

# 10-Acetyl-10-hydroxy-1,6-dimethyl-1,2,10,11-tetrahydrophenanthro[1,2-*b*]-furan-11-one

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## Key indicators

Single-crystal X-ray study  
 $T = 295\text{ K}$   
 Mean  $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$   
 $R$  factor = 0.065  
 $wR$  factor = 0.179  
 Data-to-parameter ratio = 8.5

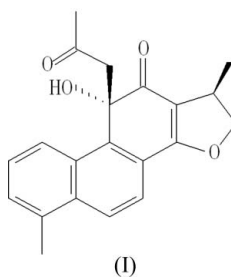
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the crystal structure of the title compound,  $\text{C}_{21}\text{H}_{20}\text{O}_4$ , an intramolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond is found between a hydroxyl H atom and a carbonyl O atom.

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## Comment

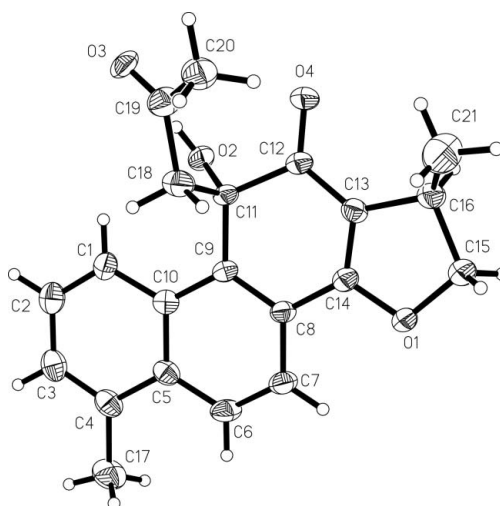
The title compound (danshenol A), (I), was extracted from *Salvia yunnanensis* (Xu *et al.*, 2006) and recrystallized from ethanol. As it shows strong aldose reductase (AR) inhibitory activity (Tezuka *et al.*, 1997) we have determined its structure.



The molecular structure of (I) is shown in Fig. 1. All four rings are coplanar and an intramolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond is found between a hydroxyl H atom and a carbonyl O atom (Fig. 2).

## Experimental

The dried and powdered roots of *S. yunnanensis* were extracted three times over 24 h with  $\text{Me}_2\text{CO}$  at room temperature and afterwards the



**Figure 1**  
 The molecular structure of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 30% probability level.

solvent was removed *in vacuo*. The residue was subjected to column chromatography over DM-130 porous resin and eluted with MeOH–H<sub>2</sub>O (1:1) and 90% MeOH–H<sub>2</sub>O (9:1). The residue of the MeOH–H<sub>2</sub>O (9:1) fraction was partitioned between H<sub>2</sub>O and EtOAc. The EtOAc part was subjected to silica-gel column chromatography. Mixtures of petroleum ether/EtOAc (1:0, 9:1, 8:2, 7:3, 6:4, 5:5 and 0:1) of increasing polarity were used as eluents. Seven fractions were collected and combined by monitoring with thin-layer chromatography. In this procedure, a mixture of danshenol A and danshenol C was obtained from the third fraction by silica-gel column chromatography using petroleum ether–CHCl<sub>3</sub>–EtOAc (70/25/5) as eluents. Both compounds were separated by semi-preparative high-performance liquid chromatography using 85% MeOH–H<sub>2</sub>O as eluant. Crystals of Danshenol A suitable for data collection were obtained by slow evaporation of an ethanol solution over a period of two weeks.

#### Crystal data

C <sub>21</sub> H <sub>20</sub> O <sub>4</sub>	$Z = 2$
$M_r = 336.37$	$D_x = 1.322 \text{ Mg m}^{-3}$
Monoclinic, $P2_1$	Mo $K\alpha$ radiation
$a = 7.5180 (15) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$b = 6.7150 (13) \text{ \AA}$	$T = 295 (2) \text{ K}$
$c = 17.161 (3) \text{ \AA}$	Column, colorless
$\beta = 102.75 (3)^\circ$	$0.30 \times 0.15 \times 0.10 \text{ mm}$
$V = 845.0 (3) \text{ \AA}^3$	

#### Data collection

MAC DIP 2030K diffractometer	1930 independent reflections
$\omega$ scans	1894 reflections with $I > 2\sigma(I)$
Absorption correction: none	$R_{\text{int}} = 0.048$
7044 measured reflections	$\theta_{\text{max}} = 27.2^\circ$

#### Refinement

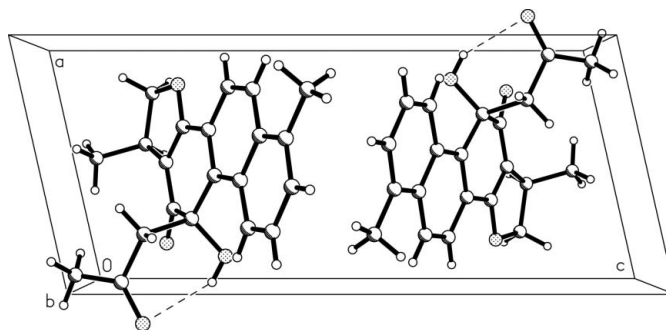
Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.1047P)^2 + 0.3521P]$
$R[F^2 > 2\sigma(F^2)] = 0.065$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.179$	$(\Delta/\sigma)_{\text{max}} = 0.002$
$S = 1.09$	$\Delta\rho_{\text{max}} = 0.28 \text{ e \AA}^{-3}$
1930 reflections	$\Delta\rho_{\text{min}} = -0.22 \text{ e \AA}^{-3}$
227 parameters	Extinction correction: <i>SHELXL97</i>
H-atom parameters constrained	Extinction coefficient: 0.081 (15)

**Table 1**

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O2-H2A\cdots O3$	0.82	2.24	2.969 (6)	148

The methyl H atoms were placed in calculated positions, with C–H = 0.96  $\text{\AA}$ , allowed to rotate but not tip and were refined using a



**Figure 2**

The molecular packing of the title compound, viewed along the  $b$  axis. Hydrogen bonds are shown as dashed lines.

riding model, with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ . All other C–H H atoms were placed in geometrically idealized positions with C–H = 0.92–0.98  $\text{\AA}$  and were refined using a riding model with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . The O–H H atom was placed in an ideal position such that the O–H vector points in the direction of the nearest acceptor atom and afterwards it was refined using a riding model with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$ . In the absence of significant anomalous scattering effects, Friedel pairs were averaged.

Data collection: *DENZO* (Otwinowski & Minor, 1997); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *SCALEPACK*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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#### References

- Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Tezuka, Y., Kasimu, R., Basnet, P., Namba, T. & Kadota, S. (1997). *Chem. Pharm. Bull.* **45**, 1306–1311.
- Xu, G., Peng, L. Y., Lu, L., Weng, Z. Y., Zhao, Y., Li, X. L., Zhao, Q. S. & Sun, H. D. (2006). *Planta Med.* **72**, 84–86.