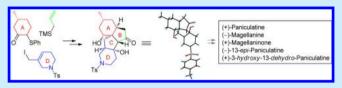


Collective Total Synthesis of Tetracyclic Diquinane Lycopodium Alkaloids (+)-Paniculatine, (-)-Magellanine, (+)-Magellaninone and **Analogues Thereof**

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Supporting Information

ABSTRACT: The collective total synthesis of tetracyclic diquinane Lycopodium alkaloids, (+)-paniculatine, (-)-magellanine, (+)-magellaninone, and two analogues (-)-13-epipaniculatine and (+)-3-hydroxyl-13-dehydro-paniculatine, has been accomplished. By logic-guided addition of a strategically useful hydroxyl group at C-3 of paniculatine, the formidable



tetracyclic core was rapidly synthesized utilizing a site-specific and stereoselective aldol cyclization, thus making the ABD → ABCD tetracyclic approach to diquinane Lycopodium alkaloids attainable for the first time.

Lycopodium alkaloids have recently received much attention due to their diverse skeletal features and broad ranging biological activities. In this context, numerous innovations in synthetic strategies and tactics have been introduced in a cornucopia of reports of total synthesis.² In contrast to the well investigated fawcettimine type,³ tetracyclic diquinane Lycopodium alkaloids still represent a formidable synthetic challenge owing to the unusually compacted polycyclic framework that bears up to six contiguous stereocenters, most of which, including an all-carbon quaternary, around a diquinane core (rings B, C) (Figure 1).

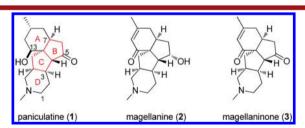


Figure 1. Tetracyclic diquinane Lycopodium alkaloids.

Tetracyclic diquinane Lycopodium alkaloids usually refer to the alkaloids of (+)-paniculatine, (-)-magellanine, and (+)-magellaninone which were isolated by Castillo et al. in the 1970s. 4 Since their isolation, extensive synthetic efforts have been made to conquer them. 5,6 In the known successful total synthesis reports,⁵ as shown in Figure 2, synthetic strategies of the challenging tetracyclic core are typified by the liner construction of ABC \rightarrow ABCD, and the basic piperidine D ring was formed either by oxidative cleavage of cyclopentene and then insertion of a N-atom (Overman, ^{5a} Liao, ^{5e} and Mukai ^{5g}) or by cyclization between a N-containing branch and another carbogenic side chain (Paquette, Sb,c Sha, Sd Ishizaki Sf). The late-stage installation

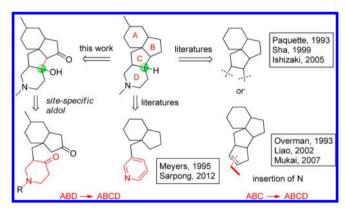


Figure 2. Strategies comparison for the ABCD tetracyclic formation.

of the D ring was often adopted because of the difficulties in the construction of the complex framework and controlling the highly basic tertiary amine group of these alkaloids. In recognition of this challenge, other groups pursued an alternative disconnection strategy. Two reports regarding the tetracyclic formation with a preinstalled D ring, namely ABD → ABCD, were published first by Meyers^{6c} in 1995 and then Sarpong^{6e} in 2012, both using pyridine as the D ring surrogate. Given the troubles encountered in the further functionalization, the core structures rather than natural products have been finished so far. Therefore, development of an unconventional ABD → ABCD approach to complete the tetracyclic diquinane Lycopodium is still an unsolved challenge. Here, we present a successful solution to this convergent ABD \rightarrow ABCD formation, which was guided by the logic of retrosynthetic analysis producing the greatest

Received: September 10, 2014 Published: October 9, 2014

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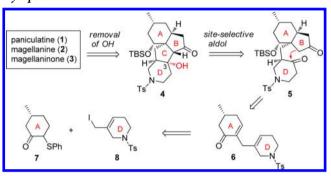
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molecular simplification, and culminating in the collective total synthesis of *Lycopodium* alkaloids (+)-paniculatine, (-)-magellanine, (+)-magellaninone, and two analogues, (-)-13-epi-paniculatine, (+)-3-hydroxyl-13-dehydro-paniculatine.

To simplify the complexity, the intelligent selection of transformations is of utmost importance. After an evolution of retrosynthetic thinking, the derivative of C-3 hydroxyl paniculatine 4 was finally singled out to be the intermediate precursor of the tetracyclic diquinane alkaloids (in Scheme 1).

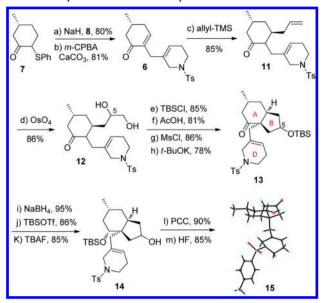
Scheme 1. Retrosynthetic Analysis of Tetracyclic Diquinane Lycopodium Alkaloids



Adding an extra hydroxyl group at C-3 is strategically useful, because it not only could be easily removed later, delivering the desired diquinane tetracycle, but also can simplify and accelerate the retrosynthetic disconnection of the targets by enlisting a site-specific and stereoselective aldol cyclization of two carbonyl groups, which is the essence of our new ABD \rightarrow ABCD strategy (Figure 2). The ABD tricyclic backbone 5 was in turn assembled by the annulation of the B ring into AD ring system 6, utilizing the enone functionality of the A ring. Ultimately, the AD ring was stitched convergently from two readily available starting materials, thioether 7^9 and allylic iodide 8.

Our synthesis commenced with the AD ring formation (Scheme 2). Known 7 was alkylated with 8 (prepared from commercial arecoline; see the Supporting Information (SI)) thermodynamically. The product was oxidized with m-CPBA, and the resulting sulfoxide was subjected to elimination, furnishing AD bicyclic cyclohexenone 6. The annulation of the B ring into the AD bicyclic system, exploiting the chemistry of the enone functionality of cyclohexenone, was not trivial. Pdcatalyzed [3 + 2] trimethylenemethane 9 cycloaddition, developed by Trost,10 was initially attempted and found extremely sluggish (Scheme 3). Using bisliane 10 under Lewis acid conditions, 11 a Sakurai reaction product (not shown) was obtained and the intermediate was advanced to close the B ring, furnishing the ABD tricyclic framework 13' with a methylene group at C-5 that required cleavage. Unfortunately, in that case, the ensuing oxidative cleavage predominately occurred in the D ring's internal double bond. That problem was circumvented by using allyltrimethylsilane. The allylic group was selectively attached to cyclohexenone 6, affording 11 which was selectively oxidized with dihydroxylation, giving a mixture of inconsequential ca. 1:1 isomeric diol 12 at C-5 (Scheme 2). To form the B ring, the primary alcohol should be converted into a leaving group for cyclization of an intramolecular alkylation. To this end, a practical procedure including bis-silylation, desilylation, mesylation, and alkylation was enlisted uneventfully to provide an inseparable ABD tricycle 13. This cyclization to form the B ring is stereospecific, leading to a cis-fused AB ring system. 12

Scheme 2. Synthesis of ABD Tricyclic Framework^a



"Reagent and conditions: (a) NaH, DMF, 8, rt, 80%; (b) *m*-CPBA, DCM, -78 °C to rt; CaCO₃, toluene, reflux, 81%; (c) TiCl₄, allyl-TMS, DCM, -78 °C to rt, 85%; (d) OsO₄, NMO, acetone/THF/*t*-BuOH/H₂O, 86%; (e) TBSCl, imidazole, DMF, rt, 85%; (f) AcOH/H₂O/THF, 70%, or 81% brsm; (g) MsCl, 4-DMAP, DCM, rt, 86%; (h) *t*-BuOK, THF, rt, 78%; (i) NaBH₄, MeOH, 0 °C, 95%; (j) TBSOTf, Et₃N, DCM, 0 °C, 86%; (k) TBAF, THF, rt, 85%; (l) PCC, Celite, rt, 90%; (m) HF-Py, THF, rt, 85%.

Scheme 3. Attempted Synthesis of ABD via [3 + 2] Cycloaddition

Therefore, 13 is still a pair of inconsequential isomers at the C-5, but both with the desired quaternary stereocenter at C-12. Considering that cyclohexanone would interfere with the upcoming pivotal aldol cyclization during ABD \rightarrow ABCD, reduction of the cyclohexanone with NaBH4 was necessary, and it provided a completely $\alpha\text{-selective}$ secondary alcohol that was protected as the TBS ether. Due to the steric difference, the sec-TBS ether at C-5 is much easier to remove. Then selective desilylation with TBAF afforded tricyclic alcohol 14. Oxidation of 14 and removal of TBS provided a crystalline compound 15, which was subjected to X-ray analysis. At this point, the complete structure of the ABD tricycle with four stereocenters was unequivocally confirmed.

To prepare the aldol cyclization precursor diketone 5, hydroboration—oxidation of alkene 14 and further oxidation of the two hydroxyl groups into diketone were employed (Scheme 4). Homogenous 5 was difficult to acquire presumably due to incomplete facial selectivity of hydroboration or facile epimerization of the α -position of the D-ring ketone, yielding a ca. 2:1 epimeric diketone mixture at C-10. With diketone in hand, the

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Scheme 4. Synthesis of ABCD Tetracyclic Diquinane Core^a

^aReagent and conditions: (a) BH₃·Me₂S, THF, −5 °C, 3 days, 83%; (b) PCC, 4 Å M.S., DCM, rt, 78%; (c) *t*-BuOK, toluene, 0 °C, 68%; (d) HF·Py, THF, rt, 82%; (e) Burgess reagent, toluene, 4 Å M.S., 80 °C, 80%; (f) H₂, Pd/C (10%), EtOAc, rt, 88%.

stage was set for the most critical intramolecular aldol process. Gratifyingly, after treatment of 5 with potassium tert-butoxide in toluene, a smooth aldol cyclization occurred and yielded the ABCD tetracycle 4 with the anticipated site- and stereoselectivities. The isolated 68% yield is good enough to advance our synthesis. 13 The newly formed C-4 stereocenter was assigned according to the apparent ROSY signal correlation with the methine at C-13. More importantly, after removal of the TBS group, a crystalline compound 16 was grown and subjected to Xray analysis, which gave a clear-cut validation of the ABCD tetracyclic system. This result has constituted, for the first time, a compound with the complete skeleton of tetracyclic diquinane through an ABD → ABCD synthetic approach. According to our collective synthesis proposal, 14 removal of the C-3 hydroxyl group was pursued. Deoxygenation through Barton-McCombie radical reaction¹⁵ was impeded because the thiocarbonyl intermediate could not be formed in such a highly congested environment. Luckily, dehydration of 4 with the Burgess reagent¹⁶ gave a sole product as tetrasubstituted alkene 17 in 80% yield. 17 could be smoothly saturated via hydrogenation, 6c yielding a cis-fused tetracyclic diquinane 18. Pleasingly, 18 is a crystalline compound whose complex stereochemistry around the diquinane core was confirmed again via X-ray single crystal analysis. The highly facial selectivity of hydrogenation probably attributed to the shielding effect to the top-side exerted by the bulking TBS group and bowl-shaped conformation of the substrate as well, leading to bottom-side approach. ¹⁷ So far, the most striking structural elements of tetracyclic diquinane Lycopodium alkaloids, except the stereochemistry of C-13 and N-methyl group, have been established.

To continue our total synthesis odyssey, as shown in Scheme 5, the tetracycle **16** was oxidized with PCC and deprotection of *p*-toluenesulfonamide followed by in situ *N*-methylation was executed, furnishing 3-hydroxy-13-dehydro-paniculatine (**20**).

Scheme 5. Collective Synthesis of Tetracyclic Diquinane Lycopodium Alkaloids^a

"Reagent and conditions: (a) PCC, 4 Å M.S., DCM, rt, 85%; (b) Na-Naph, HCHO, NaBH₃CN, overnight, 60%; (c) Na-Naph, HCHO, NaBH₃CN, overnight, 70%; (d) HF·Py, CH₃CN/THF/acetone (1:1:1), rt, 86%; (e) NaBH₄, EtOH, 0 °C, 99%; (f) MOMCl, iPr₂NEt, rt, 96%; (g) TBAF, THF, 70 °C, 88%; (h) PCC, 4 Å M.S., DCM, rt, 92%; (i) Na-Naph, HCHO, NaBH₃CN, overnight, 62%; (j) LDA, THF, -20 °C, 1 h, Cl(Ph)S=N^tBu, HMPA, -20 \rightarrow 0 °C; (k) 10% HCl, MeOH, 65% 2 steps; (n) L-Selectride, THF, -10 °C.

Similarly, after TBS protected tetracycle 18 was subjected to in situ N-methylation and desilylation with hydrogen fluoride pyridine, 13-epi-paniculatine (22) was obtained smoothly. Tetracycle 18 was reduced with NaBH4 to selectively afford sec-alcohol which was immediately protected with a MOM ether. After subjecting MOM ether to TBAF promoted desilylation of TBS and then oxidation of the new hydroxyl group at C-13, the resulting tetracyclic ketone was converted into the methylpiperidine D ring via the in situ N-methylation protocol, delivering a known compound 23, with 48% overall yield in 5 steps, which was used to furnish the (+)-paniculatine, (-)-magellanine, and (+)-magellaninone by Mukai et al. ^{5g} Although the structure of 23 was not assigned in Mukai's synthesis (our synthetic 23 was elucidated by 2D-NMR; see the SI), treatment of 23 via the Mukaiyama method¹⁸ provided the enone functionality of the A ring, and then hydrolysis of the MOM ether afforded the penultimate compound 24 en route to (-)-magellanine and (+)-magellaninone through one step of Mitsunobu inversion and Dess-Martin oxidation, respectively. In addition, according to Mukai's streamlined protocol (5 steps, one purification), obtaining 23 also means the completion of the formal synthesis of (+)-paniculatine. To our surprise, in our hands, the first step of

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L-Selectride reduction of the 13-ketone always gave a 2:1 mixture favoring the α -configured isomer, ¹⁹ which could be separated after benzoylation, and then parallel advancing two isomers eventually furnished (+)-paniculatine (1) and 13-*epi*-paniculatine (22). Our NMR data and optical rotations ²⁰ of synthetic magellanine, magellaninone, and paniculatine are all in agreement with those from literature.

In summary, we present a new strategy for the synthesis of title compounds of tetracyclic diquinane Lycopodium alkaloids. Highlights include the rapid and convergent construction of the ABD tricyclic backbone from simple starting materials thioether 7, allylic iodide 8, and allyltrimethylsilane. The formidable ABCD tetracycle including complex stereochemistry was constructed through a site-specific and stereoselective aldol cyclization, which was designed by the logic-guided addition of a strategically useful hydroxyl group at C-3 of paniculatine, making the ABD \rightarrow ABCD tetracyclic formation leading to diquinane Lycopodium alkaloids attainable for the first time. The stereochemistry of the compacted framework was scrutinized by X-ray analysis 21 or 2D-NMR. The logic-guided retrosynthetic disconnection producing unconventional strategies 22 could be further applied to other natural products synthesis.

ASSOCIATED CONTENT

Supporting Information

Experimental procedures, CIF files of compounds 15, 16, 18, and copies of the NMR spectra of all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

This work was supported by NSFC (21072200, 21472200) and The Hundred Talents Program of CAS.

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- (20) The optical rotations of our synthetic paniculatine, magellanine, and magellaninone are as follows: $[\alpha]_D^{22} = +83$ (c=0.18, in CHCl₃), $[\alpha]_D^{25} = -8.4$ (c=0.18, in CHCl₃), and $[\alpha]_D^{24} = +73$ (c=0.11, in CHCl₃). Mukai et al.^{5g} documented the following: $[\alpha]_D^{30} = +75$ (c=0.17, in CHCl₃), $[\alpha]_D^{30} = -20$ (c=0.14, in CHCl₃), and $[\alpha]_D^{24} = +82.6$ (c=0.16, in CHCl₃).
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