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### 7,8-Secolignans from the fruits of *Schisandra neglecta*

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## 7,8-Secolignans from the fruits of *Schisandra neglecta*

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Two new 7,8-secolignans, neglectahenols E and F (**1** and **2**), together with four known 7,8-secolignans (**3–6**), were isolated from the fruits of *Schisandra neglecta*. The structures were elucidated by spectroscopic methods, including extensive 1D and 2D NMR techniques. Compounds **1–6** were tested for their anti-tobacco mosaic virus (anti-TMV) activities at the concentration of 20  $\mu$ M. Compounds **1** and **6** showed high anti-TMV activities with inhibition rates of 38.2% and 32.7%, respectively. These rates are higher than that of a positive control. Compounds **2–5** also showed modest anti-TMV activities with inhibition rates in the range of 22.8–28.7%. These rates are close to that of a positive control.

**Keywords:** *Schisandra neglecta*; 7,8-secolignans; anti-tobacco mosaic virus (anti-TMV) activities

### 1. Introduction

The stems and fruits of plants in the genus *Schisandra* are used commonly in Traditional Chinese Medicine for their diverse beneficial bioactivities [1,2]. Previous studies have shown that these species are rich in lignans and triterpenoids, especially dibenzocyclooctadiene lignans, which have been found to possess some potentially beneficial activities, including anti-hepatotoxic, anti-HIV, antioxidant, antitumor, and cytotoxic effects [3–5].

*Schisandra neglecta* A.C. Smith (Schisandraceae) is a climbing plant distributed mainly in southwest of mainland China. In previous studies, several new dibenzocyclooctadiene lignans were isolated from the stems of *S. neglecta* [6–8]. In our continuing efforts to identify bioactive natural products from the medicinal plants of the family Schisandraceae, a chemical investigation on the fruit of *S. neglecta*

was carried out, which were collected from the Xichang Prefecture, Sichuan Province, China. As a result, two new (**1** and **2**) and four known (**3–6**) 7,8-secolignans were isolated from this species. Described in this paper are the structure elucidation of compounds **1–6** and their anti-tobacco mosaic virus (anti-TMV) activities.

### 2. Results and discussion

The fruits of *S. neglecta* were extracted with 70% acetone. The extract produced was subjected repeatedly to column chromatography on silica gel, Sephadex LH-20, RP-18, and RP-HPLC, to afford two 7,8-secolignans, neglectahenols E and F (**1** and **2**), together with four known 7,8-secolignans (**3–6**). The structures of compounds **1–2** are shown in Figure 1, and the <sup>1</sup>H and <sup>13</sup>C NMR data of compounds **1** and **2** are listed in Table 1.

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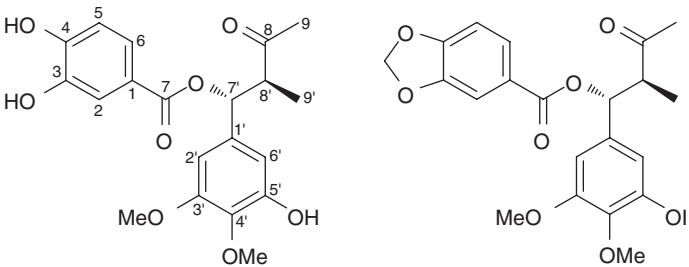


Figure 1. The structures of compounds 1–2.

The known compounds were identified as marphenol A (**3**) [9], 7,8-secoholostylone B (**4**) [10], neglectahenol A (**5**) [11], and neglectahenol C (**6**) [11], by comparing their spectroscopic data with values reported in the literature.

Compound **1** was obtained as a white amorphous powder, and was assigned the molecular formula  $C_{20}H_{22}O_8$  by HRE-SIMS at  $m/z$  391.1399  $[M + H]^+$ . Its  $^1H$

and  $^{13}C$  NMR (Table 1) spectra showed signals of 22 protons and 20 carbons, respectively, corresponding to a 1',3',4',5'-tetra-substituted benzyl ( $\delta_C$  133.3 s, 102.9 d, 152.3 s, 139.8 s, 148.8 s, 109.1 d;  $\delta_H$  6.88 s, 7.26 s) [9], a 1,3,4-tri-substituted benzyl ( $\delta_C$  126.3 s, 108.1 d, 150.9 s, 147.2 s, 109.9 d, 125.3 d;  $\delta_H$  7.68 s, 6.83 d  $J = 8.2$ , 7.80 d  $J = 8.2$ ) [9], two carbonyl carbons ( $\delta_C$  165.0 s, 209.0 s), two methyl

Table 1.  $^1H$  and  $^{13}C$  NMR spectral data of **1** and **2** ( $\delta$  ppm,  $J$  Hz) measured in pyridine- $d_5$ .

No.	<b>1</b>		<b>2</b>	
	$\delta_C$ (m)	$\delta_H$ (mult, $J$ ,Hz)	$\delta_C$ (m)	$\delta_H$ (mult, $J$ ,Hz)
1	126.3 s		125.2 s	
2	108.1 d	7.68, s	111.5 d	7.72, d, $J = 1.9$
3	150.9 s		148.5 s	
4	147.2 s		151.0 s	
5	109.9 d	6.83, d, $J = 8.2$	109.0 d	7.04, d, $J = 8.2$
6	125.3 d	7.80, d, $J = 8.2$	124.5 d	7.85, dd, $J = 8.2, 1.9$
7	165.0 s		165.3 s	
8	209.0 s		208.7 s	
9	29.7 q	2.23, s	29.3 q	2.25, s
1'	133.3 s		136.0 s	
2'	102.9 d	6.88, s	105.2 d	6.73, s
3'	152.3 s		154.9 s	
4'	139.8 s		139.3 s	
5'	148.8 s		154.9 s	
6'	109.1 d	7.26, s	105.2 d	6.73, s
7'	78.9 d	6.42, d, $J = 9.9$	79.1 d	6.40, d, $J = 9.9$
8'	51.3 d	3.44–3.46, m	53.0 d	3.44–3.46, m
9'	14.3 q	1.01, d, $J = 7.1$	13.6 q	1.06, d, $J = 7.1$
OMe-3'	56.0 q	3.73, s	55.8 q	3.76, s
OMe-4'	60.7 q	3.79, s	60.8 q	3.83, s
OMe-5'			55.8 q	3.76, s
Ar–OH-3		10.80, br s		
Ar–OH-4		10.68, br s		
Ar–OH-5'		11.10, br s		
–OCH <sub>2</sub> O–			101.6 t	5.92, 5.97, s

groups ( $\delta_C$  29.7 q, 14.3 q;  $\delta_H$  2.23 s, 1.01 d  $J = 7.1$ ), one methine group ( $\delta_C$  51.3 d;  $\delta_H$  3.45 m), one oxidated methine group ( $\delta_C$  78.9 d;  $\delta_H$  6.42 d  $J = 9.9$ ), two methoxyl groups ( $\delta_C$  56.0 q, 60.7 q;  $\delta_H$  3.73 s, 3.79 s), and three phenolic hydroxy groups ( $\delta_H$  10.80 br s, 10.68 br s, 11.10 br s), which were in accordance with the molecular formula  $C_{20}H_{22}O_8$ . Strong absorption bands accounting for hydroxy group ( $3445\text{ cm}^{-1}$ ), carbonyl group ( $1745$ ,  $1709\text{ cm}^{-1}$ ), and aromatic group ( $1638$ ,  $1582$ ,  $1518$ ,  $1485\text{ cm}^{-1}$ ) could also be observed in its IR spectrum. The UV spectrum of **1** showed absorption maximum at 272 nm which confirmed the existence of the aromatic functions. The results of HMBC experiment (Figure 2) supported the presence of ketone ( $\delta_C$  209.0), methyl ( $\delta_C$  29.7), and veratryl ester ( $\delta_{C=O}$  165.0) groups. The HMBC experiment also showed that the carbonyl carbon C-8 ( $\delta_C$  209.0) was correlated to both H<sub>3</sub>-9 ( $\delta_H$  2.23, s) and H<sub>3</sub>-9' ( $\delta_H$  1.01, d,  $J = 7.1$ ). Furthermore, the prominent absorption bands at  $1709\text{ cm}^{-1}$  observed in the IR spectrum were in accordance with ketone and conjugated ester carbonyl groups in the molecule. A  $^1\text{H}$ - $^1\text{H}$  COSY experiment showed that H-7' ( $\delta_H$  6.42, d,  $J = 9.9$ ) was coupled to H-8' ( $\delta_H$  3.45, m) which, in turn, was also coupled to H<sub>3</sub>-9'. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR data of **1** were very similar to those of known compound, marphenol B [9], except for the appearance

of two phenolic hydroxy signal at ( $\delta_H$  10.80 and 10.68) and the disappearance of a methylenedioxy signal at ( $\delta_C$  102.5 t;  $\delta_H$  5.99, 6.01, each s). These indicated the methylenedioxy group in marphenol B was replaced by two phenolic hydroxy groups in **1**. These two phenolic hydroxy groups located at C-3 and C-4 were also supported by the HMBC correlations of one phenolic hydroxy proton signal ( $\delta_H$  10.80) with C-2 ( $\delta_C$  108.1), C-3 ( $\delta_C$  150.9), and C-4 ( $\delta_C$  147.2); and of another phenolic hydroxy proton signal ( $\delta_H$  10.68) with C-3 ( $\delta_C$  150.9), C-4 ( $\delta_C$  147.2), and C-5 ( $\delta_C$  109.9), respectively. Considering the magnitude of the coupling constant between H-7' and H-8' ( $J = 9.9$ ) as well as the similar  $^1\text{H}$  and  $^{13}\text{C}$  NMR data and optical rotation with those of reported marphenol B, the structure of **1** could be established as (7'*R*, 8'*S*)-7,8-secolignans [9]. Thus, the structure of **1**, neglectahenol E, was established.

Neglectahenol F (**2**), obtained as a white amorphous powder solid, had the molecular formula  $C_{22}H_{24}O_8$  as revealed by its HRESIMS at  $m/z$  439.1363  $[\text{M} + \text{Na}]^+$ . The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were very similar to those of marphenol A (**3**) [9]. Analysis of HSQC,  $^1\text{H}$ - $^1\text{H}$  COSY, and HMBC spectra of **2** showed that obvious chemical-shift differences were resulted from the substituents in aromatic ring, and a phenolic hydroxy group at C-5' in marphenol A was replaced by a methoxyl group in **2**. The appearance of typical aromatic proton signals ( $\delta_H$  6.73, s, 2H) and the two equivalent methoxyl proton signals ( $\delta_H$  3.76 s) also supported the 3',4',5'-trimethoxyl substituted aromatic for ring B. Compound **2** is therefore the 5'-methoxyl derivative of marphenol A.

Because certain 7,8-secolignans exhibit potential anti-virus activities [9,11], compounds **1**-**6** were tested for their anti-TMV activities. The anti-TMV activities were tested using the half-leaf method [12,13]. Ningnanmycin (a commercial product for plant disease in China), was

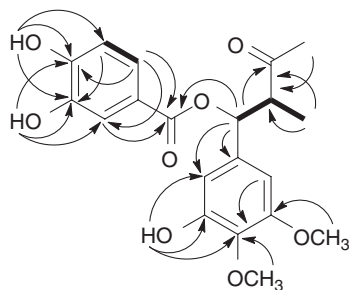


Figure 2. Key  $^1\text{H}$ - $^1\text{H}$  COSY (—) and HMBC (↷) correlations of **1**.

Table 2. TMV infection inhibition activities of compounds **1**–**6**.

Compounds	Inhibition rate (%)	Compounds	Inhibition rate (%)
<b>1</b>	38.2 ± 3.5	<b>5</b>	25.5 ± 3.0
<b>2</b>	25.6 ± 3.4	<b>6</b>	32.7 ± 2.5
<b>3</b>	22.8 ± 2.5	Ningnanmycin	31.8 ± 3.0
<b>4</b>	28.7 ± 2.7		

All results are expressed as mean ± SD;  $n = 3$  for all groups.

used as a positive control. Their antiviral inhibition rates at the concentration of 20  $\mu\text{M}$  were listed in Table 2. Compounds **1** and **6** showed high anti-TMV activities with inhibition rates of 38.2% and 32.7%, respectively. These rates are higher than that of a positive control. Compounds **2**–**5** also showed modest anti-TMV activities with inhibition rates in the range of 22.8–28.7%. These rates are close to that of a positive control.

### 3. Experimental

#### 3.1 General experimental procedures

Optical rotations were measured with a Horiba SEPA-300 polarimeter (Horiba, Tokyo, Japan). UV spectra were obtained using a Shimadzu UV-2401A spectrophotometer (Shimadzu, Kyoto, Japan). A Tensor 27 spectrophotometer (Bruker Optics, Ettlingen, Germany) was used for scanning IR spectroscopy with KBr pellets. 1D and 2D NMR spectra were recorded on DRX-500 spectrometers (Bruker Optics, Ettlingen, Germany) with TMS as an internal standard. Unless otherwise specified, chemical shifts ( $\delta$ ) were expressed in ppm with reference to the solvent signals. HRESIMS was performed on an API QSTAR time-of-flight spectrometer (Applied Biosystems, Foster City, American) and a VG Autospec-3000 spectrometer (VG, Manchester, England), respectively. Preparative HPLC was performed on a Shimadzu LC-8A preparative liquid chromatograph (Shimadzu, Kyoto, Japan) with a ZORBAX PrepHT GF (21.2 mm  $\times$  25 cm, 7  $\mu\text{m}$ ) column or a

Venusil MP C18 (20 mm  $\times$  25 cm, 5  $\mu\text{m}$ ) column. Column chromatography was performed with Si gel (200–300 mesh; Qing-dao Marine Chemical, Inc., Qingdao, China), Lichroprep RP-18 gel (40–63  $\mu\text{m}$ ; Merck, Darmstadt, Germany), and MCI gel (75–150  $\mu\text{m}$ ; Mitsubishi Chemical Corporation, Tokyo, Japan). The fractions were monitored by TLC, and spots were visualized by heating Si gel plates sprayed with 5%  $\text{H}_2\text{SO}_4$  in EtOH.

#### 3.2 Plant material

The fruits of *Schisandra neglecta* were collected in Xichang Prefecture, Sichuan Province, China, in September 2011. The identification of the plant material was verified by Prof. Xi-Wen Li of Kunming Institute of Botany, Chinese Academy of Sciences. A voucher specimen (KIB 11-9-38) has been deposited in the State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences.

#### 3.3 Extraction and isolation

The air-dried and powdered fruits of *S. neglecta* (5.0 kg) were extracted four times with 70% aqueous  $\text{Me}_2\text{CO}$  (4  $\times$  8 L) at room temperature and filtered, with the filtrate evaporated under reduced pressure and partitioned with EtOAc (4  $\times$  5 L). The EtOAc partition (357 g) was applied to silica gel (200–300 mesh) column chromatography, eluting with a  $\text{CHCl}_3$ –MeOH gradient system (20:1, 9:1, 8:2, 7:3, 6:4, 5:5), to give six fractions A–F. The further

separation of fraction B (9:1, 52.5 g) by silica gel column chromatography, eluted with  $\text{CHCl}_3$ –acetone (20:1–2:1), yielded mixtures B1–B7. Fraction B2 (9:1, 38.4 g) was subjected to silica gel column chromatography using petroleum ether–acetone and preparative HPLC (65%  $\text{MeOH}$ – $\text{H}_2\text{O}$ , flow rate 12 ml/min) to give **2** (16.4 mg, Rt 12 min) and **6** (14.8 mg, Rt 17 min). Fraction B3 (8:2, 26.7 g) was subjected to silica gel column chromatography using petroleum ether–acetone and preparative HPLC (60%  $\text{MeOH}$ – $\text{H}_2\text{O}$ , flow rate 12 ml/min) to give **3** (12.8 mg, Rt 13 min), **4** (15.2 mg, Rt 19 min), and **5** (18.6 mg, Rt 25 min). Fraction B4 (7:3, 22.8 g) was subjected to silica gel column chromatography using petroleum ether–acetone (20:1, 9:1, 8:2, 7:3, 6:4, 5:5) and preparative HPLC (55%  $\text{MeOH}$ – $\text{H}_2\text{O}$ , flow rate 12 ml/min) to give **1** (14.8 mg, Rt 12 min).

### 3.3.1 *Neglectahenol E (1)*

$\text{C}_{20}\text{H}_{22}\text{O}_8$ , obtained as a white amorphous powder;  $[\alpha]_{\text{D}}^{24.8} + 28.6$  ( $c$  0.20,  $\text{MeOH}$ ); UV ( $\text{MeOH}$ )  $\lambda_{\text{max}}$  (log  $\epsilon$ ): 272 (3.48), 356 (1.24) nm; IR (KBr)  $\nu_{\text{max}}$ : 3445, 3078, 2936, 2288, 1745, 1709, 1638, 1582, 1518, 1485, 1423, 1367, 1321, 1236, 1122, 1052, 1008, 874, 768  $\text{cm}^{-1}$ ; for  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data ( $\text{C}_5\text{ND}_5$ , 500 and 125 MHz), see Table 1; ESIMS (positive ion mode):  $m/z$  391  $[\text{M} + \text{H}]^+$ ; HRESIMS (positive ion mode):  $m/z$  391.1399  $[\text{M} + \text{H}]^+$  (calcd for  $\text{C}_{20}\text{H}_{23}\text{O}_8$ , 391.1393).

### 3.3.2 *Neglectahenol F (2)*

$\text{C}_{22}\text{H}_{24}\text{O}_8$ , obtained as a white amorphous powder;  $[\alpha]_{\text{D}}^{24.8} + 25.6$  ( $c$  0.20,  $\text{MeOH}$ ); UV ( $\text{MeOH}$ )  $\lambda_{\text{max}}$  (log  $\epsilon$ ): 273 (3.42), 355 (1.08) nm; IR (KBr)  $\nu_{\text{max}}$ : 3457, 3072, 2918, 2856, 1743, 1712, 1623, 1571, 1518, 1468, 1432, 1368, 1330, 1249, 1133, 1049, 1021, 923, 875  $\text{cm}^{-1}$ ; for  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data, see Table 1; ESIMS (positive ion mode):  $m/z$  439  $[\text{M} + \text{Na}]^+$ ; HRE-

SIMS (positive ion mode):  $m/z$  439.1363  $[\text{M} + \text{Na}]^+$  (calcd for  $\text{C}_{22}\text{H}_{24}\text{O}_8\text{Na}$ , 439.1369).

## 3.4 Anti-TMV assays

The anti-TMV activity was tested using the half-leaf method [14]. The inhibitory activity of the compounds against TMV replication was tested using two approaches. First, the half-leaf method was used to test the antiviral activity in the local lesion host *Nicotiana glutinosa in vivo*. Then, the leaf-disk method was used to evaluate the antiviral activity of the compounds in the systemic infection host *Nicotiana tabacum* cv. K326. Ningnanmycin (20  $\mu\text{M}$ ), a commercial product for plant disease in China, was used as a positive control.

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