

Swietemahalactone, a rearranged phragmalin-type limonoid with anti-bacterial effect, from *Swietenia mahagoni*†

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Swietemahalactone (1), a novel rearranged phragmalin-type limonoid, was isolated from *Swietenia mahagoni*. The structure of 1 was established on the basis of extensive spectroscopic and X-ray crystallographic methods. Swietemahalactone exhibited antibacterial activity using the agar diffusion method. The key biogenetic pathway of 1 is the semipinacol rearrangement.

Swietenia mahagoni (Linn.) Jacq. (Meliaceae), a large, deciduous, and economically important timber tree native to the West Indies, is mainly cultivated in tropical areas, such as southern China, Malaysia, and India.¹ Traditionally, various parts of this plant have been used in the treatment of fever, diabetes, malaria, hypertension and tuberculosis, and as an abortifacient, antiseptic, astringent, depurative, purgative, and tonic.² Modern pharmacological studies have implied that genus *Swietenia* had anti-diabetes, anti-bacterial, and cytotoxic activities.³

The diversified structures and significant biological activities of limonoids from the Meliaceae plant have prompted continuous studies on this kind of metabolite.⁴ Previous chemical investigations of this genus led to the isolation of more than 60 limonoids.⁵ Some of the limonoids from this genus showed anti-inflammatory, antifungal, antioxidant, and antifeeding effects.⁶

To further discover structurally diverse and biologically significant compounds from *S. mahagoni*, we examined its leaves and branches, which led to the isolation of swietemahalactone (**1**), a novel rearranged phragmalin-type limonoid. The structure of **1** was confirmed by the 1D and 2D NMR of triacetylate (**1a**) and the X-ray crystallographic analysis of the *p*-bromobenzoate derivative (**1b**) (Fig. 1). Herein we describe the structure elucidation and propose a biosynthetic route of **1**. Swietemahalactone selectively

exhibited significant antibacterial activity against Gram-negative bacteria *Escherichia coli* (minimal inhibitory concentration (MIC) 0.01 μ M).

The air-dried powder of leaves and branches of *S. mahagoni* (dry weight 5 kg) was extracted with MeOH by refluxing and concentrated *in vacuo* to give a crude extract (500 g), which was then partitioned in succession with H₂O, *n*-hexane, EtOAc, and then *n*-BuOH. The EtOAc fraction (100 g) was repeatedly chromatographed on silica gel, RP-18 and Sephadex LH-20 (MeOH) to yield swietemahalactone (**1**, 250 mg) and 1-*O*-deacetyl-2 α -hydroxykhayanolide E (**2**, 100 mg) (for details see the ESI†).

Compound **1** was assigned a molecular formula of C₂₇H₃₀O₁₀, which was confirmed by high resolution electrospray ionization mass spectroscopy (HRESIMS, m/z [M + H]⁺ 515.1910, calcd. for C₂₇H₃₁O₁₀, 515.1917) and NMR data (Table 1). Its IR spectrum showed the presence of a hydroxyl group (3444 cm⁻¹), carboxylate (1728 cm⁻¹) and α,β -unsaturated lactone (1643 cm⁻¹).

Its ¹H NMR spectrum (Table 1) showed three tertiary methyl signals at δ_{H} 1.90 (s), 1.64 (s) and 1.25 (s), one methoxyl signal at δ_{H} 2.86 (s), and four sp² methine signals at δ_{H} 6.17, 6.54, 7.68 and 7.75. The ¹³C NMR-DEPT spectra (Table 1) displayed 27 carbon signals, including five methyl groups (one methoxyl group at δ_{C} 52.4), three methylene groups, nine methine groups (two oxygenated methines at δ_{C} 83.0 and 71.8, and four sp² methines

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† Electronic supplementary information (ESI) available: Experimental procedures; 1D and 2D NMR spectra of swietemahalactone (**1**) and 1,3,6-*O*-triacetyl-swietemahalactone (**1a**); X-ray data of 1,3,6-tris(4-bromobenzoyl) swietemahalactone (**1b**). CCDC 900980. See DOI: 10.1039/c3ra23401k

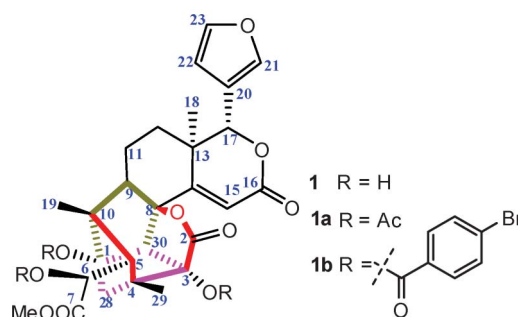


Fig. 1 The structures of **1**, **1a**, and **1b**.

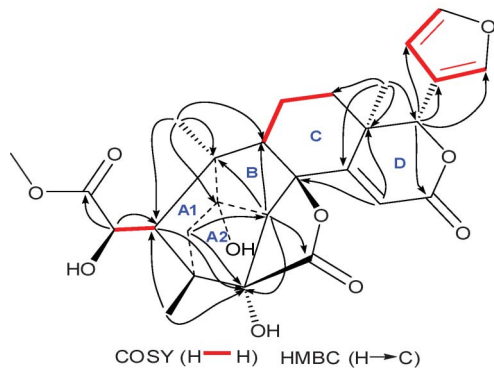
Table 1 ^1H and ^{13}C NMR-DEPT data for **1** (600 Hz and 150 Hz, respectively, in pyridine- d_5)

No.	δ_{C} multi	δ_{H} (multi; J in Hz)	No.	δ_{C} multi	δ_{H} (multi; J in Hz)
1	91.9 s		15	115.4 d	6.17 (s)
2	177.2 s		16	165.1 s	
3	87.7 s		17	83.0 d	5.22 (s)
4	51.3 s		18	20.9 q	1.25 (s)
5	46.3 d	2.84 (d; 1.7)	19	15.9 q	1.64 (s)
6	71.8 d	4.93 (d; 3.3)	20	121.3 s	
7	176.6 s		21	142.5 d	7.75 (s)
8	91.0 s		22	111.1 d	6.54 (d; 1.2)
9	48.6 d	2.76 (dd; 12.1, 8.2)	23	144.1 d	7.68 (t; 1.6)
10	54.0 s		28	45.3 t	3.30 (d; 9.4); 2.86 (dd; 9.3, 1.3)
11	16.7 t	2.28 (dt; 12.7, 12.4, 9.8); 1.81–1.85 (m)	29	12.7 q	1.90 (s)
12	27.2 t	1.94–1.97 (m); 1.26–1.28 (m)	30	62.7 d	3.69 (s)
13	40.1 s		OMe	52.4 q	3.86 (s)
14	164.8 s				

at δ_{C} 144.1, 142.5, 115.4 and 111.1), and nine quaternary carbons (three oxygenated quaternary carbons at δ_{C} 91.9, 91.0 and 87.7, one sp^2 quaternary carbon at δ_{C} 164.8 and three carboxylate groups at δ_{C} 177.2 (s), 176.6 (s) and 165.1 (s)).

The above data show that the structure of **1** is similar to khayanolide C.⁷ The A1, B, C, D and furan-ring fragments were deduced as those of khayanolide C by comparing the ^{13}C NMR data of **1** with khayanolide C, together with the analysis of the HMQC, HMBC, and ^1H - ^1H COSY data of **1** (Fig. 2).

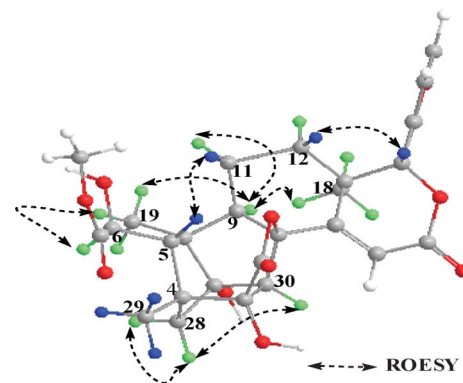
However, the ^{13}C NMR spectrum of **1** showed an ester carbonyl signal at δ_{C} 177.2 and a quaternary-carbon signal at δ_{C} 87.6 instead of one carboxyl signal (C-2) and one oxygenated methine signal (C-3) as in khayanolide C. In the HMBC spectrum, the correlations from δ_{C} 87.7 to H₃-29, H₂-28, H-5, and H-30 determined that the signal of δ_{C} 87.7 belonged to C-3. Furthermore, the correlations between H-30 and δ_{C} 177.2, C-3, C-8, C-9, and C-10 in the HMBC spectrum indicated that the signal of δ_{C} 177.2 pertained to C-2. Thus, the lactone in **1** may be located at C-2/C-8 or C-2/C-1. But it was difficult to determine whether it was 2,8-lactone or 2,1-lactone in **1** only by the 1D and 2D NMR spectra. Consequently, a triacetyl product, 1,3,6-triacetyl swietemahalactone (**1a**), was obtained by acetylation of **1** (for details see the ESI†).

**Fig. 2** Key COSY and key HMBC correlations of **1**.

The signals of H₃-18, H₃-19, and H₃-29 at δ_{H} 1.22, 1.07 and 1.13, respectively, were assigned by the HMQC and HMBC spectra of **1a** (for its 1D and 2D NMR data see the ESI†). Following that, H₃-18, H₃-19 and H₃-29 were found to be correlated with δ_{C} 82.3, 93.9 and 88.5 in the HMBC spectrum, respectively, which indicated that the above carbon signals belonged to C-17, C-1, and C-3, respectively. Finally, in the HMBC spectrum, two of the acetyl methyl signals were correlated with C-6 and C-3, and the other one was correlated with C-1. Therefore, three acetyl groups were assumed to be attached to OH-1, OH-3 and OH-6, respectively. Thus, compound **1** was a 2,8-lactone and the planar structure of **1** was deduced.

The relative configuration of **1** was determined on the basis of a ROESY experiment of **1a** (Fig. 3). H-9 α showed remarkable correlations with H-11, H₃-18 α , and H₃-19 α , and H-5 β exhibited correlation with H-11 β . However, it was hard to determine the absolute configuration of C-6 and the relative configurations of O-8 and OH-3 in **1** by a ROESY experiment.

None but the single-crystal X-ray would be helpful to confirm the stereochemistry of **1**. However, compounds **1** and **1a** were not easy to crystallize in most kinds of organic solvents. Fortunately, the *p*-bromoxybenzoylation of **1** afforded 1,3,6-tris(4-bromobenzoyl) swietemahalactone (**1b**) which crystallized in the CH₃OH-

**Fig. 3** Key ROESY correlations of **1a**.

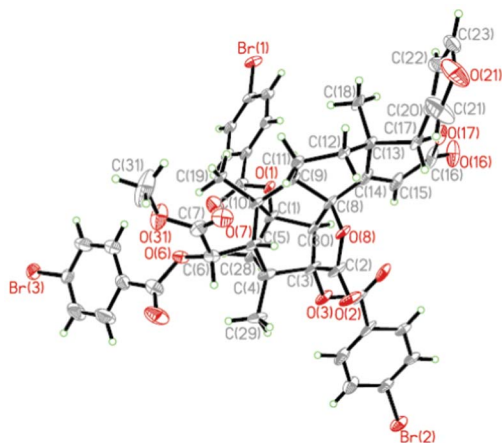
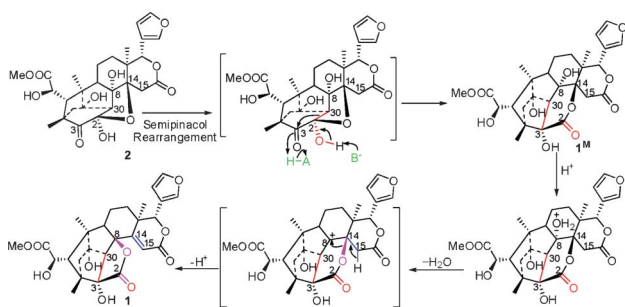


Fig. 4 Single-crystal X-ray structure of **1b**.

CHCl_3 (1 : 1) solvent.⁸ Consequently, a single X-ray crystallographic diffraction of **1b** was conducted as shown in Fig. 4. Thus, the absolute configuration of C-6 was deduced as *S* and the relative configurations of OH-3 and O-8 were determined as α and β , respectively, by the standard anomalous scattering method. Finally, the structure of compound **1** was deduced and named as swietemahalactone.

Swietemahalactone (**1**) is a novel rearranged phragmalin-type limonoid. A plausible biosynthetic origin of **1** from 1-*O*-deacetyl-2 α -hydroxykhayanolide **2** was proposed (Scheme 1). **2** was found to be widely distributed in the leaves and branches of *S. mahagoni* and therefore considered to be a plausible biosynthetic precursor of **1**.

Semipinacol rearrangement, widely used in natural product synthesis, is a special type of pinacol rearrangement in which an electrophilic carbon center is vicinal to an oxygen-containing carbon and can drive the 1,2-migration of a C–C or C–H bond to terminate the process, generating a carbonyl group.⁹ This type of reaction was proposed to occur during the biosyntheses of various secondary metabolites, including aurachin B, aflatoxin B1, (+)-liphagal, (+)-asteltoxin, brevianamides, paraherquamide B, versicolamide B, and notoamides.¹⁰ Considering that C-3 in **2** is an electrophilic carbon center and C-2 in **2** is an oxygen-containing carbon, compound **2** could be transformed into **1^M** by the



Scheme 1 Proposed biosynthetic pathway of **1**.

Table 2 Antimicrobial activities of compound **1**

	Zones of inhibition [mm]/MIC [μM]				
	<i>E. coli</i>	<i>S. aureus</i>	<i>P. aeruginosa</i>	<i>S. epidermidis</i>	<i>B. subtilis</i>
1	20/0.010	15/0.160	16/0.130	12/0.380	12/0.380
2^a	30/0.0004	25/0.0031	24/0.0111	26/0.0045	22/0.0120

^a Norfloxacin, as a positive control.

semipinacol rearrangement. Then, the **1^M** intermediate undergoes an acid catalyzed dehydration to produce **1**.

The insecticidal¹¹ and cytotoxic¹² activities of **1** were tested *in vitro*. However, **1** did not display cytotoxicity against HL-60, SMMC-7721, A-549, MCF-7, and SW480 cell lines ($\text{IC}_{50} > 40 \mu\text{M}$), and did not exhibit insecticidal activity using *Artemia salina* L ($\text{LD}_{50} > 100 \mu\text{g ml}^{-1}$).

Considering that some limonoids display anti-bacterial activity,¹³ compound **1** was further evaluated for antimicrobial activity against Gram-positive and negative bacteria *via* a microdilution assay (for details see the ESI†). From the antimicrobial test results (Table 2), **1** selectively exhibited significant antibacterial activity against Gram-negative bacteria *Escherichia coli* (ATCC 25922).

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Notes and references

- S. K. Chen, B. Y. Chen and H. Li, in *Flora of China (Zhongguo Zhiwu Zhi)*, Science Press, Beijing, 1997, vol. 43, pp. 44–46.
- (a) *Dr Duke's Phytochemical and Ethnobotanical Databases*, 2008, URL: <http://www.ars-grin.gov/cgi-bin/duke/ethnobot.pl>; (b) X. Y. Chen, X. N. Wang, C. Q. Fan, S. Yin and J. M. Yue, *Tetrahedron Lett.*, 2007, **48**, 7480–7484.
- (a) M. Haque, M. O. Ullah and K. Nahar, *Pak. J. Biol. Sci.*, 2009, **12**, 599–602; (b) G. Sahgal, S. Ramanathan, S. Sasidharan, M. N. Mordi, S. Ismail and S. M. Mansor, *Molecules*, 2009, **14**, 4476–4485; (c) A. Maiti, S. Dewanjee, M. Kundu and S. C. Mandal, *Nat. Prod. Sci.*, 2007, **13**, 295–299.
- Q. G. Tan and X. D. Luo, *Chem. Rev.*, 2011, **111**, 7437–7522.
- (a) J. Q. Liu, C. F. Wang, J. C. Chen and M. H. Qiu, *Nat. Prod. Res.*, 2012, **26**, 1887–1891; (b) B. S. Mootoo, A. Ali, R. Motilal, R. Pingal, A. Ramlal, A. Khan, W. F. Reynolds and S. McLean, *J. Nat. Prod.*, 1999, **62**, 1514–1517; (c) R. Segura-Correa, R. Mata, A. L. Anaya, B. Hernandez-Bautista, R. Villena, M. Soriano-Garcia, R. Bye and E. Linares, *J. Nat. Prod.*, 1993, **56**, 1567–1574; (d) Y. Y. Chen, X. N. Wang, C. Q. Fan, S. Yin and J. M. Yue, *Tetrahedron Lett.*, 2007, **48**, 7480–7484.
- (a) J. J. Chen, S. S. Huang, C. H. Liao, D. C. Wei, P. J. Sung, T. C. Wang and M. J. Cheng, *Food Chem.*, 2010, **120**, 379–384; (b) T. R. Govindachari, G. Suresh, B. Banumathy, S. Masilamani, G. Gopalakrishnan and G. N. K. Kumari, *J. Chem. Ecol.*, 1999, **25**, 923–933; (c) S. Falah, T. Suzuki and T. Katayama, *Pak. J. Biol. Sci.*, 2008, **11**, 2007–2012; (d) R. Fowles, B. Mootoo,

- R. Ramsewak, A. Khan, A. Ramsubhag, W. Reynolds and M. Nair, *Pest Manage. Sci.*, 2010, **66**, 1298–1303.
- 7 S. A. M. Abdelgaleil, H. Okamura, T. Iwagawa, A. Sato, I. Miyahara, M. Doe and M. Nakatani, *Tetrahedron*, 2001, **57**, 119–126.
- 8 (a) M. Isaka, C. Suyarnsestakorn and M. Tanticharoen, *J. Org. Chem.*, 2002, **67**, 1561–1566; (b) S. V. Mudur, D. C. Swenson, J. B. Gloer, J. Campbell and C. A. Shearer, *Org. Lett.*, 2006, **15**, 3191–3194; (c) J. Qi, A. B. Beeler, Q. Zhang and J. A. Porco Jr., *J. Am. Chem. Soc.*, 2010, **132**, 13642–13644.
- 9 Z. L. Song, C. A. Fan and Y. Q. Tu, *Chem. Rev.*, 2011, **111**, 7523–7556.
- 10 (a) Y. Katsuyama, K. Harmrolfs, D. Pistorius, Y. Li and R. Müller, *Angew. Chem., Int. Ed.*, 2012, **51**, 9437–9440; (b) H. Kato, Y. Nakamura, J. M. Finefield and H. Umaoka, *Tetrahedron Lett.*, 2011, **52**, 6923–6926; (c) S. Li, J. M. Finefield, J. D. Sunderhaus, T. J. McAfoos, R. M. Williams and D. H. Sherman, *J. Am. Chem. Soc.*, 2012, **134**, 788–791.
- 11 (a) F. Hiramatsu, T. Murayama, T. Koseki and Y. Shiono, *Phytochemistry*, 2007, **68**, 1267–1271; (b) M. M. Cao, Y. Zhang, H. P. He, S. F. Li, S. D. Huang, D. Z. Chen, G. H. Tang, S. L. Li, Y. T. Di and X. J. Hao, *J. Nat. Prod.*, 2012, **75**, 1076–1082.
- 12 C. F. Wang, J. Q. Liu, Y. X. Yan, J. C. Chen, Y. Lu, Y. H. Guo and M. H. Qiu, *Org. Lett.*, 2010, **12**, 1656–1659.
- 13 (a) A. K. M. S. Rahman, A. K. A. Chowdhury, H.-A. Ali, S. Z. Raihan, M. S. Ali, L. Nahar and S. D. Sarker, *J. Nat. Med.*, 2009, **63**, 41–45; (b) B. D. Lin, T. Yuan, C. R. Zhang, L. Dong, B. Zhang, Y. Wu and J. M. Yue, *J. Nat. Prod.*, 2009, **72**, 2084–2090; (c) K. Pudhom, D. Sommit, P. Nuclear, N. Ngamrojanavanich and A. Petsom, *J. Nat. Prod.*, 2010, **73**, 263–266.