t, 7.7) | 7.57 (t, 7.4) |  |  |
| OBz-H-5 ${ }^{\prime}$ | 7.34 (t, 7.8) | 7.44 (d, 7.6) |  |  |
| $1-\mathrm{OH}$ |  |  |  | 6.57 (s) |
| 4-OH |  |  |  | 7.27 (s) |
| $7-\mathrm{OH}$ |  |  |  | 7.07 (s) |
| 9-OH |  |  |  | 8.64 (d, 6.1) |
| OAc | 1.99 (s) | $2.04(3 \mathrm{H}, \mathrm{s}), 1.71(3 \mathrm{H}, \mathrm{s})$ | $2.11(3 \mathrm{H}, \mathrm{s}), 2.06(3 \mathrm{H}, \mathrm{s}), 2.05(3 \mathrm{H}, \mathrm{s})$ | 2.23 (s), 2.07 (s), 2.04 (s) |

[^1]Table 5. ${ }^{13} \mathrm{C}$ NMR data of compounds $\mathbf{3 - 5}$ ( $\delta$ in ppm)

| Carbon | $3^{\text {a,b }}$ | $4^{\text {c,d }}$ | $5^{\text {a,d }}$ |
| :---: | :---: | :---: | :---: |
| C-1 | 66.7 s | 67.0 s | 80.4 s |
| C-2 | 77.5 d | 71.9 d | 84.0 d |
| C-3 | 44.0 d | 44.3 d | 48.6 d |
| C-4 | 81.6 s | 76.9 s | 82.2 s |
| C-5 | 68.9 d | 70.4 d | 71.4 d |
| C-6 | 31.4 t | 29.5 t | 33.5 t |
| C-7 | 72.8 d | 71.7 d | 73.3 d |
| C-8 | 44.5 s | 46.5 s | 46.2 s |
| C-9 | 78.7 d | 82.0 d | 80.0 d |
| C-10 | 69.1 d | 68.7 d | 76.7 d |
| C-11 | 136.3 s | 139.0 s | 137.4 s |
| C-12 | 146.7 s | 145.8 s | 137.6 s |
| C-13 | 77.5 d | 79.0 d | 72.0 d |
| C-14 | 41.7 t | 41.0 t | 37.6 t |
| C-15 | 76.2 s | 76.5 s | 43.3 s |
| C-16 | 28.1 q | 27.0 q | 27.9 q |
| C-17 | 28.0 q | 28.0 q | 23.1 q |
| C-18 | 13.5 q | 11.9 q | 15.2 q |
| C-19 | 16.4 q | 16.2 q | 16.0 q |
| C-20 | 75.4 t | 52.4 t | 74.9 t |
| OBz-C-1 ${ }^{\prime}$ | 166.3 s | 168.0 s |  |
| OBz-C-2 ${ }^{\prime}$ | 131.4 s | 129.8 s |  |
| OBz-C-3 ${ }^{\prime}$ | 130.6 d | 130.5 d |  |
| OBz-C-4 ${ }^{\prime}$ | 128.7 d | 129.4 d |  |
| OBz-C-5 ${ }^{\prime}$ | 132.9 d | 134.0 d |  |
| $\mathrm{OAc}-\mathrm{C}=\mathrm{O}$ | 170.3 s | 172.4 s | 172.1 s |
|  | 7-Oac | 171.7 s | 171.6 s |
|  |  |  | 172.1 s |
| $\mathrm{OAc}-\mathrm{CH}_{3}$ | 21.2 q | 22.3 q | 21.8 q |
|  |  | 21.8 q | 21.5 q |
|  |  |  | 21.1 q |

[^2]acetyl carbonyl carbon at $\delta 170.3$, and $\mathrm{H}-9[\delta 5.91(1 \mathrm{H}, \mathrm{d}$, $J=5.1 \mathrm{~Hz})$ ] to the benzoyl carbonyl carbon at $\delta 166.3$.

The stereochemistry of $\mathbf{3}$ was determined by application of 2D ROESY experiment (Table 6). The ROESY cross-peaks of $\mathrm{H}-2 / \mathrm{CH}_{3}-19, \mathrm{H}-7 / \mathrm{H}-3, \mathrm{H}-9 / \mathrm{CH}_{3}-19, \mathrm{H}-10 / \mathrm{CH}_{3}-18$ and $\mathrm{H}-$ $13 / \mathrm{CH}_{3} / 16$ established $\mathrm{H}-2, \mathrm{H}-7, \mathrm{H}-9, \mathrm{H}-10$ and $\mathrm{H}-13$ as in the $\beta$-, $\alpha-, \beta-, \alpha$ - and $\beta$-configuration, respectively. The $\beta$ oriented $4-\mathrm{CH}_{2} \mathrm{O}$ - was deduced from the ROESY correlation of $\mathrm{H}-20 \mathrm{a} / \mathrm{CH}_{3}-19$, whereas, the $5-\mathrm{OH}$ was assigned as $\alpha$-oriented due to the small coupling constant of $\mathrm{H}-5$ [ $\delta 4.58$ ( $1 \mathrm{H}, \mathrm{br} \mathrm{s}$ )].

Taxuyunnanine Y is the second naturally occuring rearranged taxane with a C -2(20) tetrahedrofuran ring moiety. Taxuyunnanine E, the first of this type of taxanes was isolated previously by us from the same species. ${ }^{5}$

Taxuyunnanine Z (5) was obtained as a white amorphous solid (Fig. 1) with a molecular formula of $\mathrm{C}_{26} \mathrm{H}_{38} \mathrm{O}_{11}$ based on negative FAB mass spectrometry and ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra (Tables 4 and 5), which was confirmed by HRFABMS (Found: $m / z 525.2272[\mathrm{M}-\mathrm{H}]^{-}$, calcd 525.2336). Its ${ }^{1} \mathrm{H}$ NMR spectrum showed a typical oneproton doublet due to $\mathrm{H}-3\left[\delta_{\mathrm{H}} 2.39(1 \mathrm{H}, \mathrm{d}, J=9.8 \mathrm{~Hz})\right.$, $\mathrm{CD}_{3} \mathrm{OD}$ as solvent] and four methyl singlets owing to $\mathrm{CH}_{3}-$ $16,17,18$ and 19 [ $\delta_{\mathrm{H}} 1.07,1.40,2.02$ and 1.44 , each in 3 H ], along with three acetyl methyl groups at $\delta_{\mathrm{H}} 2.05(3 \mathrm{H}, \mathrm{s})$, $2.06(3 \mathrm{H}, \mathrm{s})$ and $2.11(3 \mathrm{H}, \mathrm{s})$. These data, along with the ${ }^{13} \mathrm{C}$ NMR (including DEPT) spectra suggested that 5 was a baccatin IV derivative. However, in the HMBC spectrum of 5 (Table 6), the notable ${ }^{3} J$ correlations of $\mathrm{H}-2$ to C-20 and $\mathrm{H}_{2}-20$ to $\mathrm{C}-2$ showed that it contained a $\mathrm{C}-2(20)$

Table 6. 2D ROESY and HMBC data of compounds 3 and 5

| Proton | $3^{\text {a }}$ |  | $5^{\text {b }}$ |  |
| :---: | :---: | :---: | :---: | :---: |
|  | ROESY | HMBC | ROESY | HMBC |
| H-2 | H-19 | C-8, 14, 20 | H-19 | C-1, 3, 8, 14, 20 |
| H-3 | H-7, 14b | C-1, 2, 3, 4, 7, 8, 19 | H-7 | C-2, 8, 19 |
| H-5 | H-6 | C-7 | H-6a, 6b, 20 | $\mathrm{C}-4,6, \mathrm{OAc}^{\text {c }}$ |
| H-6a | H-5, 6b | C-7 | H-6b, |  |
| H-6b | H-5, 6a | C-7 | H-6a | C-4, 5, 7, 8 |
| H-7 | H-3, 6b | C-3, 6, 8, 9, 19, OAc ${ }^{\text {c }}$ | H-3 | C-7, 19 |
| H-9 | H-10, 19 | C-3, 8, 10, 11, 19, $1^{\prime}$ | H-19 | C-7, 8, 10, 19 |
| H-10 | H-9, 18 | C-1, 8, 9, 11, 12 | H-7, 18 | C-9, 11, 12, 15, OAc ${ }^{\text {c }}$ |
| H-13 | H-16, 14a | C- 11, 12 | H-14a, 16 | C-11, 12, OAc* |
| H-14a | H-13, 14b, 17 | C-1, 11, 12, 15 | H-13, 14b | C-1, 2, 12, 13 |
| H-14b | H-3, 14a | C-1, 15 | $\mathrm{H}-14 \mathrm{a}$ | C-1, 2, 13, 15 |
| $\mathrm{CH}_{3}$-16 | H-13 | C-1, 15, 17 | H-13 | C-1, 11, 15, 17 |
| $\mathrm{CH}_{3}-17$ | $\mathrm{H}-14 \mathrm{a}$ | C-1, 15, 16 | H-2, 9 | C-1, 11, 15, 16 |
| $\mathrm{CH}_{3}-18$ | H-10 | C-11, 12, 13 | H-10 | C-11, 12, 13 |
| $\mathrm{CH}_{3}$-19 | H-2, 9 | C-3, 7, 8, 9 | H-2, 9 | C-3, 7, 8, 9 |
| H-20a | H-3, 19, 20b | C-3, 4 | H-3, 19 | C-2, 3, 4 |
| H-20b | H-20a | C-2, 3, 4 |  |  |
| OBz-H-3' | H-4' | C- $1^{\prime}, 2^{\prime}, 3^{\prime}, 5^{\prime}$ |  |  |
| OBz-H-4' | H-3', $5^{\prime}$ | C-2', $3^{\prime}, 4^{\prime}$ |  |  |
| OBz-H-5 | H-4' | $\mathrm{C}-3^{\prime}$ |  |  |

${ }^{\text {a }}$ Recorded in pyridine- $d_{5}$.
${ }^{\mathrm{b}}$ Recorded in $\mathrm{CD}_{3} \mathrm{OD}$.
${ }^{\text {c }}$ The carbonyl carbon of acetoxy group.
tetrahedrofuran ring moiety as was the case in $\mathbf{3}$, instead of a normal C-5(20) oxetane ring as existed in baccatin IV and its derivatives. Despite the intense water peak that prevented the $\mathrm{H}-5$ signal from being seen the ${ }^{1} \mathrm{H}$ NMR spectrum of 5 conducted in $\mathrm{CD}_{3} \mathrm{OD}$, the HMBC cross peak between the concealed H-5 signal and one of the acetyl carbonyl carbons was still observed. Using pyridine- $d_{5}$ as the NMR solvent, the expected correlation for the $\mathrm{H}-5$ signal was clearly evident. The other two acetoxy groups were assigned to C10 and C-13, respectively, based on the presence of the HMBC correlations of $\mathrm{H}-10$ [ $\left.\delta_{\mathrm{H}} 6.04(1 \mathrm{H}, \mathrm{d}, J=10.4 \mathrm{~Hz})\right]$ and $\mathrm{H}-13\left[\delta_{\mathrm{H}} 5.73(1 \mathrm{H}, \mathrm{dt}, J=3.2,8.9 \mathrm{~Hz})\right]$ to two other acetyl carbonyl carbons. Since all these acetyl groups and the hydroxyls at $\mathrm{C}-1,-7$, and -9 have been unequivocally assigned, the remaining hydroxyl must be at $\mathrm{C}-4$ in $\mathbf{5}$. This was confirmed by the significant upfield chemical shift at $\delta_{\mathrm{C}}$ 82.2 (s). ${ }^{5}$

Although the relative stereochemistry of $\mathbf{5}$ could be deduced from analysis of the proton spin system, the more definitive assignments were obtained by a 2D ROESY experiment (Table 6). The presence of the NOEs of $\mathrm{H}-2 / \mathrm{H}-19, \mathrm{H}-7 / \mathrm{H}-3$, $\mathrm{H}-9 / \mathrm{H}-19, \mathrm{H}-10 / \mathrm{H}-7, \mathrm{H}-13 / \mathrm{H}-16$, and $\mathrm{H}-20 / \mathrm{H}-19$ established the orientation of $\mathrm{H}-2, \mathrm{H}-7, \mathrm{H}-9, \mathrm{H}-10, \mathrm{H}-13$ and $4-$ $\mathrm{CH}_{2} \mathrm{O}-$ as $\beta$-, $\alpha-, \beta-, \alpha-, \beta$ - and $\beta$-, respectively. The acetoxy group of $\mathrm{C}-5$ in 5 was $\alpha$-oriented according to the ROE correlations between $\mathrm{H}-5$ and $\mathrm{H}_{2}-20$, in accordance with the small coupling constant of $\mathrm{H}-5$.

10-Deacetyl-10-( $\beta$-glucopyranosyl)-yunnanxane (8) was a white amorphous solid (Fig. 1). Its molecular formula was determined as $\mathrm{C}_{35} \mathrm{H}_{54} \mathrm{O}_{13}$ by HRFABMS (Found: $\mathrm{m} / \mathrm{z}$ $681.3466[\mathrm{M}-\mathrm{H}]^{-}$, calcd 681.3486). In the ${ }^{1} \mathrm{H}$ NMR spectra, the presence of four characteristic taxane-type methyls at $\delta_{\mathrm{H}} 1.22(3 \mathrm{H}, \mathrm{s}), 1.71(3 \mathrm{H}, \mathrm{s}), 2.02(3 \mathrm{H}, \mathrm{br} \mathrm{d}$, $J=1.6 \mathrm{~Hz})$ and $0.86(3 \mathrm{H}, \mathrm{s})$ at $\mathrm{CH}_{3}-16,17,18$ and 19 , and a pair of broad singlets at $\delta_{\mathrm{H}} 5.27$ and 4.85 attributed to $\mathrm{H}_{2^{-}}$

20, suggested the compound to be a taxane diterpenoid with a C-4(20) double bond. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra also revealed that $\mathbf{8}$ had two acetoxy groups, a glucose moiety and an $\alpha$-methyl- $\beta$-hydroxyl-butanoyl residue. The $\alpha$ -methyl- $\beta$-hydroxyl-butanoyl residue was observed as a base peak at $\mathrm{m} / \mathrm{z} 117\left[\mathrm{CH}_{3} \mathrm{CH}(\mathrm{OH}) \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{COO}\right]^{-}$in the FABMS. Further study of the NMR spectra revealed $\mathbf{8}$ as a taiwanxan-type taxoid. For comparison purpose, a very similar structure compound, yunnanxane (7), ${ }^{19}$ was used to aid in the structural elucidation of $\mathbf{8}$. The NMR spectra of the two compounds were shown to be very similar, with the aglycone moiety of $\mathbf{8}$ differing from $\mathbf{7}$ only by the absence of an acetoxy group. The HMBC correlations between H-10 and the anomeric $\mathrm{C}-1^{\prime \prime}$, and between the glucose $\mathrm{H}-1^{\prime \prime}$ and C-10 revealed that the 10 -acetyl group in 7 was replaced by a glucosyl moiety in $\mathbf{8}$. The stereochemistry of $\mathbf{8}$ was confirmed by analysis of the ROESY spectral data and the coupling pattern of its ${ }^{1} \mathrm{H}$ NMR spectrum. Hence, $\mathbf{8}$ was deduced as 10 -deacetyl-10-( $\beta$-glucopyranosyl)-yunnanxane.

This majority of natural taxoid glycosides have been found as xylosides, with glycosylation exclusively at C-7. To date, only one taxoid glucoside, a 5/7/6-skeletal taxane containing a glucosyl moiety at C-5, had been reported in the literature. ${ }^{20}$ Compound $\mathbf{8}$ is the second natural taxane glucoside, but the first in a 6/8/6 taxane skeleton.

7-( $\beta$-Xylosyl)- $N$-methyl-taxol C (10) was obtained as white needles (Fig. 1). Negative FABMS showed $[\mathrm{M}-\mathrm{H}]^{-}$ion peak at $m / z$ 994, in accordance with a molecular formula of $\mathrm{C}_{52} \mathrm{H}_{69} \mathrm{NO}_{18}$, which was verified by HRFABMS (Found: $\mathrm{m} /$ $z 995.4576[\mathrm{M}-\mathrm{H}]^{-}$, calcd 995.4515). Analysis of the NMR spectra indicated that $\mathbf{1 0}$ is a derivative of $\mathbf{9}$, a paclitaxel-type taxane xyloside previously reported from the same plant, ${ }^{10}$ with their difference occurring on the C-13 side chain. In the highfield region of the NMR spectra, the
presence of the signals for a methyl $\left[\delta_{\mathrm{H}} 3.13(3 \mathrm{H}, \mathrm{s}), \delta_{\mathrm{C}} 33.8\right.$ $(\mathrm{q})$ ] and two methylene [ $\delta_{\mathrm{C}} 22.8(\mathrm{t})$ and $31.7(\mathrm{t})$ ] groups in $\mathbf{1 0}$ in comparison to 9 , suggested that $\mathbf{1 0}$ was substituted by an $N$-methyl- N -hexanoylphenylisoserine side chain as found in $N$-methyltaxol $\mathrm{C}^{21}$ instead of an N -butanoylphenylisoserine side chain in 9 . The FABMS base peak at $\mathrm{m} / \mathrm{z}$ 292, which correspond to a fragment ion of $\left[\mathrm{C}_{5} \mathrm{H}_{11}\right.$ $\left.\mathrm{CON}\left(\mathrm{CH}_{3}\right) \mathrm{CH}\left(\mathrm{C}_{6} \mathrm{H}_{5}\right) \mathrm{CH}(\mathrm{OH}) \mathrm{COO}\right]^{-}$strongly supported this deduction.

The pentyl group at $\mathrm{C}-5^{\prime}$ was established by ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY experiment, and the N -methylation was confirmed by HMBC correlations of $\mathrm{H}-3^{\prime}$ to $\mathrm{C}-\left(\mathrm{N}-\mathrm{CH}_{3}\right)$ and $\mathrm{H}-(\mathrm{N}-$ $\mathrm{CH}_{3}$ ) to $\mathrm{C}-3^{\prime}$ and $\mathrm{C}-5^{\prime}$. In addition, HMBC correlations of $\mathrm{H}-$ 10 to a carbonyl carbon at $\delta_{\mathrm{C}} 170.0$ (s), and $\mathrm{H}-7$ to the anomeric $\mathrm{C}-1^{\prime \prime \prime \prime}$ and the xylosyl $\mathrm{H}-1^{\prime \prime \prime \prime}$ to $\mathrm{C}-7$, confirmed the respective assignment of an acetoxyl and a xylosyl group to $\mathrm{C}-10$ and C-7. Compound $\mathbf{1 0}$ was thus elucidated as $7-(\beta-$ xylosyl)- $N$-methyl-taxol C. Although the paclitaxel derivatives with a C-7 xylosyl group are very common in nature, the $N$-methylated paclitaxel-equivalent compounds are rare, with only two examples being reported to date, ${ }^{21,22}$ and the co-occurrence of xylosidation and N -methylation in the same molecule has not been reported previously.

The known compounds taxuchin B (4) (Fig. 1) and yunnanxane (7) (Fig. 1) were identified by comparison of their spectroscopic data ( ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$ NMR, MS) with those reported in literatures. ${ }^{12,13}$ Taxuchin B, originally isolated from Taxus chinensis by Wei-Shuo Fang et al., ${ }^{12}$ is the only natural chlorine-containing taxane reported to date, and its existence in T. yunnanensis is being recorded for the first time.

## 3. Experimental

### 3.1. General

1D and 2D NMR experiments were performed either on a Bruker AM-400 or DRX-500 spectrometer. Unless otherwise specified, chemical shifts ( $\delta$ ) were expressed in ppm with reference to the solvent signals. FABMS and HRFABMS were taken on a VG Auto Spec-3000 or a Finnigan MAT 90 instrument. IR spectra were recorded on a Bio-Rad FTS-135 spectrometer with KBr pellets. UV spectral data were obtained on a UV 210A spectrometer. Optical rotations were carried out on a HORIBA SEPA-300 High Sensitive Polarimeter or Perkin-Elmer model 241 Polarimeter. Column chromatogrphy was performed either on silica gel (200-300 mesh, Qingdao Marine Chemical Inc., China), silica gel H (10-40 $\mu \mathrm{m}$, Qingdao Marine Chemical Inc., China), Lichroprep $\mathrm{RP}_{18}$ gel ( $40-63 \mu \mathrm{~m}$, Merck, Darmstadt, Germany), or on MCI gel (70-150 $\mu \mathrm{m}$, Mitsubishi Chemical Corporation, Tokyo, Japan). Fractions were monitored by TLC on silica gel and spots were visualized by heating plates sprayed with $10 \% \mathrm{H}_{2} \mathrm{SO}_{4}$ in EtOH . All solvents were distilled prior to use.

### 3.2. Plant material

The bark of T. yunnanensis, Cheng et L. K. Fu (Taxaceae) was collected in Lijiang Prefecture of Yunnan Province of

People's Republic of China. A voucher specimen (No. YAF-97-18) has been deposited at the Yunnan Academy of Forestry, Kunming, Yunnan, People's Republic of China.

### 3.3. Extraction and isolation

The barks of T. yunnanensis, Cheng et L. K. Fu (Taxaceae, collected in Lijiang Prefecture of Yunnan Province of People's Republic of China) were dried, milled, extracted (EtOH) and fractionated on Si gel column chromatography ( $2.0 \mathrm{~kg}, 200-300 \mathrm{mesh}$ )) by gradient elution with $\mathrm{CHCl}_{3}$ by increasing concentration of $\mathrm{Me}_{2} \mathrm{CO}\left[\mathrm{CHCl}_{3}-\mathrm{Me}_{2} \mathrm{CO} / 100: 0\right.$ (eluate F1, 25 1), 90:10 (eluates F2-F62, 600 ml ), 80:20 (eluates F63-F110, 600 ml ), 70:30 (eluates F111-F144, 600 ml ), 0:100 (eluates F145-F172, 600 ml ), respectively] to afford 172 fractions. Fraction $2(7.5 \mathrm{~g})$ was subjected to four Si gel (200-300 mesh) chromatographic columns ( 200 g Si gel each), eluting with $\mathrm{CHCl}_{3}-\mathrm{Me}_{2} \mathrm{CO}$ (7:1), petroleum ether-EtOAc (5:2), petroleum ether- $\mathrm{Me}_{2} \mathrm{CO}$ (4:1) and cyclohexane-EtOAc (2:1), respectively, to yield compound 7 ( 207 mg ). The pooled fractions F111-F118 $(7.2 \mathrm{~g})$ were rechromatographed on silica gel $(150 \mathrm{~g}, 200-$ 300 mesh), eluting with $\mathrm{CHCl}_{3}-i-\mathrm{PrOH}$ (9:1) to afford 12 fractions ( 200 ml each). Of the 12 fractions, the 5 th was subjected to column chromatography over $\mathrm{RP}_{18}$ silica gel ( $80 \mathrm{~g}, 40-63 \mu \mathrm{~m}$ ), eluting with acetonitrile $-\mathrm{H}_{2} \mathrm{O}$ (3:7), followed by preparative TLC using $\mathrm{CHCl}_{3}-i-\mathrm{PrOH}(9: 1)$ as the developing system to yield compound $\mathbf{1}(25 \mathrm{mg})$; and the 9th was chromatographed over $\mathrm{RP}_{18}$ silica gel ( $80 \mathrm{~g}, 40-$ $63 \mu \mathrm{~m}$ ) eluting with acetonitrile $-\mathrm{H}_{2} \mathrm{O}(2: 8)$, then over Si gel ( $100 \mathrm{~g}, 200-300 \mathrm{mesh}$ ) eluting with $\mathrm{CHCl}_{3}-i-\mathrm{PrOH}$ (7:1) to afford $2(3 \mathrm{mg})$. The pooled fractions F119-F143 $(27.5 \mathrm{~g})$ were chromatographed on $\operatorname{Si}$ gel ( $500 \mathrm{~g}, 200-300$ mesh) eluting with $\mathrm{CHCl}_{3}-\mathrm{CH}_{3} \mathrm{OH}(9: 1)$, on $\mathrm{RP}_{18}$ gel ( 80 g , $40-63 \mu \mathrm{~m}$ ) eluting with aqueous $\mathrm{CH}_{3} \mathrm{OH}(65 \%)$ and on Si gel ( $200 \mathrm{~g}, 200-300$ mesh) with $\mathrm{CHCl}_{3}-i-\mathrm{PrOH}(6: 1)$ to afford compounds $\mathbf{6}(4 \mathrm{mg})$ and $\mathbf{1 0}(11 \mathrm{mg})$. The $\mathrm{Me}_{2} \mathrm{CO}$ fractions (F145-F172) were combined, filtered and evaporated to dryness to afford 38 g of a residue, which was subjected to further Si gel ( $500 \mathrm{~g}, 200-300$ mesh) column chromatography separation using $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ as the eluents (gradient ratios:19:1, 15:1, 9:1, 7:1, 4:1, 1:1) in ascending order of polarity to provide 70 fractions ( 500 ml each). The 41st to 58th fractions of the 70 fractions were pooled ( 16.5 g ) and further separated using CC sequentially on Si gel ( $300 \mathrm{~g}, 200-300$ mesh) with $\mathrm{CHCl}_{3}-\mathrm{CH}_{3} \mathrm{OH}$ (7:1), on MCI gel ( 100 g ) with a gradient of $50 \%$ aqueous $\mathrm{CH}_{3} \mathrm{OH}$ to $100 \% \mathrm{CH}_{3} \mathrm{OH}$, on $\mathrm{RP}_{18}$ gel ( $80 \mathrm{~g}, 40-63 \mu \mathrm{~m}$ ) with aqueous acetonitrile ( $30 \% \mathrm{CH}_{3} \mathrm{CN} \rightarrow 70 \% \mathrm{CH}_{3} \mathrm{CN}$ ), on Si gel ( $100 \mathrm{~g}, 200-300$ mesh) with $\mathrm{CHCl}_{3}-i-\mathrm{PrOH}$ (4:1) to provide compounds $\mathbf{3}(11 \mathrm{mg}), 4(3 \mathrm{mg}), 5(8 \mathrm{mg})$ and a compound $\mathbf{8}$ containing fraction ( 110 mg ). The last fraction was subjected to preparative TLC separation using $\mathrm{CHCl}_{3}-$ $i-\mathrm{PrOH}$ (3:1) as developing solvents to afford compound $\mathbf{8}$ $(18 \mathrm{mg})$.
3.3.1. Taxuyunnanine $\mathbf{W}$ (1). White amorphous powder, $[\alpha]_{\mathrm{D}}^{14.9}=-10.71^{\circ}(c \quad 0.35, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon)=273$ (3.5), 253 (3.6), 231 (4.4), 202 (4.6) nm; IR (KBr) $\nu_{\max }=3432,2927,2862,1743,1723,1448,1404$, 1372, 1328, 1274, 1174, 1147, 1116, 1049, 1002, 976, 945, 923, 888, 851, 810, 765, $715 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR data, see Table $1 ;{ }^{13} \mathrm{C}$ NMR data, see Table 2; negative FABMS (glycerol)
$m / z(\%)=[\mathrm{M}+\mathrm{gly}-\mathrm{H}]^{-} 721(3),[\mathrm{M}-\mathrm{H}]^{-} 629$ (12), 505 (3), 339 (5), 175 (2), 121 (100), 60 (21); HRFABMS m/z $[\mathrm{M}-\mathrm{H}]^{-} 629.2546$ (calcd for $\mathrm{C}_{33} \mathrm{H}_{41} \mathrm{O}_{12}$ 629.2598).

Acetylation of 1: The sample of $\mathbf{1}(5 \mathrm{mg})$ was acetylated in 1.0 ml pyridine $-\mathrm{Ac}_{2} \mathrm{O}(1: 1)$ at room temperature for 24 h to afford 1a $(5 \mathrm{mg})$. Further acetylation of $\mathbf{1 a}$ in 1.0 ml pyridine $-\mathrm{Ac}_{2} \mathrm{O}$ (1:1) at room temperature for 20 days to provide the unchanged product. 1a: IR $(\mathrm{KBr}) \nu_{\max }=3541$, 2922, 2857, 1740, 1466, 1404, 1372, 1269, 1239, 1175, $1098,1069,1047,925,890,851,805,713,604,558 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR data, see Table 1; negative FABMS (glycerol) $\mathrm{m} / \mathrm{z}$ $(\%)=[\mathrm{M}+\mathrm{gly}]^{-} 806(2),[\mathrm{M}]^{-} 714$ (3), 671 (6), 629 (1), 490 (1), 431 (2), 339 (3), 311 (13), 277 (9), 244 (6), 184 (72), 152 (17), 121 (100), 92 (44), 59 (90).
3.3.2. Taxuyunnanine $\mathbf{X}$ (2). White amorphous powder, $[\alpha]_{\mathrm{D}}^{15.8}=-8.33^{\circ}$ (c $\left.0.15, \mathrm{MeOH}\right)$; UV (MeOH) $\lambda_{\text {max }}$ $(\log \varepsilon)=2035$ (4.6), 229 (4.3) nm; IR (KBr) $\nu_{\max }=3632$, 3614, 3446, 3415, 3120, 3108, 3064, 2981, 2930, 2860, 1717, 1609, 1447, 1398, 1373, 1278, 1174, 1148, 1115, $1070,1047,997,979,938,923,849,763,715,687 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR data, see Table 1; ${ }^{13} \mathrm{C}$ NMR data, see Table 2; negative FABMS (glycerol) $m / z(\%)=[\mathrm{M}+\mathrm{gly}]^{-} 680(15)$, $[\mathrm{M}]^{-} 588$ (31), 462 (6), 369 (10), 339 (10), 325 (12), 311 (8), 277 (32), 184 (100), 152 (11), 121 (98), 92 (48); HRFABMS $\mathrm{m} / \mathrm{z}[\mathrm{M}-\mathrm{H}]^{-} 587.2430$ (calcd for $\mathrm{C}_{31} \mathrm{H}_{39} \mathrm{O}_{11}$ 587.2492).
3.3.3. Taxuyunnanine $\mathbf{Y}$ (3). White powders, $[\alpha]_{\mathrm{D}}^{14.3}=+20.00^{\circ}\left(c \quad 0.60, \mathrm{CH}_{3} \mathrm{OH}\right) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon)=274.5$ (3.4), 231 (4.3), 202 (4.4) nm; IR $\nu_{\max }$ $(\mathrm{KBr})=3444,2982,2949,1730,1652,1447,1371,1328$, 1278, 1177, 1118, 1070, 1026, 1007, 983, 948, 883, 843, 810, 759, 716, 692, $634 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR data, see Table 4, ${ }^{13}$ C NMR data, see Table 5; negative FABMS $m / z(\%)=545$ $[\mathrm{M}-\mathrm{H}]^{-}$(13), 529 (6), 509 (3), 479 (3), 431 (5), 381 (5), 353 (6), 339 (82), 325 (100), 311 (82), 297 (32), 283 (12), 255 (26), 235 (10), 197 (5), 173 (6), 147 (8), 121 (64), 97 (11), 80 (22); HRFABMS $m / z 545.2329[\mathrm{M}-\mathrm{H}]^{-}$, calcd 545.2387.
3.3.4. Taxuchin B (4). White needles, ${ }^{1} \mathrm{H}$ NMR data, see Table 4, ${ }^{13} \mathrm{C}$ NMR data, see Table 5; negative FABMS $\mathrm{m} / \mathrm{z}$ $(\%)=623[\mathrm{M}-\mathrm{H}]^{-}(78), 581$ (12), 502 (8), 384 (6), 339 (7), 325 (10), 219 (3), 121 (100), 77 (8); HRFABMS m/z $623.2178[\mathrm{M}-\mathrm{H}]^{-}$, calcd for $\mathrm{C}_{31} \mathrm{H}_{40} \mathrm{O}_{11} \mathrm{Cl} 623.2259$.
3.3.5. Taxuyunnanine $\mathbf{Z}$ (5). White powders; $[\alpha]_{\mathrm{D}}^{14.8}=+89.37^{\circ}\left(c \quad 0.40, \mathrm{CH}_{3} \mathrm{OH}\right) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon): \mathrm{nm} ; ~ \mathrm{IR} \nu_{\max }(\mathrm{KBr})=3460,2944,1734,1656$, 1432, 1374, 1244, 1041, 1019, 954, 881, 739, 709, 688, $612,525 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR data, see Table $4,{ }^{13} \mathrm{C}$ NMR data, see Table 5; negative FABMS $m / z(\%)=525[\mathrm{M}-\mathrm{H}]^{-}(21)$, 483 (8), 465 (6), 423 (7), 406 (3), 339 (11), 311 (20), 207 (10), 152 (10), 99 (19), 59 (100); HRFABMS $m / z 525.2272$ $[\mathrm{M}-\mathrm{H}]^{-}$, calcd 525.2336.
3.3.6. 7-( $\boldsymbol{\beta}$-Xylosyl)-baccatin III (6). White powders. UV $(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon)=202(4.3), 230.5$ (4.2), 274 (3.0) nm; IR $\nu_{\text {max }}(\mathrm{KBr})=3486,2940,1719,1453,1374,1316,1247$, 1178, 1097, 1071, 1048, 1025, 979, 909, 778, $710 \mathrm{~cm}^{-1}$; positive FABMS $m / z(\%)=719[M+H]^{+}(15), 659(6), 587$
(15), 527 (10), 509 (3), 449 (2), 387 (2), 327 (6), 299 (4), 237 (6), 193 (8), 133 (9), 105 (100), 77 (15); HRFABMS $\mathrm{m} / \mathrm{z} 719.2870[\mathrm{M}+\mathrm{H}]^{+}$, calcd 719.2915.
3.3.7. Yunnanxane (7). Amorphous solid, IR $\nu_{\max }$ $(\mathrm{KBr})=3458,2975,2934,2360,2255,1732,1645,1453$, 1373, 1318, 1237, 1183, 1107, 1068, 1019, 959, 913, 878, 836, 733, 606, $562 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{H}}$ $0.77\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-19\right), 1.05\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-17\right), 1.08(3 \mathrm{H}, \mathrm{d}$, $\left.J=7.2 \mathrm{~Hz}, \mathrm{H}-5^{\prime}\right), 1.13\left(3 \mathrm{H}, \mathrm{d}, J=6.3 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right), 1.17(1 \mathrm{H}, \mathrm{br}$ $\left.\mathrm{s}, 3^{\prime}-\mathrm{OH}\right), 1.22(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7 \mathrm{a}), 1.55(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J=4.1 \mathrm{~Hz}, \mathrm{H}-$ 9a), $1.59\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-16\right), 1.74\left(2 \mathrm{H}, \mathrm{m}, \mathrm{H}_{2}-6\right), 1.83(1 \mathrm{H}, \mathrm{d}$, $J=1.4 \mathrm{~Hz}, \mathrm{H}-1), 1.88(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7 \mathrm{~b}), 1.95(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.99$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}$ ), $2.02\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-18\right.$ ), 2.11 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}$ ), 2.30 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-13 \mathrm{a}$ ), $2.31(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-9 \mathrm{~b}), 2.35\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2^{\prime}\right)$, $2.78(1 \mathrm{H}, \mathrm{dd}, J=9.2,19.0 \mathrm{~Hz}, \mathrm{H}-13 \mathrm{~b}), 2.85(1 \mathrm{H}, \mathrm{d}$, $J=6.6 \mathrm{~Hz}, \mathrm{H}-3)$, $3.80\left(1 \mathrm{H}, \mathrm{f}, \mathrm{H}-3^{\prime}\right), 4.74$ ( $1 \mathrm{H}, \mathrm{s}, \mathrm{H}-20 \mathrm{a}$ ), $4.95(1 \mathrm{H}, \mathrm{dd}, J=4.7,9.1 \mathrm{~Hz}, \mathrm{H}-14), 5.20(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-20 \mathrm{~b})$, 5.21 ( 1 H, br d, $J=6.5 \mathrm{~Hz}, \mathrm{H}-5$ ), 5.28 ( $1 \mathrm{H}, \mathrm{dd}, J=1.9,6.6 \mathrm{~Hz}$, $\mathrm{H}-2), 5.98(1 \mathrm{H}$, dd, $J=5.6,12.0 \mathrm{~Hz}, \mathrm{H}-10) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{C}} 13.7\left(\mathrm{q}, \mathrm{C}-5^{\prime}\right), 20.6\left(\mathrm{q}, \mathrm{C}-4^{\prime}\right), 20.8(\mathrm{q}$, $\mathrm{C}-18), 21.2\left(2 \mathrm{C}, \mathrm{q}, 2 \times \mathrm{OAc}-\mathrm{CH}_{3}\right), 21.7\left(\mathrm{q}, \mathrm{OAc}-\mathrm{CH}_{3}\right)$, 22.4 (q, C-19), 25.3 (q, C-16), 28.8 (t, C-6), 31.6 (q, C-17), 33.7 (t, C-7), 37.2 (s, C-15), 39.4 (t, C-13), 39.6 (s, C-8), 42.1 (d, C-3), 43.8 (t, C-9), 46.9 (d, C-2'), 59.1 (d, C-1), 69.3 (d, C-3'), 70.0 (d, C-10), 70.4 (d, C-2), 70.6 (d, C-14), 78.1 (d, C-5), 116.8 (t, C-20), 135.4 ( $\mathrm{s}, \mathrm{C}-11$ ), 134.5 ( $\mathrm{s}, \mathrm{C}-12$ ), 142.2 (s, C-4), 169.7 ( $\mathrm{s}, \mathrm{OAc}-\mathrm{C}=\mathrm{O}$ ), 169.8 ( $\mathrm{s}, \mathrm{OAc}-$ $\mathrm{C}=\mathrm{O}), 170.1(\mathrm{~s}, \mathrm{OAc}-\mathrm{C}=\mathrm{O})$, $174.5\left(\mathrm{~s}, \mathrm{C}-1^{\prime}\right)$; positive FABMS $m / z(\%)=563[\mathrm{M}+\mathrm{H}]^{+}(17), 503$ (30), 445 (4), 385 (14), 343 (19), 315 (10), 265 (17), 249 (14), 157 (62), 133 (100).
3.3.8. 10-Deacetyl-10-( $\beta$-glucosyl)-yunnanxane (8). Colorless amorphous solid, $[\alpha]_{\mathrm{D}}^{14.4}=-4.17^{\circ}\left(c 0.90, \mathrm{CH}_{3} \mathrm{OH}\right)$; UV (MeOH) $\lambda_{\max }(\log \varepsilon)=206.5 \quad(4.0) \mathrm{nm} ; ~ I R ~ \nu_{\text {max }}$ $(\mathrm{KBr})=3462,2922,1730,1638,1430,1375,1243,1197$, 1105, 1074, 1019, 976, 941, 908, 876, 835, $609 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $\left.400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta_{\mathrm{H}} 0.86\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-19\right), 1.09$ ( $3 \mathrm{H}, \mathrm{d}, J=7.1 \mathrm{~Hz}, \mathrm{H}-5^{\prime}$ ), 1.16 ( $3 \mathrm{H}, \mathrm{d}, J=6.4 \mathrm{~Hz}, \mathrm{H}-4^{\prime}$ ), 1.22 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-16$ ), 1.23 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7 \mathrm{a}$ ), 1.67 ( $1 \mathrm{H}, \mathrm{dd}, J=4.6$, $14.7 \mathrm{~Hz}, \mathrm{H}-9 \mathrm{a}$ ), 1.71 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-17$ ), 1.81 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}_{2}-6$ ), $1.87(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J=2.2 \mathrm{~Hz}, \mathrm{H}-1), 2.00(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 2.01(1 \mathrm{H}$, $\mathrm{m}, \mathrm{H}-7 \mathrm{~b}), 2.02\left(3 \mathrm{H}, \mathrm{br} \mathrm{d}, \mathrm{J}=1.6 \mathrm{~Hz}, \mathrm{CH}_{3}-18\right), 2.18(3 \mathrm{H}, \mathrm{s}$, OAc), $2.42(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-9 \mathrm{~b}), 2.43\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2^{\prime}\right), 2.48(1 \mathrm{H}, \mathrm{dd}$, $J=4.8,19.0 \mathrm{~Hz}, \mathrm{H}-13 \mathrm{a}), 2.86(1 \mathrm{H}, \mathrm{dd}, J=9.2,19.0 \mathrm{~Hz}, \mathrm{H}-$ 13b), 2.97 ( $1 \mathrm{H}, \mathrm{d}, J=6.5 \mathrm{~Hz}, \mathrm{H}-3$ ), 3.17 ( $1 \mathrm{H}, \mathrm{m}$, glc-H-5 ${ }^{\prime \prime}$ ), $3.26\left(1 \mathrm{H}, \mathrm{t}, J=7.9 \mathrm{~Hz}\right.$, glc-H-2') ), $3.31\left(1 \mathrm{H}, \mathrm{m}\right.$, glc-H-4 $\left.\mathrm{4}^{\prime \prime}\right)$, $3.32(1 \mathrm{H}, \mathrm{m}$, glc-H-3"), $3.68(1 \mathrm{H}, \mathrm{dd}, J=5.8,12.0 \mathrm{~Hz}$, glc-H-6"a), 3.86 ( $1 \mathrm{H}, \mathrm{m}$, glc-H-6"b), 3.87 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3^{\prime}$ ), 4.29 $(1 \mathrm{H}, \mathrm{d}, J=7.6 \mathrm{~Hz}$, glc-H-1"), $4.85(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-20 \mathrm{a}), 5.01(1 \mathrm{H}$, dd, $J=5.0,9.2 \mathrm{~Hz}, \mathrm{H}-14), 5.26$ ( 1 H , br t, $J=2.8 \mathrm{~Hz}, \mathrm{H}-5$ ), 5.27 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{H}-20 \mathrm{~b}$ ), 5.38 ( $1 \mathrm{H}, \mathrm{dd}, J=6.7,12.2 \mathrm{~Hz}, \mathrm{H}-10$ ), $5.41(1 \mathrm{H}, \mathrm{dd}, J=2.3,6.6 \mathrm{~Hz}, \mathrm{H}-2)$; ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta_{\mathrm{C}} 13.4\left(\mathrm{q}, \mathrm{C}-5^{\prime}\right), 20.5\left(\mathrm{q}, \mathrm{C}-4^{\prime}\right), 21.4(\mathrm{q}, \mathrm{C}-18)$, 21.5 (q, $\mathrm{OAc}-\mathrm{CH}_{3}$ ), 21.9 ( $\mathrm{q}, \mathrm{OAc}-\mathrm{CH}_{3}$ ), $23.0(\mathrm{q}, \mathrm{C}-19)$, 25.4 (q, C-17), 30.0 (t, C-6), 32.0 (q, C-16), 35.1 (t, C-7), 38.5 ( $\mathrm{s}, \mathrm{C}-15$ ), 40.7 ( $\mathrm{s}, \mathrm{C}-8$ ), 40.7 (t, C-13), 43.7 (d, C-3), 45.9 (t, C-9), 48.5 (d, C-2'), 60.9 (d, C-1), 62.7 (t, glc-C-6"), 70.2 (d, C-3'), 71.8 (d, C-14), 71.8 (d, glc-C-4"), 72.1 (d, C2), 72.8 (d, C-10), 75.0 (d, glc-C-2"), 77.9 (d, glc-C-5"), 78.4 (d, glc-C-3"), 79.9 (d, C-5), 100.1 (d, glc-C-1"), 117.4 ( $\mathrm{t}, \mathrm{C}-20$ ), 137.2 ( $\mathrm{s}, \mathrm{C}-11$ ), 138.2 ( $\mathrm{s}, \mathrm{C}-12$ ), 144.2 ( $\mathrm{s}, \mathrm{C}-4$ ),
171.7 ( $2 \mathrm{C}, \mathrm{s}, 2 \times \mathrm{OAc}-\mathrm{C}=\mathrm{O}$ ), 175.9 ( $\mathrm{s}, \mathrm{C}-1^{\prime}$ ); negative FABMS $m / z(\%)=681[M-H]^{-}(14), 325$ (23), 179 (24), 117 (100), 59 (42); HRFABMS m/z $681.3466[\mathrm{M}-\mathrm{H}]^{-}$, calcd 681.3486.
3.3.9. 7-( $\boldsymbol{\beta}$-Xylosyl)- $N$-methyl-taxol $\mathbf{C}$ (10). White needles, $\mathrm{mp} 220^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{20}=$ active $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$; UV (MeOH) $\lambda_{\text {max }}$ $(\log \varepsilon)=213.5$ (4.33), 202 (4.41) nm; IR $\nu_{\text {max }}(\mathrm{KBr})=3450$, 2937, 1722, 1603, 1448, 1399, 1371, 1313, 1247, 1177, 1106, 1069, 1047, 1025, 981, 854, 778, 710, $609 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 500 MHz , Pyridine- $d_{5}$ ) $\delta_{\mathrm{H}} 0.74(3 \mathrm{H}, \mathrm{t}, J=7.0 \mathrm{~Hz}$, $\left.\mathrm{CH}_{3}-10^{\prime}\right), 1.12\left(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-9^{\prime}\right), 1.18\left(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-8^{\prime}\right), 1.42(3 \mathrm{H}$, $\left.\mathrm{s}, \mathrm{CH}_{3}-17\right), 1.61\left(2 \mathrm{H}, \mathrm{m}, \mathrm{H}^{-} 7^{\prime}\right), 1.67\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-16\right), 2.06$ $(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{OAc}), 2.19\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-19\right), 2.32(2 \mathrm{H}, \mathrm{t}$, $\left.J=7.5 \mathrm{~Hz}, \mathrm{H}-6^{\prime}\right), 2.37\left(3 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CH}_{3}-18\right), 2.42(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-$ 6a), 2.51 ( $3 \mathrm{H}, \mathrm{s}, 4-\mathrm{OAc}$ ), $2.99(1 \mathrm{H}, \mathrm{dd}, J=6.1,15.4 \mathrm{~Hz}, \mathrm{H}-$ 14a), 3.11 (H, m, H-6b), 3.11 (H, m, H-14b), 3.13 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{N}-$ $\left.\mathrm{CH}_{3}\right), 3.61\left(1 \mathrm{H}, \mathrm{t}, J=11.0 \mathrm{~Hz}\right.$, xyl-H-5 $5^{\prime \prime \prime}$ 'a) , $3.82(1 \mathrm{H}, \mathrm{t}$, $J=8.0 \mathrm{~Hz}$, xyl-H-2 $\left.2^{\prime \prime \prime \prime}\right), 4.06\left(1 \mathrm{H}, \mathrm{t}, J=8.6 \mathrm{~Hz}\right.$, xyl $\left.-\mathrm{H}-3^{\prime \prime \prime \prime}\right)$, $4.10\left(1 \mathrm{H}, \mathrm{m}\right.$, xyl-H-4 ${ }^{\prime \prime \prime \prime}$ ), $4.25(1 \mathrm{H}, \mathrm{dd}, J=5.1,11.3 \mathrm{~Hz}$, xyl-$\mathrm{H}-5^{\prime \prime \prime} \mathrm{b}$ ), $4.35(1 \mathrm{H}, \mathrm{d}, J=6.7 \mathrm{~Hz}, \mathrm{H}-3), 4.44(1 \mathrm{H}, \mathrm{AB}$ d, $J=8.3 \mathrm{~Hz}, \mathrm{H}-20 \mathrm{a}), 4.53(1 \mathrm{H}, \mathrm{AB}$ d, $J=8.3 \mathrm{~Hz}, \mathrm{H}-20 \mathrm{~b}), 4.80$ ( $\left.1 \mathrm{H}, \mathrm{d}, J=7.4 \mathrm{~Hz}, \mathrm{xyl}-\mathrm{H}-1^{\prime \prime \prime \prime}\right), 4.87(1 \mathrm{H}, \mathrm{dd}, J=7.1,10.1 \mathrm{~Hz}$, $\mathrm{H}-7), 5.24(1 \mathrm{H}, \mathrm{d}, J=9.0 \mathrm{~Hz}, \mathrm{H}-5), 5.65(1 \mathrm{H}, \mathrm{br} \mathrm{d}$, $\left.J=3.7 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right), 6.29(1 \mathrm{H}, \mathrm{d}, J=6.7 \mathrm{~Hz}, \mathrm{H}-2), 6.72(1 \mathrm{H}, \mathrm{t}$, $J=9.4 \mathrm{~Hz}, \mathrm{H}-13), 6.78\left(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J=3.5 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 7.13(1 \mathrm{H}$, $\mathrm{s}, \mathrm{H}-10), 7.33\left(2 \mathrm{H}, \mathrm{t}, J=7.5 \mathrm{~Hz}, \mathrm{Bz}-\mathrm{H}-4^{\prime \prime}\right), 7.33(2 \mathrm{H}, \mathrm{t}$, $\left.J=8.0 \mathrm{~Hz}, \mathrm{H}-4^{\prime \prime \prime}\right), 7.41$ ( $2 \mathrm{H}, \mathrm{t}, J=7.3 \mathrm{~Hz}, \mathrm{Bz}-\mathrm{H}-5^{\prime \prime}$ ), 7.41 ( $2 \mathrm{H}, \mathrm{t}, J=7.6 \mathrm{~Hz}, \mathrm{H}-3^{\prime \prime \prime}$ ), $7.75\left(2 \mathrm{H}, \mathrm{d}, J=7.7 \mathrm{~Hz}, \mathrm{H}-2^{\prime \prime \prime}\right), 8.30$ ( $2 \mathrm{H}, \mathrm{d}, J=7.4 \mathrm{~Hz}, \mathrm{Bz}-\mathrm{H}-3^{\prime \prime}$ ); ${ }^{13} \mathrm{C}$ NMR ( 125 MHz , Pyridine$\left.d_{5}\right) \delta_{\mathrm{C}} 11.6(\mathrm{q}, \mathrm{C}-19), 14.1\left(\mathrm{q}, \mathrm{C}-10^{\prime}\right), 14.7$ (q, C-18), 21.0 (q, $10-\mathrm{OAc}-\mathrm{CH}_{3}$ ), $22.0(\mathrm{q}, \mathrm{C}-16), 22.7$ (q, 4-OAc- $\mathrm{CH}_{3}$ ), 22.8 (t, C-9'), 25.3 (t, C-7'), 26.9 (q, C-17), 31.7 (t, C-8'), 33.8 (t, C-6'), 33.8 ( $\mathrm{q}, \mathrm{N}-\mathrm{CH}_{3}$ ), 36.5 (t, C-6), 36.5 (t, C-14), 44.3 ( s , C-15), 47.8 (d, C-3), 57.9 (d, C-3'), 58.4 (s, C-8), 67.1 (t, xyl-C-5"II'), 70.7 (d, xyl-C-4 ${ }^{\prime \prime \prime \prime}$ ), 71.7 (d, C-13), 72.7 (d, C$2^{\prime}$ ), 74.6 (d, xyl-C-2"II'), 75.7 (d, C-2), 76.6 (t, C-20), 77.1 (d, C-10), 77.9 ( $\mathrm{s}, \mathrm{C}-1$ ), 77.9 (d, xyl-C-3 $3^{\prime \prime \prime}$ ), 80.4 (d, C-7), 81.5 (s, C-4), 84.4 (d, C-5), 105.4 (d, xyl-C-1 ${ }^{\prime \prime \prime \prime}$ ), 127.9 (d, C-4"I), 128.9 (d, Bz-C-4"), 128.9 (d, C-3 ${ }^{\prime \prime \prime}$ ), 129.1 (d, C-2"I'), 130.4 (d, Bz-C-3"), 131.1 (d, Bz-C-2"), 133.3 (d, Bz-C-5 ${ }^{\prime \prime}$ ), 134.2 (s, C-11), 138.7 ( $\mathrm{s}, \mathrm{C}-1^{\prime \prime \prime}$ ), 141.1 (s, C-12), 166.4 (s, Bz-C$1^{\prime \prime}$ ), 170.0 ( $\mathrm{s}, 10-\mathrm{OAc}-\mathrm{C}=\mathrm{O}$ ), 170.8 ( $\mathrm{s}, 4-\mathrm{OAc}-\mathrm{C}=\mathrm{O}$ ), 174.4 (s, C-1'), 174.4 (s, C-5'), 203.1 (s, C-9); negative FABMS $m / z(\%)=994[\mathrm{M}-\mathrm{H}]^{-}(2), 933(1), 855(1), 774$ (1), 701 (2), 477 (2), 384 (2), 292 (100), 219 (2), 121 (17), 77 (3); HRFABMS $m / z 995.4576[\mathrm{M}-\mathrm{H}]^{-}$, calcd 995.4515.

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[^0]:    Keywords: Taxus yunnanensis; taxaceae; barks; abeotaxoids; taxane glycosides; orthoester; C-2(20) epoxide; taxuyunnanines W-Z; 7-( $\beta$ -xylosyl)- $N$-methyl-taxol $\quad$ C; 10-deacetyl-10-( $\beta$-glucopyranosyl)yunnanxane.

    * Corresponding authors. Address: Program for Collaborative Research in the Pharmaceutical Sciences, Department of Medicinal Chemistry and Pharmacognosy, College of Pharmacy, University of Illinois at Chicago, 833 S. Wood Street, Chicago, IL 60612, USA. Tel.: +1-312-996-7868; fax: +1-312-413-5894; e-mail: zhanghj@uic.edu; hfong@uic.edu

[^1]:    ${ }^{\text {a }}$ Measured at 400 MHz .
    ${ }^{\mathrm{b}}$ The data were obtained in pyridine- $d_{5}$.
    ${ }^{\text {c }}$ Measured at 500 MHz .
    ${ }^{\mathrm{d}}$ The data were obtained in $\mathrm{CD}_{3} \mathrm{OD}$.
    ${ }^{\mathrm{e}}$ Recorded at $50^{\circ} \mathrm{C}$.

[^2]:    ${ }^{\text {a }}$ Measured at 100 MHz .
    ${ }^{\mathrm{b}}$ The data were obtained in pyridine $-d_{5}$.
    ${ }^{\text {c }}$ Measured at 125 MHz .
    ${ }^{\mathrm{d}}$ The data were obtained in $\mathrm{CD}_{3} \mathrm{OD}$.

