

Available online at www.sciencedirect.com



Food Chemistry

Food Chemistry 101 (2007) 365-371

www.elsevier.com/locate/foodchem

Phenolic antioxidants from Chinese toon (fresh young leaves and shoots of *Toona sinensis*)

Kai-Jin Wang, Chong-Ren Yang, Ying-Jun Zhang *

State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650204, PR China

Received 22 September 2005; received in revised form 16 January 2006; accepted 16 January 2006

Abstract

Chinese toon (the fresh young leaves and shoots of *Toona sinensis*), known as a tree vegetable, belongs to the family of Meliaceae. It is one of the most popular vegetables in China. The 80% acetone extract of Chinese toon exhibited considerable antioxidant activity in the 1,1-diphenyl-2-picryldydrazyl (DPPH) radical-scavenging assay. Bioassay-guided purification of the extract led to the isolation of 12 phenolic compounds (1–12), whose structures were determined by detailed spectroscopic analyses, including UV, IR, MS, 1D and 2D NMR techniques. Of them, compounds 3–8 and 11 were isolated from *T. sinensis* for the first time. All of the isolated compounds were examined for their antioxidant activities by the DPPH method, and the results showed that gallic acid and its derivatives, gallotannins and flavonoids were the main constituents contributing to the antioxidant activity of Chinese toon. From the health point of view, Chinese toon is an ideal dietary vegetable with natural antioxidants.

© 2006 Elsevier Ltd. All rights reserved.

Keywords: Chinese toon; Toona sinensis; DPPH radical-scavenging activity; Phenolic constituents

1. Introduction

The free radicals and reactive oxygen species are thought to be harmful to human health and trigger many diseases, such as cancer, coronary heart diseases, arteriosclerosis, inflammatory disorders, and aging processes (Aviram, 2000; Halliwell, Gutteridge, & Cross, 1992; Halliwell, 1994). Synthetic antioxidants, such as butylated hydroxyanisole (BHA), tertiary butyhydroquinone (TBHQ) and butylated hydroxytoluene (BHT) are often used in food products. However, demand for natural antioxidants has been increasing due to concerns about the safety of synthetic antioxidants (Williams, Iatropoulos, & Whysner, 1999). Natural anti-oxidative compounds from plants have aroused much attention, and more increasing efforts have been made to search for plant-derived antioxidants (Luo, Basile, & Kennelly, 2002; Mangiapane et al., 1992).

Toona sinensis (Meliaceae) is a deciduous tree distributed widely in China, whose bark, oil, seed, flower and leaf have been used in traditional Chinese medicine. The fresh young leaves and shoots, known as Chinese toon and a tree vegetable, is one of the most popular vegetables in China (Editorial Board of Zhong-Hua-Ben-Cao (China Herbal), State Administration of Traditional Chinese Medicine, 1999). The leaves of T. sinensis were used medicinally for the treatment of heliosis, vomitting, dysentery, lack of appetite and enteritis in the folk thereby of China, due to their affects in detoxification, antiinflammation. It was reported that the leaf extract of T. sinensis can induce the apoptosis of cancer cells (Chang, Hung, Huang, & Hsu, 2002) and can enhance, in vitro, lipolysis and glucose uptake in differentiated 3T3-L1 adiposities (Hsu, Yang, Hwang, & Hong, 2003; Yang, Hsu, Hwang, & Hong, 2003). Methyl gallate, from T. sinensis, was reported to have a protective effect against hydrogen peroxide-induced oxidative stress and DNA damage in MDCK cells (Hsieh et al., 2004). Previous phytochemical investigations showed

^{*} Corresponding author. Tel.: +86 871 5223235; fax: +86 871 5150124. *E-mail address:* zhangyj@mail.kib.ac.cn (Y.-J. Zhang).

^{0308-8146/\$ -} see front matter @ 2006 Elsevier Ltd. All rights reserved. doi:10.1016/j.foodchem.2006.01.044

that the leaves of *T. sinensis* were rich in flavonoids, alkaloids, terpenes and anthraquinones (Chen, Luo, Cui, Zhen, & Liu, 2000; Luo, Wu, Ma, & Wu, 2001). However, no detailed study has been reported on the antioxidant activity of *T. sinensis* and its constituents.

The 80% acetone extract of Chinese toon (the fresh young leaves and shoots of *T. sinensis*) exhibited considerable antioxidant activity, by DPPH assay, in our preliminary test. This prompted us to perform a detailed bioassay-guided chemical investigation on this plant, which led to the isolation of 12 phenolic compounds. This paper describes the structural determination of these compounds and their antioxidant activities in a DPPH radical-scavenging assay.

2. Materials and methods

2.1. General procedures

 $[\alpha]_D$ was measured on JASCO-20 polarimeter. IR spectra were recorded on a Bio-Rad FTS-135 spectrometer with KBr pellets. UV spectra were recorded on a UV 210A Shimadzu spectrometer. 1D- and 2D-NMR spectra were run on Bruker AM-400 and DRX-500 instruments with TMS as internal standard. The MS data were recorded on a VG Auto Spec-3000 spectrometer. Radical-scavenging assay was performed on an Emax precision microplate reader.

2.2. Chemicals and reagents

The DPPH (Aldrich Chem. Co.) was offered by Dr. Xing-Cong Li (National Center for Natural Products Research, University of Mississippi). Column chromatography was performed on a Diaion HP20SS (Mitsubishi Chemical Co.), using MCI-gel CHP 20P (75–150 μ m, Mitsubishi Chemical Co.), Sephadex LH-20 (25–100 μ m, Pharmacia Fine Chemical Co. Ltd.), Chromatorex ODS (100–200 mesh, Fuji Silysia Chemical Co. Ltd.), and Toyopearl HW-40F (37–70 μ m, Tosoh Co.). TLC was carried on silica gel G-precoated plates (Qingdao Haiyang Chemical Co.) with benzene–ethyl formate–formic acid (3:6:1). Spots were detected by spraying with ferric chloride (FeCl₃) and 10% sulfuric acid reagents, followed by heating.

2.3. Plant material

Chinese toon (the fresh young leaves and shoots of *T. sinensis*) was collected from the north suburb of Kunming, Yunnan, China, and identified by Prof. Chong-Ren Yang (Kunming Institute of Botany, Chinese Academy of Sciences).

2.4. Extraction and isolation

Chinese toon (the fresh young leaves and shoots of *T*. *sinensis*) (10.0 kg) was cut into small pieces and extracted with 80% aqueous acetone ($20 1 \times 3$) at room temperature.

After removal of the organic solvent, the extract was concentrated to a small volume (2 l), and then partitioned with ethyl ether and EtOAc successively. The crude extract, ethyl ether, EtOAc and aqueous fractions showed SC₅₀ values of 14.8, 172, 8.22 and 111 μ g/ml, respectively, on DPPH assay. The result suggested that the antioxidants were mainly contained in the EtOAc fraction, on which further investigation was focussed.

The EtOAc fraction (160 g) was fractionated on Diaion HP20SS eluted with H₂O-MeOH (1:0-0:1) to give eight fractions $(A_1 - A_8)$. Repeated column chromatography on Sephadex LH-20, MCI-gel CHP20P and Chromatorex ODS, eluted with H_2O -MeOH (1:0-0:1), led to the isolation of 1 (400 mg) and 2 (26 mg) from fraction A_2 (10.5 g), and 3 (24 mg), 4 (12 mg), 5 (34 mg), 6 (25 mg), and 7 (150 mg) from fraction A₃ (24.2 g). Fraction A₄(11.4 g) was subjected to Sephadex LH-20 chromatography eluted with H₂O-MeOH (1:0-0:1) and then subjected to silica gel chromatography eluted with CHCl3-MeOH-H2O (8:2:0.2) to yield 10 (30 mg). Fraction A₅ (7.5 g) was subjected to silica gel chromatography eluted with CHCl₃-MeOH-H₂O (8.5:1.5:0.1) to give 8 (30 mg) and 9 (18 mg). Fraction A_6 (11.0 g) was subjected to silica gel chromatography eluted with CHCl₃-MeOH-H₂O (9:1:0.1) to yield 11 (20 mg) and 12 (16 mg).

2.5. DPPH radical-scavenging assay

The DPPH assay was performed as described (Yoshida et al., 1989). In this assay, ascorbic acid was used as positive control, and reaction mixtures containing an ethanolic solution of 200 μ M DPPH (100 μ l) and two fold serial dilutions of sample (dissolved in 100 μ l ethanol, with amounts of sample ranging from 2 to 1000 μ g/ml) were placed in a 96 well microplate and incubated at 37 °C for 30 min. After incubation, the absorbance was read at 517 nm and the scavenging activity was determined by the following equation: % scavenging activity = $[A_{\text{control}} - A_{\text{sample}}]/A_{\text{control}} \times 100$. The SC₅₀ values were obtained through extrapolation from linear regression analysis and denoted the concentration of sample required to scavenge 50% of DPPH radicals. The antioxidant activities were evaluated by SC₅₀ value.

2.6. Statistics

The data presented are means \pm SD of three determinations.

3. Results and discussion

3.1. Identification of compounds 1-12

The 80% aqueous acetone extract of Chinese toon (the fresh young leaves and shoots of *T. sinensis*) was partitioned successively with ethyl ether and EtOAc. The crude extract as well as the EtOAc fraction, displayed obvious

antioxidant activity (SC₅₀ = 14.8 and 8.22 μ g/ml, respectively) on DPPH radical-scavenging assay.

Further purification of the EtOAc fraction was carried out with Diaion HP20SS, MCI-gel CHP 20P, Sephadex LH-20, Chromatorex ODS, and silica gel column chromatography. This led to the determination of 12 known phenolic compounds, whose structures were determined to be gallic acid (1), methyl gallate (2), trigallic acid (3), 6-O-galloyl-D-glucose (4), 1,2,3-tri-O-galloyl- β -D-glucopyranose (5), 1,2,3,6-tetra-*O*-galloyl-β-D-glucopyranose (6), 1,2,3, 4.6-penta-O-gallovl-β-D-glucopyranose (7), quercetin-3-O- β -D-glucopyranoside (8), quercetin-3-O- α -L-rhamnopyranoside (9), rutin (10), kaempferol-3-O-β-D-glucopranoside (astragalin) (11) and kaempferol-3-O- α -L-arabinopyranoside (juglalin) (12), respectively, by detailed spectroscopic analysis and comparing with the literature data. Compounds 3-8 and 11 were identified from T. sinensis for the first time. Their structures are shown in Fig. 1.

Gallic acid (1): $C_7H_6O_5$, white needle crystals, negative FAB-MS m/z: 169 $[M-H]^-$, ¹H NMR (400 MHz, CD₃OD): δ 7.07 × 2 (2H, s, H-2, 6); ¹³C NMR (100 MHz, CD₃OD): see Table 1.

Methyl gallate (2): $C_8H_8O_5$, white needle crystals, negative FAB-MS m/z: 183 $[M-H]^-$, ¹H NMR (500 MHz, CD₃OD): δ 7.04 (2H, s, H-2, 6), 3.80 (3H, s, -OCH₃); ¹³C NMR (100 MHz, CD₃OD): see Table 1 (Nishizawa, Yamagishi, & Nonaka, 1982).

Trigallic acid (3): $C_{21}H_{14}O_{13}$, white powder, negative FAB-MS m/z: 473 [M-H]⁻. ¹H NMR (500 MHz,



Fig. 1. Structures of phenolic antioxidants (1-12) from Chinese toon.

¹³ C NMF	ζ (δ in ppm,	in CD_3OD)	data for compounds 1–7					
Position	1	2	3	4		5	9	7
1	122.0 (s)	121.4 (s)	120.5–122.5 (s)	121.4 (s)		120.4, 121.1, 121.4 (each s)	119.8, 120.4, 120.8, 121.0 (each s)	119.6, 120.1, 120.2, 120.3, 120.9 (each s)
2	110.4 (d)	110.0 (d)	110.0–110.9 (d)	110.5 (d)		$110.4 (s) \times 3$	110.0, 110.2, 110.4, 110.6 (each d)	110.0, 110.3, 110.4 \times 2, 110.6 (each d)
	146.4 (s)	146.5 (s)	146.5–147.5 (s)	146.5 (s)		145.8, 146.4, 146.5 (each s)	146.3, 146.4, 146.5, 146.6 (each s)	146.3, 146.4, 146.5 \times 2, 146.6 (each s)
4	139.6 (s)	139.7 (s)	140.1–140.5 (s)	139.8 (s)		$140.1 (s) \times 3$	139.9, 140.2, 140.6, 140.8 (each s)	140.1, 140.2, 140.4, 140.8, 140.9 (each s)
5	146.4 (s)	146.5 (s)	146.5–147.5 (s)	146.5 (s)		145.8, 146.4, 146.5 (each s)	146.3, 146.4, 146.5, 146.6 (each s)	146.3 , 146.4 , 146.5×2 , 146.6 (each s)
9	110.4 (d)	110.0 (d)	110.0–110.9 (d)	110.0 (d)		$110.4 (s) \times 3$	110.0, 110.2, 110.4, 110.6 (each d)	110.0, 110.3, 110.4 \times 2, 110.6 (each d)
7	167.5 (s)	167.5 (s)	166.3, 166.6, 169.8 (each s)	168.5 (s)		166.4, 167.2, 167.8 (each s)	166.2, 166.4, 167.1, 167.3 (each s)	166.2, 166.9, 167.0, 167.3, 167.9 (each s)
-OCH ₃		52.3 (q)						
Glu				a form	β form			
1				94.0 (d)	98.3 (d)	93.9 (d)	93.9 (d)	93.8 (d)
2				71.9 (d)	74.8 (d)	72.4 (d)	72.3 (d)	72.2 (d)
3				74.1(d)	76.2 (d)	76.7 (d)	76.4 (d)	74.4 (d)
4				(p) 6.02	71.1(d)	69.3 (d)	(b) 60.9 (d)	(b) 8.69
5				73.8 (d)	75.6 (d)	77.0 (d)	76.6 (d)	74.6 (d)
9				64.9 (t)	64.9 (t)	61.9 (t)	63.9 (t)	63.1 (t)

CD₃OD): δ 7.11, 7.21 × 2, 7.23, 7.25, 7.40 (s, galloyl H-2, 6); ¹³C NMR (100 MHz, CD₃OD): galloyl: 120.5–122.5 (s, C-1), 110.0–110.9 (d, C-2, 6), 146.5–147.5 (s, C-3, 5), 140.1–140.5(s, C-4), 115.0, 117.4, 130.1, 132.5, 144.1, 151.6 (s, weak, signal for *p*- and *m*- isomers), 166.3, 166.6, 169.8 (–COO–) (Kadil, El-Sayed, Micheal, & Mabry, 1996).

6-*O*-Galloyl-D-glucopyranose (**4**): C₁₃H₁₆O₁₀, white powder, negative FAB-MS *m/z*: 331 [M–H]⁻(100), 169 [M–H-162]⁻(10); $[\alpha]_D^{19} = 28.3$ (MeOH; c 1.0), ¹H NMR (400 MHz, CD₃OD): δ 4.58 (1/2H, d, *J* = 7.0 Hz, β form H-1), 5.10 (1/2H, d, *J* = 3.5 Hz, α form H-1), 7.15 (2H, s, galloyl H); ¹³C NMR (100 MHz, CD₃OD): see Table 1 (Nonaka & Nishioka, 1983).

1,2,3-tri-*O*-Galloyl-β-D-glucopyranose (5): $C_{27}H_{24}O_{18}$, pale yellow powder, negative FAB-MS *m/z*: 635 [M–H]⁻. ¹H NMR (500 MHz, CD₃OD): δ 6.90, 7.01, 7.06 (each 2H, s, galloyl H-2, 6); glu: 6.04 (1H, d, *J* = 8.2 Hz, H-1), 5.39 (1H, dd, *J* = 8.2, 9.3 Hz, H-2), 5.51 (1H, dd, *J* = 9.3, 9.6 Hz, H-3), 3.50–4.00 (4H, m, H-4, 5, 6); ¹³C NMR (100 MHz, CD₃OD): see Table 1 (Nawwar, Hussein, & Merfort, 1994).

1,2,3,6-tetra-*O*-Galloyl-β-D-glucopyranose (6): C₃₄H₂₈O₂₂, pale yellow powder, negative FAB-MS m/z: 787 [M–H]⁻. ¹H NMR (500 MHz, CD₃OD): δ 6.96, 7.05, 7.07, 7.13 (each 2H, s, galloyl H-2, 6); glu: 6.10 (1H, d, J = 8.3 Hz, H-1), 5.44 (1H, dd, J = 8.3, 9.3 Hz, H-2), 5.51 (1H, dd, J = 9.3, 9.5 Hz, H-3), 3.50–4.00 (2H, m, H-4, 5), 4.60 (2H, m, H-6); ¹³C NMR (100 MHz, CD₃OD): see Table 1 (Nishizawa, Yamagishi, Nonaka, & Nishioka, 1983).

1,2,3,4,6-penta-*O*-Galloyl-β-D-glucopyranose (7): C₄₁H₃₂O₂₆, white powder, negative FAB-MS *m/z*: 939 [M–H]⁻. ¹H NMR (500 MHz, CD₃OD): δ 6.89, 6.94, 6.97, 7.04, 7.10 (each 2H, s, galloyl H-2, 6); glu: 6.23 (1H, d, J = 8.4 Hz, H-1), 5.58 (1H, dd, J = 8.4, 9.6 Hz, H-2), 5.90 (1H, dd, J = 9.5, 9.6 Hz, H-3), 5.60 (1H, t, J = 8.7, 9.5 Hz, H-4), 3.78 (1H, m, H-5), 4.40–4.52 (2H, m, H-6); ¹³C NMR (100 MHz, CD₃OD): see Table 1 (Nonaka, Ishimatsu, Tanaka, Nishioka, & Nishizawa, 1987).

Quercetin-3-O- β -D-glucopyranoside (**8**): C₂₁H₂₀O₁₂, yellow powder, negative FAB-MS m/z: 463 [M–H]⁻. ¹H NMR (500 MHz, DMSO- d_6) and ¹³C NMR (100 MHz, DMSO- d_6): see Table 2 (Markham, Ternai, Stanley, Geiger, & Mabry, 1978).

Quercetin-3-O- α -L-rhamopyranoside (quercetrin) (9): C₂₁H₂₀O₁₁, yellow powder, negative FAB-MS m/z: 447 [M–H]⁻. ¹H NMR (500 MHz, DMSO- d_6) and ¹³C NMR (100 MHz, DMSO- d_6): see Table 2 (Agrawal, 1989).

Rutin (10): $C_{27}H_{30}O_{16}$, yellow powder, negative FAB-MS m/z: 609 $[M-H]^-$. ¹H NMR (400 MHz, DMSO- d_6) and (100 MHz, DMSO- d_6): see Table 2 (Wenkert & Gottlieb, 1977).

Kaempferol-3-O- β -D-glucopranoside (astragalin) (11): C₂₁H₂₀O₁₁, yellow powder, negative FAB-MS m/z: 447 [M–H]⁻. ¹H NMR (400 MHz, DMSO- d_6) and ¹³C NMR (100 MHz, DMSO- d_6): see Table 2 (Agrawal, 1989).

Kaempferol-3-O- α -L-arabinopyranoside (juglalin) (12): C₂₀H₁₈O₁₀, yellow powder, negative FAB-MS m/z: 417 [M–H]⁻, ¹H NMR (500 MHz, DMSO- d_6) and ¹³C NMR (125 MHz, DMSO- d_6): see Table 2 (Gong, 1986).

3.2. DPPH radical-scavenging assay

DPPH (1,1-diphenyl-2-picrylhydrazyl) radical-scavenging assay is widely used to evaluate antioxidant capacity in a short time (Blois, 1958). The antioxidant activities of compounds 1-12 isolated from Chinese toon were determined by DPPH radical-scavenging assay and the results are shown in Table 3. Most of the isolated compounds exhibited considerable scavenging activity on DPPH assay.

Gallic acid (1) and its derivatives (2 and 3) displayed obvious antioxidant activity in the DPPH assay. Gallic acid (1) is a well-known natural antioxidant found widely in plants, and methyl gallate (2) from *T. sinensis* was reported to have protective effect against hydrogen peroxide-induced oxidative stress and DNA damage in MDCK cells (Hsieh et al., 2004). Methyl gallate (2), and other gallates, were often used in food products as additives to increase shelf life by retarding lipid peroxidation, which is one of the major reasons for deterioration of food products during processing and storage (Van der Heijden, Janssen, & Strik, 1986).

Gallotannins 4–7 showed stronger antioxidant activitites than that of ascorbic acid (SC₅₀ = 30.79 μ M) and the order of their activities was 4 < 5 < 6 < 7. This observation suggested that the antioxidant activity of gallotannin was increased when there were more galloyl groups in the molecule, which agreeds with the previous report by Zhao, Sun, Hou, Wei, and Xin (2005). Gallotannins were reported to possess inhibition activity on HIV-RT and the growth of many tumor cells (Kashiwada, Nonaka, Nishioka, Chang, & Lee, 1992).

Flavonoids (8-12), isolated from Chinese toon, displayed antioxidant activities from moderate to weak on DPPH assay. Further analysis, by comparing their activities with a reference compound, quercetin, indicated that flavonoids belonging to the quercetin class (8-10) possessed higher antioxidant activities than did kaempferol derivatives (11, 12), due to their possessing an o-dihydroxy B-ring structure, which conferred higher stability in the radical form and participating in electron delocalisation. This conclusion is consistent with those reported in the literature (Pietta, 2000; Rice-Evans, Miller, & Paganga, 1996, 1997) and confirmed that O-glycosylation at C-3 had a negative effect on antioxidant activity, and the presence of a bigger substituent at the C-3 position will reduce the antioxidant activity, probably due to steric hindrance (Braca et al., 2003; Cioffi et al., 2002). Flavonoids have been considered to be an important antioxidants for a long time and were reported to play an important role in the prevention of lipid peroxidation and cardiovascular disease (Hertog & Holmann, 1996; Lekse, Xia, Stark, Morrow, & May, 2001).

Position	8		9		10		11		12		
	$\delta_{\rm C}$	$\delta_{ m H}$	$\delta_{\rm C}$	$\delta_{ m H}$	$\delta_{\rm C}$	$\delta_{ m H}$	$\delta_{\rm C}$	$\delta_{ m H}$	$\delta_{\rm C}$	$\delta_{ m H}$	
2	156.2 (s)		156.5 (s)		156.6 (s)		156.2 (s)		156.2 (s)		
3	133.3 (s)		134.2 (s)		133.3 (s)		133.5 (s)		133.2 (s)		ĸ
4	177.4 (s)		177.7 (s)		177.4 (s)		177.5 (s)		179.8 (s)		<u>-</u> .
5	161.2 (s)		161.3 (s)		161.2 (s)		161.1 (s)		161.0 (s)		W
6	98.6 (d)	6.19 (d, 2.0)	98.8 (d)	6.18 (d, 1.2)	98.7 (d)	6.18 (d, 1.9)	98.6 (d)	6.20 (d, 2.0)	98.7 (d)	6.19 (d, 2.0)	gut
7	164.1 (s)		164.5 (s)		164.2 (s)		164.1 (s)		164.1 (s)		. et
8	91.4 (d)	6.39 (d, 2.0)	93.7 (d)	6.39 (d, 1.2)	93.6 (d)	6.37 (d, 1.9)	93.6 (d)	6.31 (d, 2.0)	93.7 (d)	6.43 (d, 2.0)	al.
9	156.3 (s)		157.2 (s)		156.6 (s)		156.2 (s)		156.9 (s)		2
10	104.0 (s)		104.0 (s)		104.0 (s)		103.9 (s)		104.0 (s)		Foo
1'	121.6 (s)		121.1 (s)		121.6 (s)		120.6 (s)		121.0 (s)		d (
2'	116.2 (d)	6.82 (d, 2.0)	115.5 (d)	7.28 (d, 1.7)	115.3 (d)	7.54 (d, 2.1)	130.9 (d)	8.03 (d, 8.8)	130.9 (d)	8.09 (dd, 1.8, 8.8)	Che
3'	144.8 (s)		145.2 (s)		144.8 (s)		115.2 (d)	6.87 (d, 8.8)	115.1 (d)	6.88 (dd, 1.8, 8.8)	mi
4′	148.4 (s)		148.5 (s)		148.5 (s)		159.9 (s)		159.9 (s)		stry
5'	115.2 (d)	6.83 (d, 6.9)	115.6 (d)	6.80 (d, 8.8)	116.3 (d)	6.82 (d, 8.1)	115.2 (d)	6.87 (d, 8.8)	115.1 (d)	6.88 (dd, 1.8, 8.8)	, 10
6'	121.2 (d)	7.57 (dd, 2.0, 6.9)	120.7 (d)	7.25 (dd, 1.7, 8.8)	121.2 (d)	7.52 (dd, 2.1, 8.1)	130.9 (d)	8.03 (d, 8.8)	130.9 (d)	8.09 (dd, 1.8, 8.8)	10
1″	100.8 (d)	5.45 (d, 7.3)	101.8 (d)	5.24 (brs)	101.2 (d)	5.32 (d, 7.6)	106.1 (d)	5.44 (d, 7.4)	100.8 (d)	5.20 (d, 6.7)	(26
2"	74.1 (d)		70.4 (d)	4.00 (brs)	74.1 (d)		72.5 (d)		74.2 (d)		07
3″	76.5 (d)		70.6 (d)	3.49 (dd, 3.2, 9.2)	76.5 (d)		74.7 (d)		76.4 (d)		ر پي
4″	69.9 (d)		71.2 (d)	3.12 (dd, 9.2, 9.2)	70.0 (d)		69.9 (d)		69.9 (d)		5
5″	77.6 (d)		70.1 (d)	3.22 (m)	75.9 (d)		66.1 (t)		77.5 (d)		-37
6″	61.0 (t)		17.5 (q)	0.80 (d, 6.0)	67.0 (t)				60.8 (t)		1
1‴					100.8 (d)	4.36 (s)					
2‴					70.4 (d)						
3‴					70.6 (d)						
4‴					71.9 (d)						
5‴					68.3 (d)						
6‴					17.8 (q)	0.97 (d, 6.1)					

Table 2 ¹H and ¹³C NMR (δ in ppm, J in Hz, in DMSO- d_6) data for compounds 8–12

Table 3

Samples	$SC_{50} \left(\mu M \right)^b$
Gallic acid (1)	12.1 ± 0.16
Methyl gallate (2)	18.5 ± 0.23
Trigallic acid (3)	10.6 ± 0.28
6-O-Galloyl-D-glucose (4)	25.2 ± 0.34
1,2,3-tri-O-Galloyl-β-D-glucopyranose (5)	17.7 ± 0.15
1,2,3,6-tetra- <i>O</i> -Galloyl-β-D-glucopyranose (6)	10.3 ± 0.19
1,2,3,4,6-penta-O-Galloyl-β-D-glucopyranose (7)	7.1 ± 0.26
Quercetin 3- <i>O</i> -β-D-glucopyranoside (8)	39.7 ± 0.48
Quercetin 3-O-a-L-rhamopyranoside (9)	52.3 ± 0.62
Rutin (10)	63.4 ± 0.55
Kaempferol-3- <i>O</i> -β-D-glucopranoside (11)	110 ± 2.01
Kaempferol-3-O-α-L-arabinopyranoside (12)	125 ± 2.55
Ascorbic acid (positive control)	30.8 ± 0.30
Quercetin (reference)	13.7 ± 0.22

^a SC₅₀: radical-scavenging activity (concentration in μ M necessary for 50% reduction of DPPH radical).

^b Values represent means \pm SD (n = 3).

As the main anti-oxidative constituents in Chinese toon were gallic acid and its derivatives, gallotannins and flavonol glycosides may play an important role in the antioxidant activity of this tree vegetable. From the health point of view, Chinese toon (the young leaves and shoots of *T. sinensis*) is an ideal dietary vegetable with natural antioxidants, and is worthy of further development as a healthypromoting food.

Acknowledgements

The authors are grateful to the staffs of the analytical group at State Key Laboratory of Phytochemistry & Plant Resources in West China, and Kunming Institute of Botany, Chinese Academy of Sciences, for measuring all the spectral data, and also to Dr. Xing-Cong Li, for kind provision of DPPH radical agent.

References

- Agrawal, P. K. (1989). Carbon-¹³NMR of flavonoids. New York: Elsevier Scientific Publishing Company (pp. 335).
- Aviram, M. (2000). Review of human studies on oxidative damage and antioxidant protection related to cardiovascular diseases. *Free Radical Research*, 33(suppl.), 85–97.
- Blois, M. S. (1958). Antioxidant determinations by the use of a stable free radical. *Nature*, 181, 1199–1200.
- Braca, A., Fico, G., Morelli, I., Simone, F. D., Tome, F., & Tommasi, N. D. (2003). Antioxidant and free radical scavenging activity of flavonol glycosides from different *Aconitum* species. *Journal of Ethnopharmacology*, 86, 63–67.
- Chang, H. C., Hung, W. C., Huang, M. S., & Hsu, H. K. (2002). Extract from the leaves of *Toona sinensis roemor* exerts potent antiproliferative effect on human lung cancer cells. *American Journal of Chinese Medicine*, 30, 307–314.
- Chen, T. S., Luo, Z. P., Cui, H. A., Zhen, X. Q., & Liu, Z. Z. (2000). Preliminary study of chemical constituents from leaves of *Toona* sinensis. Shanxi Forest Science and Technology, 20, 1–2.
- Cioffi, G., D'Auria, M., Braca, A., Mendez, J., Castillo, A., Morelli, I., et al. (2002). Antioxidant and free-radical scavenging activity of

constituents of the leaves of *Tachigalia paniculata*. Journal of Natural Products, 65, 1526–1529.

- Editorial Board of Zhong-Hua-Ben-Cao (China Herbal), State Administration of Traditional Chinese Medicine (1999). *Zhong-Hua-Ben-Cao*(*China herbal*) (Vol. 5, pp. 45–48). Shanghai, China: Shanghai Science and Technology Press.
- Gong, Y. H. (1986). ¹³C NMR of natural organic products. Yunnan, China: Yunnan Science and Technology Press (p.167).
- Halliwell, B. (1994). Free radicals, antioxidants, and human diseases: curiosity, cause, or consequence? *Lancet*, *334*, 721–724.
- Halliwell, B., Gutteridge, J. M. C., & Cross, C. E. (1992). Free radicals, antioxidants and human disease: Where are now? *The Journal Laboratory and Clinical Medicine*, 119, 598–919.
- Hertog, M. G. L., & Holmann, P. D. H. (1996). Potential health effects of the dietary flavonol quercetin. *European Journal of Clinical Nutrition*, 50, 63–71.
- Hsieh, T. J., Liu, T. Z., Chia, Y. C., Chern, C. L., Chuang, M. C., Mau, S. Y., et al. (2004). Protective effect of methyl gallate from *Toona sinensis* (Meliaceae) against hydrogen peroxide-induced oxidative stress and DNA damage in MDCK cells. *Food and Chemical Toxicology*, 42, 843–850.
- Hsu, H. K., Yang, Y. C., Hwang, J. H., & Hong, S. J. (2003). Effects of *Toona sinensis* leaf extract on lipolysis in differentiated 3T3-L1 adipocytes. *Kaohsiung Journal of Medical Sciences*, 19, 385–390.
- Kadil, F. E., El-Sayed, N. H., Micheal, M. S., & Mabry, T. J. (1996). Gallotannins and flavonoids from *Haematoxylon campechianum*. *Phytochemistry*, 42, 1243–1245.
- Kashiwada, K., Nonaka, G. I., Nishioka, I., Chang, J. J., & Lee, K. H. (1992). Antitumor Agents, 129. Tannins and related compounds as selective cytotoxic agents. *Journal of Natural Products*, 55, 1033–1043.
- Lekse, J. M., Xia, L., Stark, J., Morrow, J. D., & May, M. J. (2001). Plant catechols prevent lipid peroxidation in human plasma and erythrocytes. *Molecular Cell Biochemistry*, 226, 89–95.
- Luo, X. D., Basile, M. J., & Kennelly, E. J. (2002). Polyphenolic antioxidants from the fruit of *Chrysophyllum cainito* L. (Star Apple). *Journal of Agricultural and Food Chemistry*, 50, 1379–1382.
- Luo, X. D., Wu, S. H., Ma, Y. B., & Wu, D. G. (2001). Studies on chemical constituents of *Toona sinensis*. *Chinese Traditional and Herbal Drugs*, 32, 390–391.
- Mangiapane, H., Thomson, J., Salter, A., Brown, S., Bell, G. D., & White, D. A. (1992). The inhibition of the oxidation of low-density lipoprotein by (+)-catechin, a naturally occurring flavonoid. *Biochemistry Pharmacology*, 43, 445–450.
- Markham, K. R., Ternai, B., Stanley, R., Geiger, H., & Mabry, T. J. (1978). Carbon-13 NMR studies of flavonoids-III: naturally occurring flavonoid glycosides and their acylated derivatives. *Tetrahedron*, 34, 1389–1397.
- Nawwar, M. A. M., Hussein, S. A. M., & Merfort, I. (1994). NMR spectral analysis of polyphenols from *Punica Granatum*. *Phytochemisty*, 36, 793–798.
- Nishizawa, M., Yamagishi, T., & Nonaka, G. I. (1982). Tannins and related compounds. 5. Isolation and characterization of polygalloylglucose from Chinese gallotannin. *Journal of the Chemical Society*, *Perkin Transactions I*, 104, 2963–2968.
- Nishizawa, M., Yamagishi, T., Nonaka, G. I., & Nishioka, I. (1983). Tannins and related compounds Part 9. Isolation and characterization of polygalloylglucose from Turkish Galls (*Quercus infectoria*). Journal of the Chemical Society Perkin Transactions I, 105, 961–965.
- Nonaka, G. I., Ishimatsu, M., Tanaka, T., Nishioka, I., & Nishizawa, M. (1987). Tannins and related compounds. LVIII. Novel gallotannins possessing an α-glucose core from *Nuphar japonicum* DC. *Chemical & Pharmaceutical Bulletin, 35*, 3127.
- Nonaka, G. I., & Nishioka, I. (1983). Tannins and related compounds X. Rhubarb (2): Isolation and structures of a glycerol gallate, gallic acid glucoside gallate, galloylglucose and isolindleyin. *Chemical & Pharmaceutical Bulletin*, 31, 1652–1658.
- Pietta, P. G. (2000). Flavonoids as antioxidants. Journal of Natural Products, 63, 1035–1042.

371

- Rice-Evans, C., Miller, N. J., & Paganga, G. (1996). Structure–antioxidant relationship of flavonoids and phenolic acids. *Free Radical Biology and Medicine*, 20, 933–956.
- Rice-Evans, C., Miller, N. J., & Paganga, G. (1997). Antioxidant properties of phenolic compounds. *Trends in Plant Science*, 2, 152–159.
- Van der Heijden, C. A., Janssen, P. J. C. M., & Strik, J. J. T. W. A. (1986). Toxicology of gallates: a review and evaluation. *Food and Chemical Toxicology*, 24, 1067–1070.
- Wenkert, E., & Gottlieb, H. E. (1977). Carbon-13 nuclear magnetic resonance spectroscopy of flavonoid and isoflavonoid compounds. *Phytochemistry*, 16, 1811–1816.
- Williams, G. M., Iatropoulos, M. J., & Whysner, J. (1999). Safety assessment of butylated hydroxyanisole and butylated hydroxytoluene

as antioxidant food additives. Food and Chemical Toxicology, 37, 1027-1038.

- Yang, Y. C., Hsu, H. K., Hwang, J. H., & Hong, S. J. (2003). Enhancement of glucose uptake in 3T3-L1 adipocytes by *Toona* sinensis leaf extract. Kaohsiung Journal of Medical Sciences, 19, 327–333.
- Yoshida, T., Mori, K., Hatano, T., Okumura, T., Uehara, I., Komagoe, K., et al. (1989). Studies on inhibition mechanism of autoxidation by tannins and flavonoids. V. Radical-scavenging effects of tannins and related polyphenols on 1,1-diphenyl-2-picrylhydrzyl radical. *Chemical* & *Pharmaceutical bulletin*, 37, 1919–1921.
- Zhao, X., Sun, H., Hou, A., Wei, T., & Xin, W. (2005). Antioxidant properties of two gallotannins isolated from the leaves of *Pistacia weinmannifolia*. *Biochimica et Biophysica Acta*, *30*, 103–110.