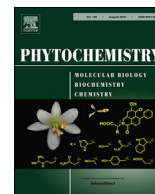




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journal homepage: [www.elsevier.com/locate/phytochem](http://www.elsevier.com/locate/phytochem)Amides and neolignans from the aerial parts of *Piper bonii*Duo-Duo Ding<sup>a,1</sup>, Yue-Hu Wang<sup>a,1</sup>, Ya-Hui Chen<sup>b</sup>, Ren-Qiang Mei<sup>a</sup>, Jun Yang<sup>a</sup>,  
Ji-Feng Luo<sup>a</sup>, Yan Li<sup>c</sup>, Chun-Lin Long<sup>a,d,\*</sup>, Yi Kong<sup>b,\*\*</sup><sup>a</sup> Key Laboratory of Economic Plants and Biotechnology, Yunnan Key Laboratory for Wild Plant Resources, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650201, People's Republic of China<sup>b</sup> School of Life Science & Technology, China Pharmaceutical University, Nanjing 210009, People's Republic of China<sup>c</sup> State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650201, People's Republic of China<sup>d</sup> College of Life and Environmental Sciences, Minzu University of China, Beijing 100081, People's Republic of China

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## ABSTRACT

Six amides, piperbonamides A–F, three neolignans piperbonins A–C, and 11 known compounds were isolated from the aerial parts of *Piper bonii* (Piperaceae). The structures of piperbonamides A–F and piperbonins A–C were elucidated based on the analysis of 1D and 2D NMR and MS data. Piperbonin A, (+)-*trans*-acuminatin, (+)-*cis*-acuminatin, (+)-kadsurenone, and pipernonaline showed weak activity against platelet aggregation with IC<sub>50</sub> values of 118.2, 108.5, 90.02, 107.3, and 116.3 μM, respectively, as compared with the positive control, tirofiban, with an IC<sub>50</sub> value of 5.24 μM. Piperbonamides A–F were inactive against five tumor cell lines at concentrations up to 40 μM.

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## 1. Introduction

Antiplatelet therapy is a milestone in the primary and secondary prevention of atherothrombotic diseases. Abundant recent data support that traditional Chinese herbs and their active constituents may be a good alternative and complementary choice for new antiplatelet drugs (Chen et al., 2015; Zhang et al., 2015). For example, ginkgolide B from the leaf of *Ginkgo biloba* L. significantly inhibited rabbit platelet aggregation induced by platelet-activating factor (PAF) with an IC<sub>50</sub> value of 0.128 μM (Hu et al., 2000).

The genus *Piper* (Piperaceae) is a medicinally important group of plants consisting of approximately 2000 species worldwide. There are approximately 60 species distributed in the tropical areas of the People's Republic of China, of which approximately 30 species have been used as traditional Chinese medicine (Tseng et al., 1999; Wang et al., 2014b). Some *Piper* species, including *Piper betle* L., *Piper*

*boehmeriifolium* (Miq.) Wall. ex C. DC., *Piper hongkongense* C. DC., *Piper kadsura* (Choisy) Ohwi, *Piper laetispicum* C. DC., *Piper macro-podium* C. DC., *Piper mutabile* C. DC., *Piper puberulilimum* C. DC., *Piper sarmentosum* Roxb., *Piper wallichii* (Miq.) Hand.-Mazz., and *Piper yunnanense* Tseng, are used for promoting blood circulation (“Huoxue” in Chinese) (Editorial Board of “Zhonghua Bencao”, 1999). Traditional Chinese medicines with functions of “Huoxue” or “Huayu” (removing blood stasis) are claimed to be useful in antiplatelet therapies and in the treatment of thrombotic diseases (Chen et al., 2015; Liu et al., 2012). Thus, from herbs used to promote blood circulation and/or remove blood stasis, active compounds against platelet aggregation might be obtained. For example, salvianolic acids A and B from *Salvia miltiorrhiza* Bunge (Huang et al., 2010; Liu et al., 2014), the dihydrochalcone loureirin A from *Dracaena cochinchinensis* (Lour.) SC Chen (Hao et al., 2015; Xin et al., 2011), and the alkaloid tetramethylpyrazine from *Ligusticum striatum* DC. were reported to possess potent antiplatelet activity (Sheu et al., 1997; Wang et al., 2013).

Amides, phenylpropanoids, lignans, and neolignans are the major constituents of *Piper* plants (Kato and Furlan, 2007; Parmar et al., 1997; Scott et al., 2008). These compounds showed various biological activities, including antiplatelet effects (Chen et al., 2014,

\* Corresponding author.

\*\* Corresponding author.

E-mail addresses: [long@mail.kib.ac.cn](mailto:long@mail.kib.ac.cn) (C.-L. Long), [yikong668@163.com](mailto:yikong668@163.com) (Y. Kong).<sup>1</sup> These authors contributed equally to this work.

2013, 2007; Fontenele et al., 2009; Iwashita et al., 2007; Li et al., 2007; Park et al., 2007; Shi et al., 2015). *Piper bonii* C. DC. is a climbing liana distributed mainly in the Chinese provinces of Guangxi, Hainan, and Yunnan, and in northern Vietnam (Tseng et al., 1999). However, the chemical constituents of *P. bonii* are unknown. In continuing efforts to search for bioactive constituents from *Piper* plants (Liu et al., 2015; Tang et al., 2011; Yang et al., 2013a, 2013b), the results of the analysis of compounds from the aerial parts of *P. bonii*, and the bioactivity of several of these compounds, are reported.

## 2. Results and discussion

Chromatographic separation of the MeOH soluble part of the aerial parts of *P. bonii* led to isolation of 20 compounds, including nine new compounds (**1–9**, Fig. 1) and 11 known compounds (**10–20**).

The molecular formula of piperbonamide A (**1**),  $C_{27}H_{33}NO_3$ , was determined by  $^{13}C$  NMR data and an HRESIMS ion at  $m/z$  442.2352  $[M+Na]^+$  (calcd for  $C_{27}H_{33}NNaO_3$ , 442.2353), and required 12 indices of hydrogen deficiency. Its IR spectrum showed absorption peaks for a secondary amide at 3396 and 1657  $cm^{-1}$ . Absorptions at 1627, 1610, 1596, 1532, 1514, and 1449  $cm^{-1}$  suggested the presence of phenyl rings and double bonds. The  $^1H$  NMR data (Table 1) indicated two *p*-disubstituted phenyl rings [ $\delta_H$  6.96 (2H, br d,  $J = 8.6$  Hz, H-2',6') and 6.68 (2H, br d,  $J = 8.6$  Hz, H-3',5'); 7.03 (2H, d,  $J = 8.6$  Hz, H-2'',6'') and 6.70 (2H, d,  $J = 8.6$  Hz, H-3'',5'')] and three double bonds, including two *E* double bonds ( $J_{2,3} = 15.1$  Hz and  $J_{4,5} = 15.1$  Hz). The  $^{13}C$  NMR data (Table 1) exhibited eight methylenes, 14 methines, two  $sp^2$  quaternary carbons [ $\delta_C$  133.1 (C-1') and 131.3 (C-1'')], two oxygenated  $sp^2$  tertiary carbons [ $\delta_C$  156.9 (C-4') and 156.5 (C-4'')], and an amide carbonyl carbon ( $\delta_C$  169.1). Comparison of its NMR data with those of known compounds from other *Piper* plants species (Tang et al., 2011; Yang et al., 2013a) suggested that the compound might be an arylalkenylacyl amide.

$^1H$ - $^1H$  COSY correlations (Fig. 2) exhibited three partial structures comprising C-2 to C-11, C-12 to C-13, and C-7'' to C-8''. On the basis of the HMBC correlations (Fig. 2) from H-3 to C-1, H-12 to C-1', H<sub>2</sub>-13 to C-11, C-2' and C-6', the acyl fragment was confirmed to be a 13-(4-hydroxyphenyl)-2,4,11-tridecatrienacyl moiety. In addition, the HMBC correlation of H<sub>2</sub>-8'' to C-1'', H<sub>2</sub>-7'' to C-2'' and C-6'', confirmed the amido fragment as a 2-(4-hydroxyphenyl)ethyl-amino moiety. The *E* geometry of the C-11–C-12 double bond was established by comparison of the  $^{13}C$  NMR chemical shifts of the allylic carbons [ $\delta_C$  39.2 (C-13) and 33.4 (C-10)] with those in the *Z* analogue, sarmentosumol A [ $\delta_C$  32.5 (C-1') and 27.2 (C-4')] (Yang et al., 2013b), and the *E* analogue, marginatine [ $\delta_C$  39.0 (C-1') and

32.7 (C-4')] (Santos et al., 1998). Therefore, the structure of compound **1** (piperbonamide A) was defined as (2*E*,4*E*,11*E*)-*N*-[(4-hydroxyphenyl)ethyl]-13-(4-hydroxyphenyl)-2,4,11-tridecatrienamide.

The molecular formula of piperbonamide B (**2**),  $C_{27}H_{35}NO_3$ , was determined by analysis of  $^{13}C$  NMR and HRESIMS data. The NMR data of **2** (Table 1) resembled those of **1** and showed that the signals for a double bond in **1** were replaced by the signals for two methylene groups in **2**. Based on its 2D NMR correlations (Fig. S1, Supplementary data), the two double bonds of **2** were located at C-2 and C-11, respectively. The *E* geometry of the double bonds was determined by the *J* values (both  $J_{2,3} = 15.5$  Hz) of the olefinic protons and the  $^{13}C$  NMR chemical shifts of the allylic carbons [ $\delta_C$  39.2 (C-13) and 33.5 (C-10)]. Thus, compound **2** (piperbonamide B) was identified as (2*E*,11*E*)-*N*-[(4-hydroxyphenyl)ethyl]-13-(4-hydroxyphenyl)-2,11-tridecadienamide.

Piperbonamide C (**3**) gave a molecular formula of  $C_{29}H_{37}NO_3$ , as deduced by  $^{13}C$  NMR and HRESIMS data. A comparison of the NMR data of **3** with those of **1** (Table 1) indicated that there were signals for two additional methylene groups in **3**. On the basis of 2D NMR correlations (Fig. S1, Supplementary data), compound **3** (piperbonamide C) was identified as (2*E*,4*E*,13*E*)-*N*-[(4-hydroxyphenyl)ethyl]-15-(4-hydroxyphenyl)-2,4,13-pentadecatrienamide, of which three double bonds were determined to be in an *E*-configuration this being based on coupling constants ( $J_{2,3} = 15.1$  Hz and  $J_{4,5} = 15.1$  Hz) of the olefinic protons, as well as the  $^{13}C$  NMR chemical shifts of the allylic carbons [ $\delta_C$  39.2 (C-13) and 33.5 (C-10)].

The molecular formula of piperbonamide D (**4**),  $C_{26}H_{31}NO_4$ , was determined by  $^{13}C$  NMR data and an HRESIMS ion at  $m/z$  444.2144  $[M+Na]^+$  (calcd for  $C_{26}H_{31}NNaO_4$ , 444.2145), indicating 12 degrees of unsaturation. The  $^1H$  NMR data (Table 2) indicated one 1,2,4-trisubstituted phenyl ring [ $\delta_H$  6.89 (br s, H-2'), 6.75 (br d,  $J = 7.9$  Hz, H-6') and 6.73 (d,  $J = 7.9$  Hz, H-5')], one *p*-disubstituted phenyl ring [ $\delta_H$  7.04 (2H, d,  $J = 8.3$  Hz, H-2'',6'') and 6.78 (2H, d,  $J = 8.3$  Hz, H-3'',5'')], two *E* double bonds [ $\delta_H$  6.83 (dt,  $J = 15.2$ , 6.9 Hz, H-3) and 5.68 (d,  $J = 15.2$  Hz, H-2); 6.27 (d,  $J = 15.8$  Hz, H-11) and 6.03 (dt,  $J = 15.8$ , 7.1 Hz, H-10)], and one methylenedioxy group [ $\delta_H$  5.93 (2H, s)]. By comparing its NMR data with those of compounds **1–3**, compound **4** should also be an arylalkenylacyl amide.

The  $^1H$ - $^1H$  COSY correlations (Fig. S1, Supplementary data) exhibited two partial structures comprising C-2 to C-11 and C-7'' to C-8''. Based on the HMBC correlations (Fig. S1, Supplementary data) from H-3 to C-1, H-10 to C-1', H-11 to C-2' and C-6', H-2' to C-4', H-5' to C-3', and  $-OCH_2O-$  to C-3' and C-4', the acyl fragment was confirmed as a (2*E*,10*E*)-11-(3,4-methylenedioxyphenyl)-2,10-undecadienacyl moiety. In addition, the HMBC correlations of H<sub>2</sub>-

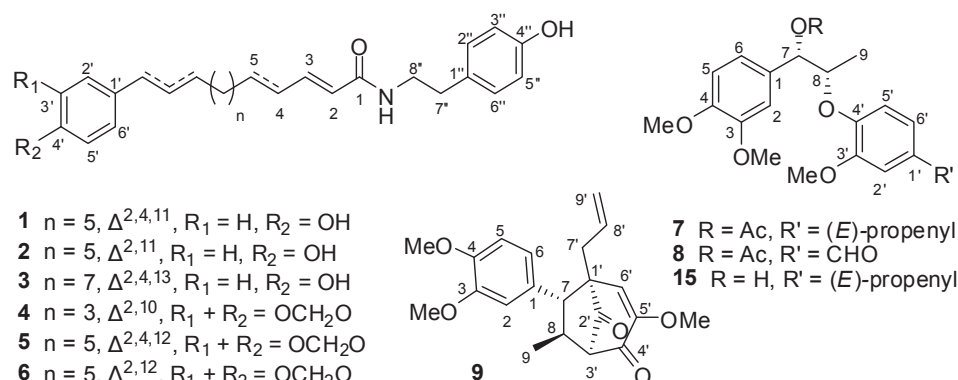


Fig. 1. Structures of compounds **1–9** and **15**.



**Table 2**  
<sup>1</sup>H (600 MHz) and <sup>13</sup>C NMR (100 MHz) NMR data of piperbonamides D–F (4–6) in CDCl<sub>3</sub>.

No.	Piperbonamide D (4)		Piperbonamide E (5)		Piperbonamide F (6)	
	$\delta_{\text{H}}$ (J in Hz)	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J in Hz)	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J in Hz)	$\delta_{\text{C}}$
1		166.3		166.6		166.3
2	5.68 d (15.2)	123.5	5.67 d (15.0)	121.5	5.68 d (15.3)	123.4
3	6.83 dt (15.2, 6.9)	145.3	7.19 dd (15.0, 9.8)	141.8	6.82 dt (15.3, 6.9)	145.4
4	2.14 m	32.1	6.09 dd (15.3, 9.8)	128.3	2.14 m	32.2
5	1.43 m	28.3	6.06 m	143.7	1.41 m	28.4
6	1.31 m	29.1	2.15 m	33.0	1.28 m	29.5 <sup>a</sup>
7	1.31 m	29.1	1.41 m	28.8	1.28 m	29.5 <sup>a</sup>
8	1.43 m	29.4	1.31 m	29.1	1.28 m	29.5 <sup>a</sup>
9	2.16 m	33.0	1.31 m	29.1	1.28 m	29.3 <sup>a</sup>
10	6.03 dt (15.8, 7.1)	129.5	1.43 m	29.5	1.44 m	29.6
11	6.27 d (15.8)	129.5	2.15 m	33.0	2.16, m	33.0
12			6.03 m	129.5	6.04 dt (15.7, 7.0)	129.6
13			6.27 d (15.8)	129.4	6.27 d (15.7)	129.4
1'		132.6		132.6		132.6
2'	6.89 br s	105.5	6.89 d (1.2)	105.5	6.89 d (1.3)	105.5
3'		148.0		148.0		148.0
4'		146.7		146.7		146.8
5'	6.73 d (7.9)	108.4	6.73 d (7.9)	108.4	6.73 d (8.1)	108.4
6'	6.75 br d (7.9)	120.3	6.75 dd (7.9, 1.2)	120.3	6.75 dd (8.1, 1.3)	120.3
1''		130.8		130.8		130.9
2'',6''	7.04 d (8.3)	130.0	7.03 d (8.4)	130.0	7.05 d (8.4)	130.0
3'',5''	6.78 d (8.3)	115.7	6.78 d (8.4)	115.7	6.78 d (8.4)	115.7
4''		154.6		154.7		154.6
7''	2.76 t (6.8)	34.9	2.67 t (6.8)	34.9	2.76 t (6.9)	34.9
8''	3.54 m	40.9	3.55 m	41.1	3.54 m	40.9
OCH <sub>2</sub> O	5.93 s	101.0	5.93 s	101.0	5.93 s	101.0
NH	5.44 t (5.5)		5.50 t (5.9)		5.42 t (5.4)	

<sup>a</sup> Data under the same entry are interchangeable.**Table 3**  
<sup>1</sup>H and <sup>13</sup>C NMR data of piperbonins A–C (7–9) in CDCl<sub>3</sub>.

No.	Piperbonin A (7)		Piperbonin B (8)		Piperbonin C (9)	
	$\delta_{\text{H}}$ (J in Hz) <sup>a</sup>	$\delta_{\text{C}}$ <sup>b</sup>	$\delta_{\text{H}}$ (J in Hz) <sup>c</sup>	$\delta_{\text{C}}$ <sup>d</sup>	$\delta_{\text{H}}$ (J in Hz) <sup>c</sup>	$\delta_{\text{C}}$ <sup>d</sup>
1		129.8		129.3		132.8
2	6.91 d (1.9)	110.6	6.91 d (1.9)	110.7	6.51 br s	111.1
3		148.8		149.0		149.0
4		148.9		149.3		148.3
5	6.82 d (8.3)	110.8	6.84 d (8.2)	111.0	6.80 d (8.2)	111.1
6	6.94 dd (8.3, 1.9)	120.0	6.95 dd (8.2, 1.9)	120.2	6.62 br d (8.2)	120.9
7	5.89 d (7.2)	78.2	5.93 d (7.1)	77.8	2.71 d (5.5)	58.0
8	4.55 m	78.1	4.74 m	77.6	2.64 m	38.8
9	1.16 d (6.4)	16.8	1.24 d (6.4)	16.6	1.08 d (7.1)	18.0
1'		132.3		130.6		56.2
2'	6.88 d (1.9)	109.5	7.42 d (1.8)	110.1		203.4
3'		150.5		150.7	3.74 d (7.6)	67.5
4'		147.0		153.6		191.3
5'	6.88 d (8.3)	117.0	7.08 d (8.1)	114.2		153.1
6'	6.83 dd (8.3, 1.9)	118.6	7.44 dd (8.1, 1.8)	126.5	6.28 s	123.0
7'	6.32 dd (15.7, 1.5)	130.5	9.85 s	191.0	2.22 dd (14.8, 7.9)	33.9
					2.05 dd (14.8, 7.0)	
8'	6.11 m	124.2			5.64 m	133.2
9'	1.85 dd (6.7, 1.5)	18.4			5.07 br d (10.0)	119.2
					5.04 br d (17.1)	
3-OMe	3.87 s	55.9	3.89 s	56.0	3.86 s	55.8
4-OMe	3.85 s	55.8	3.88 s	56.0	3.87 s	55.9
3'-OMe	3.83 s	55.8	3.91 s	56.1		
5'-OMe					3.70 s	55.6
OAc		170.0		170.1		
	1.99 s	21.1	2.00 s	21.2		

<sup>a</sup> Measured at 500 MHz.<sup>b</sup> measured at 125 MHz.<sup>c</sup> measured at 600 MHz.<sup>d</sup> measured at 100 MHz.

that it should be a dinorneolignan. On the basis of the 2D NMR correlations (Fig. S1, Supplementary data), the 2D structure of compound **8** was deduced to be 7-acetoxy-3,3',4-trimethoxy-8',9'

dinor-8,4'-oxyneolignan-7'-al. Like compound **7**, the relative configuration of **8** was also determined as *threo* ( $J_{7,8} = 7.1$  Hz). The absolute configuration of **8** (piperbonin B) was not determined as



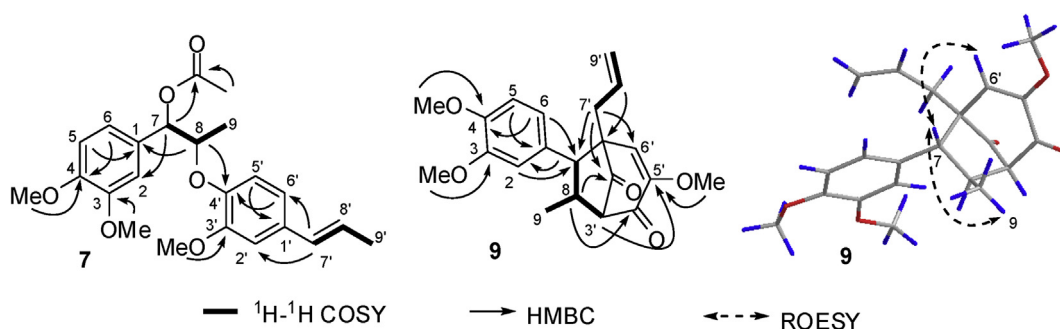


Fig. 3. Key 2D NMR correlations of compounds **7** and **9**.

yet.

The molecular formula  $C_{21}H_{24}O_5$  assigned to compound **9** was determined by  $^{13}C$  NMR data and an HRESIMS ion at  $m/z$  379.1520  $[M+Na]^+$  (calcd for  $C_{21}H_{24}NaO_5$ , 379.1516) with 10 degrees of unsaturation. The IR spectrum showed absorption bands for carbonyl (1759 and 1693  $cm^{-1}$ ), alkenyl (1638  $cm^{-1}$ ), and aromatic (1607, 1518, and 1463  $cm^{-1}$ ) functionalities. The NMR data of **9** (Table 3) clearly showed signals for two carbonyl groups ( $\delta_C$  203.4 and 191.3), one 1,2,4-trisubstituted phenyl ring [ $\delta_H$  6.80 (d,  $J = 8.2$  Hz), 6.62 (br d,  $J = 8.2$  Hz), and 6.51 (br s)], one allyl group [ $\delta_H$  5.64 (m), 5.07 (d,  $J = 10.0$  Hz), 5.04 (d,  $J = 17.1$  Hz), 2.22 (dd,  $J = 14.8$ , 7.9 Hz), and 2.05 (dd,  $J = 14.8$ , 7.0 Hz)], one trisubstituted double bond [ $\delta_H$  6.28 (s)], three methoxy groups [ $\delta_H$  3.87 (3H, s), 3.86 (3H, s), and 3.70 (3H, s)], and one methyl group [ $\delta_H$  1.08 (3H, d,  $J = 7.1$  Hz)]. By comparison of the NMR data of **9** with those of the known bicyclooctanoid [3.2.1] neolignans, denudaciones A–C (Kuroyanagi et al., 2000), compound **9** was determined to be a bicyclooctanoid neolignan. The  $^1H$ – $^1H$  COSY correlations of **9** (Fig. 3) showed the presence of two fragments C-7 to C-9 and C-7' to C-9'. The overall structure of **9** was determined to be 5-allyl-6-(3,4-dimethoxyphenyl)-3-methoxy-7-methylbicyclo[3.2.1]oct-3-ene-2,8-dione by the key HMBC correlations (Fig. 3) from H-7 to C-2 and C-6, H-8 to C-2' and C-4', H-3' to C-5', 5'-OMe to C-5', H-7' to C-7, C-2' and C-6', and H-8' to C-1'.

The relative configuration of **9** (piperbonin C) was confirmed by the key ROESY correlations (Fig. 3) of H-7/ $H_3$ -9 and H-7'/ $H_6'$ . However, the absolute configuration of compound **9** remains undefined.

The known compounds (+)-*trans*-acuminatin (**10**) (El-Feraly et al., 1982; López-Sánchez et al., 2013), (+)-*cis*-acuminatin (**11**) (López-Sánchez et al., 2013), (+)-kadsurenone (**12**) (Chang et al., 1985), (+)-burchellin (**13**) (Lima et al., 1972; Li et al., 1987), (+)-kadsurenin M (**14**) (Nascimento and Lopes, 1999), (+)-virolin (**15**) (Barata et al., 1978; Zacchino and Badano, 1985), galgravin (**16**) (Birch et al., 1958; Zhao and Ruan, 2006), 5-hydroxy-4',7-dimethoxyflavone (**17**) (Yu et al., 2005), piperonaline (**18**) (Lee et al., 2001), (+)-3',4'-dimethoxybenzoic acid-(3''4''-dimethoxyphenyl)-2-methyl-3-oxobutyl ester (**19**) (López et al., 1995; Li et al., 2013), and ferrudiol (**20**) (Kijjoa et al., 2002) were determined by comparing their NMR data (for all compounds) and optical rotation values (for the neolignans) with those reported in the literature.

The inhibitory activity of selected isolates (**1**–**18**) was evaluated against platelet aggregation induced by thrombin (Table 4). Piperbonin A (**7**), (+)-*trans*-acuminatin (**10**), (+)-*cis*-acuminatin (**11**), (+)-kadsurenone (**12**), and piperonaline (**18**) showed weak activity with  $IC_{50}$  values of 118.2, 108.5, 90.02, 107.3, and 116.3  $\mu M$ , respectively, as compared with the positive control, tirofiban, with an  $IC_{50}$  value of 5.24  $\mu M$ . (+)-Kadsurenone is a noted platelet-activating factor (PAF) receptor antagonist from *P. kadsura* (Shen

et al., 1985). It has previously been reported to inhibit the platelet release reaction induced by thrombin (Teng et al., 1991). The inhibitory activity of piperonaline against thrombin-induced platelet aggregation has previously been tested (Park et al., 2007), whereas the activities of piperbonin A, (+)-*trans*-acuminatin, and (+)-*cis*-acuminatin have only been reported in the present study. Among the dihydrobenzofuran neolignans, (+)-*trans*-acuminatin and (+)-*cis*-acuminatin with a propenyl group at C-1' were more potent than (+)-kadsurenin M with a formyl group at C-1', i.e. in eliciting an inhibitory effect of thrombin-induced platelet aggregation.

Thrombin receptor antagonists are a novel class of antiplatelet agents that inhibit thrombin-mediated platelet activation. In May 2014, the US Food and Drug Administration (FDA) approved vorapaxar, a synthetic tricyclic 3-phenylpyridine analogue of the naturally occurring alkaloid himbacine, for the prevention of thrombotic cardiovascular events in patients with a history of myocardial infarction (MI) or peripheral arterial disease (PAD). Vorapaxar functions as a thrombin receptor antagonist, working against the protease activated receptor 1 (PAR1) to inhibit platelet aggregation without affecting hemostasis (Hashemzadeh et al., 2015). The inhibition of thrombin induced platelet aggregation implies that these active compounds from *P. bonii* may be inhibitors of PAR1 or PAR4.

Some *Piper* plants are traditionally used against cancer or cancer-like symptoms, and amides are the major active constituents (Wang et al., 2014b). Therefore, the new amides (**1**–**6**) from the plant were also evaluated for their growth inhibitory activities against HL-60 (human leukemia), A-549 (human lung cancer), MCF-7 (human breast cancer), SMMC-7721 (human liver cancer), and SW480 (human rectal cancer) cell lines with the positive control *cis*-platinum with  $IC_{50}$  values of 2.38, 12.10, 18.79, 9.67, and 8.77  $\mu M$ , respectively. However, all of the tested compounds were inactive at concentrations up to 40  $\mu M$ . Piperbonamides A–F (**1**–**6**) belong to arylalkenylacyl amides with a 2-(4-hydroxyphenyl)ethylamino moiety. Previous findings showed that the presence of a pyrrolidin-1-yl or piperidin-1-yl unit is important for the

Table 4

The effect of compounds on rabbit platelet aggregation induced by thrombin (2 U/mL)<sup>a</sup>.

Compound	$IC_{50}$ ( $\mu M$ )
Piperbonin A ( <b>7</b> )	118.2
(+)- <i>trans</i> -Acuminatin ( <b>10</b> )	108.5
(+)- <i>cis</i> -Acuminatin ( <b>11</b> )	90.02
(+)-Kadsurenone ( <b>12</b> )	107.3
Piperonaline ( <b>18</b> )	116.3
Tirofiban (positive control)	5.24

<sup>a</sup> The inhibition of other tested compounds (**1**–**6**, **8**, **9**, and **13**–**17**) was less than 30% at the concentration of 100  $\mu M$ .

cytotoxicity of arylalkenylacyl amides (Wang et al., 2014a). The present results further support this conclusion.

### 3. Conclusions

In the current study, six new amides (**1–6**), three new neolignans (**7–9**), and 11 known compounds (**10–20**) were obtained from the aerial parts of *Piper bonii* for the first time. Several compounds including four neolignans (**7**, **10**, **11**, and **12**) and one amide (**18**) showed weak activity against platelet aggregation induced by thrombin. Nevertheless, in future, it may be worth searching for new platelet antagonists from the *Piper* plants or the traditional Chinese medicines with the functions of “Huoxue” or “Huayu”.

### 4. Experimental

#### 4.1. General experimental procedures

Optical rotations were recorded using a JASCO P-1020 Polarimeter (Jasco Corp., Tokyo, Japan). UV spectra were taken on a Shimadzu UV-2401 PC spectrophotometer (Shimadzu, Kyoto, Japan). Electronic circular dichroism (ECD) spectra were recorded on a Chirascan CD spectrometer (Applied Photophysics Ltd., Leatherhead, UK). IR spectra were measured on a Bruker Tensor 27 FTIR Spectrometer (Bruker Corp., Ettlingen, Germany) with KBr disks.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were collected on a Bruker AM-400, DRX-500 and Avance III-600 spectrometers (Bruker Bio-Spin GmbH, Rheinstetten, Germany) with TMS as an internal standard. ESIMS and HRESIMS analyses were performed on an API QSTAR Pulsar 1 spectrometer (Applied Biosystems/MDS Sciex, Foster City, CA, USA). HREIMS were performed on a Waters AutoSpec Premier p776 spectrometer (Waters, Millford, MA, USA). Silica gel G (80–100 and 300–400 mesh, Qingdao Meigao Chemical Co., Ltd., Qingdao, China),  $\text{C}_{18}$  silica gel (40–75  $\mu\text{m}$ , Fuji Silysia Chemical Ltd., Aichi, Japan), and Sephadex LH-20 (GE Healthcare Bio-Sciences AB, Uppsala, Sweden) were used for column chromatography (CC), and silica gel GF<sub>254</sub> (Qingdao Meigao Chemical Co., Ltd.) was used for preparative TLC as precoated plates. TLC spots were visualized under UV light at 254 nm and by dipping into 5%  $\text{H}_2\text{SO}_4$  in alcohol followed by heating. Semipreparative HPLC was performed on an Agilent 1200 series pump (Agilent Technologies, Santa Clara, USA) equipped with a diode array detector and a Waters XBridge  $\text{C}_{18}$  column (5.0  $\mu\text{m}$ ,  $\phi$  10  $\times$  250 mm) and an Agilent Zorbax SB- $\text{C}_{18}$  column (5.0  $\mu\text{m}$ ,  $\phi$  9.4  $\times$  250 mm).

#### 4.2. Plant material

Aerial parts of *P. bonii* were collected from Xishuangbanna of Yunnan Province, People's Republic of China, in June 2012. Based on the taxonomic system in *Flora of China*, the plant was identified by one of the authors (C.-L.L.), and a voucher specimen (No. 201201) was deposited at the Key Laboratory of Economic Plants and Biotechnology, Kunming Institute of Botany, Chinese Academy of Sciences.

#### 4.3. Extraction and isolation

Air-dried, powdered aerial parts of *P. bonii* (2.5 kg) were exhaustively extracted with MeOH (4  $\times$  10 L) at room temperature. The combined MeOH extract (430 g) was suspended in  $\text{H}_2\text{O}$  (2 L) and further partitioned with petroleum ether (3  $\times$  2 L) and  $\text{CHCl}_3$  (3  $\times$  2 L), respectively. The petroleum ether-soluble part (A, 51.2 g) was subjected to silica gel CC (petroleum ether/EtOAc, 20:1  $\rightarrow$  0:1, v/v) to yield six fractions (A1–A6). Fraction A2 was separated on an RP-18 silica gel column eluted with MeOH/ $\text{H}_2\text{O}$  (10% MeOH

$\rightarrow$  100%) to yield two main subfractions (A2-1 and A2-2). Subfraction A2-1 [the MeOH/ $\text{H}_2\text{O}$  (7030:) eluted portion] was purified by Sephadex LH-20 CC (MeOH), silica gel CC (petroleum ether/EtOAc, 15:1, v/v), and semipreparative HPLC (Waters XBridge  $\text{C}_{18}$  column, 10  $\times$  250 mm, MeOH/ $\text{H}_2\text{O}$ , 752, 25: mL/min) to give **11** (23.8 mg,  $t_{\text{R}}$  = 7.566 min) and **10** (4.5 mg,  $t_{\text{R}}$  = 8.162 min). Subfraction A2-2 [the MeOH/ $\text{H}_2\text{O}$  (8020:) eluted portion] was purified by Sephadex LH-20 CC (MeOH) and silica gel CC (petroleum ether/EtOAc, 15:1, v/v) to obtain **18** (32.7 mg). Fraction A3 was separated on an RP-18 silica gel column eluted with MeOH/ $\text{H}_2\text{O}$  (10% MeOH  $\rightarrow$  100%) to yield two main subfractions (A3-1 and A3-2). Subfraction A3-1 [the MeOH/ $\text{H}_2\text{O}$  (6040:) eluted portion] was purified by Sephadex LH-20 CC (MeOH), silica gel CC (petroleum ether/EtOAc, 15:1, v/v), and semipreparative HPLC (Waters XBridge  $\text{C}_{18}$  column, 10  $\times$  250 mm, MeOH/ $\text{H}_2\text{O}$ , 752, 25: mL/min) to yield **12** (334.2 mg,  $t_{\text{R}}$  = 14.189 min). Subfraction A3-2 [the MeOH/ $\text{H}_2\text{O}$  (7030:) eluted portion] was purified by Sephadex LH-20 CC (MeOH) and silica gel CC (petroleum ether/EtOAc, 15:1, v/v) to give **17** (4.5 mg). Fraction A4 was separated on an RP-18 silica gel column eluted with MeOH/ $\text{H}_2\text{O}$  (10% MeOH  $\rightarrow$  100%) to yield three main subfractions (A4-1 to A4-3). Subfraction A4-1 [the MeOH/ $\text{H}_2\text{O}$  (6040:) eluted portion] was purified by Sephadex LH-20 CC (MeOH), silica gel CC ( $\text{CHCl}_3$ /MeOH, 20:1, v/v), and semipreparative HPLC (Waters XBridge  $\text{C}_{18}$  column, 10  $\times$  250 mm, MeOH/ $\text{H}_2\text{O}$ , 7030:, 2 mL/min) to give **8** (4.5 mg,  $t_{\text{R}}$  = 10.307 min) and **9** (6.2 mg,  $t_{\text{R}}$  = 12.918 min). Subfraction A4-2 [the MeOH/ $\text{H}_2\text{O}$  (7030:) eluted portion] was isolated by Sephadex LH-20 CC (MeOH), silica gel CC ( $\text{CHCl}_3$ /EtOAc, 20:1, v/v), and further purified by semipreparative HPLC (Agilent Zorbax SB- $\text{C}_{18}$  column, 9.4  $\times$  250 mm, 2 mL/min) to yield **14** (4.0 mg; MeOH/ $\text{H}_2\text{O}$ , 7030:;  $t_{\text{R}}$  = 8.845 min), **15** (4.8 mg;  $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ , 6535:;  $t_{\text{R}}$  = 12.458 min), and **16** (4.1 mg;  $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ , 6535:;  $t_{\text{R}}$  = 16.765 min). Subfraction A4-3 [the MeOH/ $\text{H}_2\text{O}$  (8020:) eluted portion] was purified by Sephadex LH-20 CC (MeOH) and silica gel CC (petroleum ether/acetone, 10:1, v/v) to give **7** (145.5 mg).

The  $\text{CHCl}_3$ -soluble part (B, 13.5 g) was subjected to silica gel CC using  $\text{CHCl}_3$ /MeOH (30:1  $\rightarrow$  20:1, v/v) to afford a major fraction. This fraction was further separated on an RP-18 silica gel column eluted with MeOH/ $\text{H}_2\text{O}$  (10% MeOH  $\rightarrow$  100%) to yield three main subfractions (B1-1 to B1-3). Subfraction B1-1 [the MeOH/ $\text{H}_2\text{O}$  (6040:) eluted portion] was purified by Sephadex LH-20 CC (MeOH), silica gel CC ( $\text{CHCl}_3$ /MeOH, 20:1, v/v), and semipreparative HPLC (Agilent Zorbax SB- $\text{C}_{18}$  column, 9.4  $\times$  250 mm,  $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ , 702, 30: mL/min) to give **4** (5.1 mg,  $t_{\text{R}}$  = 15.900 min), **20** (3.0 mg,  $t_{\text{R}}$  = 16.893 min), **5** (4.9 mg,  $t_{\text{R}}$  = 24.073 min), and **6** (5.2 mg,  $t_{\text{R}}$  = 28.512 min). Subfraction B1-2 [the MeOH/ $\text{H}_2\text{O}$  (7030:) eluted portion] was purified by Sephadex LH-20 CC (MeOH), silica gel CC ( $\text{CHCl}_3$ /MeOH, 20:1, v/v), and semipreparative HPLC (Waters XBridge  $\text{C}_{18}$  column, 10  $\times$  250 mm, MeOH/ $\text{H}_2\text{O}$ , 802, 20: mL/min) to obtain **1** (5.2 mg,  $t_{\text{R}}$  = 18.372 min), **2** (3.5 mg,  $t_{\text{R}}$  = 21.206 min), and **3** (3.5 mg,  $t_{\text{R}}$  = 32.642 min). Subfraction B1-3 [the MeOH/ $\text{H}_2\text{O}$  (8020:) eluted portion] was purified by Sephadex LH-20 CC (MeOH), silica gel CC ( $\text{CHCl}_3$ /MeOH, 20:1, v/v), and semipreparative HPLC (Waters XBridge  $\text{C}_{18}$  column, 10  $\times$  250 mm, MeOH/ $\text{H}_2\text{O}$ , 702, 30: mL/min) to yield **19** (4.5 mg,  $t_{\text{R}}$  = 8.776 min) and **13** (5.0 mg,  $t_{\text{R}}$  = 24.290 min).

#### 4.4. Piperbonamide A (**1**)

Pale green oil; UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 261 (4.48), 226 (4.33) nm; IR (KBr)  $\nu_{\text{max}}$  3396, 2925, 2852, 1657, 1627, 1610, 1596, 1532, 1514, 1449, 1231, 999, 968, 829  $\text{cm}^{-1}$ ; For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data, see Table 1; ESIMS  $m/z$  442 [ $\text{M}+\text{Na}$ ] $^+$ ; HRESIMS  $m/z$  442.2352 [ $\text{M}+\text{Na}$ ] $^+$  (calcd for  $\text{C}_{27}\text{H}_{33}\text{NNaO}_3$ , 442.2353).

#### 4.5. Piperbonamide B (2)

Pale green oil; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 273 (3.78), 222 (4.48) nm; IR (KBr)  $\nu_{\max}$  3405, 2924, 2852, 1668, 1628, 1614, 1598, 1534, 1514, 1449, 1238, 969, 827  $\text{cm}^{-1}$ ; For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data, see Table 1; ESIMS  $m/z$  444  $[\text{M}+\text{Na}]^+$ ; HRESIMS  $m/z$  444.2512  $[\text{M}+\text{Na}]^+$  (calcd for  $\text{C}_{27}\text{H}_{35}\text{NNaO}_3$ , 444.2509).

#### 4.6. Piperbonamide C (3)

Pale green oil; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 261 (4.44), 226 (4.28) nm; IR (KBr)  $\nu_{\max}$  3417, 2920, 2850, 1655, 1626, 1613, 1596, 1544, 1514, 1465, 1447, 1262, 1239, 1005, 830  $\text{cm}^{-1}$ ; For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data, see Table 1; ESIMS  $m/z$  470  $[\text{M}+\text{Na}]^+$ , 917  $[2\text{M}+\text{Na}]^+$ ; HRESIMS  $m/z$  470.2688  $[\text{M}+\text{Na}]^+$  (calcd for  $\text{C}_{29}\text{H}_{37}\text{NNaO}_3$ , 470.2666).

#### 4.7. Piperbonamide D (4)

Pale green oil; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 300 (3.69), 263 (4.06) nm; IR (KBr)  $\nu_{\max}$  3426, 2925, 2853, 1665, 1624, 1548, 1515, 1504, 1491, 1445, 1247, 1038  $\text{cm}^{-1}$ ; For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data, see Table 2; ESIMS  $m/z$  444  $[\text{M}+\text{Na}]^+$ , 865  $[2\text{M}+\text{Na}]^+$ ; HRESIMS  $m/z$  444.2144  $[\text{M}+\text{Na}]^+$  (calcd for  $\text{C}_{26}\text{H}_{31}\text{NNaO}_4$ , 444.2145).

#### 4.8. Piperbonamide E (5)

Pale green oil; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 303 (3.71), 262 (4.57) nm; IR (KBr)  $\nu_{\max}$  3427, 2924, 2852, 1655, 1629, 1614, 1546, 1515, 1504, 1445, 1247, 1038, 995  $\text{cm}^{-1}$ ; For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data, see Table 2; ESIMS  $m/z$  470  $[\text{M}+\text{Na}]^+$ , 917  $[2\text{M}+\text{Na}]^+$ ; HRESIMS  $m/z$  470.2304  $[\text{M}+\text{Na}]^+$  (calcd for  $\text{C}_{28}\text{H}_{33}\text{NNaO}_4$ , 470.2302).

#### 4.9. Piperbonamide F (6)

Pale green oil; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 303 (3.59), 262 (4.02) nm; IR (KBr)  $\nu_{\max}$  3443, 2925, 2852, 1664, 1625, 1549, 1516, 1504, 1490, 1446, 1248, 1039  $\text{cm}^{-1}$ ; For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data, see Table 2; ESIMS  $m/z$  472  $[\text{M}+\text{Na}]^+$ , 921  $[2\text{M}+\text{Na}]^+$ ; HRESIMS  $m/z$  472.2456  $[\text{M}+\text{Na}]^+$  (calcd for  $\text{C}_{28}\text{H}_{35}\text{NNaO}_4$ , 472.2458).

#### 4.10. Piperbonin A (7)

Pale yellow oil;  $[\alpha]_D^{23} +56$  (c 0.7, MeOH);  $[\alpha]_D^{25} +47$  (c 1.0,  $\text{CHCl}_3$ ); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 309 (3.95), 278 (4.13), 230 (4.32) nm; ECD  $\Delta\epsilon$  (c 0.10, MeOH) +0.90 (232), +1.42 (204); IR (KBr)  $\nu_{\max}$  2936, 1737, 1630, 1604, 1596, 1512, 1465, 1453, 1419, 1374, 1264, 1238, 1140, 1029, 966  $\text{cm}^{-1}$ ; For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data, see Table 3; ESIMS  $m/z$  423  $[\text{M}+\text{Na}]^+$ , HRESIMS  $m/z$  423.1781  $[\text{M}+\text{Na}]^+$  (calcd for  $\text{C}_{23}\text{H}_{28}\text{NaO}_6$ , 423.1778).

#### 4.11. Piperbonin B (8)

Pale yellow oil;  $[\alpha]_D^{23} +72$  (c 0.3, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 309 (3.95), 278 (4.13), 230 (4.32) nm; ECD  $\Delta\epsilon$  (c 0.00672, MeOH) +0.27 (276), +0.67 (235), +2.10 (208); IR (KBr)  $\nu_{\max}$  2937, 1739, 1684, 1593, 1508, 1465, 1423, 1375, 1268, 1138, 1029, 813  $\text{cm}^{-1}$ ; For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data, see Table 3; ESIMS  $m/z$  411  $[\text{M}+\text{Na}]^+$ , HRESIMS  $m/z$  388.1522  $[\text{M}]^+$  (calcd for  $\text{C}_{21}\text{H}_{24}\text{O}_7$ , 388.1522).

#### 4.12. Piperbonin C (9)

Pale yellow oil;  $[\alpha]_D^{20} -70$  (c 0.3, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 280 (3.68), 228 (3.95) nm; ECD  $\Delta\epsilon$  (c 0.015, MeOH)  $-1.54$

(330), +0.30 (302),  $-0.23$  (293), +0.54 (277),  $-0.19$  (259), +0.59 (218); IR (KBr)  $\nu_{\max}$  2960, 2931, 1759, 1693, 1638, 1607, 1518, 1463, 1262, 1146, 1027  $\text{cm}^{-1}$ ; For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data, see Table 3; ESIMS  $m/z$  379  $[\text{M}+\text{Na}]^+$ ; HRESIMS  $m/z$  379.1520  $[\text{M}+\text{Na}]^+$  (calcd for  $\text{C}_{21}\text{H}_{24}\text{NaO}_5$ , 379.1516).

#### 4.13. In vitro platelet aggregation assay

New Zealand rabbits (male, 2.0–2.5 kg) were purchased from Nanjing Qinglongshan Animal Center (Nanjing, China). All animals were acclimated for at least 1 week at a temperature of  $22 \pm 2$  °C and humidity of  $55 \pm 5\%$  in a 12-h light-dark cycle with free access to food and  $\text{H}_2\text{O}$ . All protocols were approved by the Committee on the Ethics of Animal Experiments of China Pharmaceutical University and conformed to the National Institutes of Health Guideline for Care and Use of Laboratory Animals.

*In vitro* platelet aggregation was conducted using the turbidimetric method with a minor modification (Aghaloo et al., 2002; Nishikawa and Hidaka, 1982). Briefly, blood was withdrawn from the carotid artery of New Zealand rabbits after local anaesthesia with lidocaine, and collected into tubes containing 3.8% sodium citrate (1:9 citrate/blood, v/v) and centrifuged for 15 min at 950 rpm to prepare platelet-rich plasma (PRP) or 10 min at 3000 rpm to obtain platelet-poor plasma (PPP), respectively. The platelet concentration was adjusted to  $3 \times 10^8$  platelets/mL with PPP. A 270  $\mu\text{L}$  aliquot of PRP was preincubated at 37 °C for 5 min in the cuvette with 20  $\mu\text{L}$  of the sample or vehicle (saline), and then platelet aggregation was induced by addition of 10  $\mu\text{L}$  thrombin (2 U/mL, Sigma Chemical Co., St. Louis, MO, USA). The maximum platelet aggregation rate was defined as the maximal percentage increase in light transmission after addition of thrombin related to the light transmission in PPP, and the maximum aggregation rate was determined within 5 min with continuous stirring at 37 °C using a four-channel aggregometer (Beijing Steellex Science Instrument Company, China). The  $\text{IC}_{50}$  values were defined as the concentration of inhibitor that cause a 50% inhibition to the platelet aggregation, and was calculated using GraphPad Prism 6 (GraphPad Software, Inc., La Jolla, CA, USA).

#### 4.14. MTS assay for cytotoxicity

The isolated amides were tested *in vitro* for their cytotoxicity against proliferation of human leukemia HL-60, human lung cancer A-549, human breast cancer MCF-7, human liver cancer SMMC-7721, and human rectal cancer SW480 cell lines using the MTS assay (Mosmann, 1983). The 3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfophenyl)-2H-tetrazolium (MTS) (Promega, Beijing, China) mix-based cell titer assays were performed as previously described. In general, cells in log-phase of their cycle were seeded in 96-well plates (5000–10,000 cells/well, NEST Biotechnology, Wuxi, China) in a 100  $\mu\text{L}$  volume, with paired single cell suspension containing 10% fetal bovine serum (DMEM or RPMI1640, Thermo Fisher Scientific, Beijing, China). These were then treated with indicated concentrations of the compounds dissolved in DMSO, which were set the regular thickness of 40  $\mu\text{M}$  for the initial screening and five concentrations of each compound were fixed in the concentration compounds inhibition tumor cell growth about 50% to achieve a total culture medium in a volume of 200  $\mu\text{L}$ . After incubation for 48 h at 37 °C, an MTS (20  $\mu\text{L}$ ) solution and DMEM (100  $\mu\text{L}$ ) were added into the well, incubation was continued for another 1–4 h. The absorbance was measured at the detection wavelength of 490 nm ( $L_1$ ) and the reference wavelength of 680 nm ( $L_2$ ) and cytotoxicity was expressed as  $\text{IC}_{50}$  values (Reed and Muench, 1938). Cis-platinum (Sigma) was used as the positive control.



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## Appendix A. Supplementary data

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.phytochem.2016.07.004>.

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