



Racemic alkaloids from the fungus *Ganoderma cochlear*



Xin-Long Wang^{a,b,1}, Man Dou^{b,c,1}, Qi Luo^b, Li-Zhi Cheng^{b,d}, Yong-Ming Yan^b,
Rong-Tao Li^c, Yong-Xian Cheng^{a,b,*}

^a Henan University of Traditional Chinese Medicine, Zhengzhou, 450008, PR China

^b State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming, 650201, PR China

^c Faculty of Life Science and Engineering, Kunming University of Science and Technology, Kunming, 650500, PR China

^d Guangdong Pharmaceutical University, Guangzhou, 510006, PR China

ARTICLE INFO

Article history:

Received 6 September 2016

Received in revised form 11 November 2016

Accepted 19 November 2016

Available online 25 November 2016

Keywords:

Ganoderma cochlear

Alkaloids

ECD calculations

Racemates

Biosynthesis

ABSTRACT

Seven pairs of new alkaloid enantiomers, ganocochlearines C–I (1, 3–8), and three pairs of known alkaloids were isolated from the fruiting bodies of *Ganoderma cochlear*. The chemical structures of new compounds were elucidated on the basis of 1D and 2D NMR data. The absolute configurations of compounds 1, 3–10 were assigned by ECD calculations. Biological activities of these isolates against renal fibrosis were accessed in rat normal or diseased renal interstitial fibroblast cells. Importantly, the plausible biosynthetic pathway for this class of alkaloids was originally proposed.

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1. Introduction

The genus *Ganoderma* embraces several medicinal mushrooms. Of them, *G. lucidum* and *G. sinense* are recorded in the Pharmacopoeia of PR China (2015 version), highlighting their significance in traditional Chinese medicine. Due to their medicinal values, modern investigations on *Ganoderma* species have been extensively conducted, which reveals the presence of triterpenoids, polysaccharides as major components. In addition, peptides, proteins and alkaloids were also reported from this genus [1–3]. Despite that mounting studies on *Ganoderma* were documented, new chemical findings are still expectable. For example, several structurally novel meroterpenoids have been characterized from *Ganoderma* in the recent years [4–7]. In contrast to triterpenoids and emerging meroterpenoids, alkaloids present in this genus are largely unknown. In 1990, ganoines I and II were isolated from *G. capense*, to our knowledge, representing the first alkaloid example in *Ganoderma* [8]. Thereafter, studies referring to *Ganoderma* alkaloids are relatively few. During our search for meroterpenoids from *Ganoderma* fungi, two alkaloids namely ganocochlearines A and B were isolated from *G. cochlear* [9]. As a follow-up investigation on this species, seven pairs of new alkaloid enantiomers, ganocochlearines C–I (1, 3–8), together with three pairs of known alkaloids ganoapplanatumine B (2), sinsensine E (9), and

lucidimine C (10), were obtained. In this contribution, we describe their isolation, structure elucidation and biological evaluation.

2. Experimental

2.1. General

Column chromatography was performed by using silica gel (200–300 mesh; Qingdao Marine Chemical Inc., PR China), C-18 silica gel (40–60 μm; Daiso Co., Japan), MCI gel CHP 20P (75–150 μm, Mitsubishi Chemical Industries, Tokyo, Japan) and Sephadex LH-20 (Amersham Pharmacia, Sweden). Semi-preparative or analytic HPLC was carried out using an Agilent 1200 liquid chromatograph, the column used was a 250 mm × 9.4 mm, i.d., 5 μm, Zorbax SB-C18 or a 250 mm × 4.6 mm, i.d., 5 μm, Daicel Chiralpak (IC for 1–4, 7–10 and AD-H for 5–6). Optical rotations were recorded on a Horiba SEPA-300 polarimeter. UV spectra were recorded on a Shimadzu UV-2401PC spectrometer. CD spectra were measured on a Chirascan instrument. NMR spectra were recorded on a Bruker AV-600, with TMS as an internal standard. ESIMS and HRESIMS were collected by an Agilent G6230 TOF MS spectrometer.

2.2. Fungal material

The fruiting bodies of *G. cochlear* were purchased from Tongkang Pharmaceutical Co. Ltd. in Guangzhou Province, PR China, in September 2013. The material was identified by Prof. Zhu-Liang Yang at Kunming

* Corresponding author at: Henan University of Traditional Chinese Medicine, Zhengzhou, 450008, PR China.

E-mail address: yxcheng@mail.kib.ac.cn (Y.-X. Cheng).

¹ There authors contributed equally to this paper.

Table 1
¹H (600 MHz) and ¹³C NMR (150 MHz) data of compounds **1**, **3–8** (δ in ppm, J in Hz).

Position	1 ^a		3 ^b		4 ^c		5 ^d		6 ^e		7 ^f		8 ^e	
	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C
2		147.4		147.0		149.1		152.2		156.0		143.8	8.39, s	149.6
3		123.6		129.9		125.2		115.3	7.68, s	119.8	5.58, t, 7.8	78.9		134.5
4	2.62, m	28.9	3.21, m	30.3	3.02, m	28.6	3.57, m	32.1	3.07, overlap	27.4	2.73, m	34.1		147.3
	2.46, m		2.98, m		2.83, m		3.30, m		2.90, m		2.14, m			
5	2.29, m	32.8	2.81, m	34.7	2.43, m	33.5	2.57, m	35.9	2.05, m	31.1	2.97, m	28.7	5.58, t-like, 7.8	80.5
	1.55, m		2.30, m		2.16, m		2.04, overlap		1.86, m		2.88, m			
6	3.64, m	38.9	5.62, t-like, 7.8	79.3	5.04, dd, 6.5, 3.7	82.5	5.43, t-like, 6.4	74.1	4.15, m	66.6		129.4	2.82, m	35.6
													2.26, m	
7	8.56, s	145.1	8.54, s	148.9	8.48, s	146.1	8.96, s	152.5	3.07, overlap	35.5	8.28, s	150.3	3.17, m	29.9
									2.74, m				3.03, m	
8		149.2		148.6		151.3		158.0	8.24, s	147.8		132.2		135.1
9		142.8		133.4		140.4		144.5		149.0		148.3		150.8
10	5.22, dd, 13.8, 4.2	66.2	5.47, s	101.8	5.15, s	66.3		160.9		130.9	2.26, s	15.4	4.68, s	60.9
1'		125.5		123.5		124.7		121.3		120.6		123.6		124.3
2'		150.6		151.4		151.0		147.1		150.8		149.0		151.6
3'	7.16, d, 8.6	118.7	6.91, d, 8.7	118.9	6.81, d, 8.9	118.8	7.24, d, 8.8	118.7	6.73, overlap	119.4	6.83, d, 8.7	117.6	6.86, d, 8.5	119.7
4'	7.22 overlap	119.2	6.84, dd, 8.7, 3.0	118.8	6.77, dd, 8.9, 2.9	119.4	7.12, dd, 8.8, 3.0	120.5	6.73, overlap	119.4	6.71, dd, 8.7, 3.0	118.3	6.78, dd, 8.5, 2.9	119.5
5'		154.5		152.8		153.6		155.0		153.4		152.4		153.8
6'	8.45, d, 2.6	112.0	7.60, d, 3.0	110.1	7.52, d, 2.9	111.4	8.03, brs	110.3	7.24, brs	112.9	7.37, d, 3.0	109.0	7.44, d, 2.9	110.6
6-OCH ₃			3.34, s	52.8										
6-OCH ₂ CH ₃					3.67, t-like, 7.0	65.5								
6-OCH ₂ CH ₃					1.23, t, 7.0	15.7								
6-CH ₂ COCH ₃	2.87, m	49.2												
	2.53, m													
6-CH ₂ COCH ₃		207.3												
6-CH ₂ COCH ₃	2.09, s	30.4												
10-OCH ₃				3.37, s	53.3									

^a ¹H and ¹³C NMR in pyridine-*d*₆, ^b ¹H and ¹³C NMR in CDCl₃, ^c ¹H and ¹³C NMR in methanol-*d*₄, ^d ¹H and ¹³C NMR in acetone-*d*₆, ^e ¹H and ¹³C NMR in DMSO-*d*₆.

Institute of Botany, Chinese Academy of Sciences, and a voucher specimen (CHYX0570) was deposited at the State Key Laboratory of Photochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences.

2.3. Extraction and isolation

The powders of fruiting bodies of *G. cochlear* (100 kg) were extracted by reflux with 70% EtOH (300 L. × 48 h × 6) at room temperature and concentrated under reduced pressure to give a crude extract (10 kg), which was suspended in water followed by partition with ethyl acetate to afford an ethyl acetate soluble extract (2 kg). This extract was divided into seven parts (Fr.1–Fr.7) by using a silica gel column eluted with increasing MeOH in CHCl₃ (100:1–1:1). Fr.5 (440 g) was further separated by a MCI gel CHP 20P column washed with aqueous MeOH (30%–100%) to yield 7 portions (Fr.5.1–Fr.5.7). Fr.5.3 (34 g) was submitted to gel filtration on Sephadex LH-20 (MeOH) column followed by RP-18 column chromatography eluted with gradient aqueous MeOH (30%–60%) to yield 8 fractions (Fr.5.3.1–Fr.5.3.8). Of which, Fr.5.3.3 (970 mg) was separated by Sephadex LH-20 (MeOH) followed by preparative TLC (CHCl₃/MeOH, 12:1) to give Fr.5.3.3.1 and compound **5** (2.6 mg). Fr. 5.3.3.1 (16 mg) was purified by semi-preparative HPLC (MeOH/H₂O, 60%–90%) to afford compounds **2** (2.3 mg, t_R = 16.2 min) and **3** (1.3 mg, t_R = 21.8 min). Fr.5.3.4 (560 mg) was submitted to preparative TLC (CHCl₃/MeOH, 12:1) followed by semi-preparative HPLC (MeOH/H₂O, 65%) to produce compounds **1** (3.4 mg, t_R = 8.2 min) and **4** (2.1 mg, t_R = 12.4 min). Fr.5.4 (32 g) was fractionated by a MCI gel CHP 20P column eluted with gradient aqueous MeOH (40%–100%) to provide 6 portions (Fr.5.4.1–Fr.5.4.6). Fr.5.4.3 (1.2 g) was separated by a RP-18 column eluted with gradient aqueous MeOH (30%–60%) followed by preparative TLC (CHCl₃/Me₂CO, 6:1) to afford compound **6** (5.1 mg). Fr.5.4.4 (5.2 g) was fractionated by a combination of Sephadex LH-20 (MeOH), RP-18 (MeOH/H₂O, 40%–60%), and preparative TLC (CHCl₃/Me₂CO, 6:1) to yield compounds **8** (16.3 mg) and **9** (5.9 mg). Fr.5.4.5 (12 g) was passed through a MCI gel CHP 20P column eluted with gradient aqueous MeOH (40%–80%) followed by a combination of Sephadex LH-20 filtration (MeOH), RP-18 (50%–65%) and semi-preparative HPLC (MeOH/H₂O, 55%) to yield compounds **7** (1.9 mg, t_R = 15.7 min) and **10** (4.9 mg, t_R = 19.1 min).

Racemic or partially racemic compounds were submitted to chiral phase on Daicel Chiralpak IC (**1–4**, **7–10**) or AD-H (**5**, **6**) to afford their enantiomers (Table 2).

Table 2
Chiral separation of racemic mixtures of **1–10**.

Compd	Chiral phase	Mobile phase	Flow rate (mL/min)	Retention time (min)	Amount (mg)	Peak area ratio
1	1a	IC	<i>n</i> -hexane/ethanol, 94:6	15.0	0.83	0.65:1
				20.4	1.16	
2	2a	IC	<i>n</i> -hexane/ethanol, 80:20	15.0	0.80	1.08:1
				21.7	0.60	
3	3a	IC	<i>n</i> -hexane/ethanol, 85:15	7.8	0.34	0.97:1
				8.5	0.33	
4	4a	IC	<i>n</i> -hexane/ethanol, 90:10	13.9	0.62	1.38:1
				15.9	0.55	
5	5a	AD-H	<i>n</i> -hexane/ethanol, 70:30	15.7	0.60	1.77:1
				17.2	0.55	
6	6a	AD-H	<i>n</i> -hexane/ethanol, 70:30	18.0	2.33	1.08:1
				29.0	2.15	
7	7a	IC	<i>n</i> -hexane/ethanol, 75:25	7.6	0.38	1.33:1
				9.4	0.32	
8	8a	IC	<i>n</i> -hexane/ethanol, 75:25	12.3	1.86	1.88:1
				15.8	1.78	
9	9a	IC	<i>n</i> -hexane/ethanol, 75:25	22.9	2.02	0.94:1
				31.9	1.86	
10	10a	IC	<i>n</i> -hexane/ethanol, 85:15	10.9	1.04	0.93:1
				14.8	1.26	

Ganocochlearine C (**1**): white solid; $\{[\alpha]_D^{19}\}$ unstable; CD (MeOH) $\Delta\epsilon_{203} + 1.25$, $\Delta\epsilon_{235} - 1.31$, $\Delta\epsilon_{257} + 0.70$, $\Delta\epsilon_{286} - 0.94$, $\Delta\epsilon_{340} + 0.33$; **1a**; $\{[\alpha]_D^{19}\}$ unstable; CD (MeOH) $\Delta\epsilon_{198} - 2.29$, $\Delta\epsilon_{237} + 0.82$, $\Delta\epsilon_{254} - 0.58$, $\Delta\epsilon_{287} + 0.75$, $\Delta\epsilon_{347} - 0.37$; **1b**; UV (MeOH) λ_{max} (log ϵ): 342 (3.52), 276 (3.63), 235 (3.81), 214 (4.09), 203 (4.11); ¹H and ¹³C NMR data, see Table 1; HRESIMS (positive) m/z 296.1282 [M + H]⁺ (calcd for C₁₈H₁₈NO₃, 296.1281).

Ganocochlearine D (**3**): white solid; $\{[\alpha]_D^{20}\}$ = +58.6 (c 0.70, MeOH); CD (MeOH) $\Delta\epsilon_{219} + 16.48$, $\Delta\epsilon_{262} - 0.58$, $\Delta\epsilon_{295} + 6.06$, $\Delta\epsilon_{355} - 2.17$; **3a**; $\{[\alpha]_D^{19}\}$ = -194.8 (c 0.55, MeOH); CD (MeOH) $\Delta\epsilon_{218} - 19.31$, $\Delta\epsilon_{267} + 0.66$, $\Delta\epsilon_{294} - 6.54$, $\Delta\epsilon_{352} + 2.30$; **3b**; UV (MeOH) λ_{max} (log ϵ): 411 (1.91), 349 (3.66), 286 (3.74), 236 (3.94), 203 (4.19); ¹H and ¹³C NMR data, see Table 1; HRESIMS (positive) m/z 300.1234 [M + H]⁺ (calcd for C₁₇H₁₈NO₄, 300.1230).

Ganocochlearine E (**4**): white solid; $\{[\alpha]_D^{20}\}$ = +45.1 (c 1.24, MeOH); CD (MeOH) $\Delta\epsilon_{202} + 5.76$, $\Delta\epsilon_{232} - 3.49$, $\Delta\epsilon_{259} + 1.75$, $\Delta\epsilon_{292} - 2.14$, $\Delta\epsilon_{346} + 0.98$; **4a**; $\{[\alpha]_D^{20}\}$ = -16.0 (c 2.75, MeOH); CD (MeOH) $\Delta\epsilon_{198} - 8.91$, $\Delta\epsilon_{232} + 2.34$, $\Delta\epsilon_{258} - 1.66$, $\Delta\epsilon_{294} + 1.63$, $\Delta\epsilon_{346} - 0.77$; **4b**; UV (MeOH) λ_{max} (log ϵ): 343 (3.87), 275 (3.99), 236 (4.15), 204 (4.42); ¹H and ¹³C NMR data, see Table 1; HRESIMS (positive) m/z 284.1283 [M + H]⁺ (calcd for C₁₇H₁₈NO₃, 284.1281).

Ganocochlearine F (**5**): white solid; $\{[\alpha]_D^{20}\}$ unstable; CD (MeOH) $\Delta\epsilon_{203} + 18.45$, $\Delta\epsilon_{241} - 1.66$, $\Delta\epsilon_{254} + 1.30$, $\Delta\epsilon_{278} - 7.12$; **5a**; $\{[\alpha]_D^{20}\}$ unstable; CD (MeOH) $\Delta\epsilon_{205} - 5.52$, $\Delta\epsilon_{238} + 0.53$, $\Delta\epsilon_{256} - 0.25$, $\Delta\epsilon_{276} + 2.32$; **5b**; UV (MeOH) λ_{max} (log ϵ): 325 (3.77), 270 (4.13), 225 (4.69), 207 (4.64); ¹H and ¹³C NMR data, see Table 1; HRESIMS (positive) m/z 270.0761 [M + H]⁺ (calcd for C₁₅H₁₂NO₄, 270.0761).

Ganocochlearine G (**6**): white solid; $\{[\alpha]_D^{16}\}$ = +33.9 (c 1.45, MeOH); CD (MeOH) $\Delta\epsilon_{222} + 1.15$, $\Delta\epsilon_{240} + 0.19$, $\Delta\epsilon_{266} + 1.04$; **6a**; $\{[\alpha]_D^{16}\}$ = -88.1 (c 1.25, MeOH); CD (MeOH) $\Delta\epsilon_{218} - 2.18$, $\Delta\epsilon_{235} - 0.42$, $\Delta\epsilon_{266} - 1.95$; **6b**; UV (MeOH) λ_{max} (log ϵ): 343 (3.89), 264 (4.19), 213 (4.46), 202 (4.45); ¹H and ¹³C NMR data, see Table 2; HRESIMS (positive) m/z 258.1126 [M + H]⁺ (calcd for C₁₅H₁₆NO₃, 258.1125).

Ganocochlearine H (**7**): white solid; $\{[\alpha]_D^{20}\}$ = +181.9 (c 1.40, MeOH); CD (MeOH) $\Delta\epsilon_{219} + 24.20$, $\Delta\epsilon_{263} - 1.22$, $\Delta\epsilon_{294} + 9.53$, $\Delta\epsilon_{341} - 3.03$; **7a**; $\{[\alpha]_D^{19}\}$ = -212.3 (c 1.10, MeOH); CD (MeOH) $\Delta\epsilon_{219} - 19.25$, $\Delta\epsilon_{263} + 1.02$, $\Delta\epsilon_{295} - 7.59$, $\Delta\epsilon_{342} + 2.41$; **7b**; UV (MeOH) λ_{max} (log ϵ): 344 (3.84), 277 (3.85), 229 (4.58), 213 (4.62); ¹H and ¹³C NMR data, see Table 2; HRESIMS (positive) m/z 240.1022 [M + H]⁺ (calcd for C₁₅H₁₄NO₂, 240.1019).

Ganocochlearine I (**8**): white solid; $\{[\alpha]_D^{20}\}$ = +100.9 (c 1.86, MeOH); CD (MeOH) $\Delta\epsilon_{220} + 15.21$, $\Delta\epsilon_{263} - 0.95$, $\Delta\epsilon_{295} + 6.01$, $\Delta\epsilon_{349} - 2.18$; **8a**; $\{[\alpha]_D^{19}\}$ = -172.3 (c 1.78, MeOH); CD (MeOH) $\Delta\epsilon_{219} - 19.43$, $\Delta\epsilon_{265} + 0.71$, $\Delta\epsilon_{295} - 7.61$, $\Delta\epsilon_{352} + 2.15$; **8b**; UV (MeOH) λ_{max} (log ϵ): 347 (3.73), 277 (3.77), 233 (4.14), 204 (4.34); ¹H and ¹³C NMR data, see Table 2; HRESIMS (positive) m/z 256.0974 [M + H]⁺ (calcd for C₁₅H₁₄NO₃, 256.0968).

Sinensine E (**9**): white solid; $\{[\alpha]_D^{19}\}$ = +1.3 (c 1.84, MeOH); CD (MeOH) $\Delta\epsilon_{200} - 4.62$, $\Delta\epsilon_{233} + 2.13$, $\Delta\epsilon_{260} - 1.66$, $\Delta\epsilon_{286} + 1.80$, $\Delta\epsilon_{345} - 0.72$; **9a**; $\{[\alpha]_D^{18}\}$ = -8.2 (c 1.82, MeOH); CD (MeOH) $\Delta\epsilon_{197} + 6.47$, $\Delta\epsilon_{226} - 3.10$, $\Delta\epsilon_{258} + 1.38$, $\Delta\epsilon_{289} - 2.12$, $\Delta\epsilon_{345} + 0.35$; **9b**; UV (MeOH) λ_{max} (log ϵ): 343 (3.85), 274 (3.98), 233 (4.18), 203 (4.44); HRESIMS (positive) m/z 256.0972 [M + H]⁺ (calcd for C₁₅H₁₄NO₃, 256.0968).

Lucidimine C (**10**): white solid; $\{[\alpha]_D^{20}\}$ unstable; CD (MeOH) $\Delta\epsilon_{217} + 19.24$, $\Delta\epsilon_{254} - 5.24$, $\Delta\epsilon_{290} + 5.45$, $\Delta\epsilon_{328} - 3.26$; **10a**; $\{[\alpha]_D^{20}\}$ unstable; CD (MeOH) $\Delta\epsilon_{218} - 13.39$, $\Delta\epsilon_{251} + 2.52$, $\Delta\epsilon_{292} - 4.89$, $\Delta\epsilon_{332} + 0.71$; **10b**; UV (MeOH) λ_{max} (log ϵ): 331 (3.90), 268 (4.16), 213 (4.53); HRESIMS (positive) m/z 270.1129 [M + H]⁺ (calcd for C₁₆H₁₆NO₃, 270.1130).

2.4. Cell viability assay

Rat normal renal interstitial fibroblast cells (NRK-49F) were purchased from ATCC (American Type Culture Collection) and routinely grown and maintained in DMEM with 10% FBS and with 100 units/mL penicillin and 100 µg/mL streptomycin (all obtained from Gibco BRL,

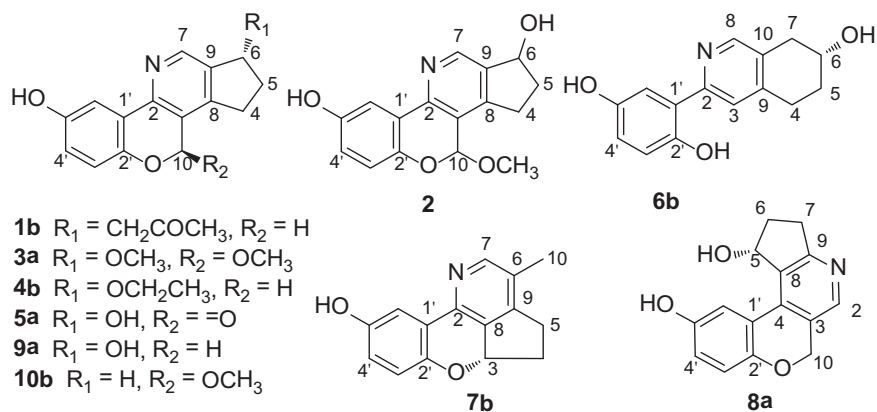


Fig. 1. Chemical structures of compounds 1–10.

Grand Island, NY). Cells were incubated in an incubator with 5% CO₂ in air at 37 °C. Cell viability was determined by the MTT assay. Cells in log phase were seeded at a density of 5000 cells/well of 96-well flat-bottomed microtiter plates and treated with compounds (20 μM) or solvent control. After 24 h of incubation, 20 μL of 5 mg/mL MTT were introduced to each well and incubated for 4 h of exposure. The plates were centrifuged and medium was decanted. Cells were subsequently dissolved in 150 μL DMSO with gentle shaking for 15 min at room temperature followed by measurement of absorbance at 490 nm.

2.5. TGF-β1-stimulated proliferation inhibitory assay

NRK-49F cells in log phase were growing by 5000 cells/well of 96-well flat-bottomed microtiter plates. 24 h before compounds (20 μM) treatment, cells were treated with TGF-β1 (5 ng/mL). After 24 h of incubation, 20 μL of 5 mg/mL MTT were added to each well and incubated for 4 h of exposure. The plates were centrifuged and medium was decanted. Cells were subsequently dissolved in 150 μL DMSO with gentle shaking for 15 min at room temperature followed by measurement of absorbance at 490 nm.

3. Results and discussion

Compound **1** has a molecular formula C₁₈H₁₇NO₃, as deduced by analysis of its positive HRESIMS at *m/z* 296.1282 [M + H]⁺ (calcd for 296.1281, C₁₈H₁₈NO₃), ¹³C NMR and DEPT spectra, having eleven degrees of unsaturation. The ¹³C NMR and DEPT spectra (Table 1) show 18 carbon resonances ascribe to one methyl, four methylene (including one oxygenated), five methine (four sp² and one sp³), and eight quaternary carbons (one carbonyl and seven olefinic). These data resemble those of known compound **9**, differing in that an OH group attached at

C-6' is replaced by a CH₃COCH₂ moiety. The ¹H–¹H COSY correlations of H-4/H-5/H-6/6-CH₂COCH₃ and HMBC correlations of 6-CH₂COCH₃/6-CH₂COCH₃, 6-CH₂COCH₃, 6-CH₂COCH₃/C-5, C-6, C-9, H-6/6-CH₂COCH₃ unambiguously clarify the substituted group at C-6. In addition, the ¹H NMR signals [δ_H 8.45 (1H, d, *J* = 2.6 Hz, H-6'), δ_H 7.16 (1H, d, *J* = 8.6 Hz, H-3'), δ_H 7.22, (1H, overlap, H-4')] support the presence of an ABX spin system in the structure. There is one chiral center in the structure of **1**, therefore, it makes sense to determine its absolute configuration. **1** was isolated as a partially racemic mixture. Separation by using chiral HPLC afforded **1a** and **1b**, the absolute configuration of **1a** was assigned as 6*R* by comparison of its calculated electronic circular dichroism (ECD) with that of experimental data (Supplementary Information). As a result, the structure of **1** was deduced and named as ganocochlearine C.

The molecular formula of compound **3** was assigned as C₁₇H₁₇NO₄ by the positive HRESIMS at *m/z* 300.1234 [M + H]⁺ (calcd for 300.1230), having ten degrees of unsaturation. The ¹H and ¹³C NMR data (Table 1) of **3** are very similar to those of known compound **2**. The difference between **2** and **3** is that an OH at C-6 in **2** is replaced by an OCH₃ in **3**, which is supported by the HMBC correlation of OCH₃/C-6. **3** was also isolated as a racemic mixture, subsequent chiral HPLC separation afforded two enantiomers. Their absolute configurations were determined by the mentioned computational methods. It was found that the ECD spectrum of 6*R*,10*S* instead of 6*S*,10*S* agrees well with the experimental data of **3b**, naturally clarifying the absolute configuration of **3b** as 6*R*,10*S* (Supplementary Information). Therefore, the structure of **3** was established and named as ganocochlearine D.

The molecular formula of **4** was deduced as C₁₇H₁₇NO₃ by the combination of its HRESIMS and NMR data. Comparison of NMR data (Table 1) of **4** with those of **9** indicated their structures were similar, except for the presence of an OCH₂CH₃ group in **4**. The ¹H–¹H COSY correlations of

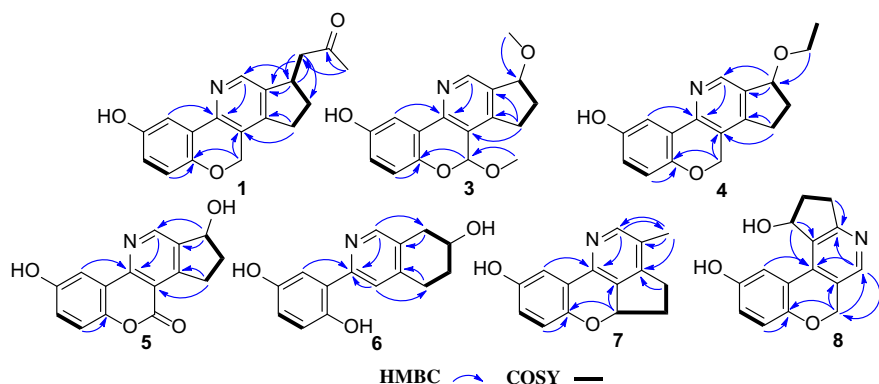


Fig. 2. Key HMBC and COSY correlations of compounds 1, 3–8.

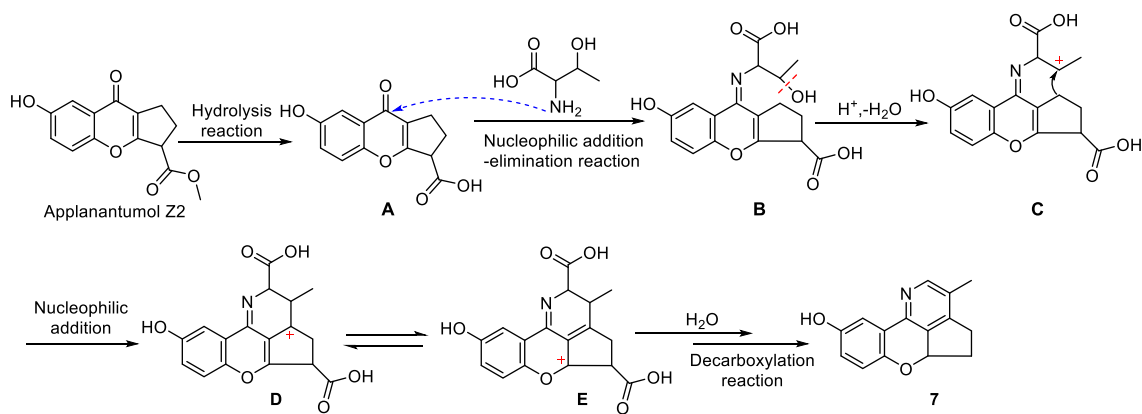


Fig. 3. A plausible biosynthetic pathway for 7.

6-OCH₂CH₃/6-OCH₂CH₃ and the HMBC correlation of 6-OCH₂CH₃/C-6 support that 6-OCH₂CH₃ is attached to C-6. Compound **4** is not enantiomerically pure, chiral HPLC separation afforded two enantiomers (**4a** and **4b**). The absolute configuration of **4a** was assigned to be 6*S* by using ECD calculations (Supplementary Information). As a result, the structure of **4** was identified as named as ganocochlearine E.

Compound **5** has the molecular formula C₁₅H₁₁NO₄ as determined by its HRESIMS and NMR data. The ¹H NMR spectrum of **5** (Table 1) contains resonances for a typical ABX spin system [δ_{H} 8.03 (1H, brs, H-6'), δ_{H} 7.24 (1H, d, $J = 8.8$ Hz, H3'), δ_{H} 7.12, (1H, d, $J = 8.8, 3.0$ Hz, H-4')]. The ¹³C NMR and DEPT spectra (Table 1) show 15 carbons, including two methylene, five methane (four sp² and one sp³), and eight quaternary carbons (one carbonyl and seven olefinic). The ¹H and ¹³C NMR data of **5** are similar to those of **9**, indicating they are analogues. The difference is that C-10 in **9** is oxidized into a ketone in **5** evident from the chemical shift of C-10 (δ_{C} 160.9). Chiral HPLC analysis indicated partially racemic nature of **5**, successive separation on chiral phase by HPLC afforded a pair of enantiomers. Likewise, the absolute configuration of **5b** was determined to be 6*S* by utilizing computational methods (Supplementary Information). Consequently, the structure of **5** was identified and named as ganocochlearine F.

Analysis of HREIMS, ¹³C NMR and DEPT spectra of compound **6** reveals that it has the molecular composition C₁₅H₁₅NO₃, suggesting nine degrees of unsaturation. The ¹H NMR spectrum of **6** (Table 1) exhibits signals [δ_{H} 7.24 (1H, brs, H-6'), δ_{H} 6.73 (2H, overlap, H3', 4')], suggesting the presence of an ABX spin system in **6**, and signals [δ_{H} 7.68 (1H, s, H3), δ_{H} 8.24 (1H, s, H-8)], indicating the possibility of a 2,4,5-pyridyl ring. The ¹³C NMR and DEPT spectra of **6** contains resonances for fifteen carbons, including three methylene, six methine, six quaternary carbons. These data imply that **6** and **9** have similar structures, differing in two places. On the one hand, a six-membered ring in **6** is replaced by a five-membered ring in **9**; On the other hand, an oxygenated methylene attaching to C-3 in **9** is absent in **6**. The ¹H–¹H COSY correlations of H-4/H-5/H-6/H-7 and HMBC correlations of H-4/C-3, C-9, C-10, H-5/C-9, H-7/C-8, C-9, C-10 indicate the presence of a six-membered ring as shown (Fig. 1). Compound **6** was isolated as a racemic mixture. Chiral HPLC separation afforded **6a** and **6b**. The absolute configuration of **6b** was determined to be 6*R* by using computational methods (Supplementary Information). Thus far, the structure of **6** was identified and named as ganocochlearine G (See Figs. 2 and 3).

Compound **7** has a molecular formula C₁₅H₁₃NO₂ (ten degrees of unsaturation), as deduced by analysis of its positive HRESIMS at m/z 240.1022 [M + H]⁺ (calcd for 240.1019, C₁₅H₁₄NO₂). ¹³C NMR and DEPT spectra. The ¹H NMR spectrum of **8** contains a typical ABX coupling system [δ_{H} 6.83 (1H, d, $J = 8.7$ Hz, H3'), δ_{H} 6.71 (1H, d, $J = 8.7, 3.0$ Hz, H-4'), δ_{H} 7.37 (1H, d, $J = 3.0$ Hz, H-6')], and one proton signal at δ_{H} 8.28 (1H, s, H-7). The ¹³C NMR and DEPT spectra give 15 carbons ascribe to one methyl, two methylene, five methine (four olefinic and

one oxygenated aliphatic), and seven olefinic quaternary carbons. These data disclose that compound **7** is an analogue of lucidimine D [10], differing in that a CH(OCH₃)₂ group at C-6 of lucidimine D is replaced by a methyl of **7**. HMBC correlations of H₃–10/C-6, C-7, C-9 and H-7/C-10 strongly support the above conclusion. Like compound **6**, **7** is partially racemic. Successive chiral HPLC provided enantiomers **7a** and **7b**. Of which, the absolute configuration of **7a** was assigned as 6*R* by ECD calculations (Supplementary Information). Taken together, the structure of **7** was identified and named as ganocochlearine H.

The molecular composition of **8** was determined to be C₁₅H₁₃NO₃ (ten degrees of unsaturation) by analysis of its positive HRESIMS, ¹³C NMR and DEPT spectra. Similar NMR data between **8** and **9** reveal that they are analogues. The difference is mainly at rings C and D. The ¹H–¹H COSY spectrum displays interactions between H-5 (δ_{H} 5.58)/H-6/H-7. HMBC correlations of H-6'/C4, H-5/C4, C-8 (δ_{C} 135.1), C-9 (δ_{C} 150.8), H-7/C-8, C-9 indicate that ring D is connected with C4 via C4–C-8. In addition, HMBC correlations of H-10/C-2 (δ_{C} 149.6), C-3 and H-2 (δ_{H} 8.39)/C-10 indicate that C-3 is attached to C-2. In addition to rings A, B, and D, two double bonds between C-3 and C4, C-8 and C-9, accounting for 8 degrees of unsaturation, the olefinic nature of C-2 and the requirement of remaining 2 degrees of unsaturation suggests the presence of a C=N group and an additional ring as shown, in accordance with biogenetic origin of such type of alkaloids. Compound **8** is partially racemic, which was subjected to chiral HPLC to afford a pair of enantiomers (**8a** and **8b**). The absolute configuration of **8a** was assigned as 4*R* by utilizing computational methods (Supplementary Information). Therefore, the structure of compound **8**, named as ganocochlearine I, was determined.

Known compounds were respectively identified as ganoapplanatumine B (**2**) [6], sinensine E (**9**) [11], and lucidimine C (**10**) [10] by comparison of their NMR data with literature data. However, it was found that these three compounds are all racemic or partially racemic. Successive chiral HPLC separation afforded their respective enantiomers (Table 2). For compounds **9** and **10**, computational methods were used to clarify their stereochemistry at the chiral center. The absolute configurations of **9b** and **10a** were therefore assigned as 6*S* and 6*R*, respectively (Supplementary Information). For compound **2**, there exist two chiral carbons in the structure, theoretically, there should have two pairs of diastereoisomers. Fortunately, we obtained two enantiomers from racemic **2**. As previously described [6], ECD calculations were not helpful for differentiating the absolute configurations of diastereoisomers due to ECD curves of ganoapplanatumine B and *epi*-ganoapplanatumine B were quite similar.

Alkaloids similar to the present findings have been characterized from *Ganoderma* [6,9–11]. Investigations on the biosynthetic pathway of this class of alkaloids are of interest. To be associated with meroterpenoids isolated from this genus, we proposed that these alkaloids are probably biosynthesized via meroterpenoidal intermediates

and threonine. In detail, hydrolysis of applanatumol Z2 forms monomer A [6], where the meroterpenoid A reacts with threonine to give intermediate B. Compound 7 can be formed by nucleophilic addition and decarboxylation of B. The biosynthetic pathway for the other compounds are given in Supplementary Information.

Notably, all the isolates in this study are racemic or partially racemic mixtures. As mentioned above, similar structures were also characterized from *Ganoderma* species by the other investigators [6,9–13]. Unfortunately, racemic nature of such alkaloids was not pointed out. Based on our observations, we highly suspect that the isolated alkaloids in the previous reports are also racemic or partially racemic, which could be concluded from their small values of optical rotations. As such, our present results might subvert the previous reports on this class of alkaloids. In general, natural compounds as an enzyme-catalyzed consequence should be optically pure, the reason for racemic nature of such alkaloids might be associated with their minor abundance in Nature. Although these alkaloids are from similar or same biogenetic origins, the acetonide residue is incorporated into the backbone of **1**, which makes the structure of **1** unusual. In contrast to compounds **1–5** and **7–10**, migration of a methyl followed by ring extension in **6** makes this structure novel. Increasing meroterpenoids present in the genus *Ganoderma* with a 1,4-dihydroxyl hybridized with a terpenoid have been characterized by us [3–7,14–17]. Without exception, all the alkaloids isolated in the present study bear a common 1,4-dihydroxyl motif, which is indeed an intermediate of shikimic acid pathway, prompting us to speculate that meroterpenoids and alkaloids in *Ganoderma* fungi might partly share a common biogenetic pathway.

Multiple factors are involved in renal fibrosis. Whereas, activation and proliferation of interstitial fibroblasts and consequent accumulation of extracellular matrix components is implicated in renal fibrogenesis. Therefore, interstitial fibroblasts are considered as target cells for the treatment of renal fibrosis. Inhibiting proliferation of fibroblasts might be beneficial for the prevention and treatment of renal fibrosis [18–20]. In this study, the selected compounds were evaluated for their inhibition on proliferation of normal (with exception of **7a/7b**) or TGF- β 1-induced (with exception of **3a/3b** and **7a/7b**) fibroblasts NRK-49F. Unfortunately, it was found that all the tested compounds are inactive towards these assays even at the concentration of 20 μ M (data not shown), indicating that these alkaloids are neither effective against diseased cells nor toxic towards normal cells.

Acknowledgments

We thank National Science Fund for Distinguished Young Scholars (81525026), NSFC-Joint Foundation of Yunnan Province (U1202222), National Natural Science Foundation of China (21472199), and a project from State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany (P2016-ZZ02) for their financial supports.

Appendix A. Supplementary data

Supplementary data to this article can be found online at <http://dx.doi.org/10.1016/j.fitote.2016.11.011>.

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