## 大紫丹参的多酚类化合物\*

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摘要:从云南丽江产大紫丹参 ( Solvia przewalskii Maxim.) 的根部分离得到 11 个多酚类化台物,其中 8 个鉴定为已知的原儿茶 醛、原儿茶酸、咖啡酸、R-(+)-3-D-(3,4-二羟基苯基)-乳酸、迷迭香酸、迷迭香酸甲酯、紫草酸和紫草酸 B. 另外 3 个为紫草酸 B 的甲酯化衍生物,即紫草酸 B 二甲酯、9"-紫草酸 B 单甲酯和 9"-紫草酸 B 甲单酯。它们的结构通过波谱方法得到鉴定。研究结果表明,大紫丹参含有与正品丹参相似的酚类化合物。

关键词:<u>鼠尾草属:</u>大紫丹参;多酚类化台物 化多成分

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## Polyphenolic Constituents of Salvia przewalskii

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**Abstract:** Three lithospermic acid B ester derivatives, dimethyl lithospermate B. 9" – methyl lithospermate B and 9" – methyl lithospermate B together with nine known polyphenol compounds; protocatechualdehyde, protocatechualde acid, caffeic acid,  $R - (+) - \beta - D - (3, 4 - dibydroxyphenyl) – lactic acid, rosmarinic acid, methyl rosmarinate, lithospermic acid and lithospermic acid B were isolated from the dried roots of$ *Salvia przewalskii*.

Key words; Salvia; S. przewalskii; Polyphenols

Salvia przewalskii Maxim. distributes in Gansu, Sichuan and Yunnan Provinces in western China. In northwest Yunnan, its roots have been used as a substitute of "Dan – Shen", a commonly used crude material of traditional Chinese medicine (Jiangsu College of New Medicine, 1979). Previously studies on this species have mainly focused on the constituents of lipophilic diterpenoid quinones (Yang et al., 1984, 1981). As a continuation of our chemical research on the genus Salvia (Tanaka et al., 1996, 1997; Wu et al., 1999), we isolated eleven polyphenols from this plant collected from Lijiang County of Yunnan Province, We report here the isolation and structural determination of these compounds.

Eleven polyphenol compounds  $(1 \sim 11)$  were isolated from the MeOH extracts of dry roots of S.

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Compound 9 was obtained as a yellow amorphous powder, which is in a dark green color with Fe-Cl<sub>3</sub> reagent on TLC . It showed a quasimolecular ion peak at m/z 745 [  $M(C_{38}H_{34}O_{16}) - H$ ]  $^$ ative FAB mass spectrum. The  $^{-13}$ C NMR spectrum showed the signals of a double bond at  $\delta$  143.96 and 116.48; four carbonyl groups at  $\delta$  172.18 (2 × C), 171.19 and 167.96; three oxo – methines at  $\delta$  88.16, 74.75 and 75.61; a methine at  $\delta$  57.46; two methylenes at  $\delta$  37.39 and 37.78; two methoxy groups at  $\delta$  52.65 and 52.79; and 24 aromatic carbons at  $\delta$  113.37 ~ 149.00. Its  $^{-1}$ H NMR spectrum showed the signals of a pair of doublet peaks due to trans – olefinic protons at  $\delta$  6.26, 7.63 (each 1H, d, J = 15.9Hz); three oxo – methine protons at  $\delta$  4.39, 4.96 and 5.20; two methoxy groups at  $\delta$  3.65 and 3.64; two benzoylic methylenes in  $\delta$  2.83 ~ 3.02; and the signals at  $\delta$  6.20 ~ 7.20 belonging to four aromatic ring protons. A comparison of the <sup>1</sup>H and <sup>13</sup>C NMR spectral data of 9 with that of compound 8, showed that 9 was quite similar to 8. The difference between them is that there are only more two methoxy group signals appeared in 9. Moreover, the carbonyl groups of  $C-9^n$ and C-9''' at  $\delta$  173.52 and  $\delta$  172.17 in 8 were upfield shifted to  $\delta$  172.18 ( - 1.34) and  $\delta$  171.19  $\langle -0.98 \rangle$  in 9, respectively. It indicated that two additional methoxy groups were linked at the position carbonyl C - 9'' and C - 9''' in 9. Therefore, the structure of 9 was identified as dimethyl lithospermate B.

Compound 10 and 11 were obtained as yellow powders, both had the same quasimolecular ion peak at m/z 731 [ $M(C_{37}H_{32}O_{16}) - H$ ]<sup>-</sup> in their negative FAB mass spectra, which were 14 mass units larger than that of compound 8. By comparison the spectral data of UV, IR, <sup>1</sup>H and <sup>13</sup>C NMR, compounds 10 and 11 were closely similar to those of 8 and 9. This suggests that both 10 and 11 are monomethyl ester of 8. By observation of the <sup>13</sup>C NMR spectral data of 10 together with that of 8, the C = 9"carbonyl group at  $\delta$  173.52 was upfield shifted to  $\delta$  172.20 (-1.32) in 10. It suggests that an additional methoxy group links C = 9" carbonyl group and the structure of compound 10 could be 9" = methyl lithospermate B. In the same way, by comparison the <sup>13</sup>C NMR spectrum of 11 with that of 8, only the C = 9" carbonyl group at  $\delta$  172.17 was upfield shifted to  $\delta$  171.20 (-0.97), while the other carbon signals were almost unaffected. It indicats that this additional methoxy group links C = 9" carbonyl group. Therefore, the structure of 11 was established as 9" = methyl lithospermate B.

It is the first isolation of mono-and di-methoxy esters of lithospermic acid B in natural forms.

## **Experimental**

The instruments and chromatographic materials used throughout this work are the same as described in the reference (Wu et al., 1999).

The dried roots of Salvia przewalskii Maxim. (10kg) which collected in Lijiang County, Yunnan Province, were extracted with 60% acetone at room temperature ( $4 \times 20L$ ), then concentrated in vacuum to evaporate the acetone. Keeping the water solution for one day, after filtration, the filtrate was acidified with 10% HCl and then extracted with EtOAc. The 100 g of EtOAc extract (230g) was chromatographed on a silica gel column ( $200 \sim 300$  mesh) with benzene: ethyl acetate; formic acid (5:4:1) as developed solvent system and giving three fractions (Fr. A-C). Protocatechualdehyde (1,100mg), protocatechuic acid (2,200mg), caffeic acid (3,200mg) and methyl rosmarinate (6,50mg) were obtained from fraction A (5g), after subjected to Sephadex LH-20 ( $30\% \sim 60\%$  acetone) and MCI – gel CHP20P ( $30\% \sim 50\%$  acetone) column chromatographies. R – (+) –  $\beta$  – D – (3,4 – dihy roxyphenyl) – lactic acid (4,2g), rosmarinic acid (5,7g), dimethyl lithospermate B (9,150mg), 9% – methyl lithospermate B (10,100mg) and 9% – methyl lithospermate B (11,200mg) were obtained from fraction B (30g), after separation by Sephadex LH –  $20(30\% \sim 50\%$  acetone) and MCI – gel CHP20P ( $30\% \sim 60\%$  acetone) column chromatographies. The fraction C (20g) was subsequently chromatographed over MCI – gel CHP20P ( $30\% \sim 50\%$  acetone) and Rp – 8 gel ( $30\% \sim 40\%$  acetone) column to afford lithospermic acid (7,1.5g) and lithospermic acid B (8,4.5g).

For the spectral data of compound  $1 \sim 8$ , see reference (Wu et al., 1999).

**Dimethyl lithospermate B** (9): A yellow amorphous powder; FAB-MS m/z: 745 [M( $C_{38}H_{34}O_{16}) - H$ ]<sup>-</sup>; [ $\alpha$ ]<sub>D</sub>(23.8°C) = +93.43°(c = 0.59, MeOH); IR $\nu_{max}^{KBr}$  cm<sup>-1</sup>: 3404, 2957, 1734, 1611, 1521, 1446, 1365, 1287, 1178, 1114, 1070, 1043, 978, 934, 867, 811, 779, 732, 687, 589; UV (MeOH)  $\lambda_{max}$  nm: 206, 255, 289, 304, 332; H and H3C NMR data in Table 1 and 2.

9" - methyl lithospermate B (10); A yellow amorphous powder, FAB - MS m/z; 731 [M( $C_{37}$  H<sub>32</sub>O<sub>16</sub>) - H]<sup>-</sup>; [ $\alpha$ ]<sub>D</sub> (24.4°C) = +98.23° (c = 0.68, MeOH); IR $\nu_{max}^{KBr}$  cm<sup>-1</sup>; 3387, 2958, 1731, 1611, 1519, 1446, 1363, 1178, 1114, 1072, 1043, 976, 935, 867, 811, 779, 724, 589; UV (MeOH)  $\lambda_{max}$  nm; 206, 254, 289, 306, 329; <sup>1</sup>H and <sup>13</sup>C NMR data in Table 1 and 2.

9" - methyl lithospermate B (11); A yellow amorphous powder; FAB - MS m/z; 731 [M ( $C_{37}H_{32}O_{16}$ ) - H]<sup>-</sup>; [ $\alpha$ ]<sub>D</sub> (24.50°C) = +98.20° ( $\alpha$  = 0.65, MeOH);  $R_{max}^{KBr} = 1$ ; 3386,

2958, 1732, 1611, 1520, 1446, 1364, 1178, 1114, 1072, 1043, 977, 935, 867, 811, 779, 724, 589; UV (MeOH)  $\lambda_{max}$  nm; 206, 254, 289, 305, 328; H and H3C NMR data in Table 1 and 2.

Table 1  $^{-13}$ C NMR spectral data of compounds 8, 9, 10 and 11 (100MHz, CD<sub>3</sub>OD)

C	8	9	10	11
1	124.58	124.60	124.60	124.60
2	126.14	126.22	126.17	126.17
3	148.90	149.00	148.97	148.97
4	144.95	145.15	145.09	145 00
5	118.35	118.67	118.37	118.37
6	122.09	121.96	122.05	121.98
7	143.46	143.96	143.65	143.78
8	116.45	116.48	116.46	116.46
9	167.96	167.96	168.04	167.91
1'	133.50	133.49	133.57	133.48
2'	113.36	113.37	113.37	113.37
3′	146.49	146.70	146 46	146.64
4′	145.68	145.25	145.99	145.84
5'	116.45	116.48	116.46	116.46
6'	118.35	118.35	118.37	118.37
7'	88.12	88.16	88.17	88.17
8'	57.69	57.46	57.78	57.45
9'	172.44	172.18	172.47	172.47
i"	129.18	128.54	128.93	129.27
2"	117.54	117.49	117.29	117.51
3*	144.95	145.15	145.00	145.09
4"	146.32	146.51	146.64	146.46
5"	116.45	116.48	116.46	116.46
6"	122,21	122,16	122.05	122.23
7"	37.67	37.78	37.76	37.76
8"	74.51	74.75	74.68	74.68
9"	173.52	172.18	172.20	173.49
1'*	128.84	128.54	128.76	128.53
2‴	117.27	117.31	117.29	117.51
3'"	144.95	145.15	145.00	145.09
4‴	145.68	145.99	145,84	145.99
5‴	116.45	116.48	116.46	116.46
6‴	121.73	121.83	121.98	121.77
7‴	37.28	37.39	37.41	37.41
8‴	75.40	75.61	75.53	75.53
9‴	172.17	171,19	172.20	171.20
9° - OCH3		52.79	52.82	
9''' - OCH <sub>3</sub>		52.65		52.66

Table 2 <sup>1</sup>H NMR spectral data of compounds 8, 9, 10 and 11 (400MHz, CD<sub>3</sub>OD)

H	8	9	10	11
5	6.85(tH,d,8.4)	6.85(1H,d,8.5)	6.84(1H,d.8.4)	6.84(1H.d.8.4)
6	7.15(1H,d,8.5)	7.18(1H,d,8.5)	7.15! (H,d,# 4)	7.17(1H.d.8.3)
7	7.53(1H,d,15.9)	7.63(1H,d,15.9)	7.53(1H,d, 6)	7 57(1H,d,16 0)
8	6.24(1H, d, 15.9)	6.26(1H,d,15.9)	6.23(1H, d, 6)	6 30(1H,d,16.0)
2'	6.79(1H, d, 2.1)	$6.78(1H,d,2.0)^{a,b}$	$6.77(1H, d0)^{a,b}$	6 77(1H,d,2.0) <sup>a1</sup>
5'	6.77(1H, d, 8.2) <sup>a)</sup>	6.71(1H,d,8.5)a,cl	6.76(1H, d, fl.1)a)	6.77(1H,d,8.21 <sup>a)</sup>
6'	6.67(1H, dd, 1.8, 8.0) 4.bl	6.65(1H,dd,2.0,8.0)**,d	6.67(1H, dd 2.0, 8.1)	6.67(1H, dd, 1.8, 8.0)
7'	5.87(1H, d, 4.6)	5.83(1H,d,4.4)	5.80(1H,d,-1.5)	5 87(1H,d,4.8)
8,	4.38(1H,d,4.7)	4.39(1H,d,4.5)	4.38(1H,d,-1.6)	4 38(1H,d,4.6)
2°	6.77(1H,d,2.0)41	6.76(1H,d,2.0) <sup>a,b)</sup>	6 76(1H, d.,! 0)a,b)	6.77(1H,d,2.0) <sup>a)</sup>
5"	6.72(1H, d.8.2)	6.70(1H,d,8.5)**,cl	6.71(1H,d.8.3)	6.70(1H.d,8.0)
6"	6.64(1H,dd,2.0,8.1)**bj	6.64(1H,dd,2.1,8.2)a,d)	6.63(1H,dd 2.0,8.0)a)	6.64(1H, dd, 2.0, 8.0)
7'	3.08(2H,dd,4 0,8.0)44c)	2.98(2H, m) <sup>a)</sup>	2.94(2H, m)*)	2.98(2H,m) <sup>al</sup>
B"	5.21(1H, dd, 3.7, 7.0)	5.19(1H <sub>1</sub> m)	5.18(1H,m)	5.19(1H, m)
2**	6.54(1H, d, 2.1)	6.57(1H,d,1.5)	6.54(1H, d,2.1)	6.54(1H, d.2.1)
5*	6.60(1H, d, 8.0)	6.60(1H,d,8 1)	6.61(1H,d,8 4)*	6.60(1H.d.B.0)
6*	6.33(1H,dd,2.0,8.0)	6 38(tH,m)	6.30(1H <sub>1</sub> m)	6.33(1H,dd,2.0,8.0)
7*	3.04(2H,dd,4.0,8.0)*,c)	2.98(2H,m)a)	2.94(2H,m) <sup>a)</sup>	2.98(1H.m) <sup>al</sup>
8*	5.12(1H,m)	4.96(1H, s. br. )	4.95(1H, s, lo.)	4.96(1H, 4, br.)
9" - OCH3		3.64(3H,s)	3.64(3H,s)	
9" - OCH3		3.65(3H,s)		3.65(3H,s)

a) Overlapping with other signals; b - d) Assignments may be interchanged in each column

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