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One chloro-furoeremophilanoid and two new natural dimers from Ligularia atroviolacea

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Abstract

Three new compounds, including one chloro-furoeremophilanoid (1), one eremophiladiolide (2), a rare dimer of nor-furoeremophilanoid, ligulatrovine A (3), and a known furoeremophilanolide (4), were isolated from *Ligularia atroviolacea*. The structures of compounds 1-4 were elucidated by spectroscopic methods including 1D and 2D NMR experiments as well as X-ray diffraction study.

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Ligularia species (Asteraceae) are mainly distributed in damp, shaded areas of western China, especially the provinces of Gansu, Sichuan and Yunnan [1]. Recent investigation on the roots of Ligularia atroviolacea (Franch.) Hand.-Mazz. has led to the isolation and identification of four eremophilendiolides [2]. Careful examination of the minor constituents of this plant led to the isolation of three new eremophilane derivatives 1–3, as well as one known furanoeremophilane (4) [3]. This article described the isolation, structural elucidation of compounds 1–4.

Compound 1 was obtained as colorless needles with a melting point of 107-108 °C, $[\alpha]_D^{20} = +0.9$ (c 0.2, MeOH). The molecular formula of 1, $C_{19}H_{25}ClO_5$ was deduced based on its ESIMS (m/z 369 for $[M+H]^+$ ion peak), elemental analysis, and ^{13}C NMR data (Table 1). The IR absorption bands of 1 suggested the existence of a hydroxyl (3500 cm⁻¹) group, an ester carbonyl (1736 cm⁻¹) group, and an α -furan keton moiety (1680, 1634, 1560 cm⁻¹). Furthermore, the ^{1}H NMR spectrum of 1 indicated the presences of three methyl groups (δ 1.05, 1.14 and 1.94) and

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Table 1	
NMR spectral data of 1, 2 and 3 (¹ H, 400 MHz and	¹³ C, 100 MHz, TMS, δ ppm, J Hz)

Position	1 ^a (CDCl ₃)		Position	2 (Me ₂ CO-d ₆)		3 (CDCl ₃)	
	δ_{C}	$\delta_{ m H}$		$\delta_{\rm C}$	δ_{H}	$\delta_{\rm C}$	δ_{H}
1	62.0	3.94 br s	1, 1' ^b	21.2	2.17 m 2.34 m	31.0	2.76 (t, 7.2)
2	24.3	1.76 (dd, 14, 2.0) 2.59 (t, 14.0)	2, 2'	21.9	1.69 m 2.05 m	23.8	2.24 m
3	23.3	1.44 (d, 13.6) 2.47 (t, 14.0)	3, 3'	136.7	6.83 (t, 3.2)	138.9	7.10 (dt, 12.8, 6.4)
4	31.9	1.05 m	4, 4'	129.9	_	134.4	_
5	50.1	_	5, 5'	44.3	_	127.1	_
6	68.2	7.06 s	6, 6'	82.9	5.12 br s	117.6	_
7	139.6	_	7, 7'	157.4	_	142.3	7.48 (d, 1.6)
8	145.9	_	8, 8'	87.3	_	155.4	_
9	186.3	-	9, 9'	35.2	1.31 m 2.39 m	108.6	7.19 (d, 1.6)
10	80.5	_	10, 10'	32.1	2.58 m	136.9	_
11	121.8	_	11, 11'	128.5	_	169.6	_
12	147.4	7.50 s	12, 12'	170.9	_	18.3	2.38 (d, 1.2)
13	8.53	1.94 s	13, 13'	10.1	1.97 (d, 2.4)	129.7	_
14	16.0	1.05 s	14, 14'	26.2	1.48 s	11.1	2.51 s
15	15.9	1.14 (d, 7.6)	15, 15'	168.4	77 II II		-

^a 6-Isobutyroxy: 1.28 (d, 7.2, H-3'), 1.30 [d, 7.2 Hz, H-4'], 2.73 (qq, 7.2, 7.2 Hz, H-2'); ¹³C NMR: 176.5 (C-1'), 34.1 (C-2'), 18.6 (C-3'), 19.5 (C-4').

^b Positions indicated with prime symbols are applicable to dimers 2 and 3.

one isobutyroxy group (δ 1.28, 1.30 and 2.73) (Table 1). Except for the isobutyroxy moiety, the ¹³C NMR and DEPT spectra of 1 demonstrated 15 carbon signals, including six quaternary carbons, four methines, two methylenes and three methyls (Table 1). From above information, 1 was suggested to be a furanoeremophilane similar to a known compound, 6β -angeloyloxy-9-oxo- 1α , 10β -dihydroxy-furanoeremophilane [4]. However, the differences between 1 and above known furanoeremophilane could be found on the substituents at C-1 and C-6 (Table 1) [4]. The HMBC spectrum of 1 exhibited the correlation between H-6 [δ 7.06 (s, 1H)] and C-1' (δ 176.5) and thus suggested the location of the isobutyroxy group at C-6. Therefore, the chlorine atom was suggested to present at C-1 position. Furthermore, Me-14 and Me-15 could be assigned to be β orientation based on general biogenetical consideration [5] and supported by the NOESY experiment. In NOSEY spectrum, the correlations between H-1 and H-14, H-4 and H-6 indicated their *cis* relationships. Compound 1 was therefore deduced as 1α -chloro- 6β -isobutyroxy-9-oxo- 10β -hydroxy-furanoeremophilane. The 3D structure of 1 was constructed by X-ray diffraction experiment (Fig. 1).

The HRESIMS of 2 exhibited a quasimolecular ion peak appeared at m/z 536.2276 [M+NH₄]⁺ (calc. for 536.2284), which indicated its molecular formula to be $C_{30}H_{30}O_8$. Furthermore, the IR absorption bands appeared at 1765, 1746, and 1675 cm⁻¹ suggested the presences of γ -lactones in compound 2. However, the base peak (m/z 259) revealed that

Fig. 1. Structures of compounds 1, 2 and X-ray diagram of compound 1.

compound 2 was composed of two halves possessing identical molecular weights. Furthermore, the 13 C NMR and DEPT spectra of 2 exhibited only 15 resonances for two methyls, three methylenes, three methines and seven quaternary carbons (Table 1). This implied that compound 2 was formed with two symmetric eremophilanolide units which were directly linked to each other with a C–C bond [6]. Structural elucidation of the half of compound 2 was accomplished by the analyses of its 1 H and 13 C NMR spectral data (Table 1). The presences of a tertiary methyl group [δ 1.48 (s), δ 26.2 (CH₃)] and an olefinic methyl group [δ 1.97 (d), δ 10.1 (CH₃)] were characteristic of an eremophilanolide. Therefore, the half of 2 was elucidated as an eremophilanolide with a structure similar to 8βH-eremophil-3,7(11)-dien-12,8α;15,6α-diolide [7], except for the absence of a methine signal [δ 4.68 (dd), δ 77.4 (CH)]. Thus, the structure of 2 was determined as a dimer of this known diolide. Furthermore, the diagnostic sp³ signal appeared at δ 87.3 in the case of 2 implied that the two halves joined at C-8/C-8′ positions. Comprehensive analyses of the NMR data of 2 indicated a H-8β stereochemistry, which was confirmed by the homoallylic coupling between Me-13 and H-6 [7] (Table 1). Therefore, the structure of 2 was elucidated as 8β-[eremophil-3',7'(11')-dien-12',8'α;15',6'α-diolide]-eremophil-3,7(11)-dien-12,8α;15,6α-diolide.

The molecular formula of ligulatrovine A (3), $C_{28}H_{28}O_6$ was deduced by its HR-ESI MS (m/z 461.1957 for [M+H]⁺ ion peak, calc. for 461.1964). The IR spectrum of 3 showed absorption bands for a carboxyl group (3161, 1726 cm⁻¹), double bond (1649 cm⁻¹), and an aromatic ring (1596, 1540, 1456 cm⁻¹). However, the ¹³C NMR and DEPT spectra of 3 displayed only 14 carbon signals including two methyls, two methelenes, three methines and seven quaternary carbons (Table 1). This suggested that 3 is another symmetric dimer. A methyl group at δ 2.38 [d, H-12/H-12'; δ 18.3 (CH₃)], an olefinic methyl group at δ 2.51 [s, H-14/H-14'; δ 11.1 (CH₃)] were observed in the upfield region of the NMR spectra of 3 (Table 1). Two allylic methylenes were demonstrated by the signals appeared at δ 2.76 [t, H-1/H-1'] and 2.24 [m, H-2/H-2'] in the ¹H NMR spectrum of 3. Furthermore, the downfield part of 3 showed the presence of a 1,2,3,5-tetra substituted aromatic ring at δ 7.19 [d, H-9/H-9'], 7.48 [d, H-7/H-7'] and δ 127.1, 117.6, 142.3, 155.4, 108.6, 136.9. A conjugated olefinic proton at δ 7.10 (dt, H-3/H-3') and their corresponding carbon signals at δ 138.9, 134.4, as well as a carbon signal due to a carboxyl group at δ 169.6 were also observable in the downfield part of the NMR spectra of 3. Moreover, it was suggested that the two halves were linked to each other with a C–C double bond, according to a single olefinic carbon signal appeared at δ 129.7 [C-13/C-13'] in its ¹³C NMR spectrum. Additionally, the ¹H-¹H COSY spectrum of 3 showed correlations of H-2 to H-1 and H-3, and H-7 to H-9, respectively (Fig. 2). Combined with the observed HMBC data, the structure of 3 could be unambiguously deduced as shown.

Compound 4 had an identical ¹H NMR data to that of an artifact furanoeremophil-3-en-15,6 α -olide synthesized by Kuroda and co-workers [3]. It is isolated as a natural product for the first time. Biogenetically, furoeremophilanolide 4 is proposed to be the possible precursor of dimer 2, both existed in the title plant. Though a acid-base catalysed

Fig. 2. Structures of 3 and 4; key ¹H-¹H COSY (bold) and HMBC (arrow) correlations of 3.

Scheme 1. Possible biogenetic path of 2 via a free radical mechanism from 4.

mechanism for C-8 dimer of eremophiladiolides was suggested by Zhao et al. [8], another reasonable free radical mechanism of forming this kind of C-8 dimer is hypothesized as shown in Scheme 1 [9].

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References

- [1] S.W. Liu, Flora of China, vol. 77, Science Press, Beijing, 1989, p. 41.
- [2] S.Y. Shi, D.Y. Wu, X. Gong, et al. Chin. Chem. Lett. 18 (2007) 59.
- [3] C. Kuroda, T. Murae, M. Tada, et al. Bull. Chem. Soc. Jpn. 55 (1982) 1228.
- [4] D.L. Cheng, J.J. Gao, L. Yang, Chem. J. Chin. Univ. 13 (1992) 781.
- [5] K. Naya, R. Kanazawa, M. Sawada, Bull. Chem. Soc. Jap. 48 (1975) 3220.
- [6] Y.C. Lin, T. Jin, X.M. Wu, et al. J. Nat. Prod. 60 (1997) 27.
- [7] Y.F. Han, D.Q. Fei, K. Gao, et al. Chin. Chem. Lett. 16 (2005) 1053.
- [8] Y. Zhao, H. Jiang, M. Macleod, et al. Chem. Biodiver. 1 (2004) 1546.
- www.cmki.net [9] S.K. Bagal, R.M. Adlington, J.E. Baldwin, et al. Tetrahedron Lett. 46 (2005) 4633.