

Phragmalin Limonoids from the Stem Barks of *Chukrasia tabularis* var. *velutina*

Authors

Jun-Lin Yin^{1,2}, Xin Fang^{1,2}, En-De Liu³, Chun-Mao Yuan^{1,4}, Shi-Fei Li^{1,2}, Yu Zhang^{1,2}, Hong-Ping He¹, Shun-Lin Li¹, Ying-Tong Di¹, Xiao-Jiang Hao¹

Affiliations

The affiliations are listed at the end of the article

Key words

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Correspondence

Prof. Dr. Xiao-Jiang Hao
Kunming Institute of Botany
Chinese Academy of Sciences
132 Lanhei Road
650201 Kunming
P.R. China
Phone: + 86 8 71 65 22 32 63
Fax: + 86 8 71 65 22 30 70
haoxj@mail.kib.ac.cn

Correspondence

Prof. Dr. Ying-Tong Di
Kunming Institute of Botany
Chinese Academy of Sciences
132 Lanhei Road
650201 Kunming
P.R. China
Phone: + 86 8 71 65 22 32 63
Fax: + 86 8 71 65 22 30 70
diyt@mail.kib.ac.cn

Abstract

Seven new phragmalin limonoids, chukvelutilides I–O (**1–7**), were isolated from the stem barks of *Chukrasia tabularis* var. *velutina*. Their structures were elucidated by extensive spectroscopic analy-

Introduction

Phragmalin limonoids possess characteristic rings of A and B tricyclo[3.3.1^{2,10}.1^{1,4}]decane or tricyclo[4.2.1^{10,30}.1^{1,4}]decane, some of which also bear an orthoester group at positions 1, 8, 9; 8, 9, 11; 8, 9, 14; or 8, 9, 30 [1]. Bioassays on these structurally diverse molecules have shown them to have a broad range of activities [1–3]. *Chukrasia tabularis* var. *velutina*, a timber tree, grows mainly in tropical areas of Asia, and its stem bark has been traditionally used as astringent, antidiarrheal, and anti-influenza agent in China. As part of our continuing search for novel limonoids from the Meliaceae family with bioactivity [4–8], seven new phragmalin derivatives, chukvelutilides I–O (**1–7**), were isolated from the air-dried stem barks of *C. tabularis* from Yuxi, in the Yunnan Province of China. Among them, compounds **1–5** were 15-acyl phragmalins with an 1,8,9-orthoacetate group and a C-16/30 δ -lactone ring, while compounds **6** and **7** featured an 8,9,30-orthoacetate group and a C-16/17 δ -lactone ring (○ **Fig. 1**). Moreover, a brine shrimp (*Artemia sinica*) lethality test was performed to assay the biological activity [9]. Herein, the isolation, structural elucidation, and bioassay results of these compounds are presented.

sis. Among them, compound **1** showed moderate lethal activity against brine shrimp larvae, with an LC₅₀ value of 84.1 μ M.

Supporting information available online at <http://www.thieme-connect.de/products>

Results and Discussion

The MeOH extract from the dried stem barks of *C. tabularis* was suspended in water and partitioned successively with petroleum ether and EtOAc. The EtOAc-soluble extract was subjected to column chromatography followed by RP-HPLC to yield seven new limonoids. The structures of the new compounds were identified by spectroscopic analysis.

Chukvelutilide I (**1**), a white amorphous powder, had the molecular formula of C₄₃H₅₂O₂₀, as determined by HRESIMS at *m/z* 911.2964 [M + Na]⁺ (calcd. 911.2949), with 18 degrees of unsaturation. The ¹H and ¹³C NMR data of **1** (○ **Table 1** and **2**) indicated the presence of five acetyl groups, one β -furanyl, one β -dicarbonyl, one orthoacetate group, one ester carbonyl, one methoxyl, two tertiary methyls, three methylenes (one oxygenated), seven methines (five oxygenated), and seven quaternary carbons (four oxygenated). Ten degrees of unsaturation were occupied by seven carbonyl ester groups and three carbon-carbon double bonds, and the remaining eight degrees were accounted for by the octacyclic core.

Extensive analysis of the 2D NMR spectra of compound **1**, especially HMBC (○ **Fig. 2**), suggested that this compound was a 15-acyl phragmalin limonoid with a 1,8,9-orthoacetate and a C-16/30 δ -lactone ring [10]. The five acetoxyl groups were located at C-2, C-3, C-12, C-17, and C-19 by the HMBC correlations of OAc-2 (δ_{H} 2.16, s)/C-2 (δ_{C} 79.1), H-3 (δ_{H} 5.34, s)/OAc-3 (δ_{C} 168.4), H-12 (δ_{H} 4.33, br s)/OAc-12 (δ_{C} 168.8), H-17 (δ_{H} 5.70, s)/

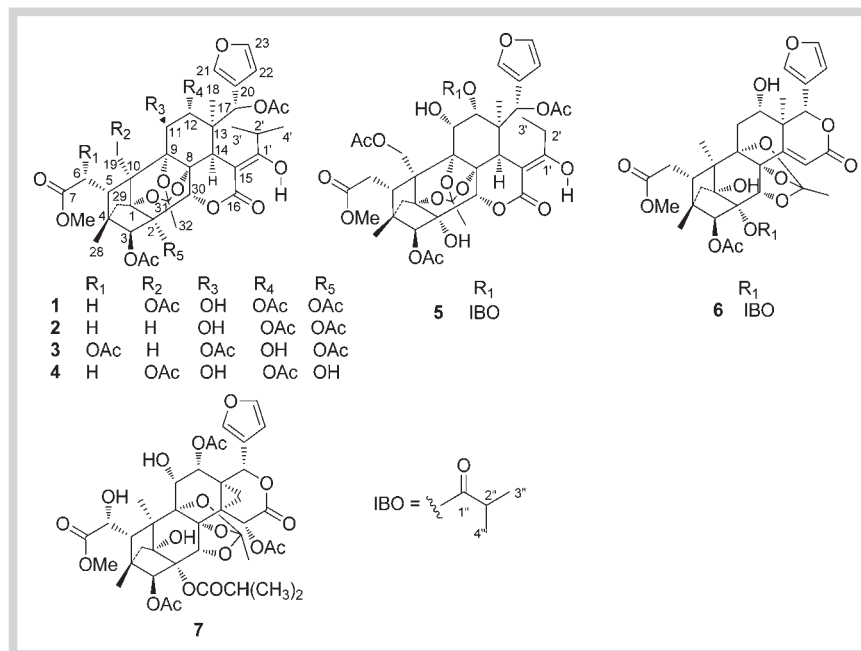


Fig. 1 Chemical structures of compounds 1–7.

OAc-17 (δ_C 168.5), and Ha-19 (δ_H 4.68, d, $J = 11.4$ Hz)/OAc-19 (δ_C 169.8). Therefore, the planar structure of compound **1** was established as shown, and the relative configuration of **1** was determined to be the same as that of chukvelutillide C by the ROESY spectrum and 3D computer modelling (Fig. 2) [10].

The structures of **2–4** were determined to be polyoxygenated phragmalins [10–13]. They all exhibited the same skeleton as **1** but with different substitution patterns, as determined by spectroscopic data analysis and comparison of their NMR data with those of **1**.

The 1H and ^{13}C NMR data of **2** were similar to those of **1**, except for the presence of an additional methyl group [CH_3 -19 (δ_H 1.16, s)] and the absence of the *O*-bearing CH_2 and CH_3CO groups in **1**, thus indicating that **2** was a 19-deacetoxy analogue of **1** (Table 1 and 2). This conclusion was further confirmed by HRESIMS and 2D NMR data.

The molecular formula of chukvelutillide K (**3**) was determined to be $C_{43}H_{52}O_{20}$ at m/z 887.2981 [$M - H$] $^-$ by HRESIMS. The 1H and ^{13}C NMR spectroscopic data suggested that **3** possessed the same skeleton as **1** but with different substitution patterns (Table 1 and 2). The 2D NMR data, especially the HMBC correlations, indicated that OAc-6, OAc-11, and OH-12 of **3** replaced H-6, OH-11, and OAc-12 of **1**. The relative configuration of compound **3** was elucidated to be the same as that of **1** by the ROESY spectrum (see Supporting Information).

Chukvelutillide L (**4**) gave the molecular formula $C_{41}H_{50}O_{19}$ from HRESIMS, indicating that it possessed one acetyl group less than **1**. The large chemical shift discrepancy at C-2 (ca. $\Delta -2.9$) and at C-3 (ca. $\Delta +3.4$) between **4** and **1** along with the aforementioned information implied that **4** was the 2-deacetyl analogue of **1** (Table 1 and 2). The remaining planar and relative configurations of **4** were identical to those of **1**, as determined by the HMBC and ROESY data.

Chukvelutillide M (**5**) had a molecular formula $C_{42}H_{51}O_{19}$, as determined by the negative HRESIMS ion at m/z 859.3029 [$M - H$] $^-$ (calcd. 859.3024). The 1H and ^{13}C NMR data of **5** resembled closely those of **4** (Table 1 and 2) except that an isobutyryloxy and an ethyl group in **5** replaced an acetoxy group at C-12 and an

isopropyl group at C-1' in **4**. The HMBC correlations from H-12 (δ_H 4.43, br s), Me-3'', and Me-2'' to the carbonyl group (δ_C 174.3), along with the 1H - 1H COSY cross-peaks of Me-3''/H-2'' and Me-4''/H-2'', placed the isobutyryloxy group at C-12. In addition, the HMBC correlations of Me-3' with C-1' (δ_C 178.6), and H₂-2' with C-1' (δ_C 178.6) and C-15 (δ_C 93.1), as well as the 1H - 1H COSY correlations of Me-3'/H₂-2' indicated that the ethyl was linked to C-1'. Thus, the planar structure of **5** was established as shown. The relative configurations at the chiral centres of **5** were the same as those of **2** on the basis of ^{13}C NMR shifts and ROESY data (see Supporting Information).

Chukvelutillide N (**6**), a white solid, had a molecular formula of $C_{35}H_{42}O_{14}$, as determined by HRESIMS at m/z 709.2484 [$M + Na$] $^+$ (calcd. 709.2472). Analysis of its 1H and ^{13}C NMR data (Table 1 and 2) indicated that compound **6** is a congener of tabularin, which possessed a phragmalin skeleton with a 8,9,30-orthoacetate [11]. In the HMBC spectrum of **6**, the observed correlations of one OH (δ_H 3.62, s) to C-1 (δ_C 83.6), the other OH (δ_H 5.11, d, $J = 4.9$ Hz) to C-12 (δ_C 64.4), and H-3 (δ_H 5.00, s) to an ester carbonyl (δ_C 168.7 s) indicated that the two hydroxyl groups were located at C-1 and C-12, respectively, and that the acetoxy group was placed at C-3. The location of the isobutyryloxy moiety at C-2 was supported by the HMBC correlations of H-3 (δ_H 5.00, s), H-30 (δ_H 5.18, s), and Me-3'' (δ_H 1.01, d, $J = 7.4$ Hz) to the ester carbonyl at C-1'' (δ_C 174.3). The structure of **6** was established as shown.

Analysis of the NMR data of chukvelutillide O (**7**) (Table 1 and 2) showed that it was a derivative of tabularin C [12]. The only difference was that the OH group at C-6 of **7** replaced the acetoxy group at the same position of the latter. This conclusion was confirmed by HRESIMS and 2D NMR data (Table 1 and 2). The structure of **7** was thus established as shown.

Compounds **1–7** were evaluated for their lethal activities against brine shrimp larvae (Table 3). Among them, compound **1** was the most lethal to brine shrimp, with an LC_{50} value of 84.1 μM .

Table 1 ^1H NMR data of **1–7** (J in Hz).^{a, b}

Position	1 ^c	2 ^d	3 ^d	4 ^a	5 ^c	6 ^c	7 ^c
3	5.34 s	5.51 s	5.43 s	4.67 s	4.67 s	5.00 s	5.02 s
5	2.90 t (10.5)	3.03 t (10.5)	3.36 br. s	2.98 t (9.2)	3.01 overlap	2.05 overlap	2.46 br. s
6a	2.87 overlap	2.67 d (17.4)	6.17 br. s	2.92 d (17.2)	2.95 br. d (17.7)	2.52 m	4.79 d (5.8)
6b	2.70 dd (17.9, 10.5)	2.42 dd (17.4, 10.5)		2.69 dd (17.2, 10.9)	2.70 dd (17.7, 10.8)		
11a	4.86 d (4.1)	4.38 d (1.6)	4.48 overlap	4.89 br. s	4.87 br. s	2.05 m	4.06 d (4.7)
11b						1.79 t (13.7)	
12	4.33 br. s	4.54 d (1.6)	4.50 overlap	4.35 br. s	4.43 br. s	3.55 overlap	5.00 overlap
14	3.55 s	3.47 s	3.47 s	3.54 s	3.56 s		
15						6.37 s	6.75 s
17	5.70 s	5.89 s	5.87 s	5.68 s	5.73 s	5.70 s	6.33 s
18a	1.46 s	1.56 s	1.56 s	0.86 s	1.47 s	1.27 s	2.84 dd (6.5, 2.1)
18b							1.35 overlap
19a	4.68 d (11.4)	1.16 s	1.25 s	4.67 d (11.5)	4.65 d (11.4)	1.20 s	
19b	4.13 (d 11.4)			4.14 d (11.5)	4.16 d (11.4)		
21	7.53 s	7.60 s	7.64 s	7.55 s	7.56 s	7.49 s	7.55 s
22	6.45 br. s	6.41 br. s	6.43 br. s	6.46 br. s	6.46 br. s	6.46 br. s	6.63 br. s
23	7.57 br. s	7.27 br. s	7.27 br. s	7.56 br. s	7.56 br. s	7.60 br. s	7.68 br. s
28	0.87 s	1.00 s	1.16 s	0.86 s	0.90 s	0.59 s	0.78 s
29a	1.90 d (12.4)	2.02 d (11.2)	2.14 d (11.2)	1.85 d (11.3)	1.88 overlap	1.97 d (11.8)	2.25 d (10.7)
29b	1.86 d (12.4)	1.86 d (11.2)	1.87 d (11.2)	1.69 d (11.3)	1.72 d (11.2)	1.51 d (11.8)	1.47 overlap
30	5.55 s	5.73 s	5.60 s	5.27 s	5.32 s	5.18 s	4.83 s
32	1.54 s	1.65 s	1.64 s	1.57 s	1.47 s	1.62 s	1.62 s
2'	2.97 m	3.01 m	3.00 m	2.96 m	2.54 overlap		
3'	1.20 d (6.8)	1.31 d (6.5)	1.28 d (6.6)	1.21 d (6.7)	1.13 m		
4'	1.06 d (6.5)	1.65 br. d	1.16 d (6.6)	1.05 d (6.7)			
1-OH						2.06 s	3.62 s
2-OH				4.73 s	4.72 s		
6-OH							6.15 d (5.7)
11-OH	5.56 d (4.4)	2.63 s		5.50 d (4.3)	5.47 d (4.1)		5.11 d (4.9)
12-OH			2.51 s			4.32 d (5.25)	
1'-OH	13.78 s	13.96 s	13.90 s	13.73 s	13.51 s		
7-OMe	3.61 s	3.74 s	3.77 s	3.63 s	3.64 s	3.66 s	
2-OAc	2.16 s	1.92 s	2.17 s				
3-OAc	1.94 s	2.15, s	2.17, s	2.19, s	2.19, s	1.92, s	1.50, s
6-OAc			2.11, s				
11-OAc			1.66, s				
12-OAc	1.55, s	2.40, s		1.56, s			2.14, s
15-OAc							2.05, s
17-OAc	1.84, s	1.15, s	1.93, s	1.85, s	1.88, s		
19-OAc	1.98, s			1.97, s	1.97, s		
2''					2.08, m	2.60, m	2.53, m
3''					0.86, d (7.1)	1.01, t (7.4)	1.07, d (7.0)
4''					0.80, d (7.1)	1.01, t (7.4)	1.05, d (7.0)

^a Signals were assigned on the basis of 1H-1H COSY, HSQC, HMBC, and ROESY experiments; ^b spectra were measured at 500 MHz; ^c data measured in DMSO- d_6 ; ^d data measured in CDCl_3

Materials and Methods

General experimental procedures

NMR spectra were measured in a CDCl_3 or DMSO- d_6 solution and recorded on a Bruker DRX-400 or DRX-500 instrument using TMS as an internal standard. The optical rotations were recorded on a Perkin-Elmer model 241 polarimeter or a Horiba SEPA-300 polarimeter. IR spectra were measured in a Bio-Rad FTS-135 spectrometer as KBr pellets. MS was performed on a VG Auto Spec-3000 or a Finnigan MAT 90 instrument. UV data were measured on a UV 210A spectrometer. Column chromatography was performed on an MCI gel CHP20P (75–150 μm , Mitsubishi Chemical Corporation), a C_{18} reversed-phase silica gel (40–75 μm , Fuji Silysia Chemical Ltd.), and a Si-gel (300–400 mesh) from Qingdao Marine Chemical Factory. Semi-preparative HPLC was performed

on an Agilent 1100 apparatus with a chromolith RP-18e (Merck, 100 \times 10 mm) column.

Material

The stem barks of *C. tabularis* var. *velutina* were collected in Yuxi, Yunnan Province, China, in August 2008 and identified by En-De Liu (Biogeography and Ecology, Kunming Institute of Botany). A voucher specimen (KIB HXJ08003) was deposited at the Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences.

Extraction and isolation

The air-dried and crashed stem barks (13 kg) of *C. tabularis* var. *velutina* were extracted with MeOH (20 L \times 3) at room temperature and concentrated to give a residue (320 g) *in vacuo* at 60 °C.

Table 2 ¹³C NMR data of 1–7 (ppm).^a

Position	1 ^b	2 ^c	3 ^c	4 ^b	5 ^b	6 ^b	7 ^b
1	84.1	84.7	84.6	84.1	84.0	83.6	84.1
2	79.1	78.7	83.2	76.2	76.2	83.1	83.5
3	79.9	80.1	80.5	83.3	83.3	85.0	85.9
4	45.8	46.3	46.5	45.3	45.3	47.4	43.7
5	35.0	35.6	40.0	36.0	35.9	39.8	43.3
6	32.1	33.0	71.4	32.2	32.1	32.6	69.4
7	173.1	172.6	170.0	173.1	173.3	173.8	176.4
8	82.6	83.2	78.9	79.4	79.4	83.4	86.0
9	83.0	83.3	83.2	83.6	83.6	85.5	84.8, s
10	47.5	46.4	47.2	46.8	46.8	47.4	48.7
11	68.9	68.9	68.7	68.9	68.8	36.8	66.3
12	71.7	72.1	72.0	71.7	71.1	64.4	68.3
13	43.6	44.8	46.5	43.8	44.3	43.7	29.3
14	41.7	42.9	42.8	41.8	42.1	153.0	24.5
15	91.5	90.9	91.0	91.9	93.1	122.5	70.5
16	170.4	170.3	170.3	170.8	170.5	163.4	165.6
17	69.9	69.9	70.1	70.0	70.3	79.8	71.7
18	17.4	17.9	18.0	17.3	17.6	13.2	16.5
19	65.6	16.4	13.8	65.8	65.9	15.0	84.8
20	122.5	122.0	121.9	122.6	122.7	121.8	122.8
21	140.2	141.4	141.4	140.3	140.1	141.3	141.0
22	109.8	109.9	109.9	109.9	110.0	110.7	110.0, d
23	143.4	142.6	142.6	143.3	143.0	142.9	144.0
28	14.0	14.5	15.9	13.7	13.8	14.3	15.3
29	38.8	40.3	40.6	38.6	38.6	39.6	40.1
30	73.4	73.5	73.5	73.9	74.0	73.5	74.9
31	118.9	119.0	119.0	119.2	119.3	119.3	115.6
32	20.2	20.1	20.6	20.6	20.6	16.1	15.3
1'	182.5	183.5	183.6	181.6	178.6		
2'	29.5	30.0	30.0	29.6	25.0		
3'	18.2	18.5	18.5	18.2	11.2		
4'	20.0	20.5	20.5	20.5			
7-OMe	51.5	51.9	53.2	51.6	51.5	52.0	
2-OAc	168.4	169.3	169.5				
	20.8	20.9	20.9				
3-OAc	168.4	170.1	169.6	169.3	169.3	168.3	168.7
	21.8	21.8	21.2	20.7	20.7	21.1	19.7
6-OAc			169.9				
			21.8				
11-OAc			169.4				
			20.0				
12-OAc	168.8	169.4		168.8			170.2
	20.4	21.0		20.3			20.5
15-OAc							167.9
							20.4
17-OAc	168.5	168.8	168.8	168.6	168.6		
	20.6	20.6	20.8	20.6	20.6		
19-OAc	169.8			169.9	169.7		
	21.0			21.0	20.9		
1''					174.3	175.8	175.2
2''					33.1	33.7	33.7
3''					18.6	18.6	18.6
4''					17.9	18.6	18.6

^a Spectra were measured at 100 MHz; ^b data measured in DMSO-*d*₆; ^c data measured in CDCl₃

The residue was dispersed in water (4.0 L) and extracted with petroleum ether and then EtOAc to give the EtOAc-soluble fraction (E-Fraction, 190 g). The E-Fraction was processed with a silica gel column (0.2 m × 1.0 m, 100 to 200 mesh, 1.0 kg) and eluted with CHCl₃-MeOH (1% MeOH, 3% MeOH, and 10% MeOH, each 5 L). The 3% MeOH eluent (120 g) was subjected to an MCI gel column (5.5 cm × 48 cm, 0.5 kg) eluted with MeOH-H₂O (60% MeOH, 80%

MeOH, 90% MeOH, each 2 L) to give three fractions, termed Fr. A–C. Fr. B (89 g) was subjected to a column of C₁₈ reversed-phase silica gel (4 cm × 48 cm) eluted with MeOH-H₂O (40% MeOH, 50% MeOH, 60% MeOH, 70% MeOH, 80% MeOH, 90% MeOH, each 2 L) to yield five fractions, termed Fr. B1–B5. Fr. B1 was purified by a silica gel column (2.5 cm × 18 cm, 300 to 400 mesh, 50 g) and eluted with CHCl₃-MeOH (75:1) to yield compounds **1** (10 mg)

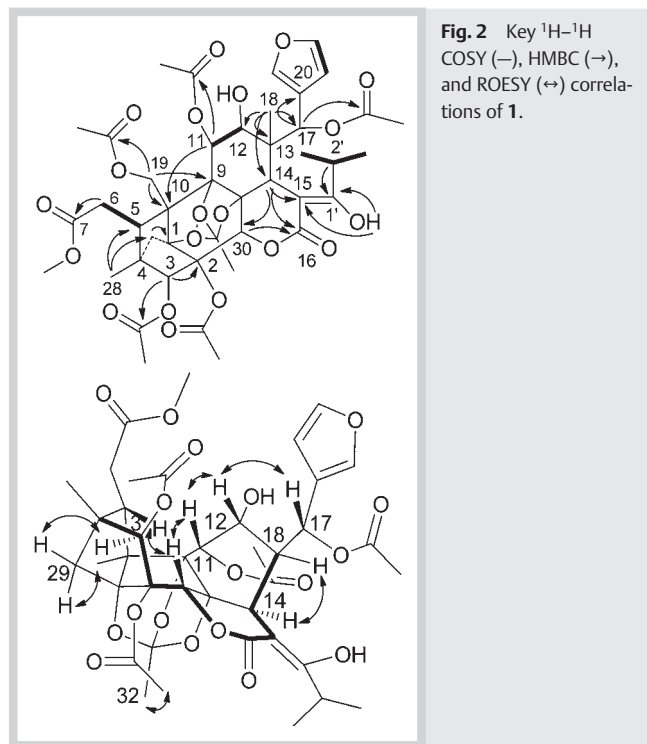


Fig. 2 Key ^1H - ^1H COSY (—), HMBC (---), and ROESY (↔) correlations of **1**.

and **3** (7 mg). Fr. B2 was separated with a silica gel column (2.5 × 18 cm, 300 to 400 mesh, 50 g), eluted with CHCl_3 -MeOH (100:1), and further purified by semi-preparative HPLC (45% MeOH) to give compounds **2** (6 mg), **4** (7 mg), and **7** (13 mg). Fr. B5 was separated with a silica column (2.5 × 18 cm, 300 to 400 mesh, 50 g), eluted with CHCl_3 -MeOH (200:1), and then purified with semipreparative HPLC (50% MeOH) to yield compounds **5** (7 mg) and **6** (7 mg). The purities of compounds **1**–**7** were 95%, as determined by TLC and HPLC-ELSD.

Isolates

Chukvelutilide I: white amorphous powder; $\text{C}_{43}\text{H}_{52}\text{O}_{20}$; $[\alpha]_{\text{D}}^{27} -92.7$ (c 0.10, MeOH); IR (KBr) ν_{max} : 3441, 2936, 1747, 1631, 1370, 1225 cm^{-1} ; UV (MeOH) λ_{max} (log ϵ) 210 (3.96), 272 (3.91) nm; positive ion FABMS m/z 889 $[\text{M} + \text{H}]^+$; positive ion HRESIMS m/z 911.2964 $[\text{M} + \text{Na}]^+$ (calcd. for $\text{C}_{43}\text{H}_{52}\text{O}_{20}\text{Na}$: 911.2949); ^1H (DMSO- d_6 , 500 MHz) and ^{13}C (DMSO- d_6 , 100 MHz) NMR data: see **Table 1** and **2**.

Chukvelutilide J: white amorphous powder; $\text{C}_{41}\text{H}_{50}\text{O}_{18}$; $[\alpha]_{\text{D}}^{21} -30.8$ (c 0.41, MeOH); IR (KBr) ν_{max} : 3450, 2936, 1744, 1639, 1402, 1373, 1238 cm^{-1} ; UV (MeOH) λ_{max} (log ϵ) 202 (3.69), 270 (3.67) nm; positive ion FABMS m/z 831 $[\text{M} + \text{H}]^+$; negative ion HRESIMS m/z 829.2918 $[\text{M} - \text{H}]^-$ (calcd. for $\text{C}_{41}\text{H}_{49}\text{O}_{18}$: 829.2918); ^1H (CD_3OD , 500 MHz) and ^{13}C (CD_3OD , 100 MHz) NMR data: **Table 1** and **2**.

Chukvelutilide K: white amorphous powder; $\text{C}_{43}\text{H}_{52}\text{O}_{20}$; $[\alpha]_{\text{D}}^{26} -38.8$ (c 0.28, MeOH); UV (MeOH) λ_{max} (log ϵ) 211 (3.77), 275 (3.81) nm; IR (KBr) ν_{max} : 3460, 2968, 1751, 1641, 1605, 1437, 1373, 1227 cm^{-1} ; FABMS m/z 887 $[\text{M} - \text{H}]^-$; HRESIMS m/z 887.2981 $[\text{M} - \text{H}]^-$ (calcd. for $\text{C}_{43}\text{H}_{51}\text{O}_{20}$: 887.2973); ^1H NMR (CD_3OD , 500 MHz) and ^{13}C (CD_3OD , 100 MHz) data: see **Table 1** and **2**.

Chukvelutilide L: white amorphous powder; $\text{C}_{41}\text{H}_{50}\text{O}_{19}$; $[\alpha]_{\text{D}}^{26} -47.6$ (c 0.14, MeOH); UV (MeOH) λ_{max} (log ϵ) 211 (3.64), 272 (3.64) nm; IR (KBr) ν_{max} : 3442, 2920, 1744, 1630, 1402, 1371,

Table 3 LC_{50} values of compounds **1**–**7** against *Artemia sinica*.

Compounds	$\text{LC}_{50} \pm \text{SD}$ (μM)
1	84.1 ± 4.0
2	203.2 ± 10.0
3	172.3 ± 9.0
4	227.9 ± 11.0
5	143.3 ± 8.0
6	229.1 ± 10.9
7	193.2 ± 10.2
Deltamethrin ^a	72.0 ± 2.9

^a Positive control

1233 cm^{-1} ; positive ion ESIMS m/z 847 $[\text{M} + \text{H}]^+$; positive ion HRESIMS m/z 869.2848 $[\text{M} + \text{Na}]^+$ (calcd. for $\text{C}_{41}\text{H}_{50}\text{O}_{19}\text{Na}$: 869.2841); ^1H (DMSO- d_6 , 500 MHz) and ^{13}C (DMSO- d_6 , 100 MHz) NMR data: see **Table 1** and **2**.

Chukvelutilide M: white amorphous powder; $\text{C}_{42}\text{H}_{52}\text{O}_{19}$; $[\alpha]_{\text{D}}^{25} -27.6$ (c 0.26, MeOH); UV (MeOH) λ_{max} (log ϵ) 213 (3.48), 273 (3.33) nm; IR (KBr) ν_{max} : 3442, 2957, 2919, 2850, 1738, 1640, 1452, 1252 cm^{-1} ; negative ion FABMS m/z 859 $[\text{M} - \text{H}]^-$; negative ion HRESIMS m/z 859.3029 $[\text{M} - \text{H}]^-$ (calcd. for $\text{C}_{42}\text{H}_{51}\text{O}_{19}$: 859.3024); ^1H (DMSO- d_6 , 500 MHz) and ^{13}C (DMSO- d_6 , 100 MHz) NMR data: see **Table 1** and **2**.

Chukvelutilide N: white amorphous powder; $\text{C}_{35}\text{H}_{42}\text{O}_{14}$; $[\alpha]_{\text{D}}^{26} +47.8$ (c 0.07, MeOH); UV (MeOH) λ_{max} (log ϵ) 212 (4.27) nm; IR (KBr) ν_{max} : 3439, 2934, 1732, 1630, 1382, 1234 cm^{-1} ; negative ion FABMS m/z 685 $[\text{M} - \text{H}]^-$; positive ion HRESIMS m/z 709.2484 $[\text{M} + \text{Na}]^+$ (calcd. for $\text{C}_{35}\text{H}_{42}\text{O}_{14}\text{Na}$: 709.2472); ^1H (DMSO- d_6 , 500 MHz) and ^{13}C (DMSO- d_6 , 100 MHz) NMR data: see **Table 1** and **2**.

Chukvelutilide O: white amorphous powder; $\text{C}_{39}\text{H}_{46}\text{O}_{19}$; $[\alpha]_{\text{D}}^{26} +10.2$ (c 0.24, MeOH); UV (MeOH) λ_{max} (log ϵ) 211 (3.73) nm; IR (KBr) ν_{max} : 3440, 2919, 1764, 1737, 1630, 1375, 1211 cm^{-1} ; negative ion FABMS: m/z 817 $[\text{M} - \text{H}]^-$; negative ion HRESIMS m/z 817.2552 $[\text{M} - \text{H}]^-$ (calcd. for $\text{C}_{39}\text{H}_{45}\text{O}_{19}$: 817.2555); ^1H (DMSO- d_6 , 500 MHz) and ^{13}C (DMSO- d_6 , 100 MHz) NMR data: see **Table 1** and **2**.

Brine shrimp toxicity assay

Compounds **1**–**7** were evaluated for their lethal activity against brine shrimp larvae based on an established 96-well plate protocol [9]. In this test, a drop of DMSO was added to the test and control vials to enhance the solubility of the compounds. Deltamethrin (Sigma, purity ≥ 98%) was used as a positive control.

Supporting information

The MS, IR, 1D, and 2D NMR spectra of compounds **1**–**7** are available as Supporting Information.

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Conflict of Interest



Prof. Xiao-Jiang Hao and Prof. Ying-Tong Di initiated the project. Ms. Jun-Lin Yin, Ms. Chun-Mao Yuan, Ms. Shi-Fei Li, Dr. Yu Zhang, Dr. Hong-Ping He, Dr. Shun-Lin Li performed the extraction, isolation, and structural identification of the compounds. Dr. Xin Fang performed the brine shrimp toxicity assay experiment. Dr. En-De Liu collected and identified the plant material. All authors approved the final version of the manuscript. We declare that all authors have read the manuscript conscientiously, and all agree with its publication. The authors have no conflict of interest to report.

Affiliations

- ¹ State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming, P.R. China
- ² University of Chinese Academy of Sciences, Beijing, P.R. China
- ³ Department of Biogeography and Ecology, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming, P.R. China
- ⁴ Key Laboratory of Structure-Based Drug Design and Discovery, Ministry of Education, Shenyang Pharmaceutical University, Shenyang, P.R. China

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