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香港坚木中的达玛烷型三萜* ダッチ6 メデ

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摘要 从楝科植物香港坚木 (Dysoxylum hongkongense (Tufch.) Merr) 的树叶中分得了 4 个新的达 玛烷型三萜,基于详细的光谱分析及与同类化合物的光谱数据比较鉴定了其结构,分别命名为: 20R, 24R - 表 - 25 - 达玛烷烯 - 3 - 酮 (3), 16β - 羟基达玛烷 - 20 (22),25 - 二烯 - 3 - 酮 (4),26 - 羟基达玛烷 - 20, 24 - 二烯 - 3 - 酮 (5) 和 3β - 乙酰基 - 7α , 21S, 25 - 三羟基 - 21S, 23R - 表 - 9 (11) - 达玛烷烯 (8)。

关键词 20R, 24R - 表 - 25 - 达玛烷烯 - 3 - 酮, 16β - 羟基达玛烷 - 20 (22), 25 - 二烯 - 3 - 酮, 26 - 羟基达玛烷 - 20, 24 - 二烯 - 3 - 酮, 3β - 乙酰基 - 7α, 21S, 25 - 三羟基 - 21S, 23R - 表 - 9 (11) - 达玛烷烯, 香港坚木, 棟科 分类号 Q 946

Dammarane Triterpenoids from Dysoxylum hongkongense

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Abstract Four new dammarane type triterpenoids, named (20R, 24R) – epoxy – 25 – dammaren – 3 – one; 16β – hydroxy – dammara – 20(22), 25 – dien – 3 – one(4); 26 – hydroxy – dammara – 20, 24 – dien – 3 – one(5); 7α , 21S, 25 – trihydroxy – 3β – acetoxy – 21S, 23R – epoxy – 9(11) – en – dammarane(8) together with four known compounds, ergosterol peroxide, dammara – 20, 24 – dien – 3β – ol, 20 – R – form hydroxy-dammaranone and cycloart – 23 – ene – 3β , 25 – diol were isolated from the leaves of *Dysoxylum hongkongense*. The proposed structures were established by spectral analysis and comparisons with closely related compounds. **Key words** – *Dysoxylum hongkongense*, Meliaceae, dammarane triterpenoids, (20R, 24R) – epoxy – 25 – dammaren – 3 – one, 16β – hydroxy – dammara – 20(22), 25 – dien – 3 – one, 26 – hydroxy – dammara – 20(24) – dien – 3 – one, 26 – hydroxy – dammara – 20(24) – dien – 20 – one, 200, 201, 202, 203, 203, 203, 203, 204 – dien – 205, 205 – trihydroxy – 206, 207, 208, 208, 209, 20

INTRODUCTION

Dysoxylum is a large genus with about 140 species distributed in South East Asia and Australia (Govindachari et al., 1994). In Yunnan Province of China about 10 species occur of which D. hongkongense (Tufch.) Merr. is a high value

^{*} This work was suppurted by grants from Natural Science Foundation of China and of Yunnan.

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Received: 1998 - 03 - 17, Accepted: 1998 - 04 - 28

leafy timber tree, endemic to Xishuangbanna, which grows in lowland forests (Yunnan Institute of Botany, 1984). The leaves and bark are used as a medicine by the indigenous people for treating malaria (Yunnan Institute of Tropical Botany, 1983). The chemistry of *D. hongkongense* has not been previously studied. We describe the isolation and structure elucidation of four novel dammarane triterpenoids and four known compounds from the leaves of *D. hongkongense*.

RESULTS AND DISCUSSION

The crude petrol – soluble fraction was fractionated and purified to yield eight compounds. Four of these were ultimately identified as the known componds: 5α , 8α – epidioxyergosta – 6, 22 – dien – 3β – ol, dammara – 20, 24 – dien – 3β – ol, 20 – R – form – hydroxydammaranone and cycloart – 23 – ene – 3β , 25 – diol. Six of these compounds had very similar 1 H – NMR and 13 C – NMR (Table 1) spectra, showing the presence of a series of structurally similar compounds. Furthermore, these compounds all give a positive Liebermann – Burchard test indicating the presence of triterpenoids.

Compound 3 was obtained as colourless needles. The molecular formula (C₁₀H₄₈O₂) can be deduced from its EIMS and 13 C - DEPT spectra. The 13 C - DEPT spectra of 3 exhibited the presence of 30 carbon atoms with two double bond carbons (\$ 147.61, s; 110.42, t), a ketone carbonyl carbon (\$ 217.78,s) and two oxygen bearing carbons (74.57, s; 76.08, d). From the molecular formula, 3 was considered to be a pentacyclic triterpenoid, with a cyclic ether moiety. The ¹H - NMR spectrum revealed the presence of a methylene group [δ 4.74 and 4.88 (each 1H, each s)], seven tertiary methyl groups [δ 0.80, 0.85, 0.91, 0.96, 1.00, 1.07, 1.66 (vinyl methy)] and a ketonic group (δ 2.40, 2H, m). It also showed signals due to an oxygen bearing methine [δ 3.95 (1H, dd, $J_1 = 7.08$, $J_2 = 4.76$)]. The remaining 22 alphatic hydrogen atoms resonated in complex, overlapping multiplet between δ 1.07 and 1.83 ppm. The IR spectrum of 3 showed the presence of a ketone carbonyl [1690 cm⁻¹(vs, C = 0)] and olefinic groups [1640 cm⁻¹(w, C = C)]. The electron impact mass spectral fragmentation pattern of 3 was characteristic of a dammarane skeleon. Namely, 3 gave a significant fragment ion at m/z 205 (C14H21O) due to cleavage of the C - ring of the dammarane skeleton and showed a fragment at m/z 125 (C₈H₁₃O) ascribed to the side chain moiety as a result of C - 17/20 bond fission. It also showed peaks at m/z 399 (m - C₃H₅, loss of an isopropenyl group) and m/z 315 (m - C₈H₁₃O, loss the side chain). The stereochemistry of Me - 20 and H - 24 was arranged by comparison of the ¹³C - NMR spectrum of 3 with those of richenone and richenol, previously reported by Aalberskerg and Singh (1991). The configuration of C - 17 can be determined as S by comparison of the 13 C - NMR data with those of salvilymitone and salvilymitol. The structures of salvilymitone and salvilymitol gaven by Pedreros and Rodriguez et al (1990) have been revised on the basis of X - ray crystallographic analysis. Thus, compound 3 is (20R, 24R) - epoxy - 25 - dammaren - 3 - one.

Compound 4 was obtained as colourless needles. The molecular formula ($C_{90}H_{48}O_2$) was determined by analysis of its EIMS and ^{13}C – DEPT spectra. The EIMS spectrum of 4 showed peaks at m/z 440 (m^+), 422($M-H_2O$), 407($M-H_2O-CH_3$), 366($M-H_2O-C_4H_8$). It also gave a significant fragment ion at m/z 205 ($C_{14}H_{21}O$) due to cleavage of the C-ring of the dammarane skeleton and showed a fragment ion at m/z 109 (C_8H_{13}) ascribed to the side chain moiety as a result of C-17/20 bond fission. The $^{13}C-DEPT$ spectra of 4 showed the presence of 30 carbon atoms with a ketone carbonyl carbon (δ 217.72, s), an isopropenyl group (δ 149.32, s, δ 110.97, t), and a trisubstituted double bond (δ 134.94, s, 127.77, d). Furthermore, down field signal at 66.89, d was observed. From the molecular formula, 4 could be deduced to be a tetracyclic dammarane triterpeniod. The 1H NMR spectrum showed five shielded methyls (δ 0.85, 0.91, 0.98, 1.01, 1.05) and two vinyl methyls (δ 1.63, 1.70). In addition, a multiplet at 2.46 (2H), a multiplet at δ 4.45 (1H, ddd, J=13.52, 8.44, 5.04), a doublet at δ 5.14 (1H, br, d, J=8.44) and two singlets at δ 4.81,

4.86, each integrating for one proton, were observed. The multiplet centred to δ 4.45 indicates the presence of a proton on a carbon bearing an oxygen function and the 1H – singlets at δ 4.81 and 4.86 suggest the presence of a exomethylene group. The IR spectrum of 4 showed the presence of hydroxyl, ketonic and vinyl [3480 cm⁻¹, (vs, 0 – H); 1035 cm⁻¹ (s, C – O), 1695 cm⁻¹(s, C = O), 1950 cm⁻¹(m, C = C), 890, 840 cm⁻¹(m)] groups. The location and stereochemistry of the hydroxyl group could be determined as $16 - \beta$ by comparison the ¹³C NMR data of 4 with those reported for elabunin, whose structure was determined by 2D – NMR experiments (Kubo *et al.*, 1990). The C – 16 bearing an α – OH resonated at δ 77.64 ppm and the H – 16 resonated at δ 4.05 ppm in elabunine, while in compound 4 the C – 16 bearing an β – OH resonated at δ 66.89 ppm and the H – 16 resonated at δ 4.45 ppm (Kubo *et al.*, 1990). The presence of isopropyl group and the fragment peaks at 366 (M – H₂O – C₄C₈) and 55 (C₄H₇) indicated a 20(22), 25 – diene system in the side chaine of 4. From all these evidence, compound 4 was identified as 16β – hydroxy – dammara – 20(22), 25 – diene – 3 – one.

The EIMS and 13 C – DEPT spectra showed compound 5 to possess the molecular formula ($C_{30}H_{48}O_2$) and showed the characteristics of a dammarane skeleton. A resonance at δ 217.81 (s) could be assigned to a ketone carbonly carbon and resonaces at δ 134.54, s 107.88, t, 128.23, d, 152.24, s could be assigned to olefinic carbons. A de – fielded resonance at δ 61.68 (t) could be assigned to a oxygen bearing carbon. An OH stretching band at 3380 cm⁻¹ in the IR spectrum of 5 indicated that there was a hydroxyl group. The ¹H NMR spectrum revealed six C – methyl groups (δ 0.86, 0.93, 0.99, 1.02, 1.06, 1.78), One of them (δ 1.78) was a vinyl methyl. Other resolved features in the ¹H – NMR spectrum included a triplet at δ 2.17 (2H, t, J = 7.40), a multiplet at δ 2.43 (4H, m), a singlet at δ 4.11 (2H, s), at δ 4.68 and 4.74 (each 1H), and a triplet at δ 5.29 (1H, t, J = 7.42). The EIMS spectrum of 5 showed peaks at m/z 440 (M⁺), 422 (M – H₂O), 409 (M – H₂O – CH₃), 357 (M – C₅H₇O), 343 (M – C₆H₉O), 315 (M – side chain, M – C₈H₁₃O), 205 (C₁₄H₂₁O, cleavage of C – ring). Analysis of these fragment ion peaks allowed the location of the double bonds and the hydroxyl group to be arranged as C₂₀ – C₂₁, C₂₄ – C₂₅ and C₂₆, respectively. By comparison of the ¹³C – NMR spectrum of 5 with that of cucurbite – 5, 24 – diene – 3 – β , 11 α , 26 – triol, previously reported (Kasai *et al.*, 1987), the double bond of C₂₄ – C₂₅ was determined as the E form. Thus, compound 5 is 26 – hydroxy – dammara – 20, 24 – dein – 3 – one.

Compound 8 was crystallized from CHCl₃ as colourless needles that gave a parent ion in the HREIMS at m/z 532.3735 Daltons, appropriate for a molecular formula of C₁₂H₅₂O₆(calcd. 532.376490). The ¹³C - DEPT spectra of 8 exhibited an ester carbonly carbon (170.65, s), two olefinic carbons (145.86, s, 118.15, d), and five down - field oxygen bearing carbons (97.29, d, 78.68, d, 78.42, d, 75.47, d, 73.25, s). From the molecular formula, compound 8 could be deduced to be a pentacyclic triterpenoids. The EIMS spectrum of 8 showed peaks at m/z 532 (M⁺), 514 (M - H₂O), $499 (M - H_2O - CH_3)$, $456 (M - H_2O - CH_3 - COCH_3)$, $439 (M - H_2O - CH_3 - CH_3COOH)$, 356 (M - side chain, M - Side Chain) $-C_8H_{15}O_3$), 341 (M - OH - CH₃ - side chain), 297 (M - OH-CH₃COOH - side chain), 293, 281 (M-OH - CH₃-CH₃COOH - side chain), 239, 159 (side chain, C₈H₁₅O₃), 59 (C₃H₇O). The fragment ion peaks at m/z 293 and 239 (as a result of a cleavage of the C - ring) showed the double bond location at C9 - C11. By analysis the EIMS fragment pattern and the ¹³C - DEPT data, 8 must be a dammarane triterpenoid with a cycle in the side chain, an acetate at the C₁, and with a hydroxy at C₇. The chemical shift value of C - 6 (\$\delta\$ 17.5) and C - 8 (\$\delta\$ 43.6) also indicated a hydroxyl group might be presented at the 7α position. The 1H - NMR spectrum of 8 showed signals at δ 5.30 (1H, br), 5.27 (1H, d, J=3.56), 4.65 (1H, br), 4.24 (1H, m), 3.20 (1H, br) as well as eight methyls (δ 0.73, 0.81, 0.86, 0.90, 0.94, 1.23, 2.02). The structure and the stereochemistry of side – chain of 8 was determined by comparison of the ¹³C – NMR data of 8 with those of 21 - 0 - acetyl - toosendantriol and 21 - 0 - methyl - toosendonpentol previously reported. The structures of 21 - 0 - acetyl - toosendantriol and 21 - 0 - methyl - toosendonpentol have been revised on the basis of

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X - ray crystallographic analysis (Inada et al., 1989, Nakanishi et al., 1986). Thus, the structure of compound 8 could be elucidated as 7α , 21S, 25 - trihydroxy - 3β - acetoxy - 21S, 23R - epoxy - 9(11) - en - dammarane.

EXPERIMENTAL

General. all mps; uncorr. TLC; silica gel precoated glass plates with petroleum ether; EtOAc(4:1, 1:1). Spots were detected by spraying with 15% H_2SO_4 (in EtOH), followed by heating at 110°C. CC: silica gel (220 ~ 300 mesh), $^1\,H$ - NMR and $^{13}\,C$ - NMR spectra were recorded at 400 MHz in CDCl $_3$ using. TMS as int. standard.

Plant materials Dysoxylum hongkongense leaves were collected in June 1995 from Xishuangbanna and identified by Prof. Wang Hong (Kumming Institute of Botany, The Academy of Sciences of China, Kumming, Yunnan, China).

Extraction and isolation of the triterpenoids Dried and powdered D. hongkongense leaves (6.0 kg) was extracted with MeOH for 15 days at room temperature. Conc. of the MeOH extract under red. pres. below $40\,\mathrm{C}$ gave a syrup which was suspended in water and the suspension was extracted successively with petrol., CH2Cl2 and n - BuOH. The pertrol. soluble material (170g) was chromatographed on silica gel using a petrol. (60 - 90°) - EtOAc gradient to give 8 Frs. Fr.2 was rechromatographed on silica gel, eluted with petrol – EtOAc (4:1) to give 5α , 8α – epidioxyergosta – 3β -6, 22 - diene (1) (21mg) and dammara - 20, 24 - dien - 3 β - ol (2) (324mg). Fr. 3 was rechromatographed on silica gel, eluted with petrol. - Me₂CO(4:1) to give (20R, 24R) - epoxy - 25 - dammaren - 3 - one (3) (38mg). Fr.4 was rechromatographed on silica gel, eluted with petrol. – $Me_2CO(4:1)$ to give $16\ \beta$ – hydroxy – dammara – 20(22), 25 – dien – 3 – one (4) (17mg) and 26 – hydroxy – dammara – 20, 24 – dien – 30 – one (5) (16mg). Fr.6 was rechromatographed on silica gel, eluted with petrol. – EtOAc (3:1) to give 20 – R – form – hydroxydammranone (6) (1137mg) and cycloart – 23 – ene – $3\ \beta$, 25 – diol (7) (19mg). Fr.7 was rechromatographed on silica gel, eluted with petrol. – EtOAc(2:1) to give 7α , 21S, 25 – trihydroxy – 3β – acetoxy – 21S, 23R – epoxy – 9(11) – en – dammarane (8) (124mg).

Table 1 ¹³C - NMR chemical shifts of compounds 1 ~ 8 (CDCl₃ solution. TMS as int. standard)

Carbon	1	2	3	4	5	6	7	8*
1	34.67, t	35.50, t	39.77, t	39.88,t	39.99, t	39.89, t	32.00, t	34.26, t
2	29.83, t	34.28, t	33.97, t	34.09, t	34.11, t	34.06, t	30.44, t	32.02, t
3	66.00, d	78.99, d	217.78, s	217.72, s	217.81, s	217.71, s	78.86, d	78.68, d
4	36.93, t	40.54, s	47.30, s	47.41, s	47.42, s	47.31, s	40.52, s	34.88, s
5	82.19, s	55.98, d	55.31, d	55.48, d	55.48, d	55.43, d	47.16, d	46.64, d
6	135.13, d	18.33, t	19.56, t	19.17, t	19.72, t	19.67, t	21.12, t	17.47, t
7	130.60, d	29.67, t	34.47, t	34.81, t	34.51, t	34.59, t	28.09, t	75.47, d
8	79.43, s	39.18, s	40.21, s	40.46, s	40.46, s	40.33, s	47.93, d	43.61, s
9	51.14, d	51.04, d	50.16, s	50.31, d	50.35, d	50.04, d	20.06, s	145.86,
10	36.77, s	37.29, s	36.74, s	36.97, s	36.99, s	36.85, s	26.01, s	36.66, s
11	23.32, t	21.42, t	21.94, t	21.93, t	21.94, t	22.04, t	26.20, t	118.15,
12	39.31, t	25.00, t	24.75, t	25.02, t	25.03, t	24.78, t	35.61, t	30.42, t
13	44.52, s	45.43, d	42.37, d	46.09, d	45.61, d	42.42, d	45.38, s	45.74, c
14	51.64, d	49.47, s	50.16,s	49.46, s	49.47, s	50.26, s	48.88, s	50.89, s
15	20.53, t	31.41, t	31.03, t	31.43, t	31.37, t	31.14, t	32.87, t	31.65, 1
16	28.49, t	27.47, t	27.39, t	66.89, d	26.59, t	27.53, t	26.51, t	27.36, 1
17	56.21, d	47.85, d	49.93, d	48.13, d	47.71, d	49.90, d	52.07, d	48.63, c
18	12.78, q	15.68, q	15.84, q	15.88, q	15.88, q	15.95, q	18.07, q	27.36, 6
19	18.06, q	15.37, q	15.07, q	15.39, q	15.38, q	15.21, q	29.94, t	23.25, 6
20	39.53, d	131.33, s	74.57, s	134.94, s	152.24, s	75.26, s	36.43, d	45.38, c
21	20.78, q	107.51, t	17.58, q	18.32, q	107.88, t	25.48,q	18.32, q	97.26,
22	135.43, d	39.01, t	36.58, t	127.77, d	34.82, t	40.52, t	39.07, t	23.85,
23	132.28, d	29.17, t	29.22, t	28.99, t	28.96, t	22.56, t	139.45, d	78.42,
24	42.72, d	124.51, d	76.08, d	42.94, t	128.23, d	124.73, d	125.63, d	22.57,
25	32.99, d	152.73, s	147.61, s	149.32, s	134.54, s	131. 48, s	70.71, s	73.25,
26	19.81, q	28.04, q	25.16, q	25.73, q	61.68, t	25.64, q	29.94, q	26.73,
27	19.52, q	17.68, q	110.42, t	110.97, t	21.33, q	17.65, q	29.94, q	26.73, 0
28	17.45, q	16.21, q	20.87, q	21.04, q	21.04, q	20.97, q	14.01, q	21.24,
29	, -1	25.64, q	26.60, q	26.80, q	26.81, q	26.72, q	19.32, q	27.36,
30		15.97, q	16.21, q	16.02, q	16.04, q	16.33, q	25.47, q	12.97, 6

^{*} The $^{13}\mathrm{C}$ - NMR chemical shifts of the acetyl group are 21.43 and 170.65 ppm.

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6.42); H - 28, 0.91, d(J = 6.64). ¹³C - NMR data; see Table 1.

Dammara - 20, 24 - dien - 3 β - ol (2). colouless needles; mp 136°C, crystallized from CHCl₃. [α]_D^{21.5} + 49° (CHCl₃, c 1.0). IRV_{max}^{KBr}cm⁻¹; 3380 (vs, O - H), 2920, 2830, 1640(w, C = C), 1450, 1440, 1390, 1365, 1030(s, C - O), 970, 890(s). EIMS; m/z (%); 426 (M⁺,99),408(M - H₂O, 10), 393(M - H₂O - CH₃,7),383(M - C₃H₇, 17), 365(6), 316(15), 299(14), 257(6), 247(8), 229(11), 218(35), 207(75), 189(55), 175(25), 161(31), 147(26), 135(46), 121(36), 109(73), 93(58), 81(49), 69(100), 55(52). ¹ H - NMR; 0.75(3H,s), 0.82(3H,s), 0.84(3H,s), 0.95(6H,s), 1.58(3H, br,s), 1.66(3H, br,s), 3.18(1H, dd, J₁ = 11.04, J₂ = 5.16), 4.67, 4.68(each 1H, each, s, H - 21), 5.16(1H, s, H - 24). ¹³C - NMR data; see Table 1.

 16β - Hydroxy - dammare - 20(22), 25 - dien - 3 - one (4). colourless needles. mp 182°C crystallized from MeOH, $[\alpha]_D^{21.5} + 58^\circ$ (CHCl₃, c 1.0). IRv_{max}^{KBr} cm⁻¹; 3480(vs, O - H), 3110(w), 2920, 2870, 1695 (vs, C = O), 1650(w, C = C), 1450, 1385, 1375, 1035 (s, C - O), 890, 840 (s). EIMS; m/z(%); 440 (M⁺, 16), 422 (M - H₂O, 57), 407 (M - H₂O - CH₃, 14), 366 (13), 356 (38), 338 (11), 313 (17), 300 (11), 288 (9), 245 (35), 229 (14), 215 (15), 205 (37), 189 (19), 175 (17), 159 (21), 147 (26), 134 (45), 119 (34), 109 (side chain, 41), 95 (42), 85 (100), 67 (35), 55 (44). 1 H - NMR; 0.85 (3H, s), 0.91 (3H, s), 0.98 (3H, s), 1.01 (3H, s), 1.05 (3H, s), 1.68 (3H, s), 1.70 (3H, s), 4.45 (1H, ddd, J₁ = 13.52, J₂ = 8.44, J₃ = 5.04, H - 16), 4. 81, 4.86 (each 1H, each, s, H - 27), 5.14 (1H, br, d, J = 8.44). 13 C NMR data see Table 1.

26 - Hydroxy - dammara - 20, 24 - dien - 3 - one (5). mp $69^{\circ}C$, $[\alpha]_{D}^{21.5} + 58^{\circ}$ (CHCl₃, c 1.0). IRv^{KBr}_{max} cm⁻¹; 3380 (vs, O - H), 2920, 2840, 1690 (vs, C = O), 1630 (m, C = C), 1440, 1370, 1000, 830 (s). EIMS; m/z (%); 440 (M⁺, 41), 422 (M - H₂O, 65), 409 (M - H₂O - CH₃, 14), 381 (10), 357 (M - C₅H₇O, 18), 343 (M - C₆H₉O, 18), 327 (M - side chain, 4), 316 (36), 300(13), 285 (9), 273 (11), 257 (8), 245 (46), 232 (19), 221 (30), 205 (82), 189 (42), 175 (35), 163 (41), 147 (43), 134 (63), 121 (55), 109 (75), 95 (100), 81 (76), 69 (66), 55(71). H - NMR; 0.86 (3H, s), 0.93 (3H, s), 0.99 (3H, s), 1.02 (3H, s), 1.06 (3H, s), 1.78 (3H, s, H - 27), 2.17 (2H, t, J = 7.40, H - 2), 2.43 (m, 4H, H - 22, H - 23), 4.11 (2H, s, H - 26), 4.68, 4.74 (each 1H, each s, H - 21), 5.29 (1H, t, J = 7.42, H - 24). ¹³C - NMR data see Table 1.

20 - R - hydroxydammara - 24 - en - 3 - one (6). colourless cystal; mp $134 \sim 136$ °C, crystallized from MeOH, $[\alpha]_D^{21.5} + 66^\circ$ (CHCl₃, c 1.2). IRv_{max}^{KBr} cm⁻¹; 3510(s, 0 - H), 2930, 2840, 1680(s, C = 0), 1440. EIMS; m/z (%); $442(M^+, 3)$, $424(M - H_2O, 86)$, $409(M - H_2O - CH_3, 9)$, $355(M - H_2O - C_5H_9, 51)$, 342(18), 327(M - side chain, 10), 313(28), 205(68), 189(27), 177(11), 163(18), 149(43), 135(41), 123(44), 109(100), 95(87), 81(67), 69(98). $^1H - NMR$; 0.84(3H, s), 0.90(3H, s), 0.96(3H, s), 1.00(3H, s), 1.04(3H, s), 1.11(3H, s), 1.59(3H, s), 1.65(3H, s), 5.08(1H, br, s), $H \sim 24$). $^{13}C - NMR$ data; see Table 1.

Cycloart - 23 - ene - 3 β , 25 - diol (7), colourless needles, mp 204°C, [α]_D^{21.5} + 38°(CHCl₃, c 0.85). IRv_{max}^{KBr} cm⁻¹; 3300 (vs, O - H), 2960, 2920, 2870, 1460, 1370, 970, 910 (s). EIMS; m/z (%); 442 (M⁺, 8), 424 (M

- H_2O , 86), 409 (M - H_2O - CH_3 , 13), 379 (7), 366 (16), 353 (19), 355 (18), 313 (M - side chain, 23), 298 (10, loss of ring A along with C_6 and C_{19}), 245 (38), 215 (16), 205 (51), 203 (53), 187 (24), 173 (16), 159 (24), 147 (37), 134 (73), 121 (62), 107 (63), 95 (72), 81 (79), 69 (68), 53 (100). 1 H - NMR; 0.31 (1H, d, J = 4.20, H - 19), 0.52 (1H, d, J = 4.20, H - 19), 0.75 (3H, s), 0.83 (3H, d, J = 10.42, H - 21), 0.94 (3H, s), 1.06 (6H, s), 1.31 (3H, s), 1.32 (3H, s), 3.26 (1H, dd, J_1 = 11.14, J_2 = 4.40, H - 3), 5. 57 (2H, dd, J_1 = 4.80, J_2 = 3.24, H - 23, H - 24). 13 C - NMR data; see Table 1.

 7α , 21S, 25 – Trihydroxy – 3β – acetoxy – 21S, 23R – epoxy – 9(11) – en – dammarane (8). mp 201 °C, crystallized from CHCl₃. [α]_D^{21.5} – 32° (CHCl₃, c, 1.0). IRv_{max}^{KBr} cm⁻¹; 3400 (vs, O – H), 2970, 1720 (vs, C = O), 1450, 1380, 1270 (vs, C – O), 1090, 1040, 900, 810. EIMS; mz/(%); 532 (M⁺, 33), 514 (M – H₂O, 74), 499 (M – H₂O – CH₃, 56), 481 (M – 2H₂O – CH₃, 19), 456 (M – H₂O – CH₃ – COCH₃, 39), 439 (M – H₂O – CH₃ – CH₃COOH, 47), 421 (M – 2H₂O – CH₃ – CH₃COOH, 29), 409 (34), 393 (28), 381 (39), 367 (26), 356 (M – side chain, 21), 341 (M – OH – CH₃ – side chain, 27), 309 (11), 297 (M – OH – CH₃COOH – side chain, 16), 281 (M – OH – CH₃ – CH₃COOH – side chain, 48), 255 (32), 241 (25), 215 (29), 199 (27), 187 (43), 184 (50), 173 (37), 161 (45), 159 (side chain, 33), 145 (46), 133 (51), 119 (53), 107 (66), 91 (56), 79 (50), 69 (52), 59 (100). ¹ H NMR; 0.73 (3H, s), 0.81 (3H, s), 0.86 (3H, s), 0.90 (3H, s), 0.94 (6H, s), 1.23 (3H, s), 2. 02 (3H, s, OAc), 3.20 (1H, br, H – 7), 4.24 (1H, m, H – 23), 4.65 (1H, br, H – 3), 5.27 (1H, d, J = 3.56, H – 21), 5.30 (1H, br, H – 11). ¹³ C NMR data see Table 1.

Acknowledgements We are grateful to Prof. Wang Hong (Kunming Institute of Botany, Academia Sinica) for identifying the specimen of the plant. We also extend our sincere appreciation to Mr. He Yi – nen for NMR data and Mrs. Liang Hui – lin for MS data.

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