

## Two novel glucosyl-fused compounds from *Curculigo crassifolia* (Hypoxidaceae)

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**Abstract**—Two novel glucosyl-fused compounds, namely crassifoside E (**1**) and crassifoside F (**2**), were isolated from the rhizomes of *Curculigo crassifolia*. Their structures were elucidated by spectral analysis. Compounds **1** and **2** displayed ACE inhibitory activity (angiotensin-converting enzyme) with an IC<sub>50</sub> value of 10 and 8.5 µg/mL, respectively, by inspecting their fluorescent light.  
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*Curculigo crassifolia* (Bak.) Hook. f. (Hypoxidaceae) is distributed in the western and southern regions of China. It is called ‘rong ye xian mao’ in Chinese and local people use the rhizomes as a folk medicine to treat child pneumonitis.<sup>1</sup> Past studies on its chemical constituents have resulted in the isolation of four new phenolic compounds.<sup>2</sup> Being interested in exploring biologically active substances from this folk medicine, we reinvestigated this plant. This study led to the isolation of two new glucosyl-fused compounds with novel skeletons, crassifoside E (**1**), and crassifoside F (**2**).

The rhizomes of *C. crassifolia* were collected in Eshan Prefecture, Yunnan Province, PR China, in October 2002. These dried rhizomes were extracted and fractionated as described previously.<sup>2</sup> Fraction 4 was chromatographed on silica gel (CHCl<sub>3</sub>–MeOH), then repeated Sephadex LH-20 gel (EtOH, EtOH–acetone) to result in the isolation of crassifoside E (**1**, 60 mg) and crassifoside F (**2**, 80 mg).

Crassifoside E (**1**) was isolated as brown powder; its molecular formula C<sub>25</sub>H<sub>26</sub>O<sub>11</sub> was established by <sup>13</sup>C NMR and HRFABMS data, corresponding to an unsaturation index of 13. The IR spectrum of compound **1** showed absorption for hydroxyl (3440 cm<sup>−1</sup>). The <sup>1</sup>H NMR<sup>3</sup> spectrum exhibited signals for three methyl pro-

tons [ $\delta_{\text{H}}$  1.46 (t,  $J = 7.30$  Hz, H-12)], four methylene protons [ $\delta_{\text{H}}$  3.30, 3.45, 3.83, 3.95 (each 1H, m, H<sub>a</sub>-5, H<sub>b</sub>-5, H<sub>a</sub>-13, and H<sub>b</sub>-13, respectively)], two oxymethine protons [ $\delta_{\text{H}}$  4.42 (dd,  $J = 11.32, 3.04$  Hz, H-6), 5.19 (s, H-7)], three low-field aromatic protons [ $\delta_{\text{H}}$  7.72, 7.35, 7.77 (each 1H, s, H-1, H-3 and H-11, respectively)] and those protons of one glucosyl moiety [ $\delta_{\text{H}}$  3.40–3.91 (6H, Glc-H-2 → H-6)] except for one anomeric proton [ $\delta_{\text{H}}$  4.95 (d,  $J = 8.10$  Hz, Glc-H-1)]. The <sup>13</sup>C NMR and DEPT data<sup>3</sup> showed one CH<sub>3</sub>, two CH<sub>2</sub>, five CH, and 11 C signals together with six carbons of one glucosyl moiety, including those for three oxygen-bearing carbons [ $\delta_{\text{C}}$  70.1 (C-13), 71.3 (C-7), and 74.7 (C-6)], five oxygen-bearing olefinic carbons [ $\delta_{\text{C}}$  122.6 (C-9), 145.8 (C-8), 146.7 (C-10), 146.9 (C-2), and 148.7 (C-4)] and nine olefinic carbons [ $\delta_{\text{C}}$  107.0 (C-3), 108.0 (C-1), 108.4 (C-11), 117.4 (C-7a), 118.4 (C-4a), 122.8 (C-7b), 123.0 (C-4b), 125.9 (C-6b) and 146.2 (C-6a)] (see Table 1).

The <sup>1</sup>H and <sup>13</sup>C NMR spectrum indicated the presence of a glucosyl moiety. The anomeric proton signal appeared as a doublet at  $\delta_{\text{H}}$  4.95 ( $J = 8.10$  Hz). Incorporating <sup>13</sup>C NMR chemical shifts it showed the presence of a  $\beta$ -D-glucosyl unit. All the carbons of the glucosyl moiety were assigned through direct <sup>1</sup>H–<sup>13</sup>C correlations in the HMQC spectrum and were situated between  $\delta$  62.5 and 81.6 except for that at the anomeric position, which was assigned to the signal at  $\delta$  94.3.

Long range <sup>1</sup>H–<sup>13</sup>C correlations (HMBC) (Fig. 2) observed as H-5/C-4a, C-4, C-4b, C-6, C-6a, C-7; H-7/

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**Table 1.**  $^1\text{H}$  and  $^{13}\text{C}$  NMR, and HMBC data for compound **1**<sup>a</sup>

Position	$\delta_{\text{H}}$	$\delta_{\text{C}}$	HMBC (H $\rightarrow$ C)
1	7.72 s	108.0 d	2, 3, 4b
2		146.9 s	
3	7.35 s	107.0 d	1, 2, 4
4		148.7 s	
4a		118.4 s	
4b		123.0 s <sup>b</sup>	
5	3.30 m (3.29–3.31) 3.46 m (3.45–3.47)	23.3 t	4, 4a, 4b, 6, 6a, 7
6	4.42 dd (3.04, 11.32)	74.7 d	Glc-C-1
6a		146.2 s	
6b		125.9 s	
7	5.19 s	71.3 d	6, 6a, 7a, 7b, Glc-C-2
7a		117.4 s	
7b		122.8 s <sup>b</sup>	
8		145.8 s	
9		122.6 s <sup>b</sup>	
10		146.7 s	
11	7.77 s	108.4 d	6b, 7a, 7b, 9
12	1.46 t (7.30)	16.0 q	13
13	3.95 m (3.93–3.97) 3.83 m (3.82–3.85)	70.1 t	4
Glc.			
1	4.95 d (8.10)	94.3 d	6, Glc-C-2
2	3.40 dd (8.10, 8.55)	81.6 d	7
3	3.63 dd (8.15, 8.60)	74.8 d	
4	3.49 m (overlapped)	72.1 d	
5	3.48 m (overlapped)	79.7 d	
6	3.91 d (11.95) 3.74 dd (4.70, 11.95)	62.5 t	

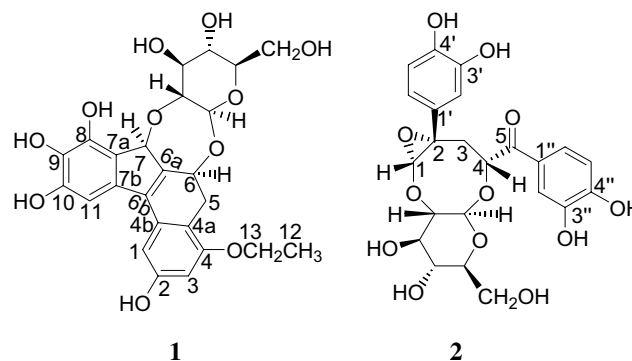
<sup>a</sup>  $J$  in hertz, in  $\text{CD}_3\text{OD}$ ,  $^1\text{H}$  and  $^{13}\text{C}$  NMR, and HMBC at 400, 100, and 500 MHz, respectively.

<sup>b</sup> Values may be interchangeable.

C-6, C-6a, C-7a, C-7b; and H-11/C-6b, C-7a, C-7b, C-9, respectively, confirmed the presence of 5, 6-2H-benzo[4a,4b]fluorene in **1**. The linkage of the ethoxyl to C-4 also can be resolved by long range  $^1\text{H}$ – $^{13}\text{C}$  correlations of H-13/C-4 observed, and can be further confirmed by NOESY correlations of H-13/H-3, H-5; and H-12/H-3. The long range  $^1\text{H}$ – $^{13}\text{C}$  correlations of Glc-H-1/C-6 and Glc-H-2/C-7 observed, confirmed that the fused glucosyl moiety was Glc-C-1 ether-linked to C-6 and Glc-C-2 to C-7. So compound **1** possessed a unique structural feature of glucosyl-fused 5,6-2H-benzo[4a,4b]fluorene in addition to a diox-seven-membered ring.

NOESY correlations of H-6/H-7, H-6/Glc-H-1 indicated the *cis* relationship of H-6 and H-7, incorporating the known stereochemistry of the  $\beta$ -D-glucosyl unit would require 6*S* and 7*S* stereochemistry in **1**. Therefore, the structure of crassifoside E was deduced, as shown in Figure 1, with a novel skeleton.

Crassifoside F (**2**) was isolated as white powder; its molecular formula  $\text{C}_{23}\text{H}_{24}\text{O}_{12}$  was established by  $^{13}\text{C}$  NMR and HRFABMS data, corresponding to an unsaturation index of 12. The IR spectrum of compound **2** showed absorption for hydroxyl ( $3429\text{ cm}^{-1}$ ) and conjugated carbonyl ( $1651\text{ cm}^{-1}$ ). The  $^1\text{H}$  NMR spectrum<sup>4</sup> exhibited signals for two methylene protons [ $\delta_{\text{H}}$  2.40 (dd,  $J = 12.00, 11.50\text{ Hz}$ ,  $\text{H}_{\text{a-3}}$ ), 3.25 (dd,  $J = 7.65,$

**Figure 1.** Structures of compounds **1** and **2**.

7.30 Hz,  $\text{H}_{\text{b-3}}$ ], two oxymethine protons [ $\delta_{\text{H}}$  4.36 (m, H-4), 5.52 (s, H-1)]. The  $^1\text{H}$  NMR spectrum<sup>4</sup> of **2** also exhibits six aromatic protons. Three of them were assigned to H-2' [ $\delta_{\text{H}}$  6.98 (d,  $J = 2.10\text{ Hz}$ )], H-5' [ $\delta_{\text{H}}$  6.69 (d,  $J = 8.55\text{ Hz}$ )], and H-6' [ $\delta_{\text{H}}$  6.74 (dd,  $J = 8.55, 2.10\text{ Hz}$ )], which suggested the existence of 1,3,4-trisubstituted benzene ring. The remaining three aromatic protons were assigned to H-2'' [ $\delta_{\text{H}}$  7.41 (d,  $J = 1.75\text{ Hz}$ )], H-5'' [ $\delta_{\text{H}}$  6.63 (d,  $J = 8.50\text{ Hz}$ )], and H-6'' [ $\delta_{\text{H}}$  7.36 (dd,  $J = 8.50, 1.75\text{ Hz}$ )] in another 1,3,4-trisubstituted benzene ring, in which H-2'' and H-6'' were shifted downfield due to an *ortho* carbonyl group (IR  $\nu_{\text{CO}}$   $1651\text{ cm}^{-1}$  and  $\delta_{\text{CO}}$  196.7). The  $^{13}\text{C}$  NMR and DEPT data<sup>4</sup> showed one methylene carbon [ $\delta_{\text{C}}$  36.0 (C-3)],

**Table 2.**  $^1\text{H}$  and  $^{13}\text{C}$  NMR, and HMBC data for compound **2**<sup>a</sup>

Position	$\delta_{\text{H}}$	$\delta_{\text{C}}$	HMBC (H $\rightarrow$ C)
1	5.25 s	97.8 d	2, 3, Glc-C-2
2		93.3 s	
3	3.25 dd (7.30, 7.65) 2.40 dd (11.50, 12.00)	36.0 t	1, 2, 4, 5, 1'
4	4.36 m (4.34–4.38)	77.4 d	2, 3, Glc-C-1
5		196.7 s	
1'		135.4 s	
2'	6.98 d (2.10)	113.0 d	2
3'		145.8 s <sup>b</sup>	
4'		146.4 s	
5'	6.69 d (8.55)	116.4 d	1', 4', 6'
6'	6.74 dd (2.10, 8.55)	117.1 d	2
1''		127.3 s	
2''	7.41 d (1.75)	119.1 d	5
3''		145.4 s <sup>b</sup>	
4''		151.6 s	
5''	6.63 d (8.50)	115.1 d	1'', 4'', 6''
6''	7.36 dd (1.75, 8.50)	126.2 d	5
Glc.			
1	4.66 d (8.10)	93.9 d	4
2	3.12 dd (8.10, 9.00)	76.9 d	1
3	3.58 dd (8.10, 8.95)	74.6 d	
4	3.30 m (3.29–3.31)	71.7 d	
5	3.38 m (3.37–3.39)	79.5 d	
6	3.84 d (11.55) 3.68 dd (4.70, 11.95)	62.4 t	

<sup>a</sup>  $J$  in hertz, in  $\text{CD}_3\text{OD}$ ,  $^1\text{H}$ , and  $^{13}\text{C}$  NMR and HMBC at 400, 100, and 500 MHz, respectively.

<sup>b</sup> Values may be interchangeable.

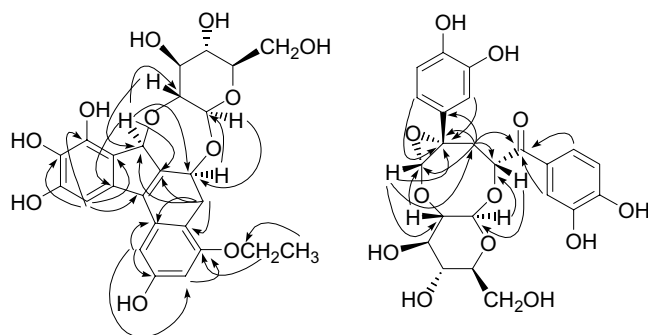


Figure 2. Key HMBC correlations of compounds **1** and **2**.

two aliphatic oxymethine carbons [ $\delta_C$  97.8 (C-1), 77.4 (C-4)], one aliphatic oxygen-substituted quaternary carbon [ $\delta_C$  93.3 (C-2)], one conjugated carbonyl carbon [ $\delta_C$  196.7 (C-5)], four oxygen bearing aromatic carbons [ $\delta_C$  145.4 (C-3''), 145.8 (C-3'), 146.4 (C-4'), and 151.6 (C-4''), six aromatic CH [ $\delta_C$  113.0 (C-2'), 115.1 (C-5''), 116.4 (C-5'), 117.1 (C-6'), 119.1 (C-2''), and 126.2 (C-6'')], two aromatic quaternary carbons [ $\delta_C$  127.3 (C-1''), 135.4 (C-1')] together with six carbons of one glucosyl moiety (see Table 2).

The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectrum indicated the presence of a glucosyl moiety. The anomeric proton signal appeared as a doublet at  $\delta_H$  4.66 ( $J = 8.10$  Hz). Incorporating  $^{13}\text{C}$  NMR chemical shifts, it showed the presence of a  $\beta$ -D-glucosyl unit. All the carbons of the glucosyl moiety were assigned through direct  $^1\text{H}$ – $^{13}\text{C}$  correlations in the HMQC spectrum and were situated between  $\delta$  62.4 and 79.5 except for that at the anomeric position, which was assigned to the signal at  $\delta$  93.9.

Long range  $^1\text{H}$ – $^{13}\text{C}$  correlations (HMBC) (Fig. 2) observed of H-1/C-2, C-3, Glc:C-1; H-3/C-1, C-2, C-4, C-5, C-1'; H-4/C-2, C-3; H-2'/C-2, H-6'/C-2, H-2''/C-5, and H-6''/C-5, respectively, together with  $^1\text{H}$ – $^1\text{H}$  COSY correlation of H-3/H-4, revealed the presence of  $-\text{C}_1-\text{C}_2\text{Ph}-\text{C}_3-\text{C}_4-\text{C}_5-\text{Ph}$  in **2**. Long range  $^1\text{H}$ – $^{13}\text{C}$  correlations was also observed for Glc:H-1/C-4 and Glc:H-2/C-1, which confirmed that the fused glucosyl moiety was Glc:C-1 ether-linked to C-4 and Glc:C-2 to C-1. So the structure of **2** possessed a novel glucosyl-fused diox-eight-membered ring moiety. According to the molecular formula ( $\text{C}_{23}\text{H}_{24}\text{O}_{12}$ ), an unsaturation index of 12 and the low-field chemical shift of C-1 and C-2, we may conclude an oxygen atom was linked to C-1 and C-2 to form a three-membered ring.

NOESY correlations of H-1/H-4, H-1/Glc:H-2 indicated the *cis* relationship of H-1 and H-4. Incorporating the known stereochemistry of the  $\beta$ -D-glucosyl unit, we may conclude the stereochemistry of **2** would require 1*R*, 2*S*, and 4*R*. Therefore, the structure of crassifoside F was deduced as shown in Figure 1, with a novel skeleton.

The two novel glucosyl-fused compounds were tested for the ACE inhibitor activities by inspecting fluorescent light, and captopril were used as positive control substance with an  $\text{IC}_{50}$  value of 27.5  $\mu\text{g}/\text{mL}$ . crassifoside E (**1**) and crassifoside F (**2**) displayed dose dependent inhibition with an  $\text{IC}_{50}$  value of 10 and 8.5  $\mu\text{g}/\text{mL}$ , respectively, but less than the positive control captopril ( $\text{IC}_{50}$ , 27.5  $\mu\text{g}/\text{mL}$ ).

To our knowledge, **1** represents the first natural occurrence of a glucosyl-fused 5,6-2*H*-benzo[4*a*,4*b*]fluorene in addition to a diox-seven-membered ring skeleton, **2** represents the first natural occurrence of a glucosyl-fused diox-eight-membered ring skeleton. The trivial name, crassifoside E and crassifoside F, are made after their plant origin.

### Acknowledgements

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### References and notes

1. Institutum Botanicum Kunmingense, Academiae Sinicae, *Flora Yunnanica*, Science Press: Beijing, 1995; Vol. 6, p 819.
2. Li, N.; Chen, J. J.; Zhou, J. *Helv. Chim. Acta* **2004**, *87*, 845–850.
3. Crassifoside E (**1**). Brown amorphous powder,  $[\alpha]_D^{24} +23.1$  (*c* 0.2, MeOH). IR (KBr)  $\nu_{\text{max}}$   $\text{cm}^{-1}$ : 3440, 2926, 2369, 1623, 1558, 1507, 1114, 1035, 615, 469; UV (MeOH)  $\lambda_{\text{max}}$  nm: 224 (4.42), 258 (4.75), 294 (4.46), 347 (3.31). FAB-MS (–)  $m/z$ : 501 [ $\text{M}^+ - \text{H}$ ], HRFAB-MS (–)  $m/z$ : Found 501.1402. Calcd 501.1397;  $\text{C}_{25}\text{H}_{25}\text{O}_{11}$  [ $\text{M}^+ - \text{H}$ ].  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR data see Table 1.
4. Crassifoside F (**2**). White amorphous powder,  $[\alpha]_D^{28} +112.9$  (*c* 0.2, MeOH). IR (KBr)  $\nu_{\text{max}}$   $\text{cm}^{-1}$ : 3429, 2925, 1651, 1606, 1558, 1521, 1436, 1363, 1294, 1131, 1032, 873, 591; UV (MeOH)  $\lambda_{\text{max}}$  nm: 196 (4.09), 206 (4.50), 232 (4.18), 283 (3.99), 315 (3.86). FAB-MS (–)  $m/z$ : 491 [ $\text{M}^+ - \text{H}$ ], HRFAB-MS (–)  $m/z$ : Found 491.1176. Calcd 491.1190;  $\text{C}_{23}\text{H}_{23}\text{O}_{12}$  [ $\text{M}^+ - \text{H}$ ].  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR data see Table 2.